

Electronic Supporting information (ESI) for

Aggregation-induced chiroptical generation and photoinduced switching of achiral azobenzene-alt-fluorene copolymer endowed with left- and right-handed helical polysilanes

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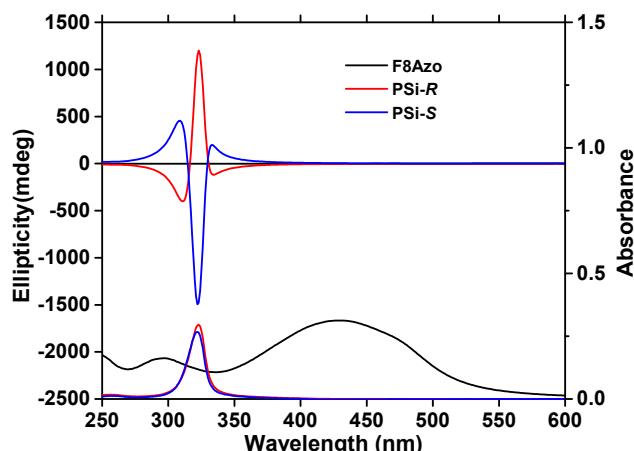


Fig. S1. CD and UV-vis spectra of PF8Azo and PSi-S/-R homo-aggregates in $\text{CHCl}_3\text{-MeOH}$ cosolvent (2.0/1.0, (v/v)). $[\text{PSi-S/-R}]_0 = [\text{PF8Azo}]_0 = 1.0 \times 10^{-5} \text{ M}$.

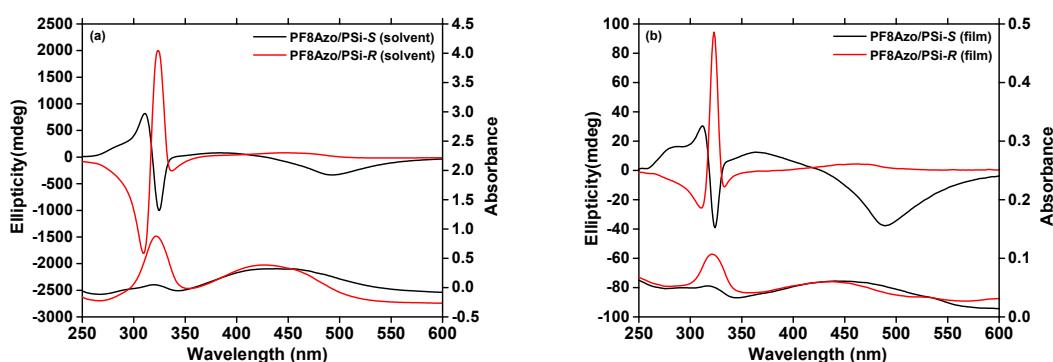


Fig. S2. (a) CD and UV-vis spectra of PF8Azo/PSi-S (black line) and PF8Azo/PSi-R (red line) hetero-aggregates in cosolvent and (b) CD and UV-vis spectra of the dried hetero-aggregate in grease onto quartz substrate. The aggregates are generated that $[\text{PSi-S/-R}]_0 = [\text{PF8Azo}]_0 = 2.0 \times 10^{-5} \text{ M}$ in chloroform-methanol (2.0/1.0, (v/v)) cosolvent.

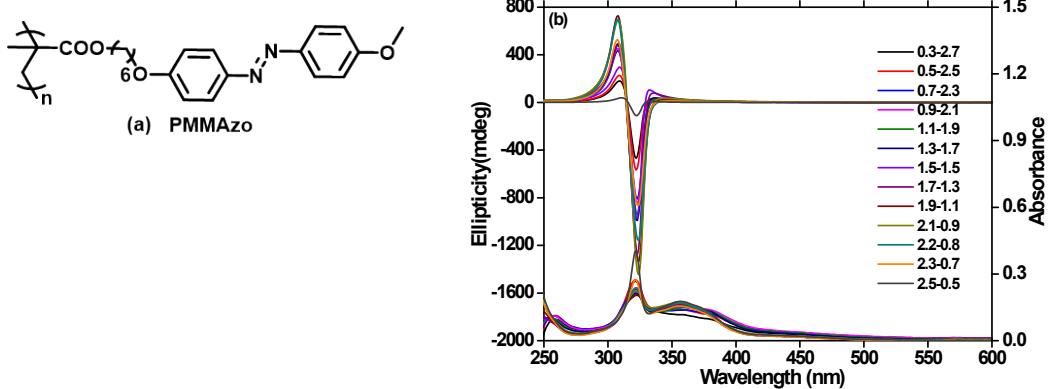


Fig. S3. (a)Chemical structure of PMMAzo ($M_n = 14,400$ Da, $M_w/M_n = 1.14$, $DP_n = 36$); (b)Changes in CD and UV-vis spectra of hetero-aggregates of PMMAzo-PSi-S that with different volume ratio of CHCl₃-MeOH. ([PSi-S]₀ = [PMMAzo]₀ = 1.0×10^{-5} M).

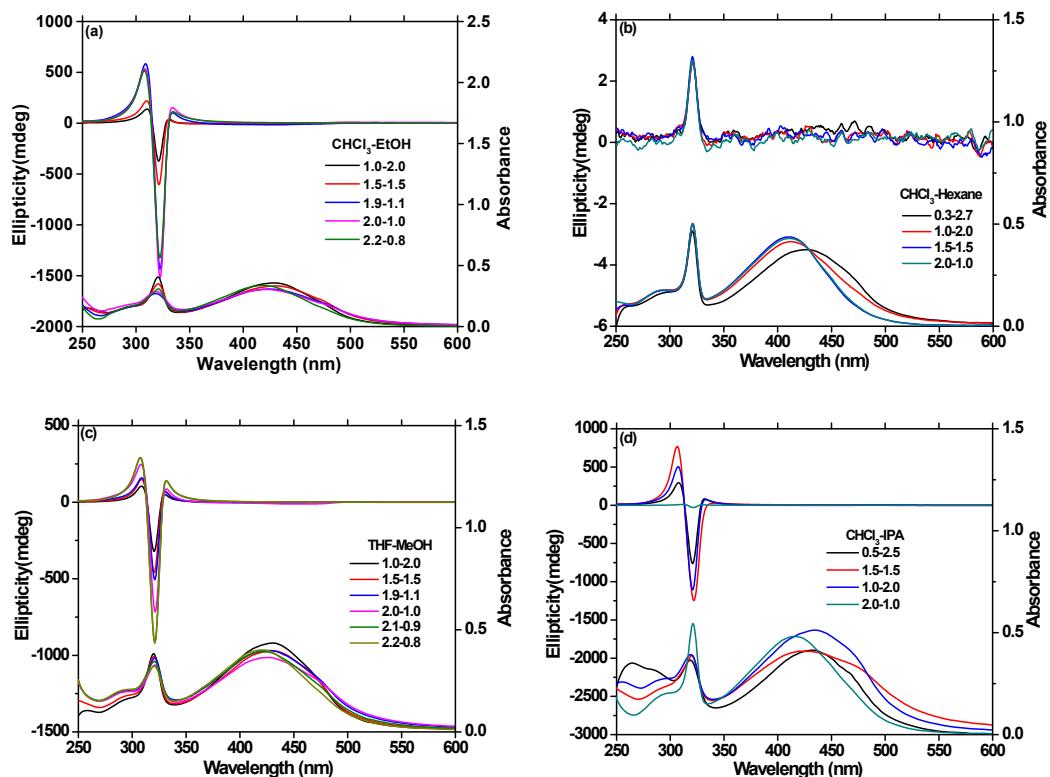


Fig. S4. CD and UV-vis spectra of hetero-aggregates including PSi-S and PF8Azo with a 1-to-1 molar ratio produced in different cosolvents with different volume ratio. [(a)chloroform-ethanol; (b)chloroform-hexane; (c)tetrahydrofuran-methanol; (d)chloroform-isopropanol.] Their repeating units are [PSi-S-R]₀ = [PF8Azo]₀ = 1.0×10^{-5} M.

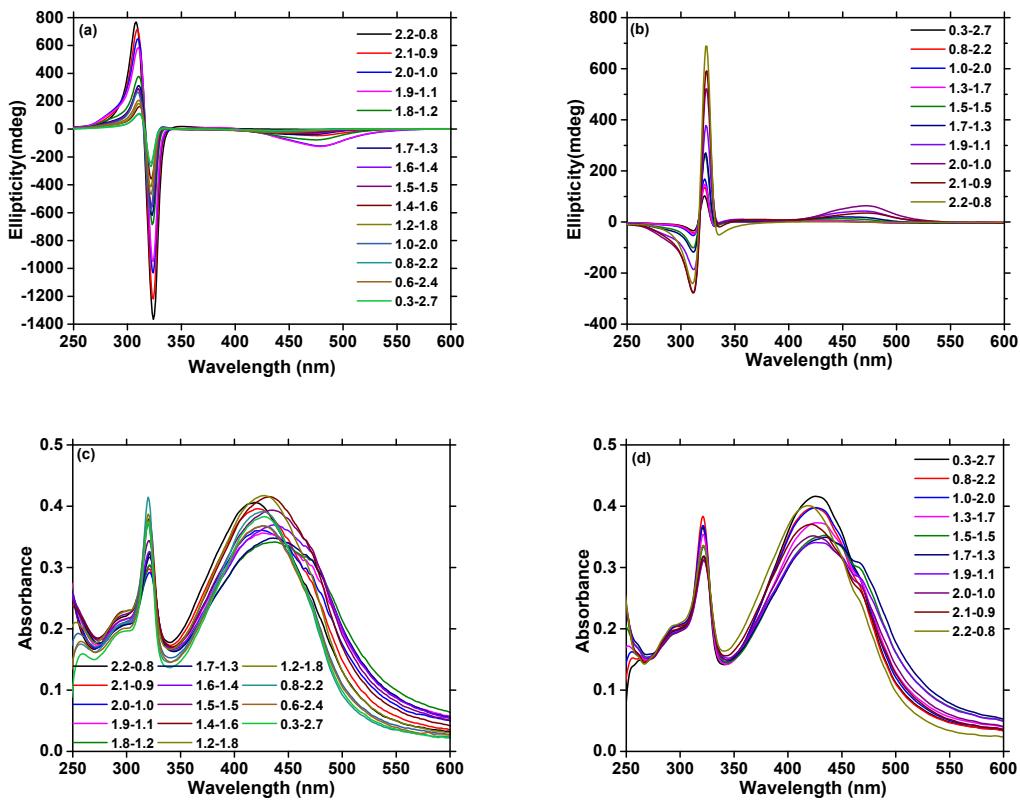


Fig. S5. Changes in CD spectra of hetero-aggregates of (a)PF8Azo/PSi-S and (b) PF8Azo/PSi-R that with different volume ratio of CHCl₃-MeOH. UV-vis spectra (c)and (d) show corresponding UV changes respectively. ([PSi-S/-R]₀ = [PF8Azo]₀ = 1.0×10⁻⁵ M).

Table S1. Changes in particle size of PSi-S homo-aggregates with different volume ratio of CHCl₃-MeOH. ([PSi-S]₀ = 1.0×10⁻⁵ M).

CHCl ₃ -MeOH	Eff.Diam.(nm)	PDI
0.3-2.7	678.3	0.21
0.5-2.5	617.8	0.18
0.8-2.2	562.7	0.18
1.0-2.0	535.3	0.20
1.3-1.7	509.2	0.13
1.5-1.5	492.0	0.19
1.7-1.3	473.4	0.10
1.9-1.1	474.3	0.16
2.0-1.0	484.2	0.12
2.1-0.9	386.4	0.14
2.2-0.8	319.6	0.09
2.3-0.7	310.8	0.09
2.4-0.6	184.7	0.13

Table S2. Changes in particle size of PF8Azo homo-aggregates with different volume ratio of CHCl₃-MeOH. ([PF8Azo]₀= 1.0×10⁻⁵ M).

CHCl ₃ -MeOH	Eff. Diam.(nm)	PDI
0.3-2.7	1776.8	0.39
0.5-2.5	969.8	0.32
0.8-2.2	794.7	0.26
1.0-2.0	685.3	0.24
1.3-1.7	629.4	0.19
1.5-1.5	471.7	0.16
1.7-1.3	314.4	0.07
1.9-1.1	284.7	0.12
2.0-1.0	263.8	0.08
2.1-0.9	250.7	0.08
2.2-0.8	318.7	0.14
2.3-0.7	230.6	0.10
2.4-0.6	225.2	0.12

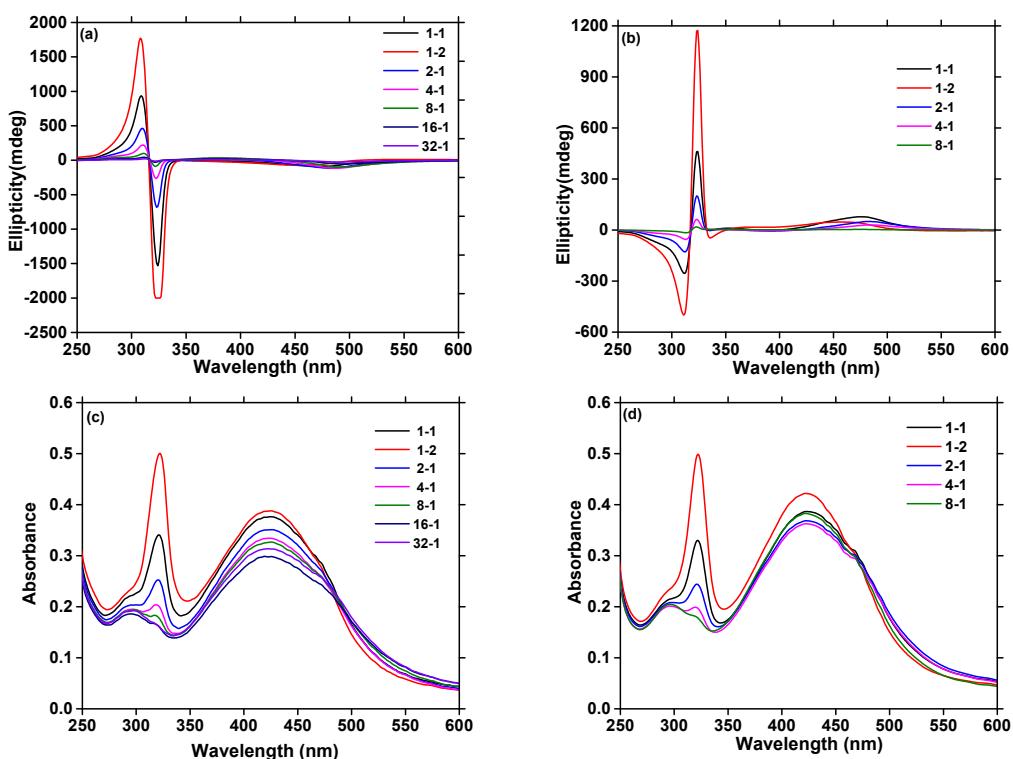


Fig. S6. Changes in CD spectra of hetero-aggregates of (a)PF8Azo/PSi-S and (b)PF8Azo/PSi-R that with different molar ratio of PF8Azo-to-PSi-S/-R ([PF8Azo]₀= 1.0×10⁻⁵ M). UV-vis spectra (c)and (d) show corresponding UV changes respectively. The volume ratio of CHCl₃-MeOH = 2-1.

Table S3. Changes in g_{CD} values at 320 nm and 475 nm of the hetero-aggregates before and after the irradiation of 313-nm light. The hetero-aggregates were prepared under the condition that the volume ratio (CHCl_3 -MeOH) was 2-to-1 and the molar ratio was 1-to-1 ($[\text{PSi-S/-R}]_0 = [\text{PF8Azo}]_0 = 1.0 \times 10^{-5} \text{ M}$).

Hetero-aggregate	$g_{CD}/10^{-3}$	Before irradiation	313 nm 5 s	313 nm 10 s	313 nm 15 s
PF8Azo/PSi-S	at 320 nm	-83.27	5.348	6.878	7.253
	at 475 nm	-12.34	-21.73	-21.14	-20.40
PF8Azo/PSi-R	at 320 nm	23.61	9.706	4.613	0.6477
	at 475 nm	6.492	9.184	8.625	8.071

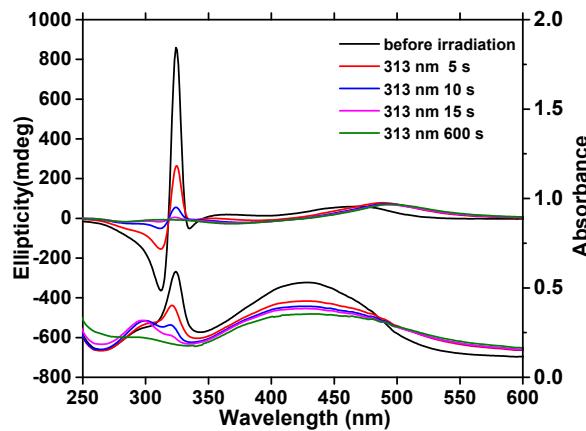


Fig. S7. Changes in CD and UV-vis spectra of PF8Azo/PSi-R hetero-aggregates before and after the irradiation of 313-nm light. The hetero-aggregates were prepared under the condition that the volume ratio (CHCl_3 -MeOH) was 2-to-1 and the molar ratio was 1-to-1 ($[\text{PSi-R}]_0 = [\text{PF8Azo}]_0 = 1.0 \times 10^{-5} \text{ M}$).

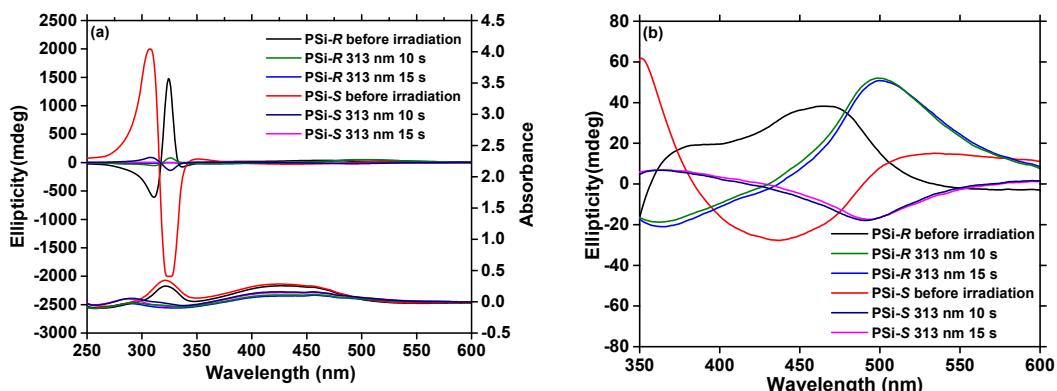


Fig. S8. (a)Changes in CD and UV-vis spectra of the PF8Azo/PSi-R and PF8Azo/PSi-S hetero-aggregates before and after the irradiation of 313-nm light. (b) is magnified part of (a).The hetero-aggregates were prepared under the condition that the volume ratio (CHCl_3 -MeOH) was 2.1-to-0.9 and the molar ratio was 1-to-1 ($[\text{PSi-S/-R}]_0 = [\text{PF8Azo}]_0 = 1.0 \times 10^{-5} \text{ M}$).

Table S4. Effective particle size of PSi-S homo-aggregates variation with time. ($[PSi-S]_0 = 1.0 \times 10^{-5}$ M, CHCl₃-MeOH = 1-2)

Time(min)	Eff. Diam.(nm)	PDI
1	421.1	0.16
2	467.3	0.14
3	524.6	0.14
4	535.3	0.20
5	597.6	0.19

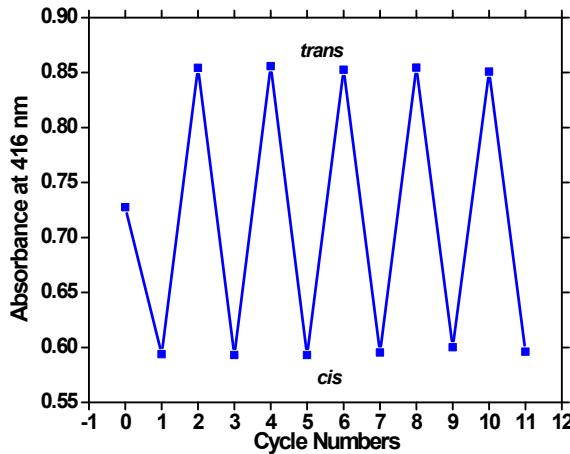


Fig. S9. Reversible photo-isomerization of non-aggregate PF8Azo dissolved in CHCl₃ upon alternating irradiation with 405-nm and 546-nm light. Irradiation time:15 min, [PF8Azo]₀ = 1.0×10⁻⁵ M.

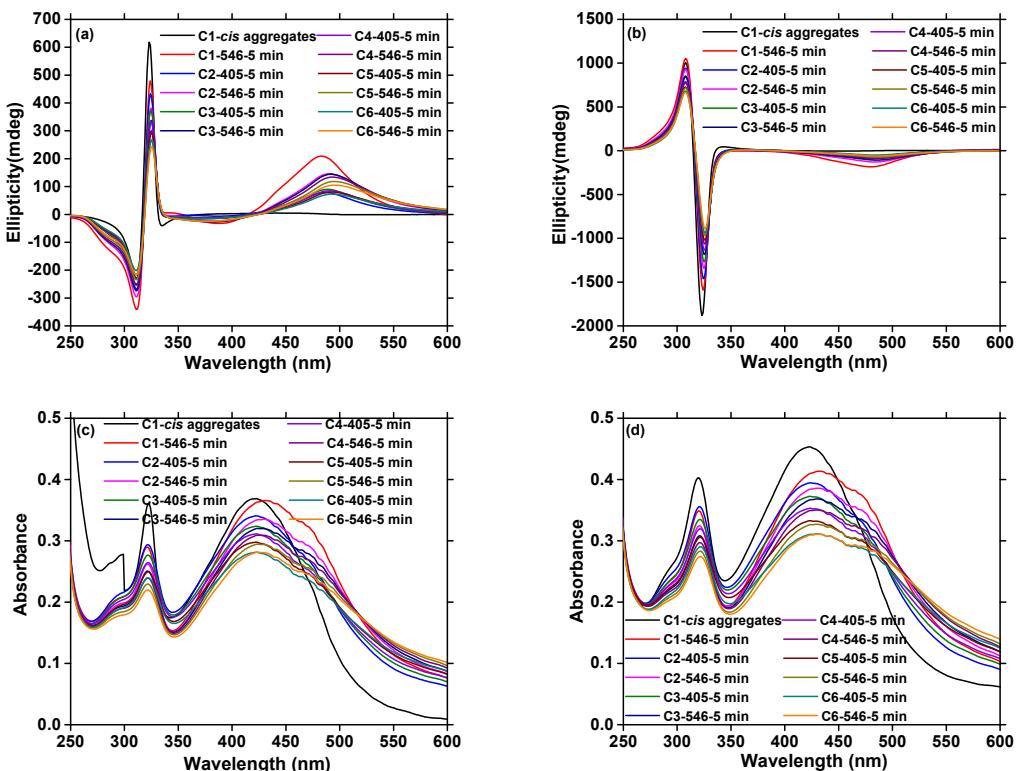


Fig. S10. Changes in CD spectra of hetero-aggregates of (a)PF8Azo/PSi-R and (b) PF8Azo/PSi-S upon alternating photoirradiation cycles with 405-nm and 546-nm light in CHCl₃-MeOH cosolvent (2.0/1.0, v/v). UV-vis spectra (c)and (d) show corresponding UV changes respectively. ($[PSi-S/-R]_0 = [PF8Azo]_0 = 1.0 \times 10^{-5}$ M). A CHCl₃ solution of PF8Azo was irradiated for 15 min with 405-nm light, then aggregate with PSi-S/-R in CHCl₃-MeOH cosolvent to obtain the *cis* aggregates.

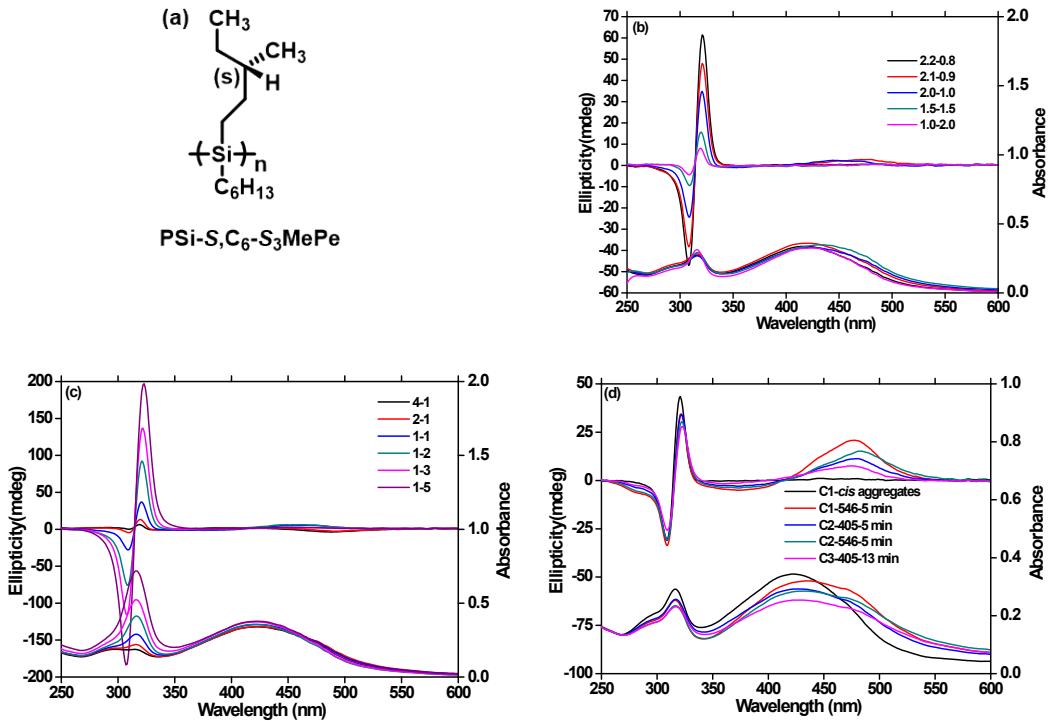


Fig. S11. (a) Chemical structure of $\text{C}_6\text{-S}_3\text{MePe}$. Changes in CD and UV-vis spectra of hetero-aggregates of (b) PF8Azo/ $\text{C}_6\text{-S}_3\text{MePe}$ that with different volume ratio of $\text{CHCl}_3\text{-MeOH}$ ($[\text{C}_6\text{-S}_3\text{MePe}]_0 = [\text{PF8Azo}]_0 = 1.0 \times 10^{-5} \text{ M}$), (c) PF8Azo/ $\text{C}_6\text{-S}_3\text{MePe}$ with different molar ratio of PF8Azo-to- $\text{C}_6\text{-S}_3\text{MePe}$ ($[\text{PF8Azo}]_0 = 1.0 \times 10^{-5} \text{ M}$). (d) Changes in CD and UV-vis spectra of reversible photoisomerization of PF8Azo/ $\text{C}_6\text{-S}_3\text{MePe}$ hetero-aggregates upon alternating 405-nm and 546-nm light. The M_n values of $\text{C}_6\text{-S}_3\text{MePe}$ and PF8Azo are 32,100 Da and 23,900 Da respectively.

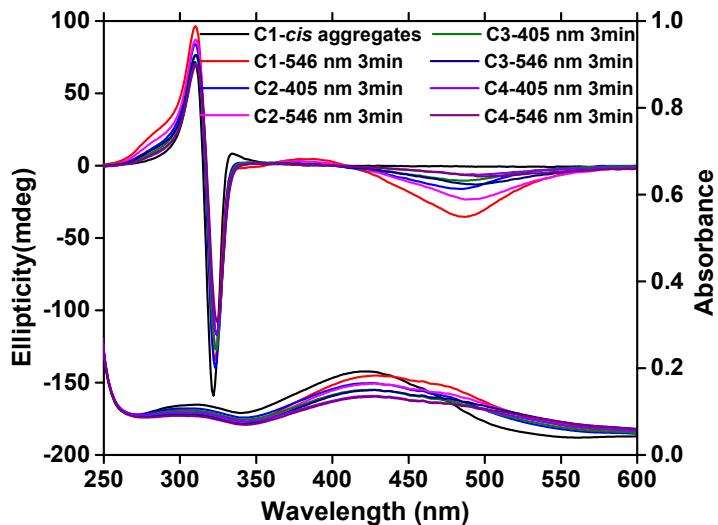


Fig. S12. Changes in CD and UV-vis spectra of PF8Azo-PSi-S hetero-aggregates upon alternating photoirradiation cycles with 405-nm and 546-nm light in $\text{CHCl}_3\text{-MeOH}$ cosolvent (2.0/1.0, (v/v)). The molar ratio of PF8Azo-to-PSi-S is 4-to-1. ($[\text{PF8Azo}]_0 = 0.5 \times 10^{-5} \text{ M}$). A CHCl_3 solution of PF8Azo was irradiated for 15 min with 405-nm light, then aggregate with PSi-S in $\text{CHCl}_3\text{-MeOH}$ cosolvent to obtain the *cis* aggregates.

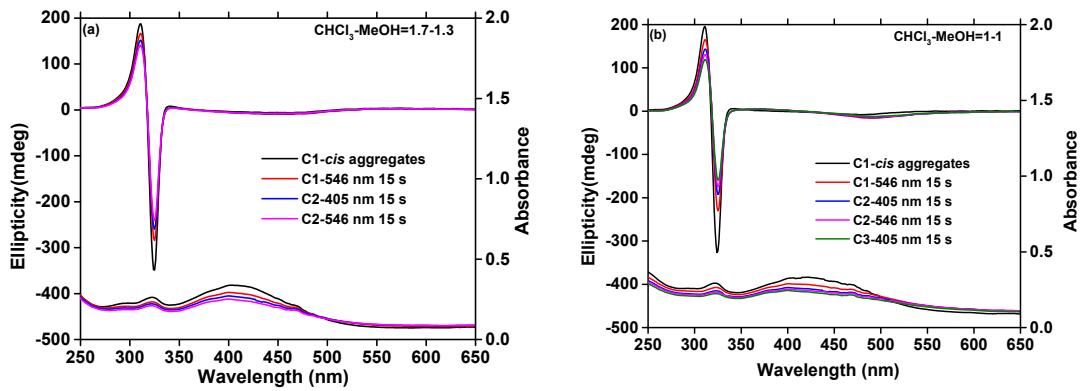
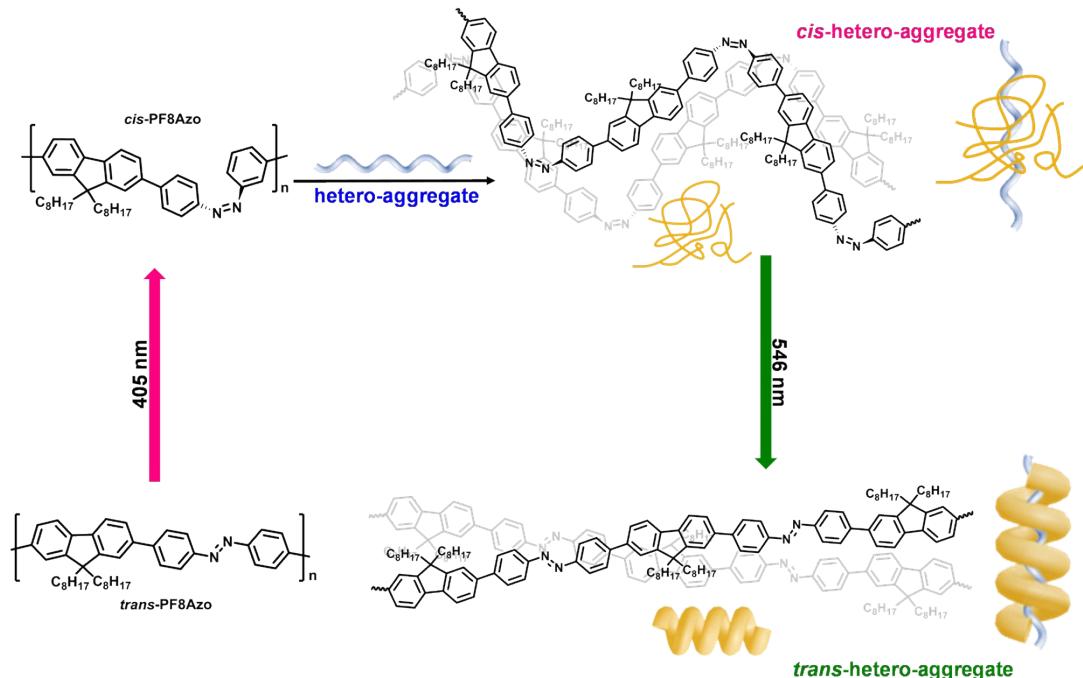
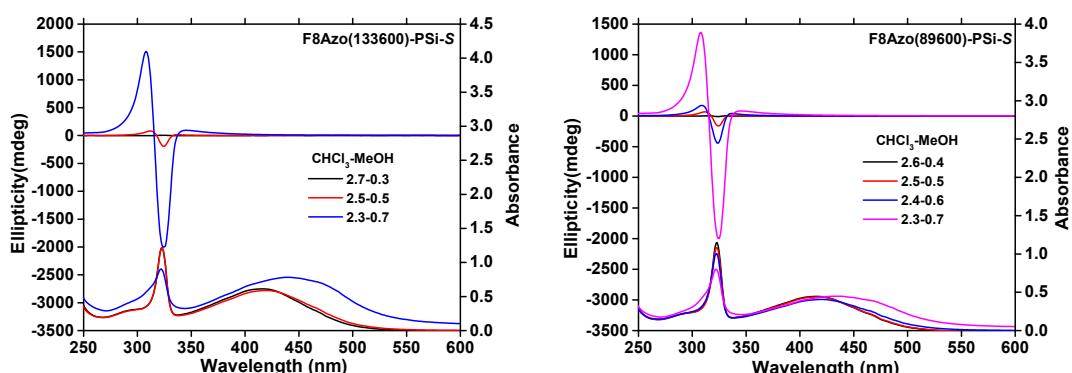
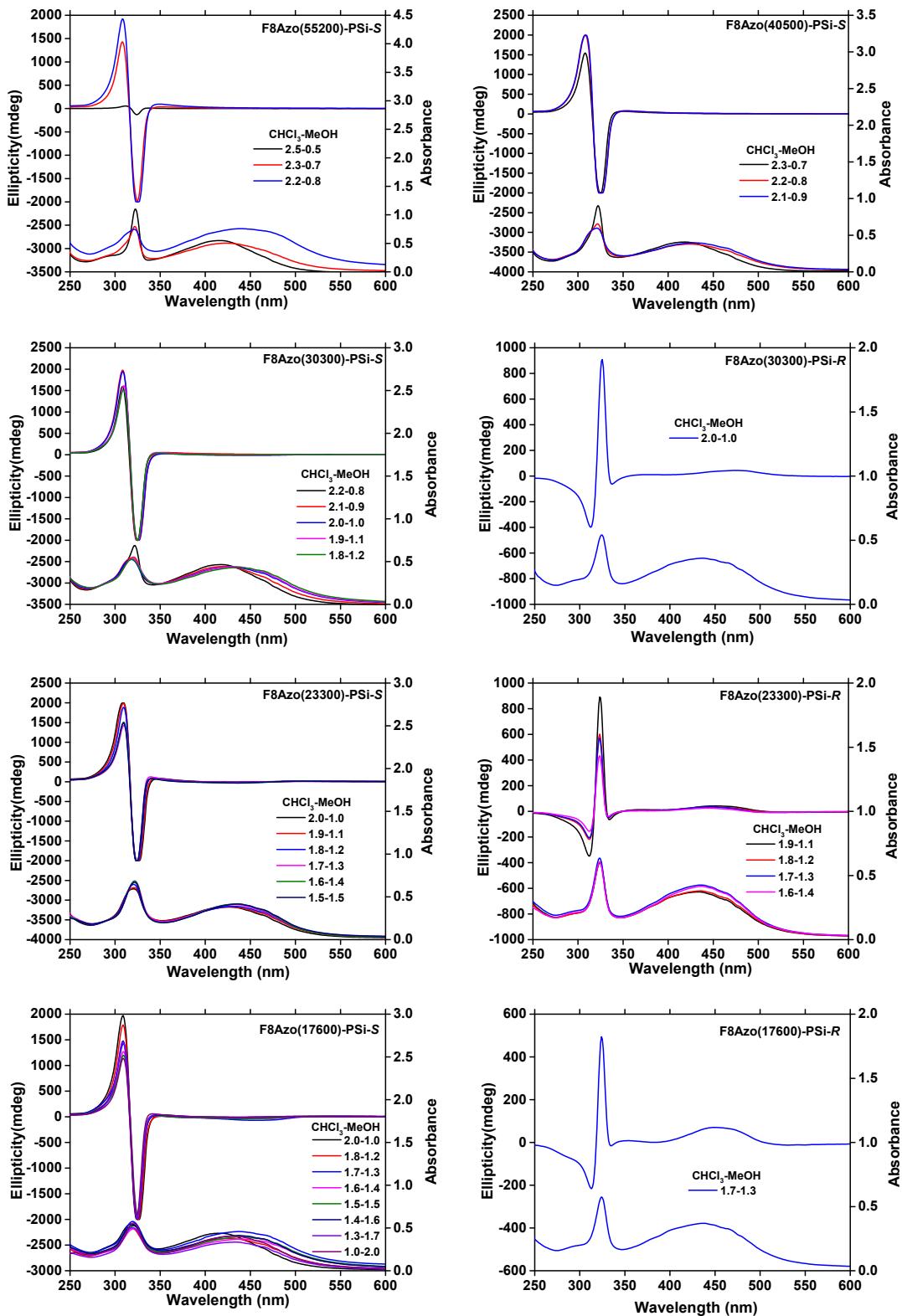


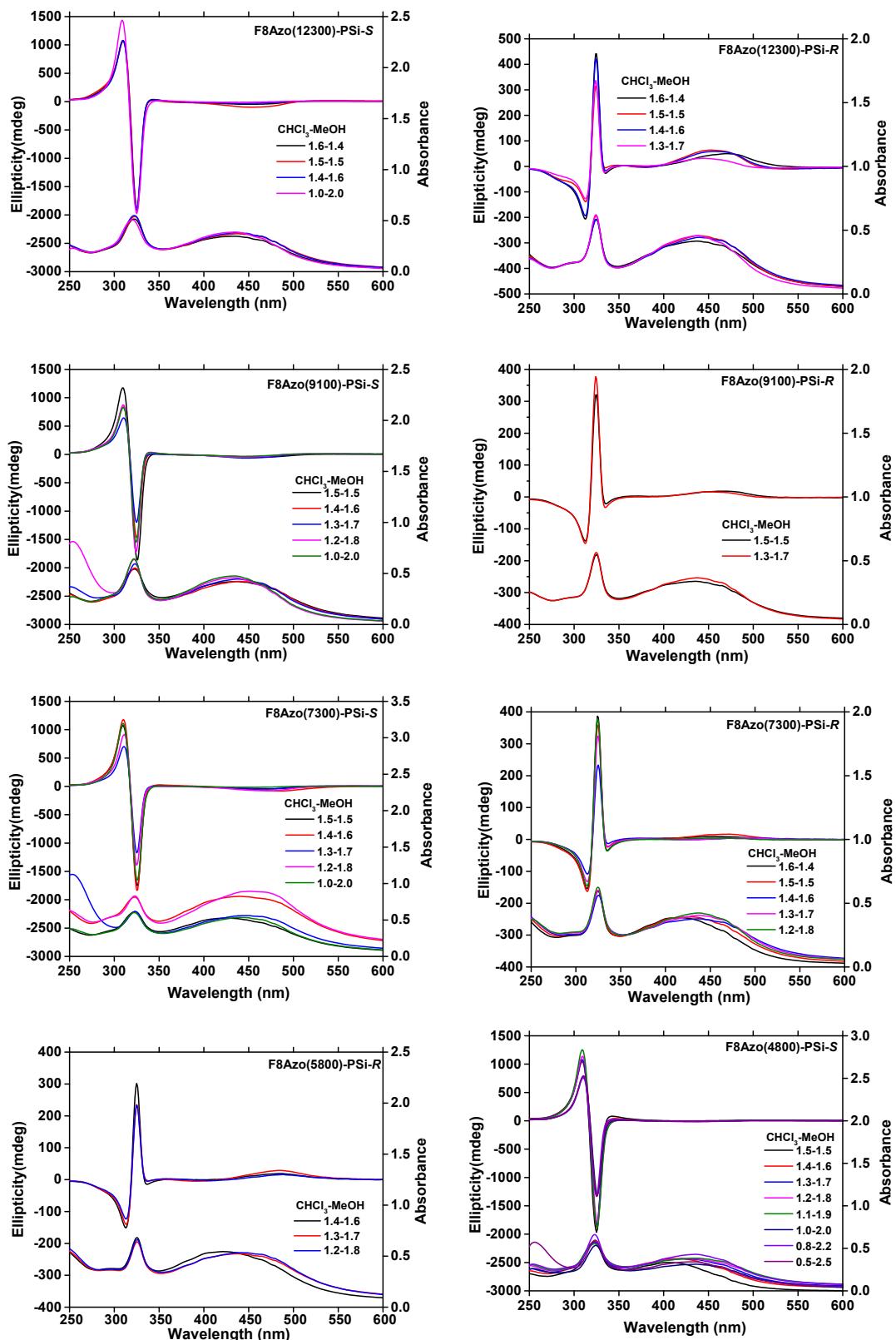
Fig. S13. Changes in CD and UV-vis spectra of PF8Azo/PSi-S hetero-aggregates upon alternating photoirradiation cycles with 405-nm and 546-nm light in CHCl₃-MeOH cosolvent [(a)1.7/1.3, (v/v); (b)1/1, (v/v)]. The molar ratio of PF8Azo-to-PSi-S is 1-to-1. ([PF8Azo]₀ = 1.0 × 10⁻⁵ M, M_n(PF8Azo) = 7,300 Da). A CHCl₃ solution of PF8Azo was irradiated for 15 min with 405-nm light, then aggregate with PSi-S in CHCl₃-MeOH cosolvent to obtain the *cis* aggregates.



Scheme S1. Proposed mechanism for the enhancement of g_{CD} upon 1st photoirradiation at 546 nm.







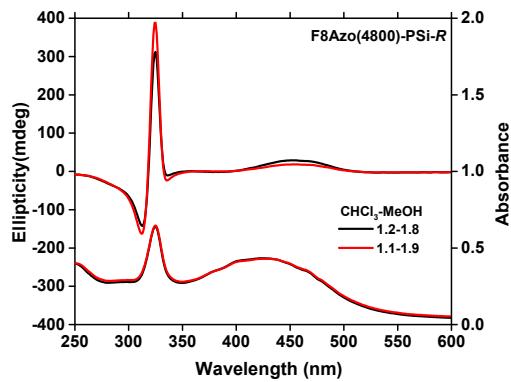


Fig. S14. Changes in CD and UV-vis spectra of PSi-S/-R hetero-aggregates with PF8Azo which has different $M_{n,\text{GPC}}$. The hetero-aggregates were prepared under the condition that the molar ratio was 1-to-1($[\text{PSi-S/-R}]_0 = [\text{PF8Azo}]_0 = 1.0 \times 10^{-5}$ M) in $\text{CHCl}_3\text{-MeOH}$ cosolvent that with different volume ratio.

Table S5. Molecular weight characteristics and number-average degree of polymerization of PF8Azo that applied for exploring molecular weight effects.

Fraction	$M_{n,\text{GPC}}$ (g/mol)	M_w/M_n	DP_n
1	133,600	1.53	234
2	89,600	1.63	157
3	55,200	1.69	97
4	40,500	1.27	71
5	30,300	1.68	53
6	23,300	1.18	41
7	17,600	1.42	31
8	12,300	1.17	22
9	9,100	1.06	16
10	7,300	1.03	13
11	5,800	1.05	10
12	4,800	1.02	8