

Supplementary information for
"Quantitative Structure-Property Relationship Study of
Reorganization Energy for *p*-Type Organic
Semiconductors"

Sule Atahan-Evrenk*

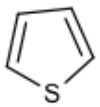
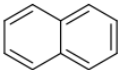
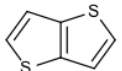
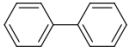
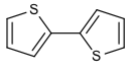
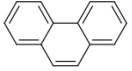
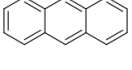
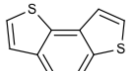
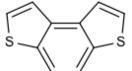
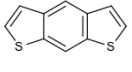
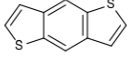
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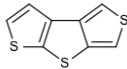
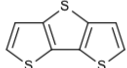
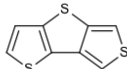
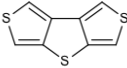
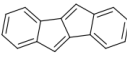
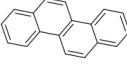
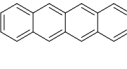
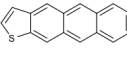
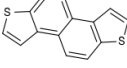
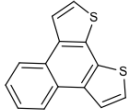
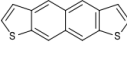
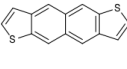
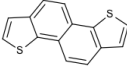
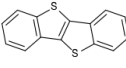
September 19, 2018

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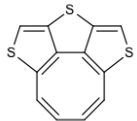
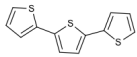
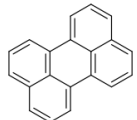
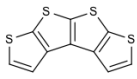
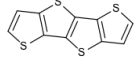
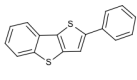
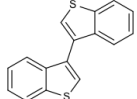
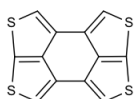
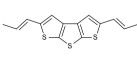
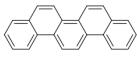
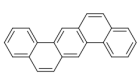
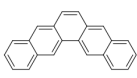
Table 1: The compound set and electronic data.

Compound	Reorganization Energy (meV)	Previous Work	ϵ_n^{homo} (eV)	ϵ_c^{homo} (eV)	IP _{adia} (eV)	IP _{vert} (eV)	References
1 	403	408 ¹	-6.35	-5.98	8.51	8.71	
2 	185	178 ² 182 ³ 186 ⁴ 189 ⁵	-5.80	-5.62	7.61	7.70	[6]
3 	409	410 ⁷	-5.86	-5.47	7.66	7.87	[8]
4 	358		-6.06	-5.71	7.61	7.80	
5 	420	365 ⁹ 424 ¹	-5.54	-5.13	7.12	7.35	[9]
6 	218	212 ²	-5.74	-5.53	7.32	7.43	[2]
7 	138	134 ² 138 ⁴ 141 ⁵	-5.24	-5.10	6.82	6.89	[10]
8 	230		-5.78	-5.55	7.40	7.51	[11]
9 	288		-5.67	-5.39	7.27	7.41	[11]
10 	108	106 ⁴	-5.50	-5.39	7.15	7.20	[12]
11 	165	166 ⁴	-5.48	-5.31	7.11	7.19	[13]

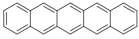
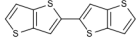
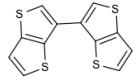
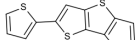
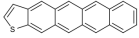
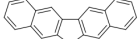
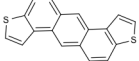
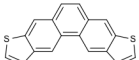
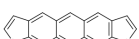
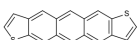
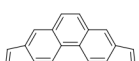
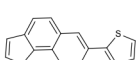
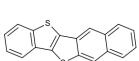
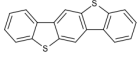
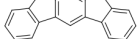
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Compound	Reorganization Energy (meV)	Previous Work	ϵ_n^{homo} (eV)	ϵ_c^{homo} (eV)	IP _{adia} (eV)	IP _{vert} (eV)	References
	193		-5.51	-5.32	7.17	7.26	[14]
	352	350 ⁷ 350 ⁴	-5.61	-5.27	7.20	7.38	[15]
	209		-5.53	-5.33	7.19	7.29	[16]
	187		-5.66	-5.48	7.36	7.45	[14]
	279		-5.42	-5.14	6.87	7.01	[17]
	165	161 ²	-5.53	-5.36	6.95	7.04	[18]
	111	109 ² 114 ⁴ 115 ⁵	-4.87	-4.76	6.28	6.34	[19]
	110		-4.99	-4.88	6.42	6.48	[20]
	243		-5.56	-5.32	6.99	7.11	[11]
	238		-5.51	-5.28	6.96	7.08	[8]
	100	100 ⁴	-5.10	-5.01	6.57	6.62	[11]
	105	106 ⁴	-5.10	-5.00	6.56	6.61	[11]
	280		-5.67	-5.41	7.10	7.24	[21]
	225		-5.59	-5.37	7.04	7.15	[22]

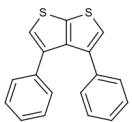
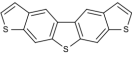
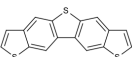
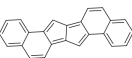
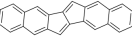
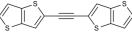
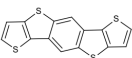
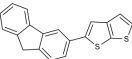
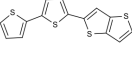
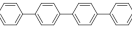
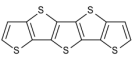
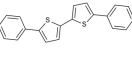
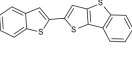
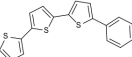
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Compound	Reorganization Energy (meV)	Previous Work	ϵ_n^{homo} (eV)	ϵ_c^{homo} (eV)	IP _{adia} (eV)	IP _{vert} (eV)	References
	378		-5.02	-4.66	6.44	6.63	[8]
	373	321 ⁹ 373 ¹	-5.20	-4.84	6.50	6.70	[9]
	145	142 ²	-4.96	-4.83	6.36	6.43	[23]
	301		-5.72	-5.42	7.19	7.34	[8]
	326	326 ⁴ 330 ⁷	-5.43	-5.11	6.85	7.02	[15]
	299		-5.49	-5.21	6.79	6.95	[8]
	302		-5.59	-5.27	6.92	7.08	[8]
	183		-5.44	-5.25	6.89	6.98	[8]
	215		-5.37	-5.20	6.71	6.82	[8]
	185	180 ² 180 ²⁴ 182 ²⁵	-5.50	-5.32	6.81	6.90	[26]
	168	168 ²⁵	-5.39	-5.23	6.70	6.78	[25]
	178	178 ¹² 177 ²⁵	-5.33	-5.16	6.61	6.70	[12]

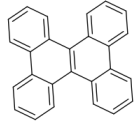
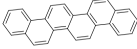
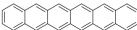
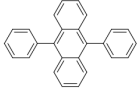
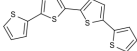
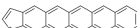
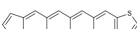
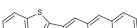
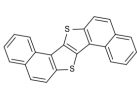
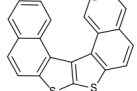
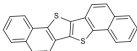
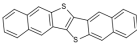
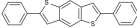
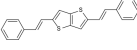
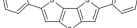
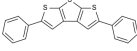

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Compound	Reorganization Energy (meV)	Previous Work	ϵ_n^{homo} (eV)	ϵ_c^{homo} (eV)	IP _{adia} (eV)	IP _{vert} (eV)	References
	93	92 ² 95 ⁵ 98 ⁴ 91 ²⁴ 92 ²⁵ 91 ²⁷	-4.61	-4.52	5.90	5.95	[28]
	365		-5.29	-4.94	6.56	6.76	[8]
	256		-5.61	-5.36	7.02	7.14	[8]
	359		-5.27	-4.92	6.55	6.75	[8]
	96	92.9 ²⁷	-4.71	-4.61	6.01	6.06	[20]
	118	116 ²⁵ 119 ¹² 115 ²⁷	-5.42	-5.31	6.77	6.83	[12]
	155		-5.23	-5.08	6.56	6.63	[11]
	200	204 ¹²	-5.35	-5.16	6.65	6.75	[12]
	94	94 ⁴ 90 ²⁷	-4.81	-4.72	6.13	6.18	[29]
	95	94 ²⁵ 96 ⁴ 96 ²² 91 ²⁷	-4.81	-4.71	6.13	6.18	[30]
	182	181 ¹²	-5.43	-5.25	6.76	6.84	[12]
	134		-5.27	-5.14	6.62	6.68	[11]
	153	148 ²⁷	-5.32	-5.17	6.66	6.73	[31]
	117	114 ²⁷	-5.58	-5.46	6.97	7.02	[32]
	87	82.5 ²⁷	-5.63	-5.55	7.01	7.05	[27]

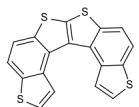
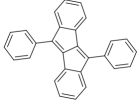
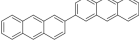
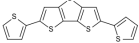
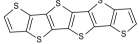
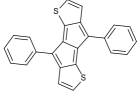
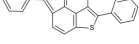
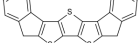
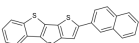
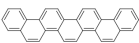
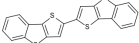
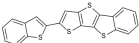
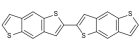
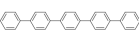
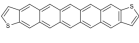
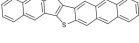
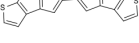
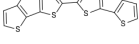
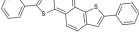
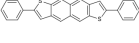
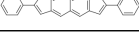
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Compound	Reorganization Energy (meV)	Previous Work	ϵ_n^{homo} (eV)	ϵ_c^{homo} (eV)	IP _{adia} (eV)	IP _{vert} (eV)	References
53 	266		-5.75	-5.52	7.12	7.26	[8]
54 	149	148 ⁴ 142 ²⁷	-5.46	-5.32	6.82	6.89	[33]
55 	118	118 ⁴ 119 ²⁵ 112 ²⁷	-5.53	-5.41	6.92	6.98	[33]
56 	208		-4.99	-4.78	6.21	6.31	[17]
57 	115		-5.24	-5.13	6.46	6.52	[17]
58 	293		-5.24	-4.96	6.46	6.61	[8]
59 	231	231 ²⁵ 221 ²⁷	-5.42	-5.19	6.75	6.86	[22]
60 	320		-5.52	-5.19	6.74	6.91	[8]
61 	362		-5.15	-4.80	6.33	6.53	[8]
62 	309		-5.62	-5.32	6.74	6.90	[34]
63 	307	306 ²² 307 ²⁵ 308 ⁴ 310 ⁷ 314 ⁵ 299 ²⁷	-5.30	-5.00	6.61	6.76	[15]
64 	318		-5.12	-4.83	6.28	6.45	[35]
65 	266		-5.39	-5.13	6.58	6.72	[8]
66 	339		-5.07	-4.75	6.21	6.39	[36]

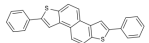
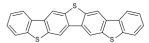
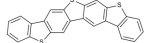
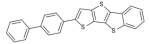
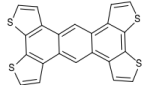
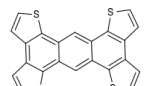
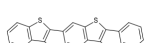
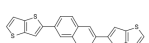
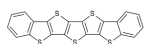
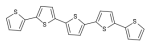
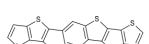
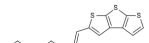
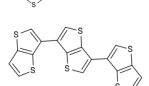
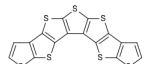
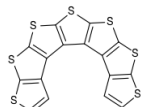
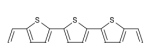
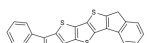
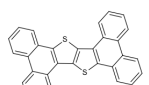
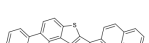
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Compound	Reorganization Energy (meV)	Previous Work	ϵ_n^{homo} (eV)	ϵ_c^{homo} (eV)	IP _{adia} (eV)	IP _{vert} (eV)	References
	193		-5.31	-5.12	6.54	6.63	[37]
	148	144 ²	-5.43	-5.28	6.64	6.72	[2]
	79	79 ²	-4.42	-4.34	5.61	5.65	[38]
	255		-5.10	-4.88	6.37	6.51	[39]
	348	293 ⁹ 345 ¹ 379 ⁵	-5.02	-4.69	6.14	6.34	[1]
	85		-4.50	-4.42	5.71	5.75	[40]
	87		-4.59	-4.50	5.80	5.85	[22]
	114		-5.02	-4.91	6.27	6.32	[31]
	196		-5.33	-5.14	6.56	6.65	[41]
	187		-5.34	-5.15	6.60	6.69	[8]
	189		-5.47	-5.29	6.68	6.77	[41]
	130	130 ²² 134 ⁵	-5.19	-5.06	6.42	6.48	[42]
	267		-5.27	-5.03	6.45	6.58	[43]
	252		-4.97	-4.74	6.06	6.18	[8]
	312		-5.19	-4.91	6.34	6.50	
	225		-5.55	-5.34	6.76	6.88	[44]
	212		-5.57	-5.36	6.85	6.96	[8]

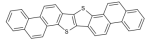
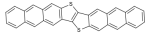
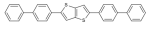
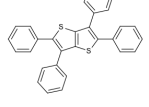
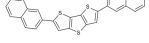
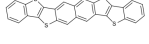
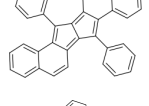
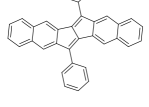
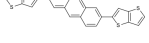
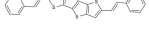
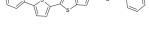
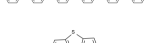

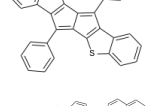
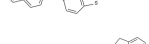

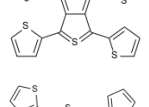
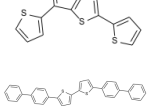

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Compound	Reorganization Energy (meV)	Previous Work	ϵ_n^{homo} (eV)	ϵ_c^{homo} (eV)	IP _{adia} (eV)	IP _{vert} (eV)	References
	211		-5.39	-5.18	6.66	6.76	[8]
	320		-5.31	-5.00	6.52	6.67	[17]
	103		-5.11	-5.00	6.23	6.28	[45]
	328		-5.07	-4.76	6.19	6.37	[46]
	291	290 ⁷	-5.21	-4.93	6.42	6.56	[47]
	414		-4.76	-4.36	5.93	6.13	[48]
	160		-4.72	-4.58	5.92	6.00	[49]
	237	245 ⁵	-5.19	-4.95	6.38	6.50	[50]
	262		-5.30	-5.05	6.42	6.55	[8]
	152	148 ²	-5.41	-5.26	6.55	6.63	[2]
	281		-5.29	-5.01	6.39	6.55	[8]
	264		-5.30	-5.04	6.41	6.55	[51]
	230		-5.31	-5.08	6.42	6.53	[52]
	288		-5.54	-5.25	6.54	6.69	[34]
	79		-4.42	-4.34	5.55	5.59	[22]
	103		-4.98	-4.88	6.14	6.19	[31]
	372		-5.39	-5.01	6.50	6.71	[53]
	337		-5.14	-4.81	6.22	6.41	[54]
	253		-5.32	-5.09	6.41	6.53	[55]
	155		-5.00	-4.87	6.18	6.25	[55]
	116		-5.02	-4.92	6.23	6.28	[55]

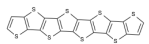
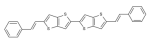
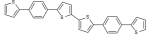
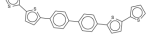
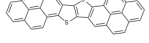
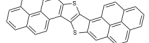
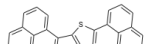

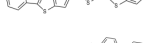
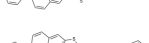




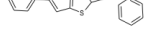
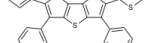
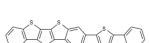
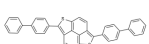
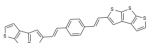
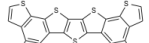

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Compound	Reorganization Energy (meV)	Previous Work	ϵ_n^{homo} (eV)	ϵ_c^{homo} (eV)	IP _{adia} (eV)	IP _{vert} (eV)	References
	232		-5.30	-5.09	6.42	6.53	[55]
	106		-5.50	-5.39	6.71	6.76	[56]
	110		-5.55	-5.43	6.75	6.81	[56]
	292		-5.29	-5.02	6.37	6.52	[8]
	132		-5.21	-5.08	6.40	6.47	[39]
	124		-5.30	-5.18	6.51	6.57	[32]
	179		-5.34	-5.16	6.50	6.58	[57]
	305		-5.26	-4.96	6.27	6.43	[58]
	241		-5.31	-5.07	6.43	6.56	[59]
	309	272 ⁹ 314 ¹	-4.90	-4.61	5.91	6.07	[60]
	236		-5.26	-5.03	6.39	6.51	[57]
	308		-5.14	-4.83	6.24	6.39	[53]
	160		-5.58	-5.42	6.79	6.87	[8]
	207		-5.55	-5.35	6.77	6.87	[8]
	210		-5.62	-5.42	6.86	6.98	[8]
	280	280 ⁴	-5.14	-4.87	6.27	6.41	[15]
	134		-4.98	-4.86	6.05	6.12	[8]
	186		-5.32	-5.14	6.42	6.51	[61]
	181		-5.33	-5.15	6.42	6.51	[61]

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	Compound	Reorganization Energy (meV)	Previous Work	ϵ_n^{homo} (eV)	ϵ_c^{homo} (eV)	IP _{adia} (eV)	IP _{vert} (eV)	References
124		181		-5.45	-5.27	6.53	6.62	[61]
125		85		-4.89	-4.81	5.96	6.00	[61]
126		311		-5.19	-4.90	6.18	6.34	[8]
127		290		-5.25	-4.97	6.32	6.47	[8]
128		261		-5.11	-4.87	6.13	6.26	
129		124		-5.12	-5.00	6.24	6.30	[31]
130		242		-4.89	-4.66	5.97	6.09	[17]
131		141		-5.06	-4.93	6.13	6.20	[17]
132		240		-5.10	-4.87	6.09	6.22	[58]
133		232		-4.91	-4.70	5.89	6.00	[8]
134		301		-4.95	-4.67	5.91	6.08	[8]
135		254		-5.49	-5.23	6.41	6.53	[34]
136		257		-5.15	-4.90	6.15	6.28	[8]
137		348		-4.71	-4.37	5.73	5.90	[48]
138		165		-4.70	-4.55	5.76	5.84	[49]
139		300		-5.05	-4.76	6.00	6.15	[8]
140		182		-4.51	-4.35	5.62	5.72	[8]
141		452		-5.22	-4.75	6.18	6.43	[8]
142		309		-5.06	-4.78	6.02	6.19	[62]

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Compound	Reorganization Energy (meV)	Previous Work	ϵ_n^{homo} (eV)	ϵ_c^{homo} (eV)	IP _{adia} (eV)	IP _{vert} (eV)	References
143 	268		-5.09	-4.83	6.15	6.28	[8]
144 	232		-4.85	-4.64	5.78	5.89	[8]
145 	323		-5.01	-4.71	5.92	6.09	[63]
146 	287		-5.12	-4.84	5.99	6.15	[64]
147 	76		-5.21	-5.14	6.27	6.31	[61]
148 	140		-4.92	-4.79	5.95	6.02	[61]
149 	123		-5.17	-5.05	6.24	6.30	[61]
150 	270		-5.15	-4.89	6.14	6.28	[51]
151 	145		-5.14	-5.01	6.19	6.26	[65]
152 	108		-5.19	-5.08	6.29	6.35	[65]
153 	258	253 ⁹ 301 ¹	-4.81	-4.56	5.74	5.87	[66]
154 	293		-5.11	-4.85	6.07	6.23	[46]
155 	305		-5.12	-4.84	6.17	6.33	[8]
156 	481		-5.11	-4.63	6.02	6.28	[8]
157 	205		-5.29	-5.09	6.30	6.40	[8]
158 	181		-4.69	-4.54	5.73	5.82	[49]
159 	257		-5.02	-4.76	5.96	6.08	[8]
160 	199		-5.22	-5.02	6.27	6.37	[67]
161 	275		-4.99	-4.73	5.92	6.06	[8]
162 	290		-4.88	-4.61	5.77	5.94	[8]
163 	83		-5.04	-4.95	5.95	5.99	[45]

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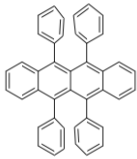
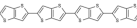
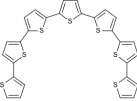
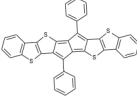

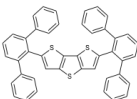
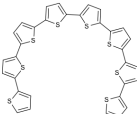
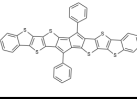
Compound	Reorganization Energy (meV)	Previous Work	ϵ_n^{homo} (eV)	ϵ_c^{homo} (eV)	IP _{adia} (eV)	IP _{vert} (eV)	References
164 	147	160 ⁵	-4.69	-4.56	5.80	5.87	[68]
165 	301		-4.94	-4.65	5.81	5.97	[8]
166 	265	273 ¹	-4.78	-4.54	5.63	5.76	[15]
167 	308		-4.72	-4.42	5.66	5.81	[48]
168 	133		-5.11	-4.98	6.12	6.19	[8]
169 	363		-5.13	-4.80	6.12	6.31	[8]
170 	268	256 ¹	-4.77	-4.52	5.54	5.68	[9]
171 	266		-4.67	-4.41	5.55	5.68	[48]

Table 2: Pairwise correlation matrix with Pearson’s correlation coefficient for the RE (λ) and electronic and structural descriptors.

	λ	E_n	ϵ_n^{homo}	ϵ_c^{homo}	ϵ_n^{lumo}	ϵ_c^{lumo}	IP_{vert}	$QPol_{ave}$	FsdRing	RotBond	Scount	vdWsa	Rcount
λ	1.00	-0.26	-0.20	0.08	0.16	0.05	0.19	-0.05	-0.46	0.44	0.34	-0.05	-0.22
E_n	-0.26	1.00	-0.32	-0.40	0.30	0.30	0.48	-0.64	-0.24	-0.43	-0.93	-0.48	-0.60
ϵ_n^{homo}	-0.20	-0.32	1.00	0.96	-0.87	-0.86	-0.94	0.64	0.20	0.35	0.13	0.55	0.52
ϵ_c^{homo}	0.08	-0.40	0.96	1.00	-0.84	-0.86	-0.90	0.64	0.08	0.47	0.23	0.54	0.47
ϵ_n^{lumo}	0.16	0.30	-0.87	-0.84	1.00	0.99	0.85	-0.65	-0.31	-0.24	-0.10	-0.50	-0.57
ϵ_c^{lumo}	0.05	0.30	-0.86	-0.86	0.99	1.00	0.84	-0.66	-0.23	-0.32	-0.10	-0.52	-0.54
IP_{vert}	0.19	0.48	-0.94	-0.90	0.85	0.84	1.00	-0.82	-0.29	-0.46	-0.22	-0.75	-0.71
$QPol_{ave}$	-0.05	-0.64	0.64	0.64	-0.65	-0.66	-0.82	1.00	0.28	0.62	0.33	0.92	0.84
FsdRing	-0.46	-0.24	0.20	0.08	-0.31	-0.23	-0.29	0.28	1.00	-0.50	0.12	0.23	0.68
RotBond	0.44	-0.43	0.35	0.47	-0.24	-0.32	-0.46	0.62	-0.50	1.00	0.26	0.65	0.21
Scount	0.34	-0.93	0.13	0.23	-0.10	-0.10	-0.22	0.33	0.12	0.26	1.00	0.13	0.30
vdWsa	-0.05	-0.48	0.55	0.54	-0.50	-0.52	-0.75	0.92	0.23	0.65	0.13	1.00	0.83
Rcount	-0.22	-0.60	0.52	0.47	-0.57	-0.54	-0.71	0.84	0.68	0.21	0.30	0.83	1.00

E_n : Total electronic energy, ϵ : Molecular orbital energy, IP_{vert} : Vertical Ionization Potential, $QPol_{ave}$: Average polarizability, FsdRing: Fused ring count, RotBond: Rotatable bond count, Scount: Sulfur atom count, vdWsa: van der Waals surface area, Rcount: Ring count Electronic descriptors calculated at the B3LYP/6-31G(d,p) level of theory.

2 Molecular transform descriptors

Following Soltzberg and Wilkins, we limit s into the interval $[1, 31]$ divided into 100 equal pieces. However, instead of the binary representation they used in which the intervals with $I(s) = 0$ were coded as “1”, and “0” otherwise, we retained the original $I(s)$ values. Figure 1 plots the comparison of $I(s)$ distributions for the pentacene with the DFT and molecular mechanics (MM) geometries.

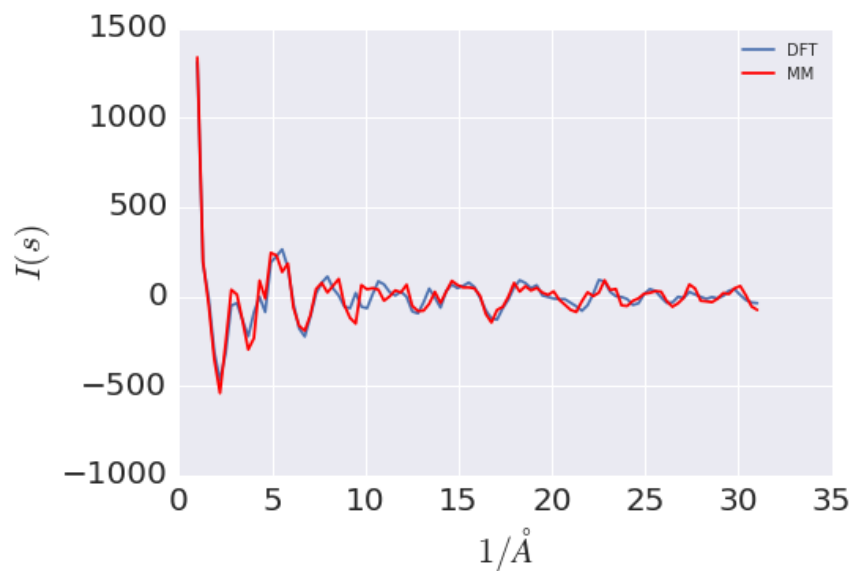


Figure 1: 3D structural descriptor based on the molecular transforms for pentacene with the DFT and MM geometries.

3 Principle Components

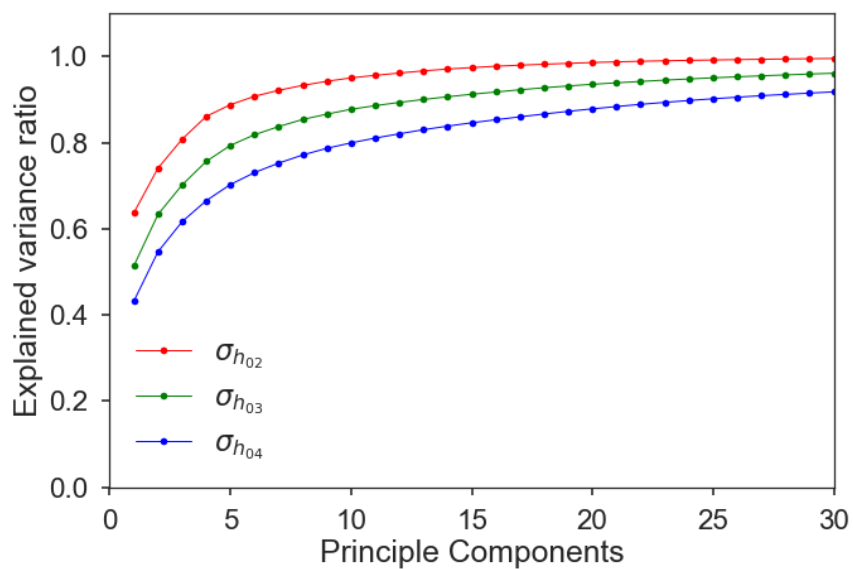


Figure 2: Explained variance ratio by the principle components in the signature (*left*) descriptor spaces

References

- (1) da Silva Filho, D. A.; Coropceanu, V.; Fichou, D.; Gruhn, N. E.; Bill, T. G.; Gierschner, J.; Cornil, J.; Brédas, J.-L. Hole-vibronic coupling in oligothiophenes: impact of backbone torsional flexibility on relaxation energies. *Philos. Trans. R. Soc., A* **2007**, *365*, 1435–1452.
- (2) Mallocci, G.; Cappellini, G.; Mulas, G.; Mattoni, A. Electronic and optical properties of families of polycyclic aromatic hydrocarbons: A systematic (time-dependent) density functional theory study. *Chem. Phys.* **2011**, *384*, 19 – 27.
- (3) Mallocci, G.; Mulas, G.; Cappellini, G.; Joblin, C. Time-dependent density functional study of the electronic spectra of oligoacenes in the charge states 1, 0, +1, and +2. *Chem. Phys.* **2007**, *340*, 43 – 58.
- (4) Coropceanu, V.; Cornil, J.; Filho, D. S.; A, D.; Olivier, Y.; Silbey, R.; Brédas, J.-L. Charge transport in organic semiconductors. *Chem. Rev* **2007**, *107*, 926–952.
- (5) Yavuz, I.; Martin, B. N.; Park, J.; Houk, K. N. Theoretical Study of the Molecular Ordering, Paracrystallinity, And Charge Mobilities of Oligomers in Different Crystalline Phases. *J. Am. Chem. Soc.* **2015**, *137*, 2856–2866.
- (6) Karl, N. Charge carrier transport in organic semiconductors. *Synthetic Metals* **2003**, *133*, 649 – 657, Proceedings of the Yamada Conference LVI. The Fourth International Symposium on Crystalline Organic Metals, Superconductors and Ferromagnets (ISCOM 2001).
- (7) Zhang, Y.; Cai, X.; Bian, Y.; Li, X.; Jiang, J. Heteroatom Substitution of Oligothiophenes: From Good p-Type Semiconductors to Good Ambipolar Semiconductors for Organic Field-Effect Transistors. *J. Phys. Chem. C* **2008**, *112*, 5148–5159.
- (8) Cinar, M. E.; Ozturk, T. Thienothiophenes, Dithienothiophenes, and Thienoacenes: Syntheses, Oligomers, Polymers, and Properties. *Chem. Rev.* **2015**, *115*, 3036–3140.
- (9) Hutchison, G. R.; Ratner, M. A.; Marks, T. J. Hopping Transport in Conductive Heterocyclic Oligomers: Reorganization Energies and Substituent Effects. *J. Am. Chem. Soc.* **2005**, *127*, 2339–2350.

- (10) Aleshin, A. N.; Lee, J. Y.; Chu, S. W.; Kim, J. S.; Park, Y. W. Mobility studies of field-effect transistor structures based on anthracene single crystals. *Appl. Phys. Lett.* **2004**, *84*, 5383–5385.
- (11) Osaka, I.; Shinamura, S.; Abe, T.; Takimiya, K. Naphthodithiophenes as building units for small molecules to polymers; a case study for in-depth understanding of structure-property relationships in organic semiconductors. *J. Mater. Chem. C* **2013**, *1*, 1297–1304.
- (12) Coropceanu, V.; Kwon, O.; Wex, B.; Kaafarani, B. R.; Gruhn, N. E.; Durivage, J. C.; Neckers, D. C.; Brédas, J.-L. Vibronic Coupling in Organic Semiconductors: The Case of Fused Polycyclic Benzene-Thiophene Structures. *Chem. Eur. J.* **2006**, *12*, 2073–2080.
- (13) Tomoya, K.; Eigo, M.; Kazuo, T. 2,6-Dialkylbenzo[1,2-b:4,5-b']dithiophenes (C_n-BDTs) as Soluble Organic Semiconductors for Solution-processed Organic Field-effect Transistors. *Chem. Lett.* **2008**, *37*, 284–285.
- (14) Janssen, M. J.; De Jong, F. Synthesis, oxidation, and electronic spectra of four dithienothiophenes. *J. Org. Chem.* **1971**, *36*, 1645–1648.
- (15) Zhang, X.; Cote, A. P.; Matzger, A. J. Synthesis and Structure of Fused alpha-Oligothiophenes with up to Seven Rings. *J. Am. Chem. Soc.* **2005**, *127*, 10502–10503.
- (16) De Jong, F.; Janssen, M. J. Synthesis of dithienothiophenes. *J. Org. Chem.* **1971**, *36*, 1998–2000.
- (17) Kawase, T.; Fujiwara, T.; Kitamura, C.; Konishi, A.; Hirao, Y.; Matsumoto, K.; Kurata, H.; Kubo, T.; Shinamura, S.; Mori, H.; Miyazaki, E.; Takimiya, K. Dinaphthopentalenes: Pentalene Derivatives for Organic Thin-Film Transistors. *Angew. Chem. Int. Ed.* **2010**, *49*, 7728–7732.
- (18) Kunugi, Y.; Ikari, M.; Okamoto, K. Chrysene Derivatives as New Organic Semiconductors for Organic Field-effect Transistors. *ECS Trans.* **2010**, *25*, 11–17.
- (19) Goldmann, C.; Haas, S.; Krellner, C.; Pernstich, K. P.; Gundlach, D. J.; Batlogg, B. Hole

- mobility in organic single crystals measured by a flip-crystal field-effect technique. *J. Appl. Phys.* **2004**, *96*, 2080–2086.
- (20) Tang, M. L.; Okamoto, T.; Bao, Z. High-Performance Organic Semiconductors: Asymmetric Linear Acenes Containing Sulphur. *J. Am. Chem. Soc.* **2006**, *128*, 16002–16003.
- (21) Shinamura, S.; Miyazaki, E.; Takimiya, K. Synthesis, Properties, Crystal Structures, and Semiconductor Characteristics of Naphtho[1,2-b:5,6-b']dithiophene and -diselenophene Derivatives. *J. Org. Chem.* **2010**, *75*, 1228–1234.
- (22) Takimiya, K.; Shinamura, S.; Osaka, I.; Miyazaki, E. Thienoacene-Based Organic Semiconductors. *Adv. Mater.* **2011**, *23*, 4347–4370.
- (23) Kim, S. H.; Yang, Y. S.; Lee, J. H.; Lee, J.-I.; Chu, H. Y.; Lee, H.; Oh, J.; Do, L.-M.; Zyung, T. Organic field-effect transistors using perylene. *Opt. Mater.* **2003**, *21*, 439 – 443.
- (24) Nguyen, T. P.; Shim, J. H.; Lee, J. Y. Density Functional Theory Studies of Hole Mobility in Picene and Pentacene Crystals. *J. Phys. Chem. C* **2015**, *119*, 11301–11310.
- (25) Mamada, M.; Yamashita, Y. *Polycyclic Arenes and Heteroarenes*; Wiley-VCH Verlag GmbH & Co. KGaA, 2015; pp 277–308.
- (26) Okamoto, H.; Kawasaki, N.; Kaji, Y.; Kubozono, Y.; Fujiwara, A.; Yamaji, M. Air-assisted High-performance Field-effect Transistor with Thin Films of Picene. *J. Am. Chem. Soc.* **2008**, *130*, 10470–10471.
- (27) Zhu, R.; Duan, Y.-A.; Geng, Y.; Wei, C.-Y.; Chen, X.-Y.; Liao, Y. Theoretical evaluation on the reorganization energy of five-ring-fused benzothiophene derivatives. *Comp. Theor. Chem.* **2016**, *1078*, 16–22.
- (28) Knipp, D.; Street, R. A.; Völkel, A.; Ho, J. Pentacene thin film transistors on inorganic dielectrics: Morphology, structural properties, and electronic transport. *J. Appl. Phys.* **2003**, *93*, 347–355.

- (29) Lehnherr, D.; Waterloo, A. R.; Goetz, K. P.; Payne, M. M.; Hampel, F.; Anthony, J. E.; Jurchescu, O. D.; Tykwinski, R. R. Isomerically Pure syn-Anthradithiophenes: Synthesis, Properties, and FET Performance. *Org. Lett.* **2012**, *14*, 3660–3663.
- (30) Laquindanum, J. G.; Katz, H. E.; Lovinger, A. J. Synthesis, Morphology, and Field-Effect Mobility of Anthradithiophenes. *J. Am. Chem. Soc.* **1998**, *120*, 664–672.
- (31) Mori, T.; Nishimura, T.; Yamamoto, T.; Doi, I.; Miyazaki, E.; Osaka, I.; Takimiya, K. Consecutive Thiophene-Annulation Approach to π -Extended Thienoacene-Based Organic Semiconductors with [1]Benzothieno[3,2-b][1]benzothiophene (BTBT) Substructure. *J. Am. Chem. Soc.* **2013**, *135*, 13900–13913.
- (32) Liu, W.-J.; Zhou, Y.; Ma, Y.; Cao, Y.; Wang, J.; Pei, J. Thin Film Organic Transistors from Air-Stable Heteroarenes: Anthra[1,2-b:4,3-b':5,6-b'':8,7-b''']tetrathiophene Derivatives. *Org. Lett.* **2007**, *9*, 4187–4190.
- (33) Wex, B.; Kaafarani, B. R.; Schroeder, R.; Majewski, L. A.; Burckel, P.; Grell, M.; Neckers, D. C. New organic semiconductors and their device performance as a function of thiophene orientation. *J. Mater. Chem.* **2006**, *16*, 1121–1124.
- (34) Gundlach, D. J.; Lin, Y.-Y.; Jackson, T. N.; Schlom, D. G. Oligophenyl-based organic thin film transistors. *Appl. Phys. Lett.* **1997**, *71*, 3853–3855.
- (35) Katz, H. E.; Lovinger, A. J.; Hong, X. M.; Dodabalapur, A.; Johnson, J.; Wang, B.-C.; Raghavachari, K. Design of organic transistor semiconductors for logic elements, displays, and sensors. *Proc. SPIE* **2001**, *4466*, 20–30.
- (36) Funahashi, M.; Kato, T. Design of liquid crystals: from a nematogen to thiophene-based π -conjugated mesogens. *Liq. Cryst.* **2015**, *42*, 909–917.
- (37) Ueda, Y.; Tsuji, H.; Tanaka, H.; Nakamura, E. Synthesis, Crystal Packing, and Ambipolar Carrier Transport Property of Twisted Dibenzo[g,p]chrysenes. *Chem. - Asian J* **2014**, *9*, 1623–1628.

- (38) Watanabe, M.; Chang, Y. J.; Liu, S.-W.; Chao, T.-H.; Goto, K.; Minarul, I.; Yuan, C.-H.; Tao, Y.-T.; Shinmyozu, T.; Chow, T. J. The synthesis, crystal structure and charge-transport properties of hexacene. *Nat Chem* **2012**, *4*, 574–578.
- (39) Brusso, J. L.; Hirst, O. D.; Dadvand, A.; Ganesan, S.; Cicoira, F.; Robertson, C. M.; Oakley, R. T.; Rosei, F.; Perepichka, D. F. Two-Dimensional Structural Motif in Thienoacene Semiconductors: Synthesis, Structure, and Properties of Tetrathienoanthracene Isomers. *Chem. Mater.* **2008**, *20*, 2484–2494.
- (40) Tang, M. L.; Mannsfeld, S. C. B.; Sun, Y.-S.; Becerril, H. A.; Bao, Z. Pentaceno[2,3-b]thiophene, a Hexacene Analogue for Organic Thin Film Transistors. *J. Am. Chem. Soc.* **2009**, *131*, 882–883.
- (41) Yamamoto, T.; Shinamura, S.; Miyazaki, E.; Takimiya, K. Three Structural Isomers of Dinaphthothieno[3,2-b]thiophenes: Elucidation of Physicochemical Properties, Crystal Structures, and Field-Effect Transistor Characteristics. *Bull. Chem. Soc. Jpn.* **2010**, *83*, 120–130.
- (42) Yamamoto, T.; Takimiya, K. Facile Synthesis of Highly π -Extended Heteroarenes, Dinaphtho[2,3-b:2',3'-f]chalcogenopheno[3,2-b]chalcogenophenes, and Their Application to Field-Effect Transistors. *J. Am. Chem. Soc.* **2007**, *129*, 2224.
- (43) Takimiya, K.; Kunugi, Y.; Konda, Y.; Niihara, N.; Otsubo, T. 2,6-Diphenylbenzo[1,2-b:4,5-b']dichalcogenophenes: A New Class of High-Performance Semiconductors for Organic Field-Effect Transistors. *J. Am. Chem. Soc.* **2004**, *126*, 5084–5085.
- (44) Shi, J.; Li, Y.; Jia, M.; Xu, L.; Wang, H. Organic semiconductors based on annelated [small beta]-oligothiophenes and its application for organic field-effect transistors. *J. Mater. Chem.* **2011**, *21*, 17612–17614.
- (45) Ito, K.; Suzuki, T.; Sakamoto, Y.; Kubota, D.; Inoue, Y.; Sato, F.; Tokito, S. Oligo(2,6-anthrylene)s: Acene-Oligomer Approach for Organic Field-Effect Transistors. *Angew. Chem. Int. Ed.* **2003**, *42*, 1159–1162.

- (46) Sun, Y.; Ma, Y.; Liu, Y.; Lin, Y.; Wang, Z.; Wang, Y.; Di, C.; Xiao, K.; Chen, X.; Qiu, W.; Zhang, B.; Yu, G.; Hu, W.; Zhu, D. High-Performance and Stable Organic Thin-Film Transistors Based on Fused Thiophenes. *Adv. Funct. Mater.* **2006**, *16*, 426–432.
- (47) Liu, Y.; Sun, X.; Di, C.-a.; Liu, Y.; Du, C.; Lu, K.; Ye, S.; Yu, G. Hexathienoacene: Synthesis, Characterization, and Thin-Film Transistors. *Chem. - Asian J.* **2010**, *5*, 1550–1554.
- (48) Dai, G.; Chang, J.; Shi, X.; Zhang, W.; Zheng, B.; Huang, K.-W.; Chi, C. Thienoacene-Fused Pentalenes: Syntheses, Structures, Physical Properties and Applications for Organic Field-Effect Transistors. *Chem. - Eur. J.* **2015**, *21*, 2019–2028.
- (49) Yoshihito, K.; Kazuo, T.; Kazuo, Y.; Yoshio, A.; Tetsuo, O. Organic Field-Effect Transistors Using Di(2-thienyl)naphthodithiophenes as Active Layers. *Chem. Lett.* **2002**, *31*, 958–959.
- (50) Afonina, I.; Skabara, P. J.; Vilela, F.; Kanibolotsky, A. L.; Forgie, J. C.; Bansal, A. K.; Turnbull, G. A.; Samuel, I. D. W.; Labram, J. G.; Anthopoulos, T. D.; Coles, S. J.; Hursthouse, M. B. Synthesis and characterisation of new diindenodithienothiophene (DITT) based materials. *J. Mater. Chem.* **2010**, *20*, 1112–1116.
- (51) Huang, P.-Y.; Chen, L.-H.; Kim, C.; Chang, H.-C.; Liang, Y.-j.; Feng, C.-Y.; Yeh, C.-M.; Ho, J.-C.; Lee, C.-C.; Chen, M.-C. High-Performance Bottom-Contact Organic Thin-Film Transistors Based on Benzo[d,d']thieno[3,2-b;4,5-b']dithiophenes (BTDTs) Derivatives. *ACS Appl. Mater. Interfaces* **2012**, *4*, 6992–6998.
- (52) Laquindanum, J. G.; Katz, H. E.; Lovinger, A. J.; Dodabalapur, A. Benzodithiophene Rings as Semiconductor Building Blocks. *Adv. Mater.* **1997**, *9*, 36–39.
- (53) Tan, L.; Zhang, L.; Jiang, X.; Yang, X.; Wang, L.; Wang, Z.; Li, L.; Hu, W.; Shuai, Z.; Li, L.; Zhu, D. A Densely and Uniformly Packed Organic Semiconductor Based on Annelated beta-Trithiophenes for High-Performance Thin Film Transistors. *Adv. Funct. Mater.* **2009**, *19*, 272–276.
- (54) Bao, Z.; A. Rogers, J.; E. Katz, H. Printable organic and polymeric semiconducting materials and devices. *J. Mater. Chem.* **1999**, *9*, 1895–1904.

- (55) Shinamura, S.; Osaka, I.; Miyazaki, E.; Nakao, A.; Yamagishi, M.; Takeya, J.; Takimiya, K. Linear- and Angular-Shaped Naphthodithiophenes: Selective Synthesis, Properties, and Application to Organic Field-Effect Transistors. *J. Am. Chem. Soc.* **2011**, *133*, 5024–5035.
- (56) Sirringhaus, H.; Friend, R. H.; Wang, C.; Leuninger, J.; Mullen, K. Dibenzothienobisbenzothiophene—a novel fused-ring oligomer with high field-effect mobility. *J. Mater. Chem.* **1999**, *9*, 2095–2101.
- (57) Seok, Y. Y.; Takuma, Y.; Chihaya, A. Organic Single-Crystal Transistors Based on π -Extended Heteroheptacene Microribbons. *Bull. Chem. Soc. Jpn.* **2012**, *85*, 1186–1191.
- (58) Kim, H.-S.; Kim, Y.-H.; Kim, T.-H.; Noh, Y.-Y.; Pyo, S.; Yi, M. H.; Kim, D.-Y.; Kwon, S.-K. Synthesis and Studies on 2-Hexylthieno[3,2-b]thiophene End-Capped Oligomers for OTFTs. *Chem. Mater.* **2007**, *19*, 3561–3567.
- (59) Okamoto, T.; Kudoh, K.; Wakamiya, A.; Yamaguchi, S. General Synthesis of Thiophene and Selenophene-Based Heteroacenes. *Org. Lett.* **2005**, *7*, 5301–5304.
- (60) Tasaka, S.; Katz, H.; Hutton, R.; Orenstein, J.; Fredrickson, G.; Wang, T. Electrical conductivity of α -quinquethiophene/stearic acid Langmuir-Blodgett films doped with iodine. *Synth. Met.* **1986**, *16*, 17 – 30.
- (61) Sokolov, A. N.; Atahan-Evrenk, S.; Mondal, R.; Akkerman, H. B.; Sánchez-Carrera, R. S.; Granados-Focil, S.; Schrier, J.; Mannsfeld, S. C. B.; Zoombelt, A. P.; Bao, Z.; Aspuru-Guzik, A. From computational discovery to experimental characterization of a high hole mobility organic crystal. *Nat. Commun.* **2011**, *2*, 437–8.
- (62) Yanagi, H.; Araki, Y.; Ohara, T.; Hotta, S.; Ichikawa, M.; Taniguchi, Y. Comparative Carrier Transport Characteristics in Organic Field-Effect Transistors with Vapor-Deposited Thin Films and Epitaxially Grown Crystals of Biphenyl-Capped Thiophene Oligomers. *Adv. Funct. Mater.* **2003**, *13*, 767–773.
- (63) Facchetti, A.; Yoon, M.-H.; Katz, H. E.; Marks, T. J. Materials for n-type organic electronics: synthesis and properties of fluoroarene-thiophene semiconductors. 2003.

- (64) Yoon, M.-H.; Facchetti, A.; Stern, C. E.; Marks, T. J. Fluorocarbon-Modified Organic Semiconductors: Molecular Architecture, Electronic, and Crystal Structure Tuning of Arene- versus Fluoroarene-Thiophene Oligomer Thin-Film Properties. *J. Am. Chem. Soc.* **2006**, *128*, 5792–5801.
- (65) Kang, M. J.; Miyazaki, E.; Osaka, I.; Takimiya, K.; Nakao, A. Diphenyl Derivatives of Dinaphtho[2,3-b:2',3'-f]thieno[3,2-b]thiophene: Organic Semiconductors for Thermally Stable Thin-Film Transistors. *ACS Appl. Mater. Interfaces* **2013**, *5*, 2331–2336.
- (66) Peng, X.; Horowitz, G.; Fichou, D.; Garnier, F. All-organic thin-film transistors made of alpha-sexithienyl semiconducting and various polymeric insulating layers. *Appl. Phys. Lett.* **1990**, *57*, 2013.
- (67) Zhang, S.; Guo, Y.; Zhang, Y.; Liu, R.; Li, Q.; Zhan, X.; Liu, Y.; Hu, W. Synthesis, self-assembly, and solution-processed nanoribbon field-effect transistor of a fused-nine-ring thienoacene. *Chem. Commun.* **2010**, *46*, 2841–2843.
- (68) Podzorov, V.; Sysoev, S.; Loginova, E.; Pudalov, V.; Gershenson, M. Single-crystal organic field effect transistors with the hole mobility $\approx 8 \text{ cm}^2/\text{Vs}$. *Appl. Phys. Lett.* **2003**, *83*, 3504.