## Supporting Information

## Enhanced oxygen reduction activity of Pt shells on PdCu truncated octahedra with different compositions

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Table S1 ICP-AES data of the $\mathrm{Pd}_{\mathrm{x}} \mathrm{Cu} @ \mathrm{Pt}$ core-shell nanocrystals

| Samples | $\mathbf{P d} / \mathbf{C u} / \mathbf{P t}$ atomic ratio | $\mathbf{w t} \%$ of $\mathbf{P t}$ |
| :--- | :--- | :--- |
| $\mathbf{P d}_{2} \mathbf{C u} @ \mathbf{P t}$ | $6.1: 3: 1$ | 18.8 |
| $\mathbf{P d C u} @ \mathbf{P t}$ | $4.5: 4.3: 1$ | 20.6 |
| $\mathbf{P d C u}_{2} @ \mathbf{P t}$ | $4.2: 7.4: 1$ | 17.4 |

Table S2 Average number (n) of Pt atomic layers and the weight percentage of Pt derived from the geometry analysis and ICP-AES data.

| Samples | n of Pt atomic <br> layers | weight percentage of Pt <br> calculated from the value of n | weight percentage of Pt <br> calculated from the ICP data |
| :---: | :---: | :---: | :---: |
| $\mathrm{Pd} \mathrm{Cu}_{2} @ \mathrm{Pt}$ | 2 | $14.36(\mathrm{n}=1)$ | 18.8 |
|  |  | $25.45(\mathrm{n}=2)$ |  |
| $\mathrm{PdCu} @ \mathrm{Pt}$ | 2 | $13.62(\mathrm{n}=1)$ | 20.6 |
|  |  | $23.39(\mathrm{n}=2)$ |  |
| $\mathrm{PdCu}_{2} @ \mathrm{Pt}$ | 1 | $17.9(\mathrm{n}=1)$ | 17.4 |
|  |  | $30.9(\mathrm{n}=2)$ |  |

Table S3 Electrochemical performances of the $\mathrm{Pd}_{\mathrm{x}} \mathrm{Cu} @ \mathrm{Pt}$ and $\mathrm{Pt} / \mathrm{C}$ catalysts

| Samples | ECSA <br> $\left(\mathbf{m}^{2} / \mathbf{g}_{\mathbf{P t}}\right)$ | $\boldsymbol{i}_{\boldsymbol{s}}$ <br> $\left(\mathbf{m A} / \mathbf{c m}^{2}\right)$ | $\boldsymbol{i}_{\boldsymbol{m}}$ <br> $\left(\mathbf{m A} / \boldsymbol{\mu g}_{\mathbf{P t}}\right)$ | $\boldsymbol{i}_{\boldsymbol{m}} / \mathbf{A D T}^{*}$ <br> $\left(\mathbf{m A} / \boldsymbol{\mu g}_{\mathbf{P t}}\right)$ | $\mathbf{d u r a b i l i t y * *}^{* *}$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| $\mathbf{P d} \mathbf{2} \mathbf{C u @ P t}$ | 129.0 | 0.46 | 0.59 | 0.42 | $71.2 \%$ |
| $\mathbf{P d C u @ P t}$ | 106.2 | 0.37 | 0.39 | 0.24 | $61.5 \%$ |
| $\mathbf{P d C u} @ \mathbf{P t}$ | 70.3 | 0.31 | 0.22 | 0.15 | $68.2 \%$ |
| $\mathbf{P t} / \mathbf{C}$ | 78.2 | 0.17 | 0.13 | 0.096 | $73.8 \%$ |

*The results of ADTs with 10,000 cycles are listed here.
**The durability is obtained by using the ratio of current densities after and before the ADTs.


Fig. S1 (a, c, e) TEM images and (b, d, f) corresponding size distributions of $\mathrm{Pd}_{2} \mathrm{Cu}$, PdCu and $\mathrm{PdCu}_{2}$ truncated octahedra.


Fig. S2 XRD patterns of $\mathrm{Pd}_{2} \mathrm{Cu}, \mathrm{PdCu}$ and $\mathrm{PdCu}_{2}$ truncated octahedra. The black and red vertical lines represent the positions of standard diffraction peaks for Pd and Cu , respectively.


Fig. S3 (a, c, e) TEM images and (b, d, f) corresponding size distributions of $\mathrm{Pd}_{2} \mathrm{Cu} @ \mathrm{Pt}, \mathrm{PdCu} @ \mathrm{Pt}$ and $\mathrm{PdCu}_{2} @ \mathrm{Pt}$ truncated octahedra.


Fig. S4 XRD patterns of $\mathrm{Pd}_{2} \mathrm{Cu} @ \mathrm{Pt}$ truncated octahedra.


Fig. S5 Morphological, structural, and compositional characterizations of truncated octahedral $\mathrm{PdCu}_{2} @$ Pt nanocrystals (a) HAADF-STEM image, (b) HRTEM image, (c) EDX mapping image, and (d) line-scanning profiles.

b


Fig. S6 (a) Pd 3d and (b) Pt 4f XPS spectra of $\mathrm{Pd}_{2} \mathrm{Cu} @ \mathrm{Pt}, \mathrm{PdCu} @ \mathrm{Pt}$ and $\mathrm{PdCu}_{2} @ \mathrm{Pt}$ truncated octahedra.


Fig. S7 Cyclic voltammetry (CV) curves of (a) $\mathrm{Pd}_{2} \mathrm{Cu} @ \mathrm{Pt}, \mathrm{PdCu} @ \mathrm{Pt}$ and $\mathrm{PdCu}_{2} @ \mathrm{Pt}$ catalysts, and (b) Pt/C catalysts in Ar-saturated $0.1 \mathrm{M} \mathrm{HClO}_{4}$ solution with a scan rate of $50 \mathrm{mV} / \mathrm{s}$.


Fig. S8 TEM images of $\mathrm{Pd}_{2} \mathrm{Cu} @ \mathrm{Pt}$ catalysts (a) before and (b) after ADTs for 10000 cycles in $\mathrm{O}_{2}$ saturated $0.1 \mathrm{M} \mathrm{HClO}_{4}$ solution.

