Supplementary information for Employing a novel O₃/H₂O₂+BiPO₄/UV synergy technique to deal with thiourea-containing photovoltaic wastewater

Text S1. HPLC Analytical Methods

The separation was performed on a 300Extend-C18 column (3.5μ m, 4.6×150 mm, Agilent, USA) with a flowrate of 1 mL/min at 30°C. The mobile phase was composed with 60% water (containing 0.1% formic acid) and 40% methyl alcohol. The sample injection volume was 10µL and the detection wavelength was set at 236nm.

Text S2. UPLC-IMS-QToF-MS Analytical Methods.

The mineralization byproducts of thiourea were analyzed via high resolution mass spectrometry analysis, carried on a Water I-Class Acquity UPLC(Waters, UK) coupled with a Vion IMS QToF(Waters, UK) using a SeQuant ZIC-HILIC column(100 mm × 2.1 mm i.d., 3.5 μ m) (Merck, Germany). The mobile phase A was 50 mM ammonium formate in water, and mobile phase B was acetonitrile. Metabolites were separated via gradient elution under the following conditions: 0–10 min, 90–50% B; 10–12 min, 50–90% B; 12–15 min, 90% B; and the column was maintained at 45 °C. The flow rate was 0.4 mL/min. The parameters of high-resolution mass spectrometry analysis on full scan mass spectrometry were as follows: MS range, m/z 50–1000; scan time, 0.3 s; CE 6 eV; desolvation temperature, 500 °C; source temperature, 120 °C; desolvation gas, 1000 L/h; cone gas, 50 L/h; capillary voltage, 2000 V. The lock correction (lock sprayer reference: mass,556.2766 m/z; interval, 0.5 min; sample time, 0.5 min; CE, 6 eV; flow

rate 10 μ L/min) enabled isotopic *m*/*z* screen, with tolerance of 3 mDa mass error, the

expected adduct -H. Data were acquired and processed using UNIFI 1.8.1.

Factors	Statistic						
	Sum of Squares	df	Mean Square	F Value	p-value Prob > F	Remark	
Model	1979809	9	219978.8	320.5533	< 0.0001	significant	
X1-H2O2	1842704	1	1842704	2685.19	< 0.0001	significant	
X ₂ -O3	16768.22	1	16768.22	24.43466	0.0017	significant	
X ₃ -pH	19296.7	1	19296.7	28.11917	0.0011	significant	
X_1X_2	19299.58	1	19299.58	28.12337	0.0011	significant	
X_1X_3	9347.308	1	9347.308	13.6209	0.0077	significant	
X_2X_3	292.0168	1	292.0168	0.425527	0.5350		
X_1^2	56907.21	1	56907.21	82.92524	< 0.0001	significant	
X_2^2	5064.43	1	5064.43	7.379892	0.0299	significant	
X_3^2	13777.19	1	13777.19	20.07613	0.0029	significant	
Pure Error	1470.797	4	367.6993				
Cor Total	1984613	16					
\mathbb{R}^2	0.99758						
Adj R ²	0.994467						
Adeq Precision	54.68905						

Table S1 Analysis of variance (ANOVA) for the prediction of TOC by the quadratic model

Table S2	Analysis	of variance	(ANOVA)	for the	prediction	of H ₂ O ₂	residue b	y the
	•		quadrati	c model	-			•

Factors	Statistic						
	Sum of Squares	df	Mean Square	F Value	p-value Prob > F	Remark	
Model	668.5874	9	74.28749	5.404299	0.0184	significant	
X1-H2O2	295.245	1	295.245	21.47861	0.0024	Significant	
X ₂ -O3	26.645	1	26.645	1.938382	0.2065		
X ₃ -pH	20.48	1	20.48	1.489888	0.2618		
X_1X_2	67.24	1	67.24	4.891605	0.0626		
X_1X_3	75.69	1	75.69	5.506329	0.0513		
X_2X_3	8.41	1	8.41	0.611814	0.4597		
X_1^2	3.260632	1	3.260632	0.237206	0.6411		
X_2^2	11.88379	1	11.88379	0.864527	0.3834		
X_3^2	165.528	1	165.528	12.0419	0.0104	significant	
Pure Error	78.512	4	19.628				
Cor Total	764.8094	16					
\mathbb{R}^2	0.874188						
Adj R ²	0.71243						
Adeq Precision	8.835728						



Figure S1 SEM images of BiPO₄



Figure S2 XRD pattern of BiPO₄



Figure S3 Validation of TOC and H_2O_2 residue model using the plot of predicted values versus observed experimental data.





Figure S4 UPLC-IMS-QToF-MS Analysis data. (a)&(b): Intermediates after O₃/H₂O₂, (c)&(d): Intermediates after BiPO₄