

Supporting Information

A convenient and efficient precursor transformation route to well-dispersed, stable, and highly accessible supported Au nanocatalysts with excellent catalytic hydrogenation performances

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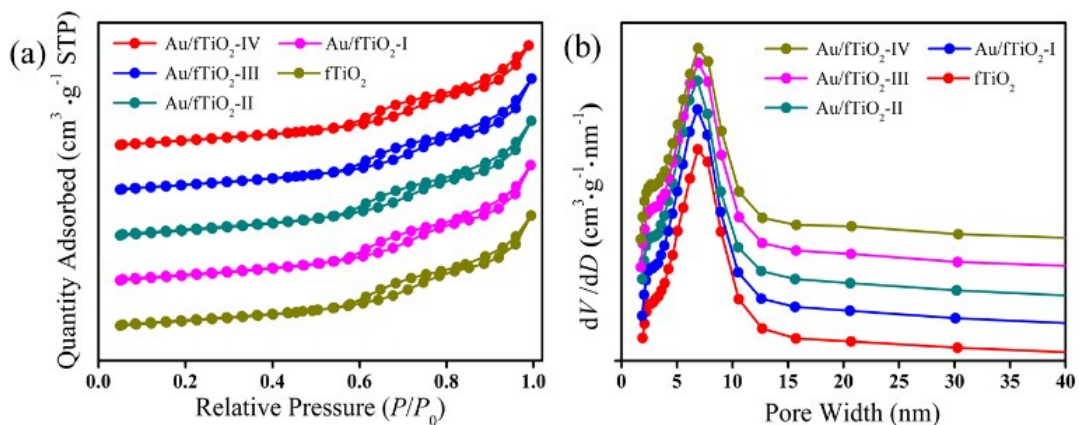


Fig. S1 (a) N_2 adsorption/desorption isothermal curves and (b) corresponding BJH pore size distribution curves of the flower-like titania samples without and with AuNPs loading.

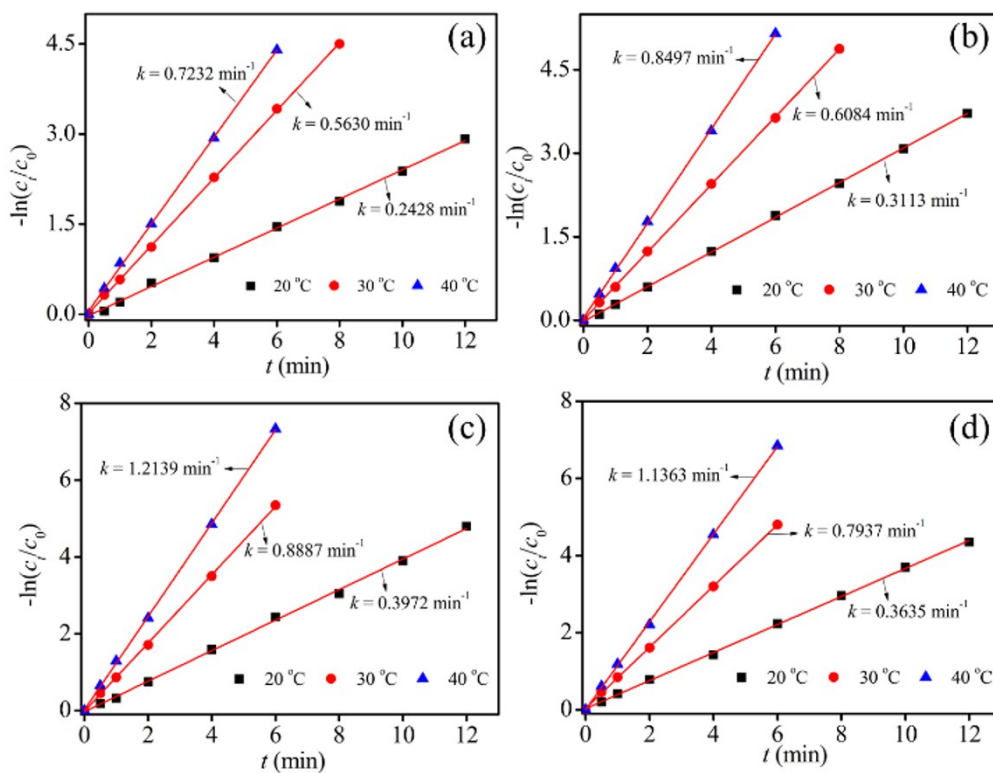


Fig. S2 Reduction kinetic curves of 4-NP catalyzed by (a) AuNPs/fTiO₂-I, (b) AuNPs/fTiO₂-II, (c) AuNPs/fTiO₂-III, and (d) AuNPs/fTiO₂-IV under different reaction temperatures.

Table S1 Comparison of the textural parameters of different Au-loaded TiO₂ nanocomposites.

Sample	Surface area (m ² ·g ⁻¹)	Pore volume (cm ³ ·g ⁻¹)	Ref.
Au-loaded TiO ₂ nanotubes	92.39	0.31	1
Au-loaded mesoporous TiO ₂ (1.0 mol% Au)	133	0.27	2
Au-loaded mesoporous TiO ₂ (2.0 mol% Au)	119	0.23	2
Au-loaded mesoporous TiO ₂ (5.0 mol% Au)	89	0.16	2
Au@TiO ₂	59	\	3
Au-loaded dendrimer derived TiO ₂	56.9	0.12	4
Au-loaded unsulphated TiO ₂	89	\	5
Au-loaded sulphated TiO ₂	132	\	5
AuNPs/fTiO ₂ -I~IV	142.3~149.3	0.39~0.43	This work

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