

Enhanced gas selectivity induced by surface active oxygen in SnO/SnO₂ Hierarchical Structure at Different Temperature

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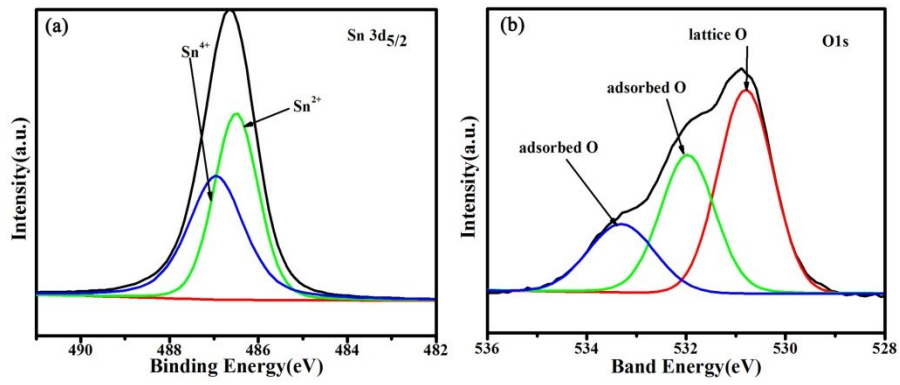


Fig. S1 High-resolution XPS spectrum of Sn 3d (a) and O1s (b) of the SnO/SnO₂ composite

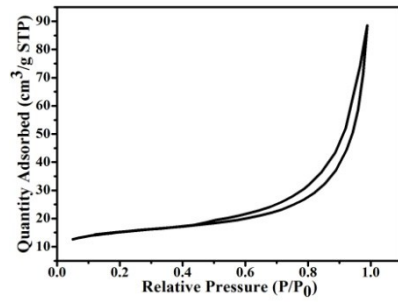


Fig. S2. The N₂ adsorption-desorption isotherms of the SnO/SnO₂ composite.

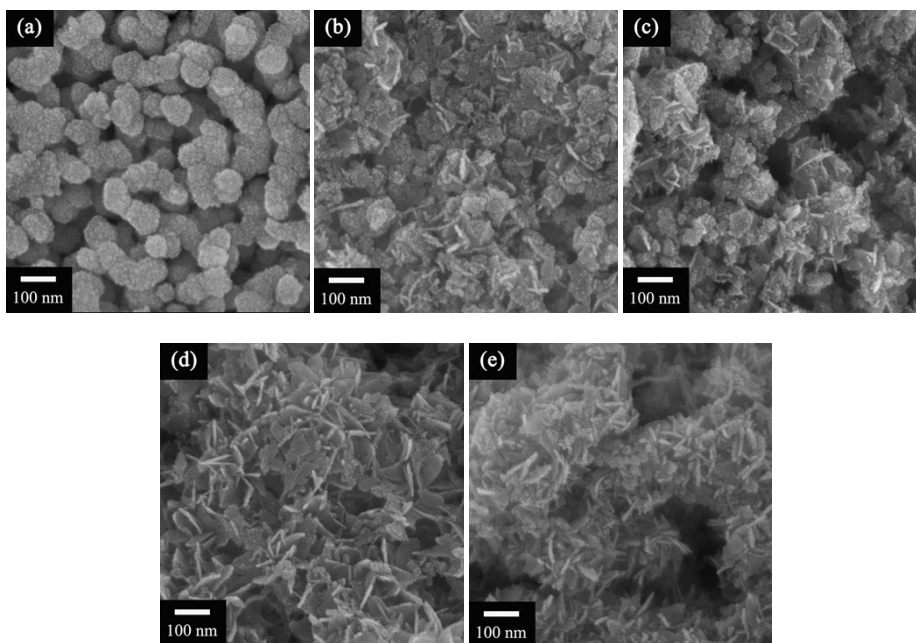


Fig. S3 SEM images with the mole ratio of HMTA : SnC₂O₄ (a) 0:1, (b) 1:4, (c) 1:2, (d) 2:1, (e) 4:1.

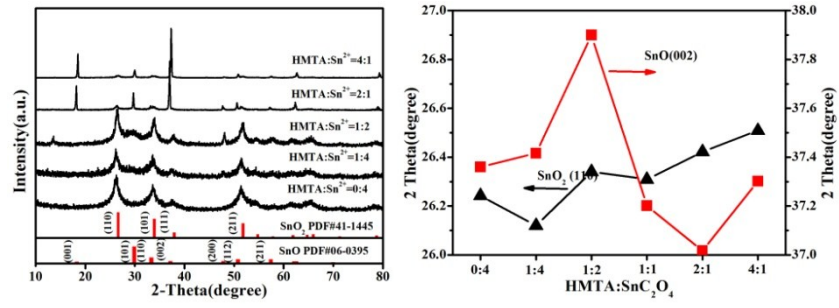


Fig. S4 XRD patterns and (b) peak position of SnO₂ (110) and SnO (002) of samples with different mole ratio of HMTA/SnC₂O₄.

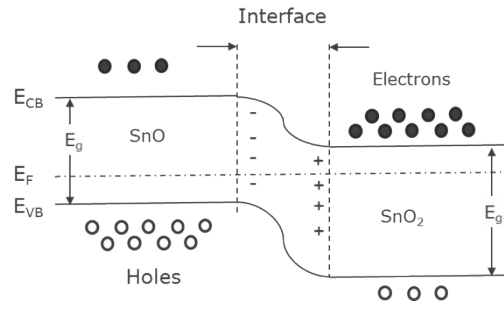


Fig. S5 Energy band diagrams of the SnO/SnO₂ heterocontact.

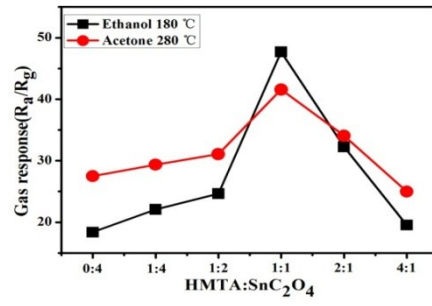


Fig. S6. Gas response of SnO/SnO₂ composite with different HMTA and SnC₂O₄ ration for 100 ppm ethanol and acetone.

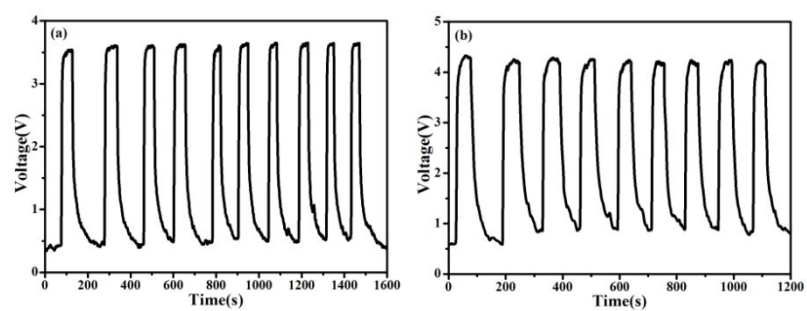


Fig. S7 Repeatability for (a) ethanol and (b) acetone at the optimum temperatures.