Electronic Supplementary Material (ESI) for RSC Advances.

Supporting Information

Catalytic oxidation of ethyl acetate over $LaBO_3$ (B=Co, Mn, Ni, Fe) perovskites supported silver catalysts

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Table S1 Specific surface areas of $Ag/LaBO_3$ (B = Co, Mn, Ni, Fe).

Sample	S_{BET} (m^2/g)
$Ag/LaCoO_3$	14.5
$Ag/LaMnO_3$	19.3
$Ag/LaNiO_3$	12.5
$Ag/LaFeO_3$	14.0

Table S2 H_2 consumption of $LaBO_3$ and $Ag/LaBO_3$ (B = Co, Mn, Ni, Fe) from H_2-TPR results.

Sample	H_2 consumption (mol/g)
$Ag/LaCoO_3$	2.04
$LaCoO_3$	1.98
$Ag/LaMnO_3$	0.61
$LaMnO_3$	0.45
$Ag/LaNiO_3$	1.80
$LaNiO_3$	1.70
$Ag/LaFeO_3$	0.05
$LaFeO_3$	0.05

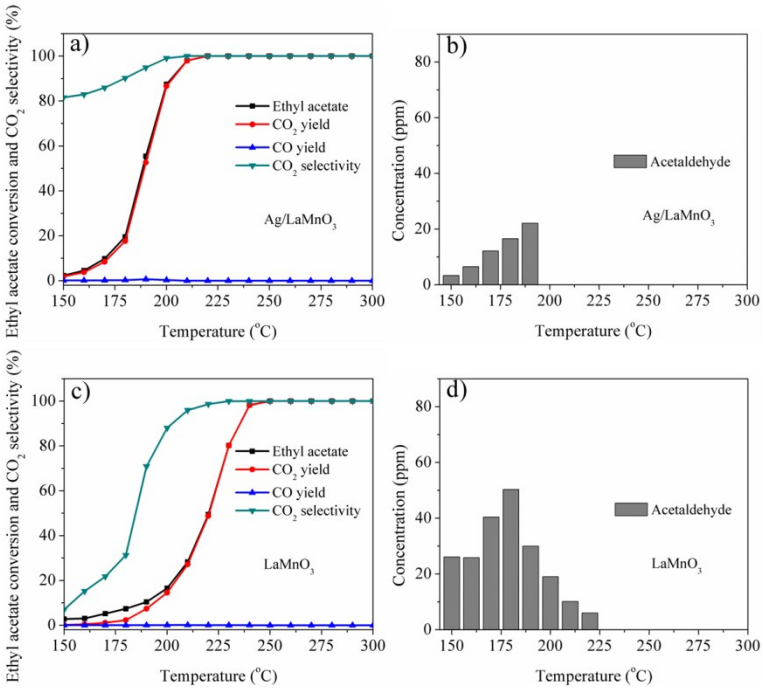

Fig. S1 Evolution of ethyl acetate conversion, CO_2 selectivity, CO_x yields, distributions of the organic byproduct in function of temperature on $Ag/LaMnO_3$ (a, b), $LaMnO_3$ (c, d). Ethyl acetate concentration: 500 ppm; ethyl acetate/O_2 molar ratio: 1/400; WHSV: 60,000 mL/(g h).

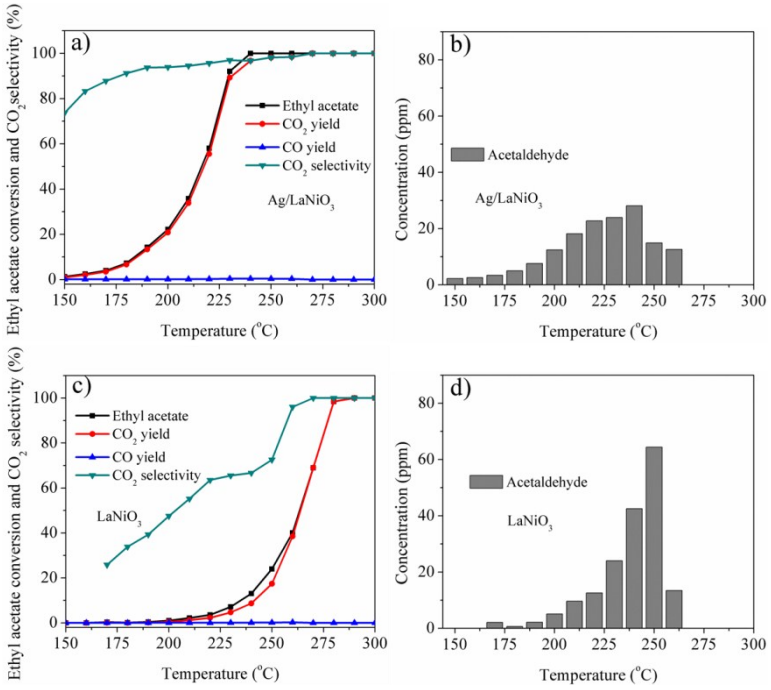

Fig. S2 Evolution of ethyl acetate conversion, CO_2 selectivity, CO_x yields, distributions of the organic byproduct in function of temperature on $Ag/LaNiO_3$ (a, b), $LaNiO_3$ (c, d). Ethyl acetate concentration: 500 ppm; ethyl acetate/O_2 molar ratio: 1/400; WHSV: 60,000 mL/(g h).

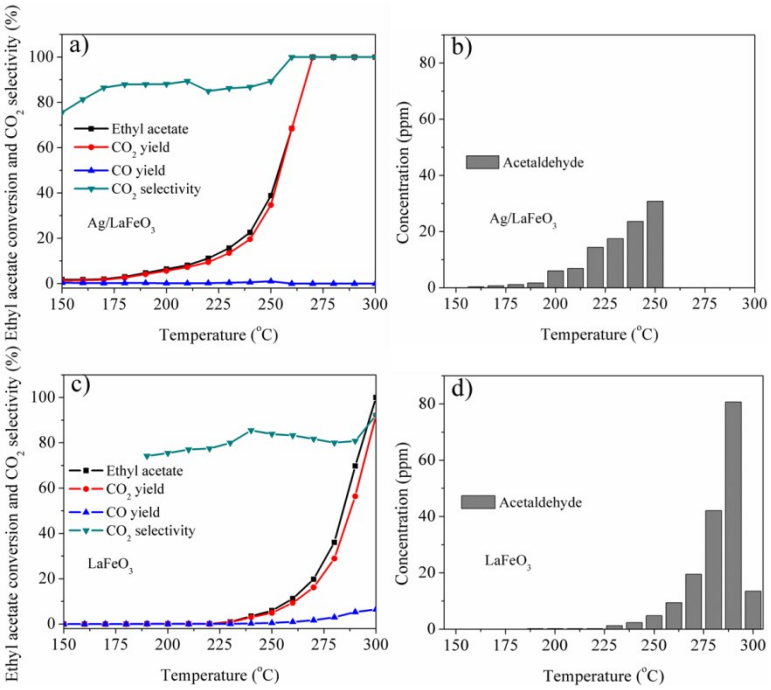

Fig. S3 Evolution of ethyl acetate conversion, CO_2 selectivity, CO_x yields, distributions of the organic byproduct in function of temperature on $Ag/LaFeO_3$ (a, b), $LaFeO_3$ (c, d). Ethyl acetate concentration: 500 ppm; ethyl acetate/O_2 molar ratio: 1/400; WHSV: 60,000 mL/(g h).