## A Synchrotron X-ray Spectroscopy Study of Titanium Co-ordination in Explosive Melt Glass Derived From the Trinity Nuclear Test: Supplementary Materials

	Desert Sand	Trinitite 1	Trinitite 2
Si	240000	260000	260000
Al	39000	51000	42000
Ca	39000	40000	46000
К	39000	24400	26100
Fe	3900	16800	17600
Na	14000	12500	11900
Mg	950	5500	5300
Ti	181	2310	2370
Ва	605	535	540
Zr	530	480	825
Р	253	514	442
Mn	87.9	356	398
Sr	180	215	235
Pb	135	15	45
Ce	205	86	65
S	53	38	35
V	6.4	43	41
Cr	5.2	35	36
Cu	4.1	29	24
Li	7.4	15.2	16.5
Zn	14.4	19.5	16.1
Ni	3	18.9	18.8
La	16.8	26.1	25.7
Y	2.3	13.9	5.9
Nb	5.2	24.4	15.2
Со	<2	6.1	6.2
Be	5.6	1.4	1.5
Hf	16.3	14.9	26.2
U	1.65	2.53	2.98

**ST1:** Raw ICP-MS data for desert sand and trinitite samples (mg.kg<sup>-1</sup>).

	Desert Sand	Trinitite 1	Trinitite 2
SiO <sub>2</sub>	75.42311	74.14623	74.09303
$AI_2O_3$	6.37887	7.56958	6.2293
CaO	8.58898	7.99392	9.18641
K <sub>2</sub> O	5.71847	3.24659	3.4703
FeO	0.6164	2.40952	2.52245
Na <sub>2</sub> O	2.68746	2.17744	2.07144
MgO	0.34499	1.81248	1.74532
TiO <sub>2</sub>	0.03338	0.38653	0.39628
BaO	0.03889	0.0312	0.03147
ZrO <sub>2</sub>	0.05128	0.04214	0.07238
$P_2O_5$	0.03605	0.06646	0.05711
MnO	0.01412	0.0519	0.05798
SrO	0.01813	0.01965	0.02147
PbO	0.00575	0.00058	0.00174
CeO <sub>2</sub>	0.01291	0.00492	0.00371
SO <sub>4</sub>	0.01459	0.00949	0.00874
$V_2O_5$	0.00055	0.00338	0.00322
$Cr_2O_3$	0.00088	0.00539	0.00554
CuO	0.00057	0.00366	0.00302
Li <sub>2</sub> O	0.00471	0.00877	0.00951
ZnO	0.00194	0.00239	0.00197
NiO	0.00045	0.00258	0.00256
$La_2O_3$	0.00053	0.00075	0.00074
$Y_2O_3$	0.00011	0.00063	0.00027
Nb <sub>2</sub> O <sub>5</sub>	0.00025	0.00105	0.00065
CoO	0.0003	0.00083	0.00084
BeO	0.00548	0.00124	0.00133
HfO <sub>2</sub>	0.00081	0.00067	0.00117
$U_3O_8$	0.00002	0.00003	0.00003
Total	99.99998	100.00000	99.99998

ST2: Composition of desert sand and trinitite samples (mol% by oxide, 5 d.p.)



**SF1:** Backscattered electron micrograph of a trinitite specimen. Numbered points indicate the location of individual spectra shown in SF2.



**SF2:** Normalised EDX spectra from point scans at locations indicated in SF1. Inset- Magnified Ti region.



**SF3:** Possible four-, five- and six-fold coordination environments: a) Tetrahedral; b) Trigonal bi-pyramidal; c) Square pyramidal; d) Octahedral; e) Trigonal prismatic.

Standard	Ti Co-	Co-ordination	Pre-edge	Pre-edge
	ordination	environment	centroid	centroid
			position (eV)	height
Ni <sub>2.6</sub> Ti <sub>0.7</sub> O <sub>4</sub>	4	Tetrahedral	4969.7	0.94
$Ni_{2.4}Ti_{0.7}Si_{0.05}O_4$	4	Tetrahedral	4969.6	0.90
β- Ba <sub>2</sub> TiO <sub>4</sub>	4	Tetrahedral	4969.4	1.00
α- Ba <sub>2</sub> TiO <sub>4</sub>	4	Tetrahedral	4969.5	1.00
CsAlTiO <sub>4</sub>	4	Tetrahedral	4969.7	0.90
Rb <sub>2</sub> TiO <sub>3</sub>	4	Tetrahedral	4969.6	1.00
$K_6Ti_2O_7$	4	Tetrahedral	4969.7	0.93
Y <sub>2</sub> TiMoO <sub>8</sub>	4	Tetrahedral	4969.9	0.74
Na <sub>4</sub> TiO <sub>4</sub>	4	Tetrahedral	4969.6	0.70
Na <sub>2</sub> TiOSiO <sub>4</sub>	5	Square pyramidal	4970.5	0.65
KNaTiO₃	5	Square pyramidal	4970.6	0.73
Ba <sub>2</sub> TiOSi <sub>2</sub> O <sub>7</sub>	5	Square pyramidal	4970.6	0.67
Sr <sub>2</sub> TiOSi <sub>2</sub> O <sub>7</sub>	5	Square pyramidal	4970.5	0.71
Ba <sub>2</sub> TiOGe <sub>2</sub> O <sub>7</sub>	5	Square pyramidal	4970.6	0.64
$Na_2Ti_4O_9$	5/6	Square	4970.5	0.43
		pyramidal/Octahedral		
K <sub>2</sub> Ti <sub>4</sub> O <sub>9</sub>	5/6	Square	4970.6	0.45
		pyramidal/Octahedral		
Rb <sub>2</sub> Ti <sub>4</sub> O <sub>9</sub>	5/6	Square	4970.5	0.47
		pyramidal/Octahedral		
$K_2Ti_2O_5$	5	Square pyramidal	4970.6	0.51
r-TiO <sub>2</sub>	6	Octahedral	4971.6	0.22
a-TiO <sub>2</sub>	6	Octahedral	4971.5	0.17
CaTiO₃	6	Octahedral	4971.6	0.11
BaTiSi₃O <sub>9</sub>	6	Octahedral	4971.2	0.04
KNa <sub>2</sub> LiFe <sub>2</sub> Ti <sub>2</sub> Si <sub>8</sub> O <sub>24</sub>	6	Octahedral	4971.1	0.32
FeTiO <sub>3</sub>	6	Octahedral	4971.7	0.22
CaTiSiO₅	6	Octahedral	4971.4	0.18
TiZrO <sub>4</sub>	6	Octahedral	4971.1	0.21
$Ca_3(Ti,Zr)_2(Si,Al_2)O_{12}$	6	Octahedral	4971.2	0.15

**ST3:** Pre-edge centroid height and energy of titanium standards, with co-ordination number and corresponding co-ordination environment. Adapted from Farges *et al* (1996).