

Supporting Information

## Molten salt construction of stable oxygen vacancies on TiO<sub>2</sub> for enhancement of visible light photocatalytic activity

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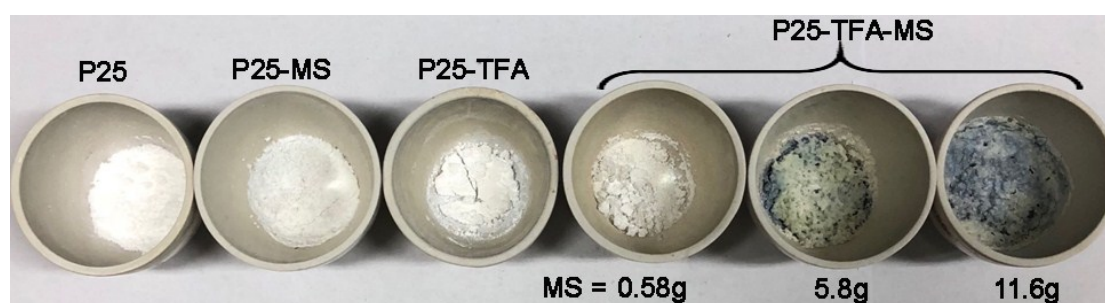


Fig. S1 Effect of MS dosages on the annealing of P25.

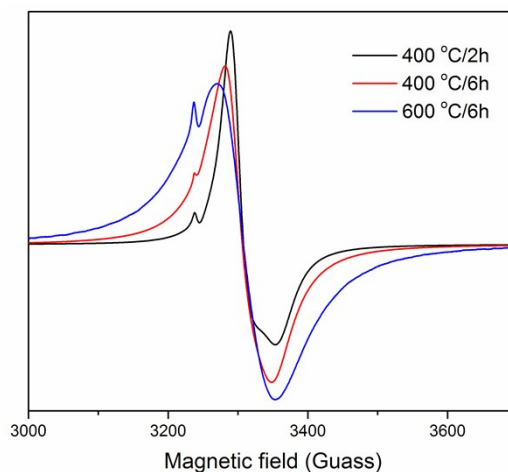


Fig.S2 X-band EPR spectra of B-TiO<sub>2</sub> synthesized with MS at different time and temperature. A longer annealing time and a higher temperature will lead to increase of the bulk Ti<sup>3+</sup> defect concentration and decrease of surface oxygen vacancy concentration.

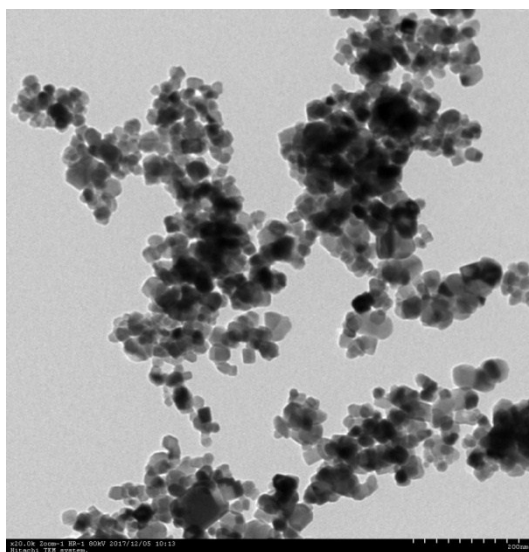


Fig. S3 TEM of  $\text{TiO}_2$  synthesized with addition of NaF via molten salt.

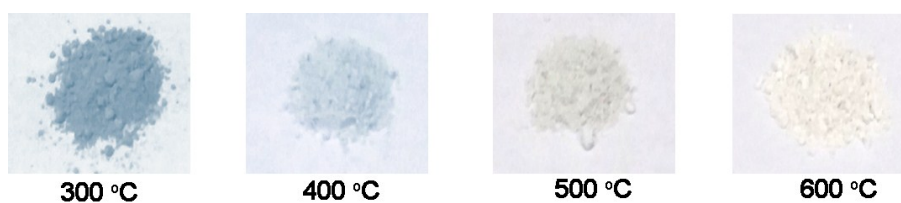


Fig. S4 Photographs of B- $\text{TiO}_2$  calcined in air at different temperature for 2h. It is found that the dark blue color fades to white gradually with increasing the calcination temperature from 300 °C to 600 °C.

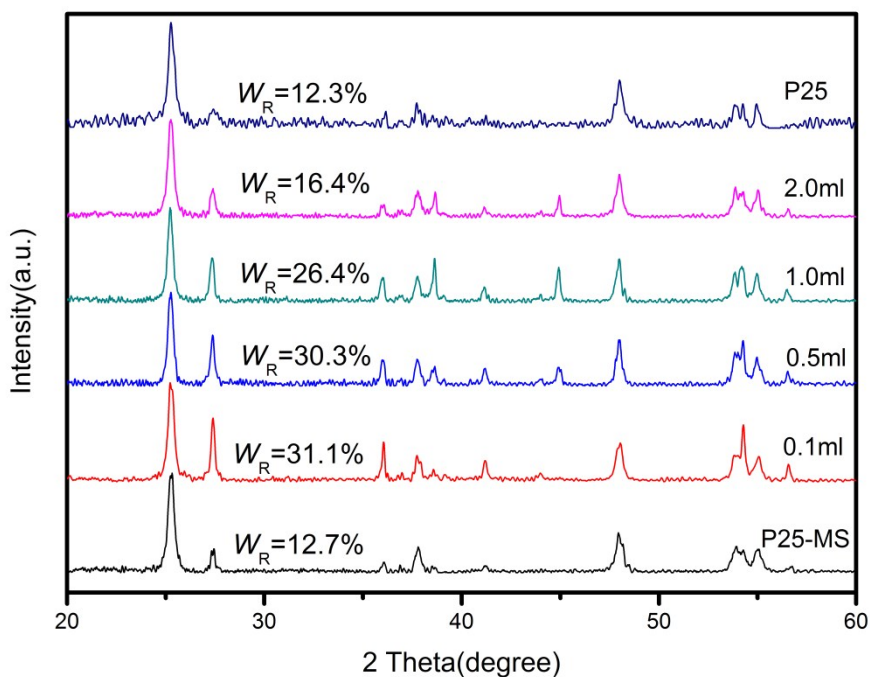


Fig. S5 XRD of  $\text{TiO}_2$  synthesized with different TFA amount via MS method.

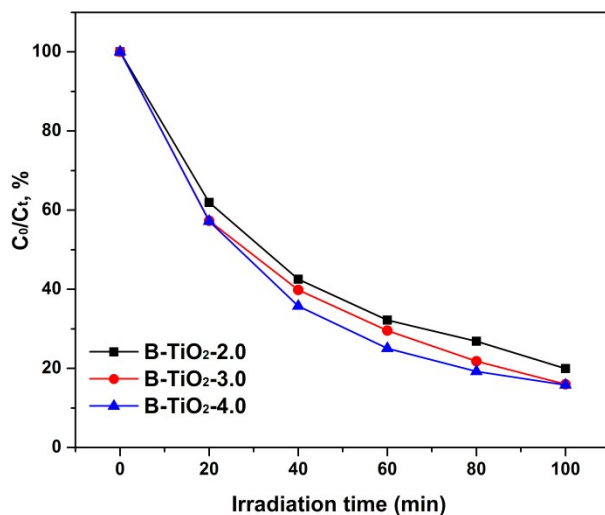


Fig. S6 Effect of TFA amount on photocatalytic activity of B-TiO<sub>2</sub>.

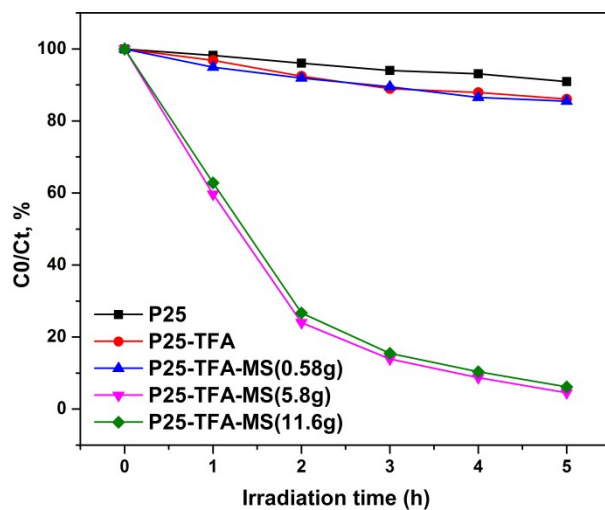


Fig. S7 Effect of molten salt dosage on photocatalytic activity.

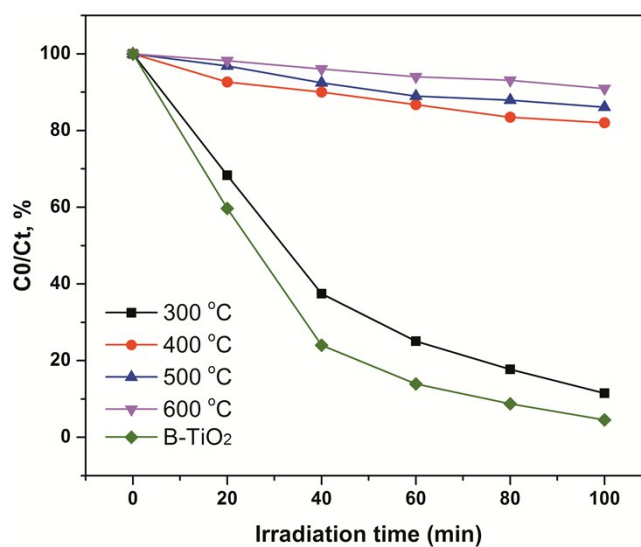
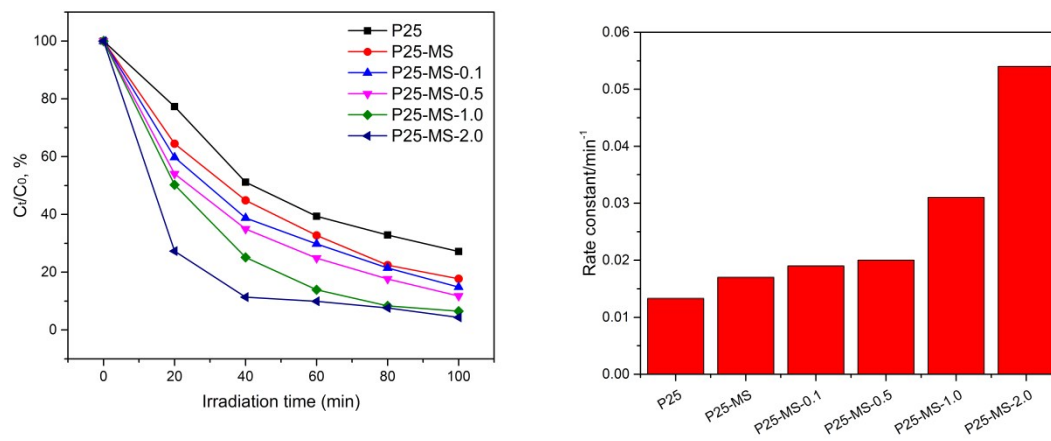


Fig. S8 Photodegradation of RhB over calcinated B-TiO<sub>2</sub> at different temperature. It is clear to see that the photocatalytic activity of calcinated B-TiO<sub>2</sub> at 300°C decreases slightly. However, the

photocatalytic activity of samples calcinated at 400°C, 500°C, and 600°C are almost vanishes.



**Fig. S9** Photodegradation of RhB under full spectrum light irradiation.