Supplementary Material

Photostability and photocatalytic degradation of ionic liquids

under solar light

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Figure S1. XRD pattern (A), -196 °C N₂ adsorption-desorption isotherm (B) and UV-vis



diffuse absorbance spectrum of the commercial TiO₂ (Aeroxide@P25).

Characterization techniques description: X-ray diffraction (XRD) pattern was recorded on a Bruker D8 diffractometer with a scintillation detector, using Cu-K α source, with a scan step of 1°·min⁻¹ between 20 and 80° of 20. N₂ adsorption-desorption at -196 °C was carried out using a Micromeritics TriStar 123 static volumetric system. The sample was previously outgassed under vacuum at 150 °C overnight in a Florprep 060 Micromeritics device. Shimadzu 2501PC UV–vis spectrophotometer was used to record UV–vis diffuse reflectance spectrum (UV-vis DRS) in the 250–800 nm region using BaSO₄ as reference material.



Figure S2. Molecular model of ion independent (C+A) and ion-pair (CA) for IL (example: [Bmim][Cl])



Figure S3. Calculated UV-Visible spectra at TD-B3LYP/6-311**G++ computational level



Figure S4. Calculated HOMO and LUMO molecular orbitals of [Bmim] cation at TD-B3LYP/6-311**G++ computational level



Figure S5. Electrostatic potential surface of [Bmim] cation and [Bmim][FAP] ionic pair in fundamental (S₀) and first excited (S₁) electronic state calculated at TD-B3LYP/6-311**G++ computational level





Figure S6. Percentage of IL adsorbed onto TiO₂ in dark tests.

Figure S7. Evolution of the TOC and the carbon amount measured from IL, organic acids and non-identified byproducts upon irradiation time: a) [Bmim][Cl] and b) [Bu₄MePy][Cl].

