

Supplementary Information

Non-photochemical catalytic hydrolysis of methyl parathion using core-shell Ag@TiO₂ nanoparticles

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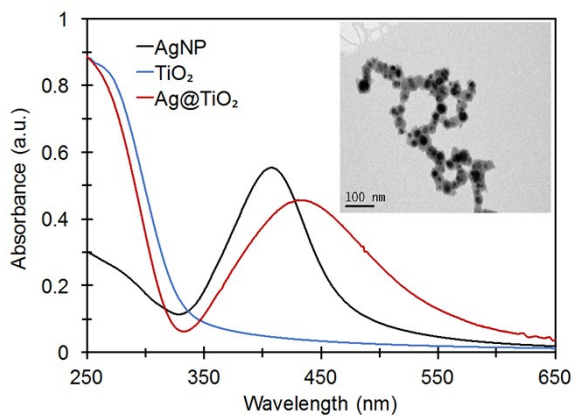


Figure S1. UV-Vis absorption spectra of Ag, TiO₂ and Ag@TiO₂ nanoparticles and TEM image of Ag@TiO₂ (inset).

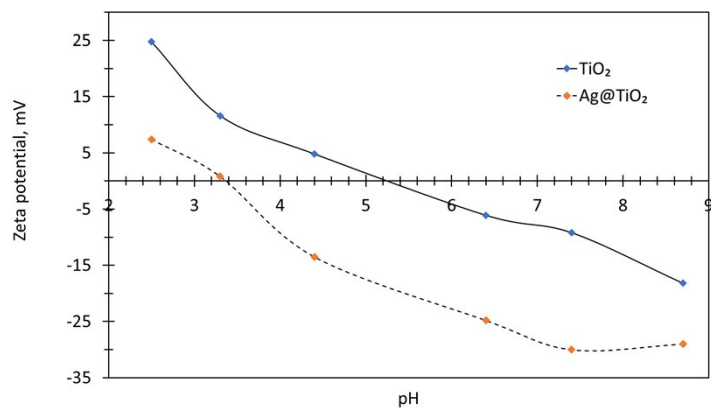


Figure S2. Zeta potential-pH relationship for TiO₂ and Ag@TiO₂.

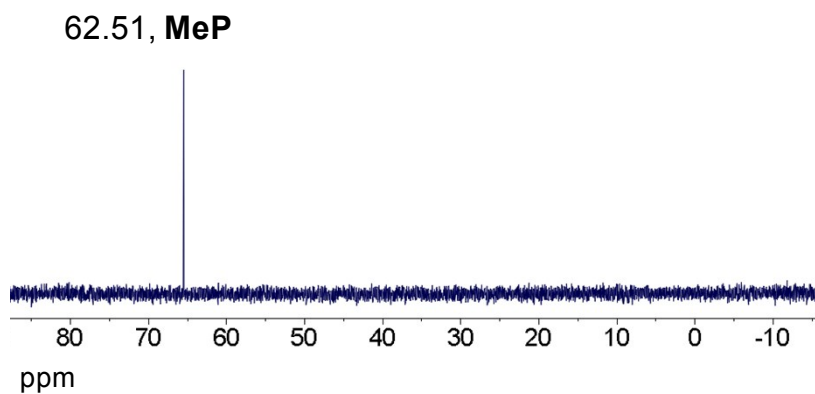


Figure S3. ³¹P NMR spectrum of methyl parathion after 24 hours (5% MeOH/MOPS, pH = 8.0) (in the absence of Ag@TiO₂), δ : 65.51 ppm (MeP). Assignments based on literature values.¹⁻²

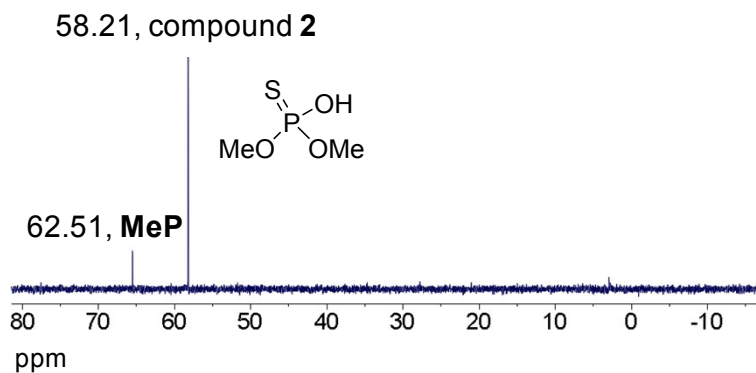


Figure S4. ^{31}P NMR spectrum of methyl parathion + Ag@TiO_2 after 24 hours (5% MeOH/MOPS, pH = 8.0). δ : 65.51 ppm (MeP), 58.21 (compound 2, Figure S1). Assignments based on literature values.¹⁻²

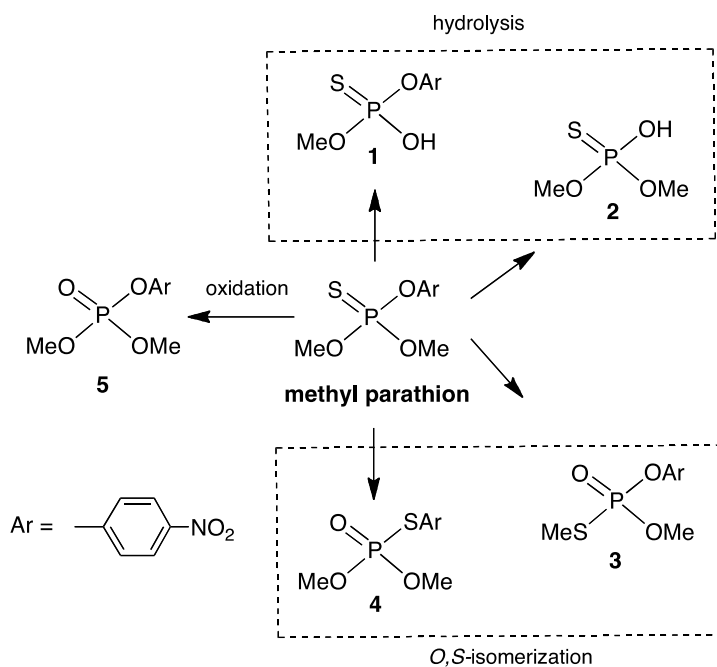


Figure S5. Non-photochemical breakdown products of methyl parathion.

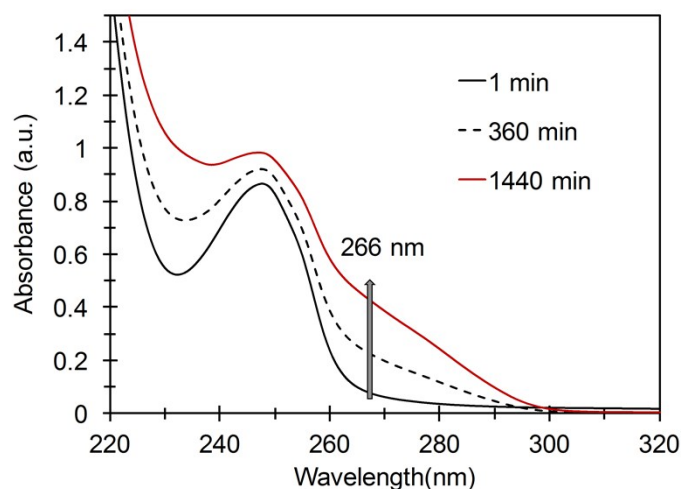
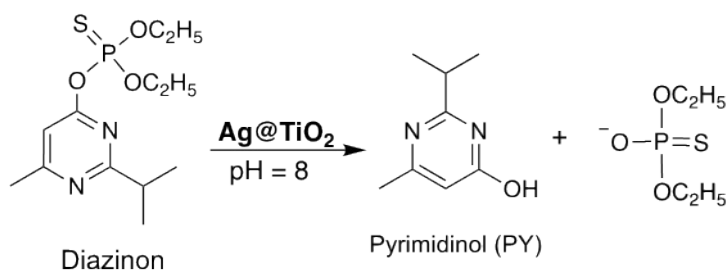


Figure S6. UV-Vis for diazinon + Ag@TiO₂ (5% MeOH/MOPS, pH = 8.0) at different interval times and after removal of the catalyst. The spectra of the catalytic mixtures after removal of Ag@TiO₂ showed a constant increase in absorbance at 266 nm indicating an increase in the concentration of pyrimidinol (PY).

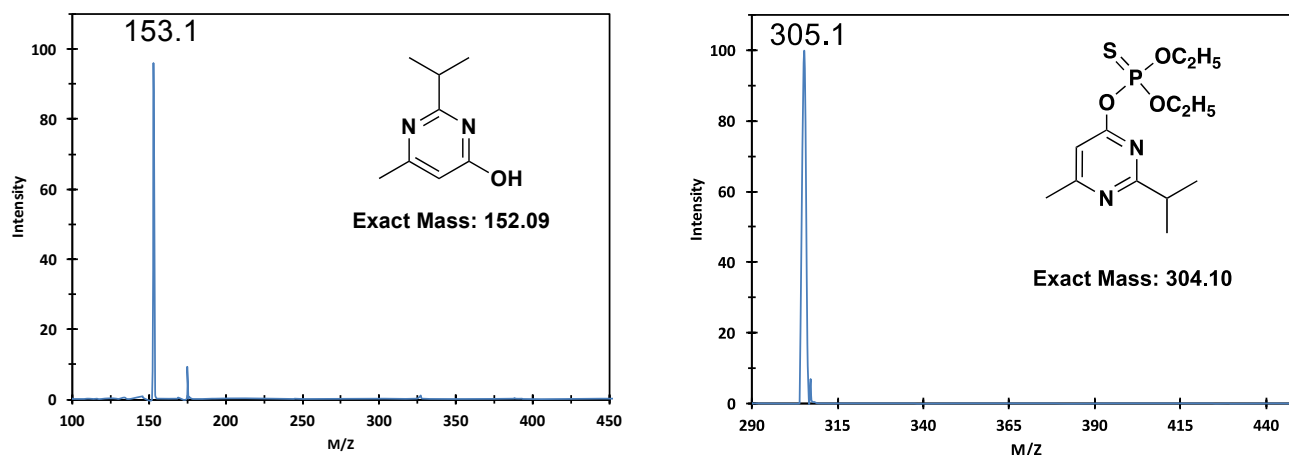


Figure S7. LC-MS analysis of diazinon + Ag@TiO₂ after 24 hours (5% MeOH/MOPS, pH = 8.0). Spectrum from APCI-MS in positive mode, (304.10 *m/z*; Calcd for [M+H]⁺; 305.1 *m/z*) and product PY (152.09 *m/z*; Calcd for [M+H]⁺; 153.1 *m/z*).³

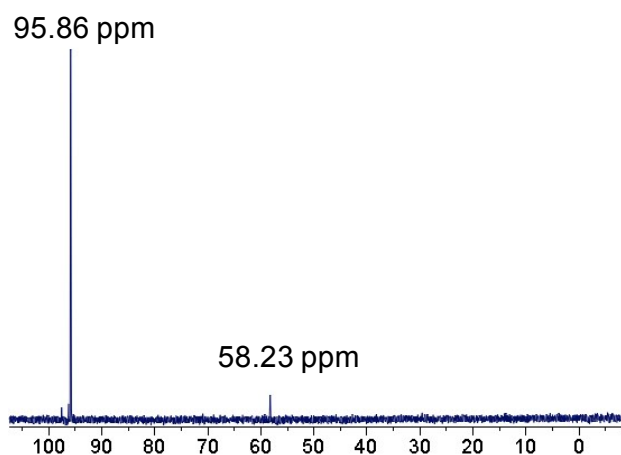


Figure S8. ³¹P NMR spectrum of malathion + Ag@TiO₂ after 24 hours (5% MeOH/MOPS, pH = 8.0). δ: 95.86 (malathion), 58.23 ([P(S)(OMe)₂(OH)]). Assignments based on literature values.⁴

References

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2. Seger, M. R.; Maciel, G. E., *J. Environ. Chem. Eng.* **2006**, 40 (2), 552-558.
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