
SUPPORT INFORMATION

A self-curing triphenol A-based phthalonitrile resin precursor acts as flexibilizer and curing agent for phthalonitrile resin

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Table 1S. The reactions conditions for preparing TPPA-Ph precursors.

Figure 1S. LC-MS of TPPA-Ph precursor and molecular ion peaks in negative mode of products.

Figure 2S. Conversion α as a function of temperature for TPPA-Ph (P4) at various heating rates.

Figure 3S. Starink plots at various degrees of conversion for TPPA-Ph (P4).

Figure 4S. Variation of $E\alpha$ versus α for TPPA-Ph (P4) (Starink method).

Scheme 1S. Synthesis of resorcinol-based phthalonitrile resin precursor (DPPH).

Figure 5S. ¹H NMR spectrum of DPPH.

Table 1S. The reactions conditions for preparing TPPA-Ph precursors.

TPPA-Ph precursors	The molar ratio of TPPA : K ₂ CO ₃ : NPh	Reaction temperature (°C)	Reaction time (h)
P1	1:2.5:3.2	80	24
P2	1:1.5:3.1	25	24
P3	1:1.5:2.05	25	24
P4	1:1.5:1.05	25	24

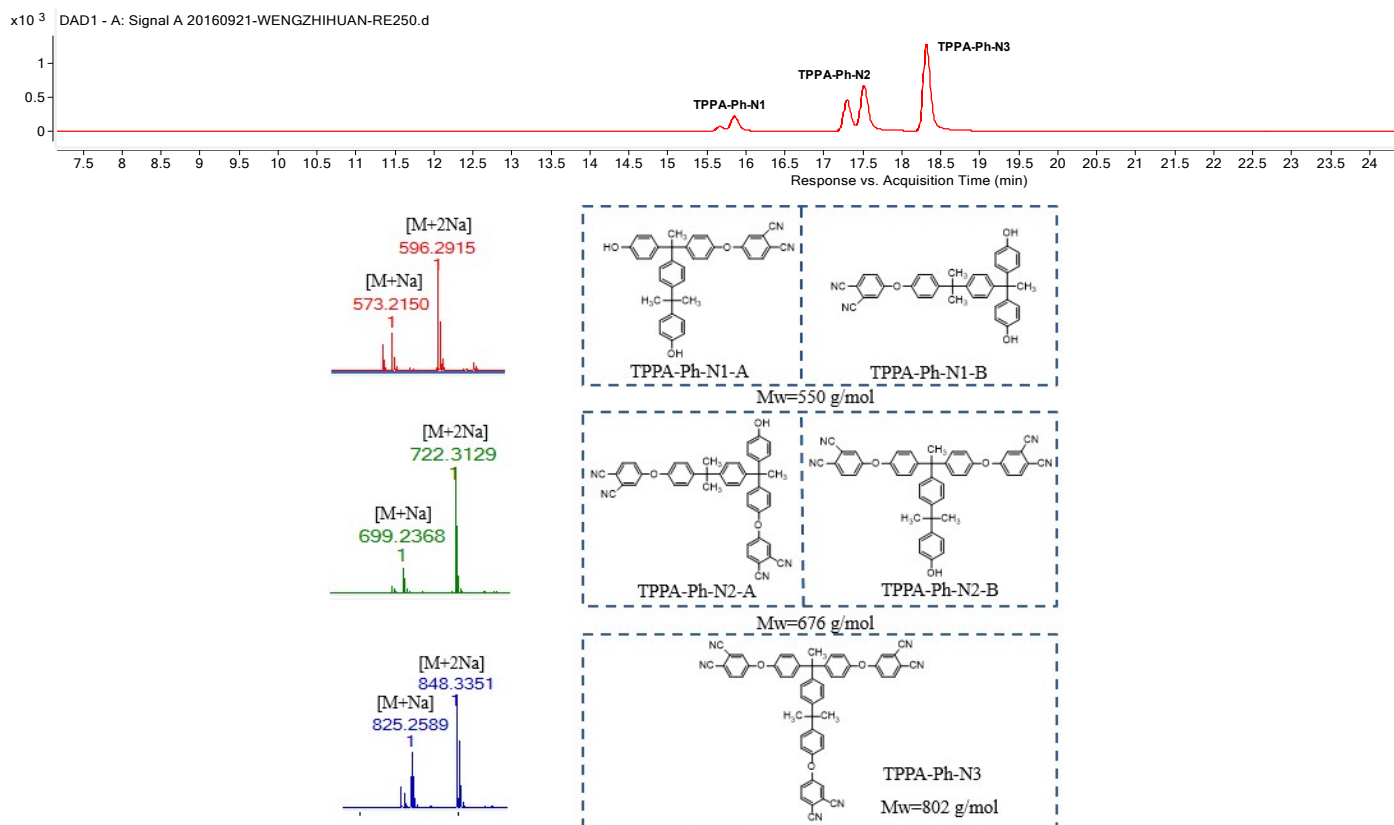


Figure 1S. LC-MS of TPPA-Ph (P2) and molecular ion peaks in negative mode of products.

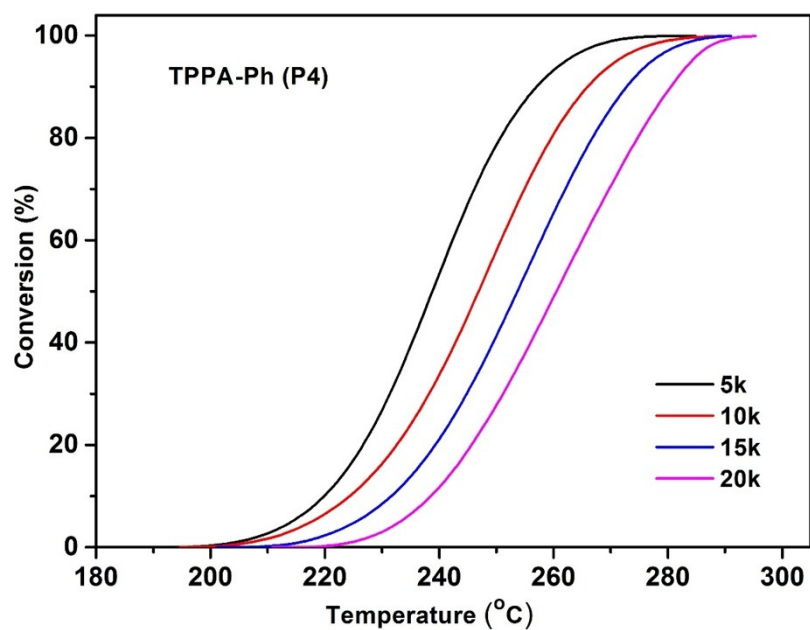


Figure 2S. Conversion α as a function of temperature for TPPA-Ph (P4) at various heating rates.

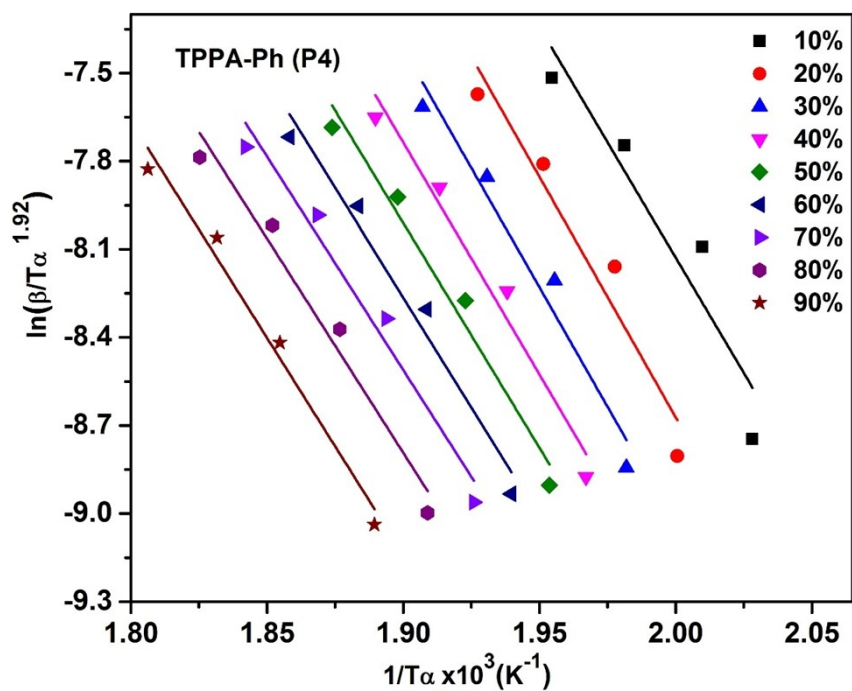


Figure 3S. Starink plots at various degrees of conversion for TPPA-Ph (P4).

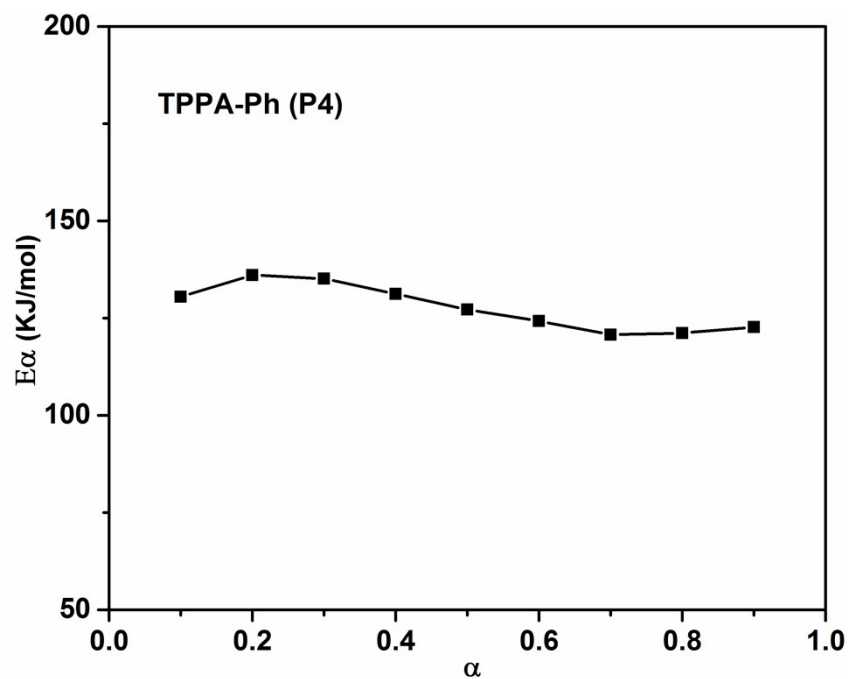
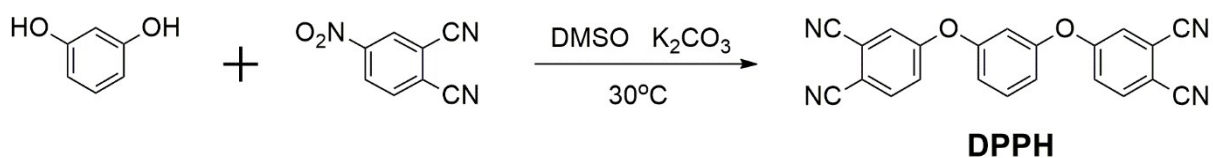


Figure 4S. Variation of $E\alpha$ versus α for TPPA-Ph (P4) (Starink method).



Scheme 1S. Synthesis of resorcinol-based phthalonitrile resin precursor (DPPH).

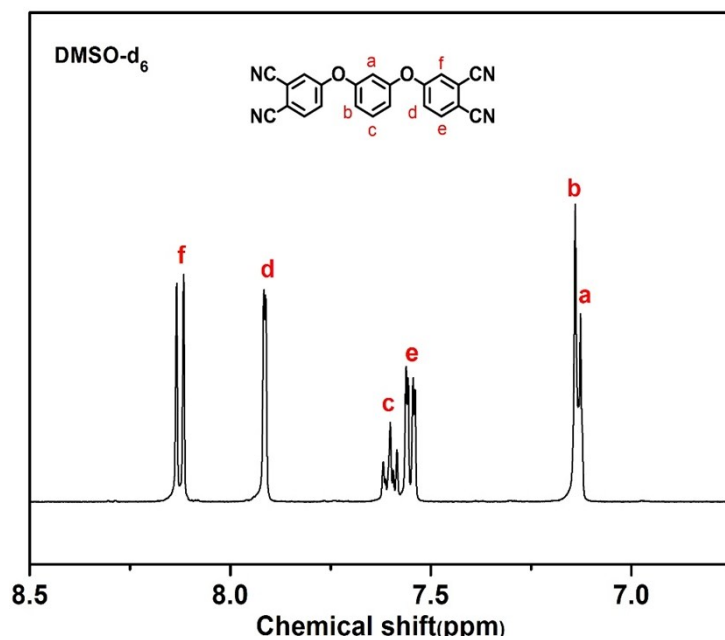


Figure 5S. ^1H NMR spectrum of DPPH.