

## **$^{79/81}\text{Br}$ Nuclear Quadrupole Resonance Spectroscopic Characterization of Halogen Bonds in Supramolecular Assemblies**

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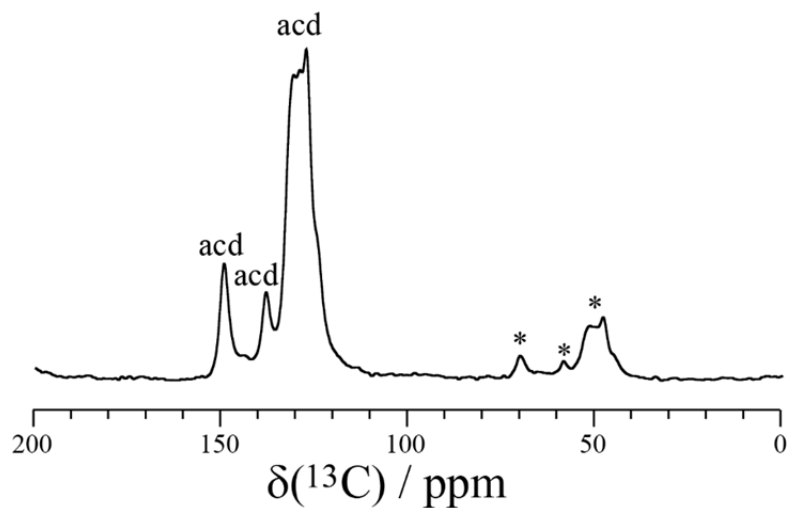
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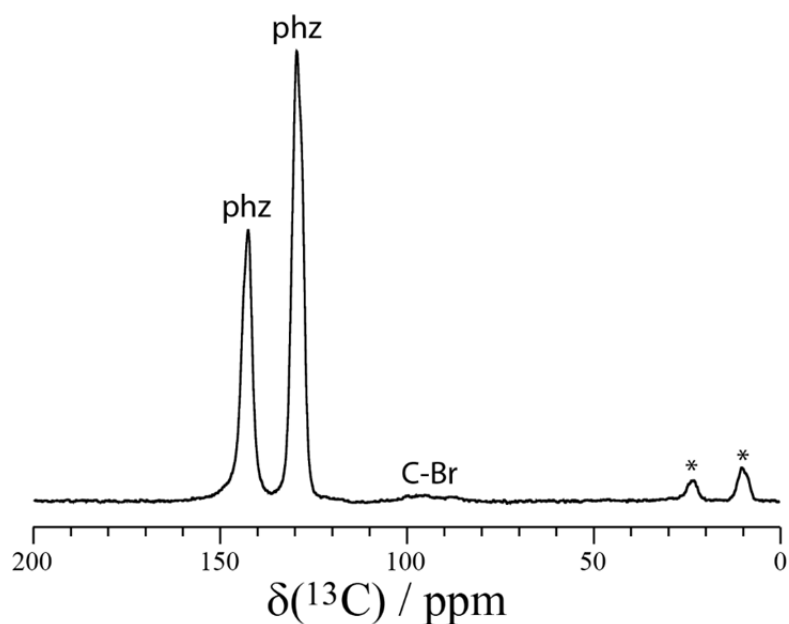
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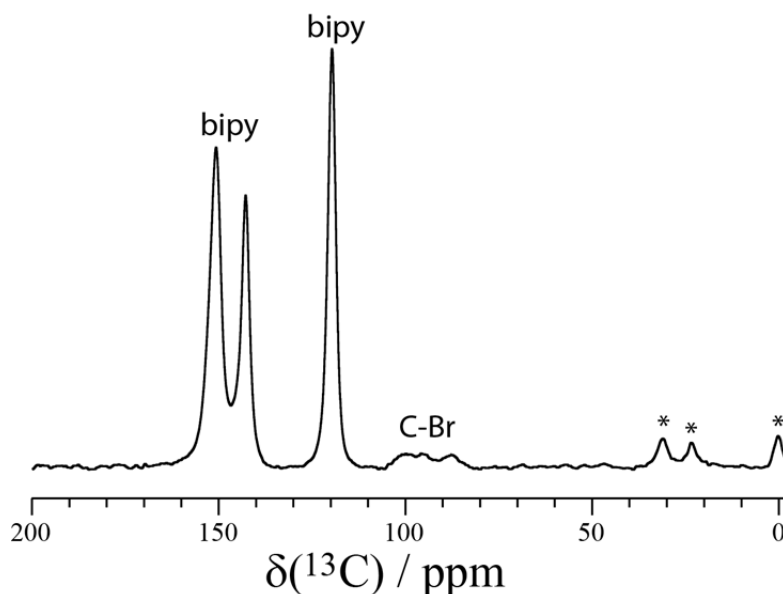
**$^1\text{H} \rightarrow ^{13}\text{C}$  Cross-Polarization Magic-Angle Spinning Solid-State NMR Spectra.  $B_0 = 9.4\text{ T}$ .**



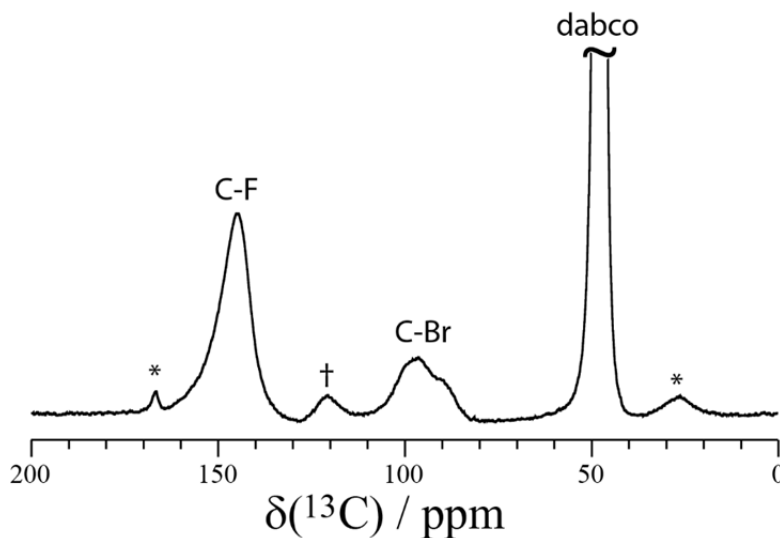
**Figure S1.** Experimental  $^{13}\text{C}$  solid-state NMR spectrum of **2** (*p*-DBrTFB)(acridine). The “acd” labels denote the resonances assigned to acridine ( $\delta(^{13}\text{C}) = 149.0 \pm 0.3\text{ ppm}$ ,  $137.6 \pm 0.3\text{ ppm}$ ,  $130.4 \pm 0.5\text{ ppm}$  to  $126.9 \pm 0.4\text{ ppm}$ , left to right). The asterisks denote spinning sidebands.



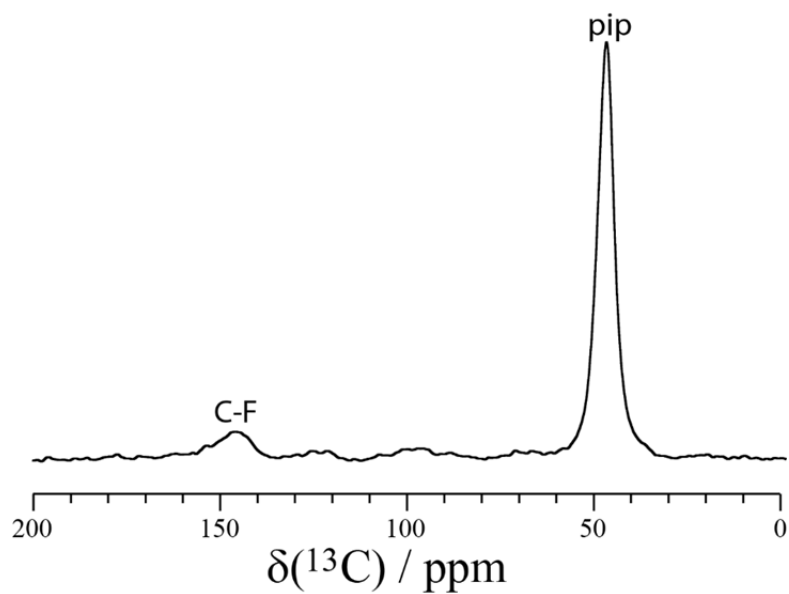
**Figure S2.** Experimental  $^{13}\text{C}$  solid-state NMR spectrum of **3** (*p*-DBrTFB)(phenazine). The “phz” labels denote the resonances assigned to phenazine ( $\delta(^{13}\text{C}) = 142.8 \pm 0.6\text{ ppm}$ ,  $129.8 \pm 0.7\text{ ppm}$ , left to right), whereas the “C-Br” label denotes the resonance assigned to the carbon covalently bonded to bromine ( $\delta(^{13}\text{C}) = 100.8 \pm 0.8\text{ ppm}$  to  $85.8 \pm 1.7\text{ ppm}$ ). The asterisks denote spinning sidebands.



**Figure S3.** Experimental  $^{13}\text{C}$  solid-state NMR spectrum of **4** (*p*-DBrTFB)(bipy). The “bipy” labels denote the resonances assigned to phenazine ( $\delta(^{13}\text{C}) = 150.9 \pm 0.4$  ppm,  $142.9 \pm 0.4$  ppm,  $119.88 \pm 0.2$  ppm, left to right), whereas the “C-Br” label denotes the resonance assigned to the carbon covalently bonded to bromine ( $\delta(^{13}\text{C}) = 103.3 \pm 1.6$  ppm to  $85.3 \pm 1.3$  ppm). The asterisks denote spinning sidebands.



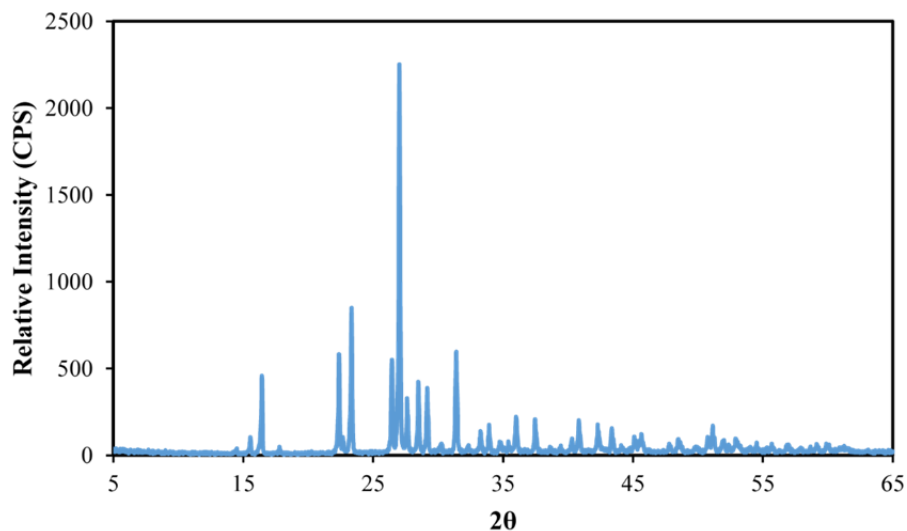
**Figure S4.** Experimental  $^{13}\text{C}$  solid-state NMR spectrum of **5** (*p*-DBrTFB)(dabco). The “dabco” label denotes the resonance assigned to dabco ( $\delta(^{13}\text{C}) = 47.6 \pm 0.1$  ppm), whereas the “C-Br” label denotes the resonance assigned to the carbon covalently bonded to bromine ( $\delta(^{13}\text{C}) = 102.6 \pm 1.3$  ppm to  $85.3 \pm 2.1$  ppm), and the “C-F” symbol denotes the carbon covalently bonded to fluorine ( $\delta(^{13}\text{C}) = 102.6 \pm 1.3$  ppm). The asterisks denote spinning sidebands. The dagger indicates a trace impurity.



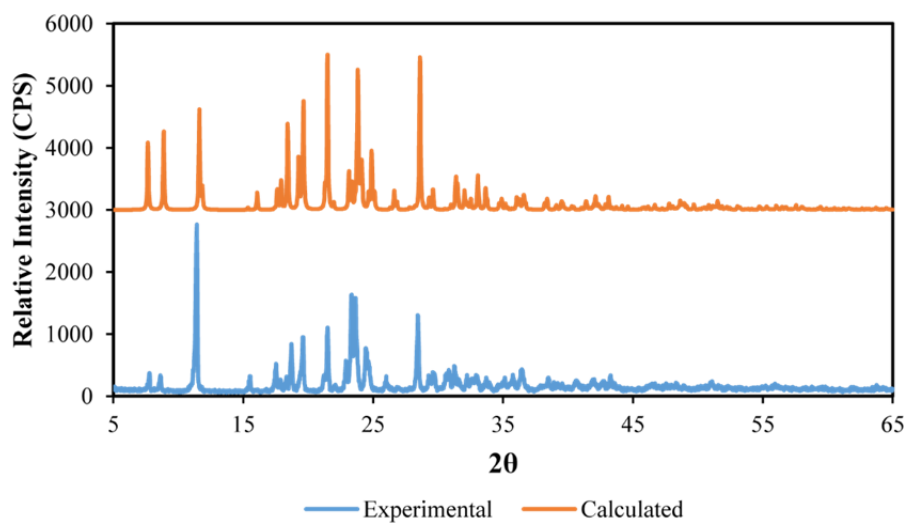
**Figure S5.** Experimental  $^{13}\text{C}$  solid-state NMR spectrum of **6** (*p*-DBrTFB)(pip). The “pip” label denotes the resonance assigned to piperazine ( $\delta(^{13}\text{C}) = 46.5 \pm 0.4$  ppm), whereas the “C-F” label denotes the resonance assigned the carbon covalently bonded to fluorine ( $\delta(^{13}\text{C}) = 145.9 \pm 2.3$  ppm). The asterisks denote spinning sidebands.

### Powder X-ray Diffraction

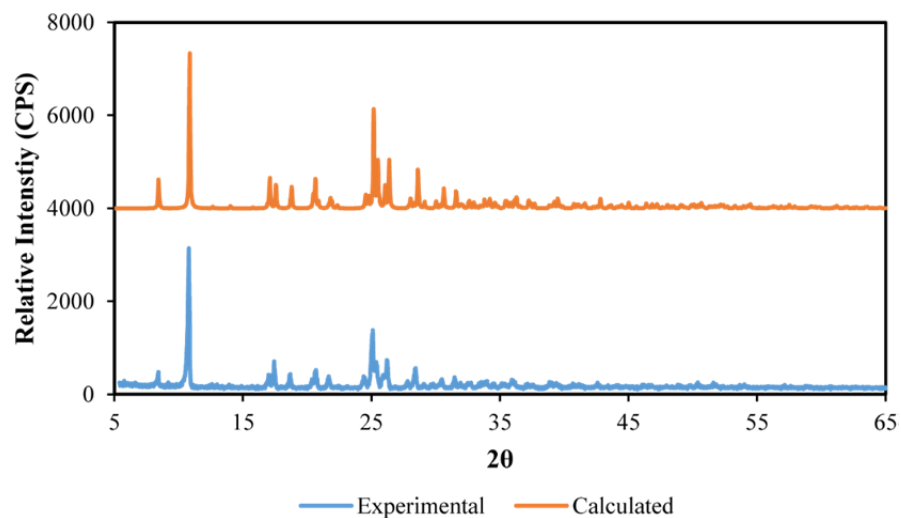
Powder X-ray diffractograms were acquired using a Rigaku Ultima IV Diffractometer operating at room temperature ( $298 \pm 1$  K) with a copper source and a diffracted beam monochromator.



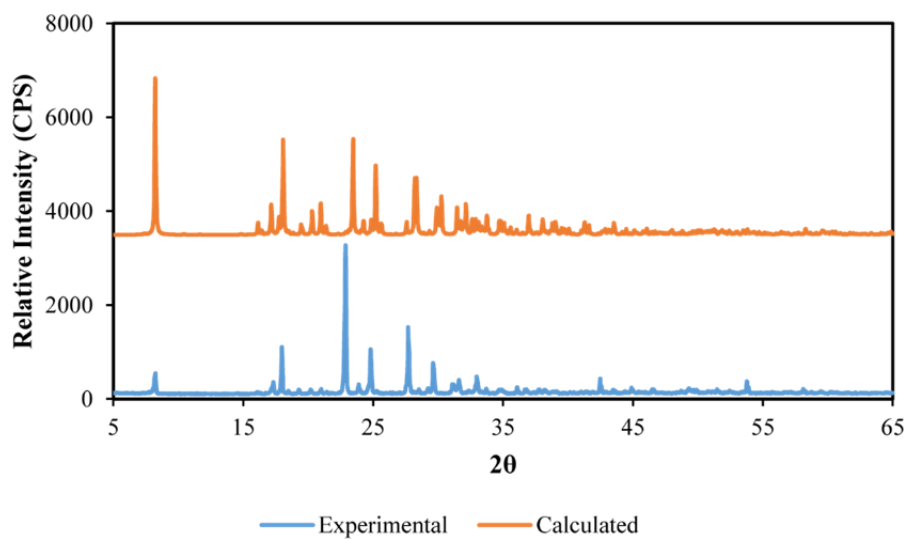
**Figure S6.** Experimental powder X-ray diffraction of sample 1.



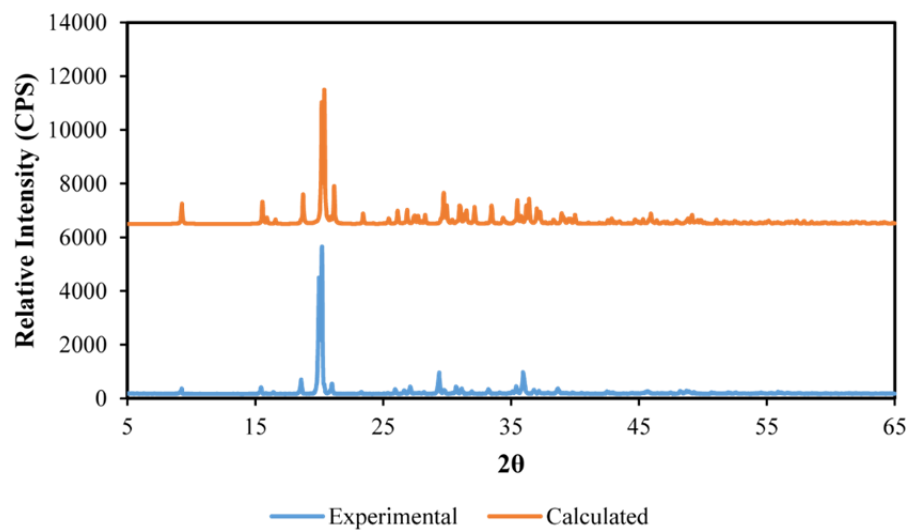
**Figure S7.** Experimental powder X-ray diffraction of sample 2 with the calculated pattern overlaid.



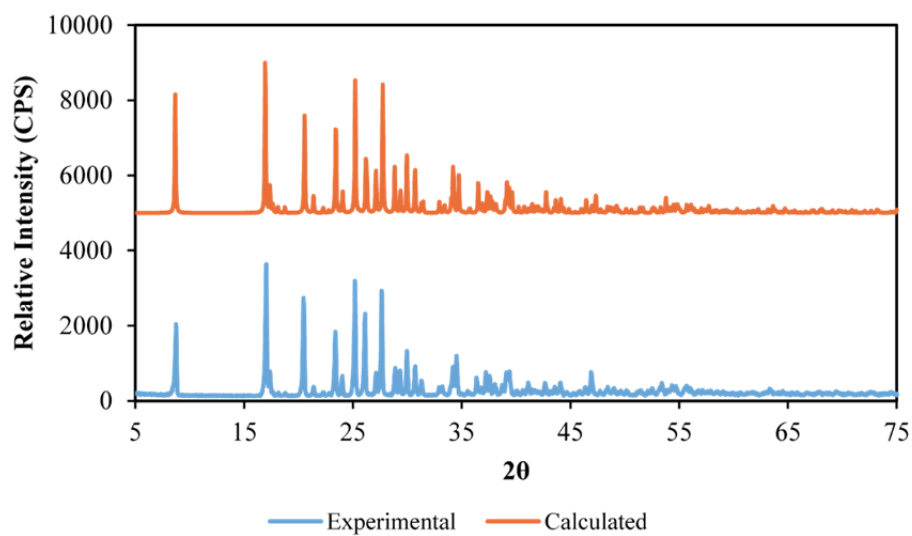
**Figure S8.** Experimental powder X-ray diffraction of sample **3** with the calculated pattern overlaid.



**Figure S9.** Experimental powder X-ray diffraction of sample **4** with the calculated pattern overlaid.



**Figure S10.** Experimental powder X-ray diffraction of sample **5** with the calculated pattern overlaid.



**Figure S11.** Experimental powder X-ray diffraction of sample **6** with the calculated pattern overlaid.