

Electronic Supplementary Information

Synthesis of 2D and 3D MOFs with tuneable Lewis acidity from preformed 1D hybrid sub-domains

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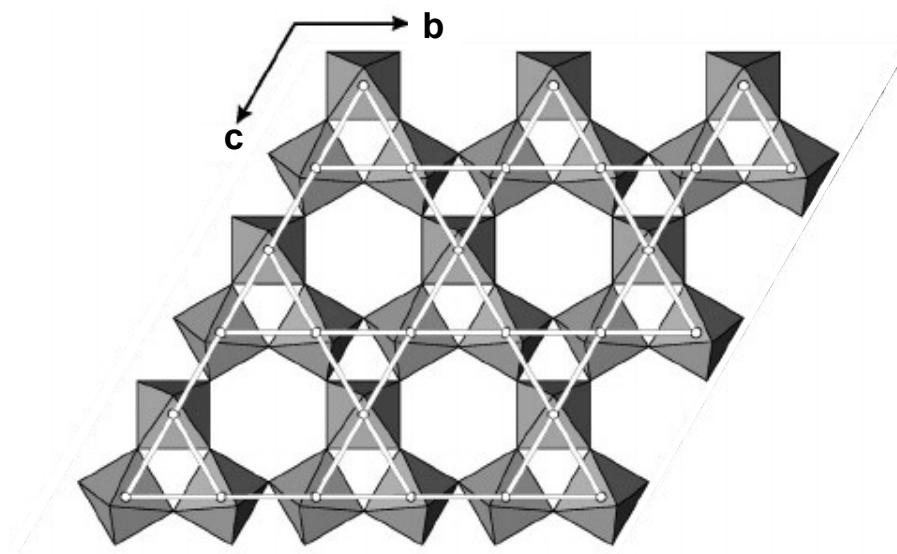


Figure S1. Representation of layers present in the solids L-MOF-EB and L-MOF-AB, based on associated aluminium octahedra following a Kagomé system. The organic spacers perpendicularly disposed to the layers were omitted for clarity. Image adapted from J. Frunzke *et al.*, *J. Mater. Chem.*, 2001, **11**, 179-185.

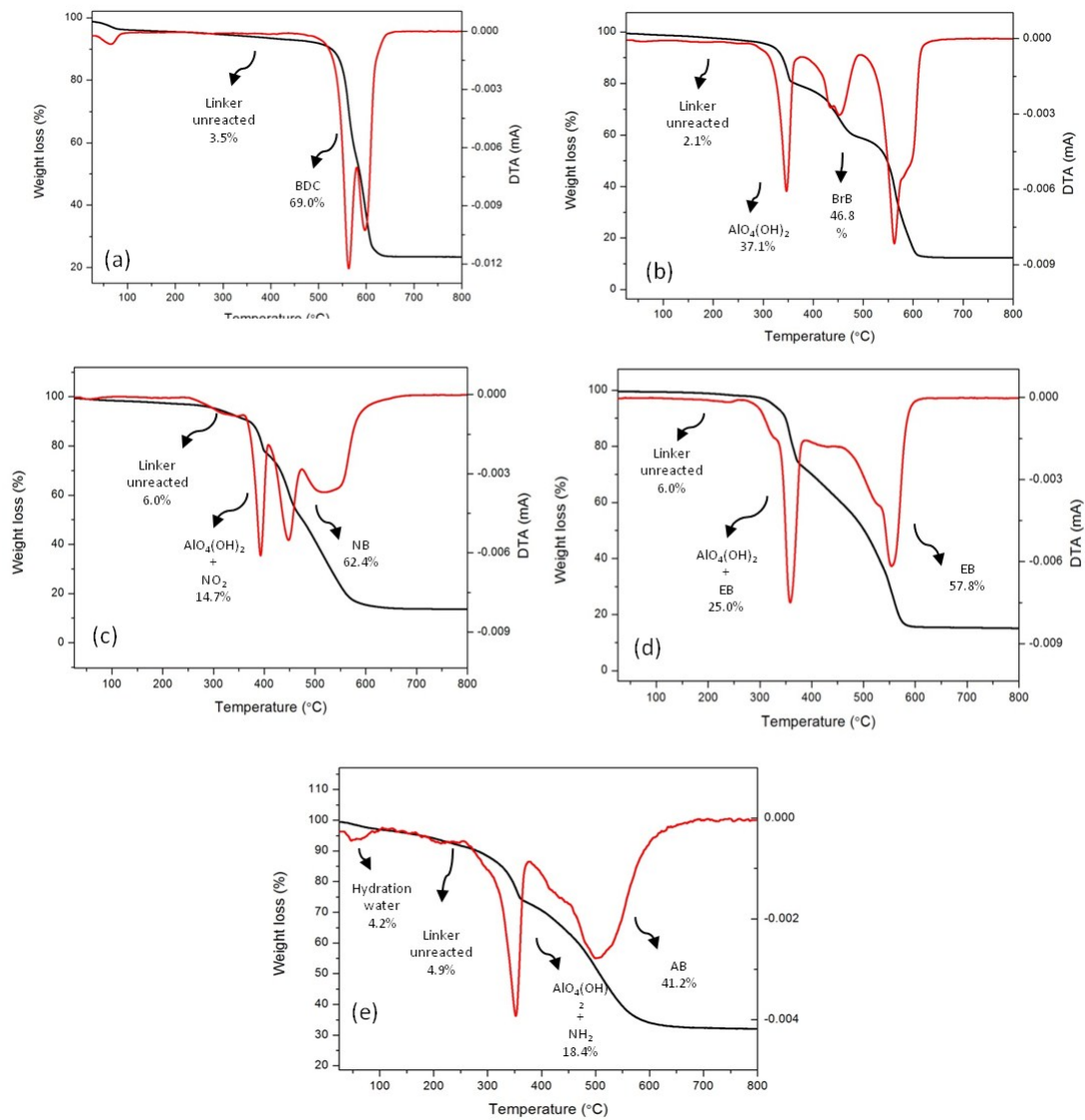


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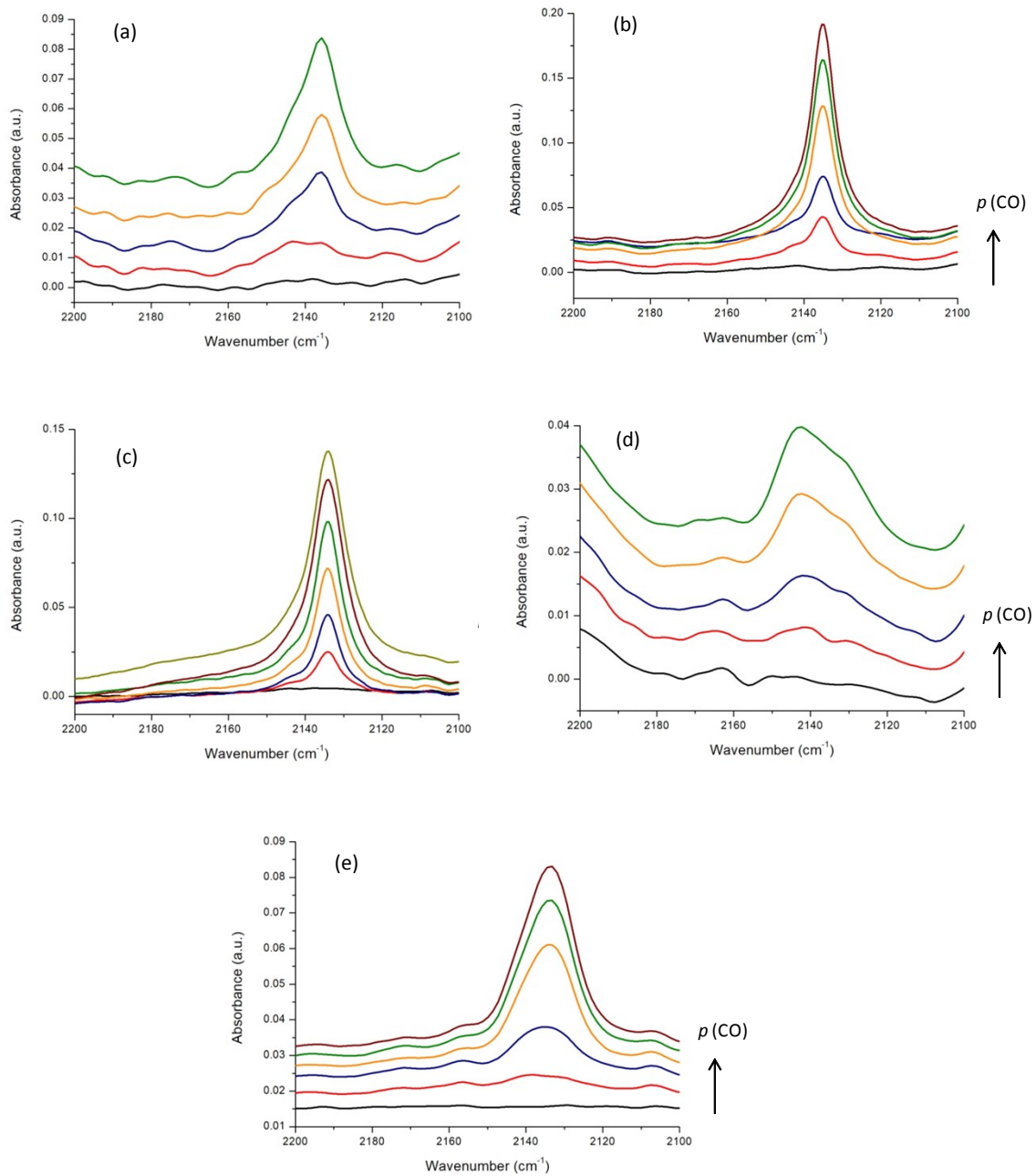


Figure S3. Difference FTIR spectra of increasing amounts of CO adsorbed at low temperature (100 K) on (a) Al-ITQ-NO₂, (b) Al/Fe-ITQ-NO₂, (c) Al-ITQ-Br, (d) L-MOF-AB and (e) L-MOF-EB (self-supporting wafers).

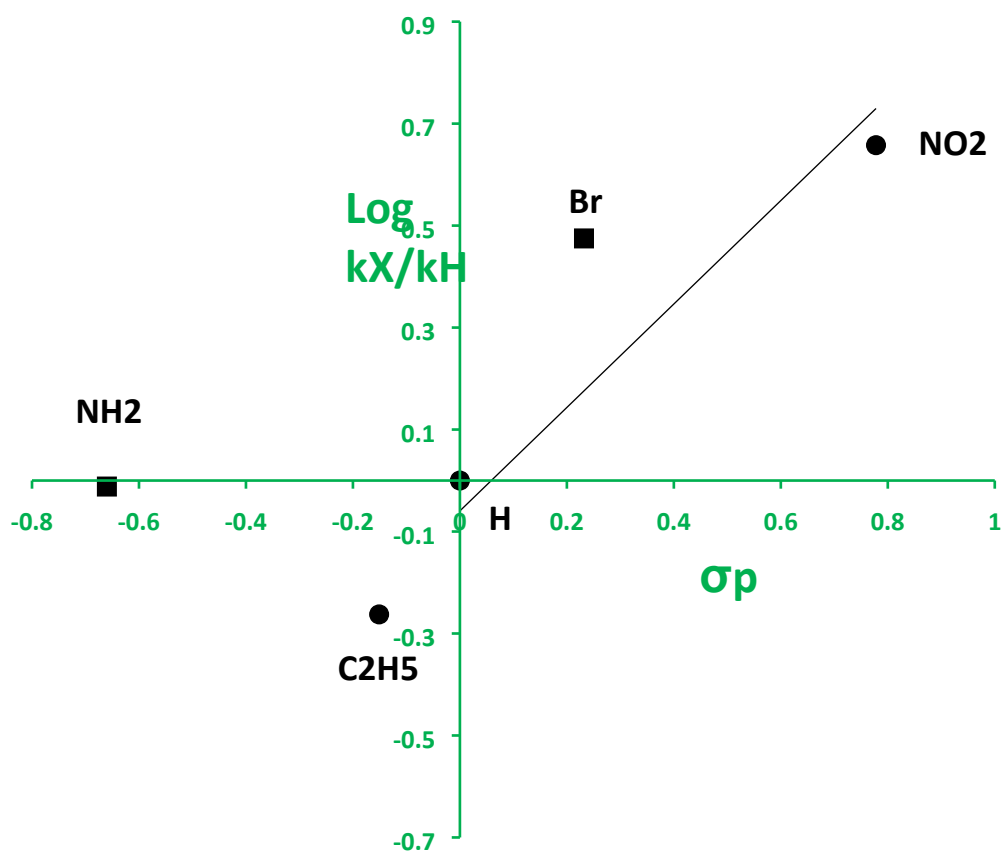


Figure S4. Hammett plot for cyanosilation of benzaldehyde with different Al-MOF type hybrid catalysts.

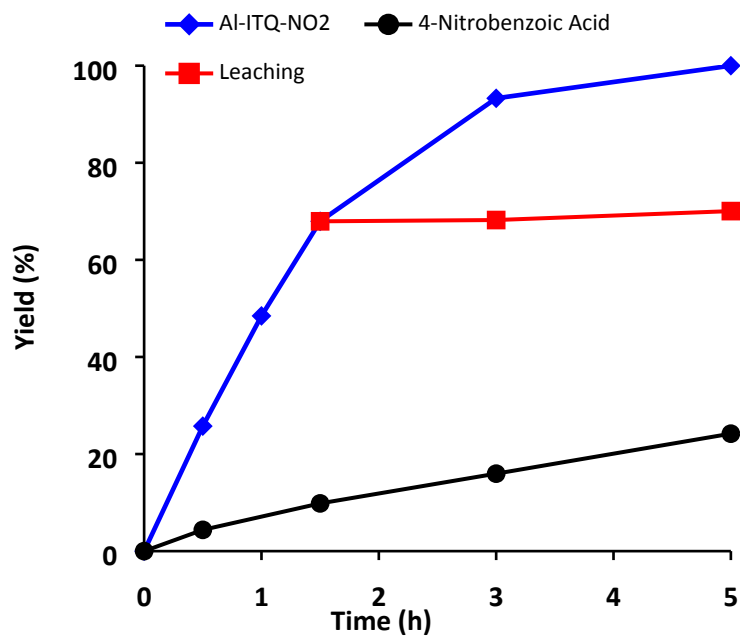


Figure S5. Kinetics for cyanohydrins production using Al-ITQ-NO₂ (0.5 mol% Al) and 4-nitrobenzoic acid (0.5 mol%) as catalysts. Test leaching was also represented.

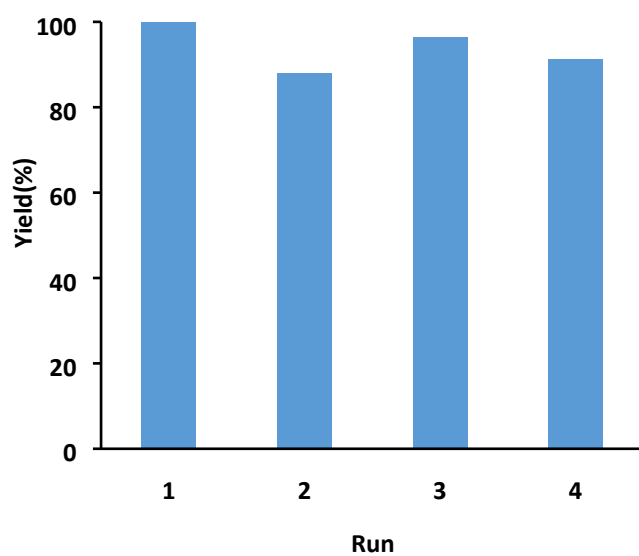


Figure S6. Yield for four consecutive uses for reaction of cyanosilylation of benzaldehyde at 5 h of reaction.

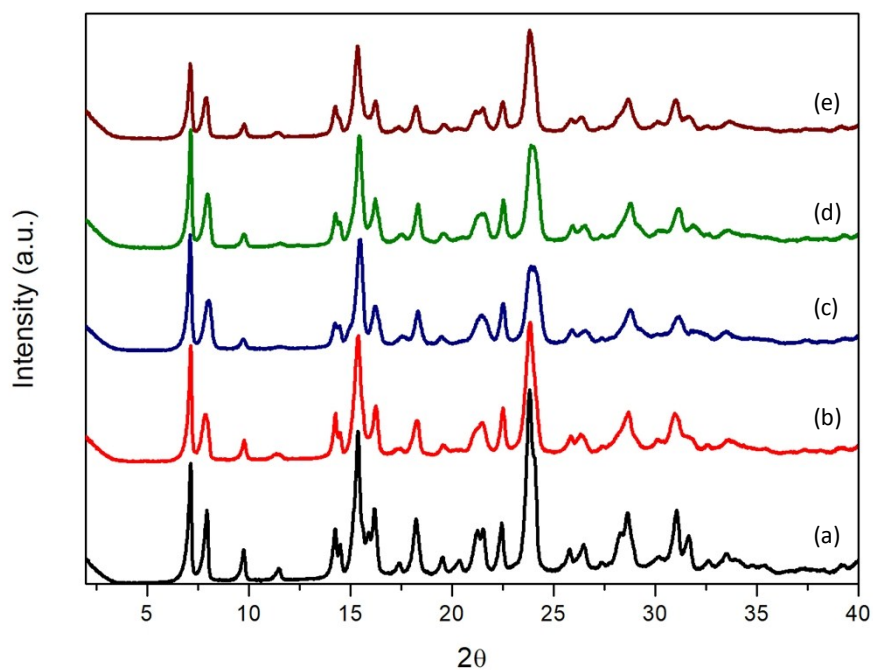


Figure S7. XRD patterns of Al-ITQ-NO₂ catalyst: (a) fresh, (b) first use, (c) second use, (d) third use and (e) fourth use.

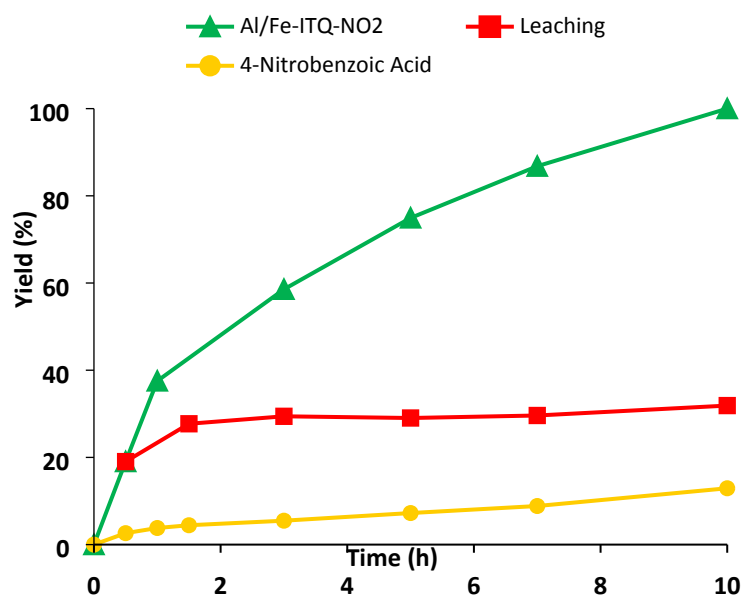


Figure S8. Kinetics for diphenyldisulfide production using Al/Fe-ITQ-NO₂ (10 mol% Fe) and 4-nitrobenzoic acid (10 mol%) as catalysts. Test leaching was also represented.

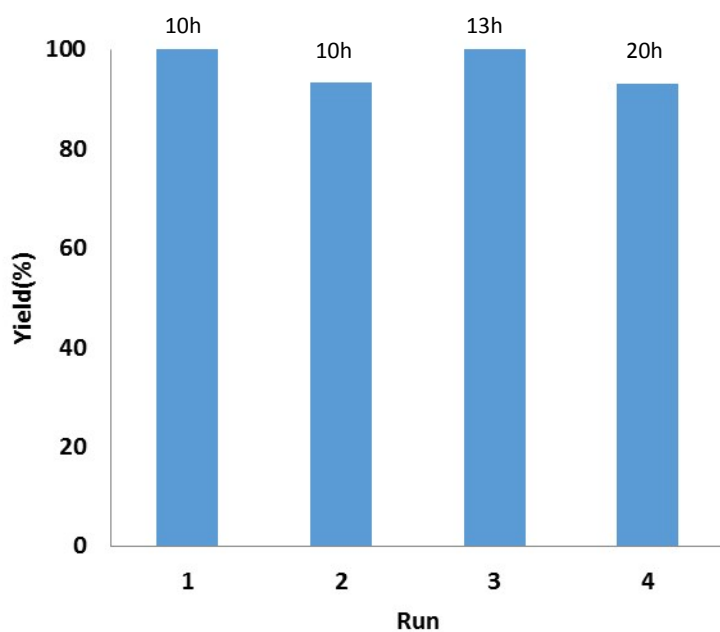


Figure S9. Yield for four consecutive uses for reaction of thiophenol oxidation.

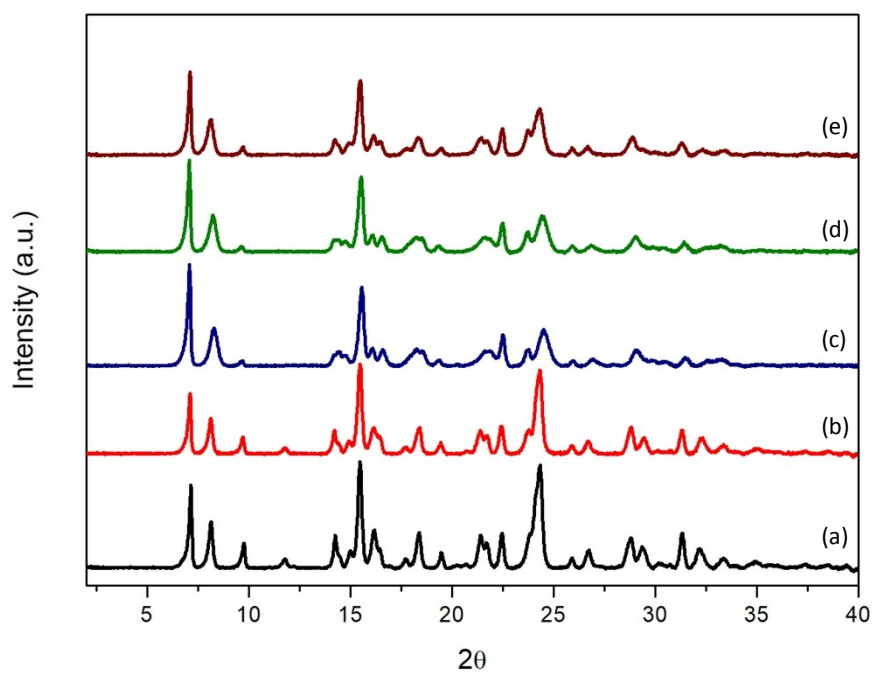


Figure S10. XRD patterns of Al/Fe-ITQ-NO₂ catalyst: (a) fresh, (b) first use, (c) second use, (d) third use and (e) fourth use.

Table S1. CO₂ uptake capacities of Al-MOF-type materials at 0°C and 1 bar.

Samples	Adsorbed Volume (cm³/g)
Al-ITQ-NO ₂	24.3
Al/Fe-ITQ-NO ₂	16.1
Al-ITQ-Br	14.7
L-MOF-EB	4.8
L-MOF-AB	6.3

Table S2. Yields and initial reaction rates for the cyanosilylation of aromatic aldehydes catalyzed by Al-ITQ-NO₂^a.

Substrate	Yield (%)	r₀ (mol·L⁻¹·h⁻¹)	TON	TOF (h⁻¹)^b	TOF (h⁻¹)^c
2-Napthaldehyde	90	3.5	180	179	134
4-Methylbenzaldehyde	99	2.9	200	148	154
4-Methoxybenzaldehyde	89	1.3	177	66	69

^a Reaction conditions: Aldehydes (5 mmol), TMSCN (6 mmol), 0.5 mol %Al, at 30°C, 5 h. ^b Calculated at initial rate. ^c Calculated at 0.5 h.

Table S3. Yields, TON and TOF values of different conventional MOFs for the preparation of cyanohydrins.

Samples	Catalyst loading (mol%)	Yield (%)	TON	TOF (h ⁻¹) ^e
MIL-47(V) ^{a [1]}	1	46	46	-
MIL-53(Al) ^{a [1]}	1	26	26	-
UiO66(Zr) ^{a [1]}	1	68	68	-
MIL-101(Cr) ^{a [1]}	1	96	96	541
In-MOF ^{a [2]}	1	72	72	204
In-MOF ^{a [2]}	0.1	80	800	1320
Zr-MCM-41 ^{b [3]}	2	28	14	2
Zr- β ^{b [3]}	1	99	99	61
Cp-Zr-binol-SBA-15 ^{b [3]}	2	95	48	0.7
Al-MCM-41 ^{b [4]}	1	99	99	108
AlPO-5 ^{b [3]}	10	0	-	-
H-Y ^{c [4]}	-	4	-	-
H-ZSM-5 ^{c [4]}	-	0	-	-
Er-MCM-41 ^{d [5]}	-	90	137	137 ^f
Al-ITQ-NO ₂ ^a	1	97	97	210
Al-ITQ-NO ₂ ^a	0.1	92	920	405

^a Reaction conditions: Benzaldehyde (1 mmol), TMSCN (1.2 mmol), 30°C, 3 h. ^b Benzaldehyde (1 mmol), TMSCN (2 mmol), 40°C, 20 h, CH₂Cl₂ (2.5 mL). ^c Benzaldehyde (1 mmol), TMSCN (5 mmol), 5 mg catalyst, 0°C, 1 h, CH₂Cl₂. ^d Benzaldehyde (0.66 mmol), TMSCN (1.3 mmol), 10 mg catalyst, room temperature, 1 h. ^e Calculated at initial reaction rate. ^f Calculated at 1 h. In blue, hybrid catalyst synthesized in this study.

Table S4. Yields, TON and TOF values of MIL-101(Cr) with different mol% of active centres for the preparation of cyanohydrins.^a.

Samples	Catalysts (mol%)	Yield (%)	Time (h)	TON	TOF (h ⁻¹)		
					^b	^c	^d
MIL-101(Cr) ^[6]	1.0	96	1.5	96	64	541	154
Al-ITQ-NO ₂	1.0	97	3.0	97	32	210	105
MIL-101(Cr) ^[6]	0.55	96	2.5	174	69	382	187
Al-ITQ-NO ₂	0.50	100	5.0	200	40	137	103
Al/Fe-ITQ-NO ₂	0.50	98	5.0	196	39	410	242
MIL-101(Cr) ^[6]	0.25	95	5.0	380	76	360	240
MIL-101(Cr) ^[6]	0.15	95	8.0	633	79	300	210
Al-ITQ-NO ₂	0.10	92	9.0	920	102	405	400

^a Reaction conditions: Benzaldehyde (1 mmol), TMSCN (1.2 mmol), at 30°C. ^b Calculated when the reaction was completed. ^c Calculated at initial rate. ^d Calculated at 0.5 h. In blue, hybrid catalysts synthesized in this study.

Table S5. Yields and initial reaction rates for the preparation of diphenyldisulfide^a.

Samples	Yield(%)	TON
Al/Fe-ITQ-NO ₂	100	10
Al-ITQ-NO ₂	33	3.3
MIL-53(Al)	11	1.1
MIL-53(Al)-NO ₂	4	0.4
4-Nitrobenzoic Acid (10 mol%)	13	-
Blank	3	-
Fe(BTC) ^{b [7]}	64	2.5
MIL-100(Fe) ^{b [7]}	81	2.3

^a Reaction conditions: Thiophenol (0.48 mmol), 10 mol %Fe, 70°C, 10 h, acetonitrile (1 mL). ^b Thiophenol (0.250 mL), 9 mg Fe(BTC) or 10 mg MIL-100(Fe), 70°C, 24 h, acetonitrile (5 mL)

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