

Supplementary Information for
Zirconia Incorporated Calcium Looping Absorbents with
Superior Sintering Resistance for Carbon Dioxide
Capture from *In-situ* or *Ex-situ* Processes

Ming Zhao ^{a, b}, Xu He ^{a, b}, Guozhao Ji ^{a, b}, Yinqiang Song ^{a, b}, Xiao Zhao ^{*, a, c}

^a School of Environment, Tsinghua University, Beijing, 100084, China.

^b Key Laboratory for Solid Waste Management and Environmental Safety, Ministry of Education Beijing 100084, China.

^c College of Water Resources & Civil Engineering, China Agricultural University, Beijing 100083, China.

*Corresponding Author. Tel: +86 10 62784701. Email: xiaozhao88@cau.edu.cn

Table S1. Crystallite sizes of CaO and CaZrO₃ of some samples.

Sample	CaO (nm)		CaZrO ₃
	(200)	(220)	(202)
SD-CaO	>100	>100	-
SD-Zr10	57.3	54.1	25.5
SD-Zr20	54.6	52.2	26.1
SD-Zr30	46.9	45.7	26.7
FD-Zr20	53.6	53.7	22.0
HD-Zr20	51.5	52.0	24.7

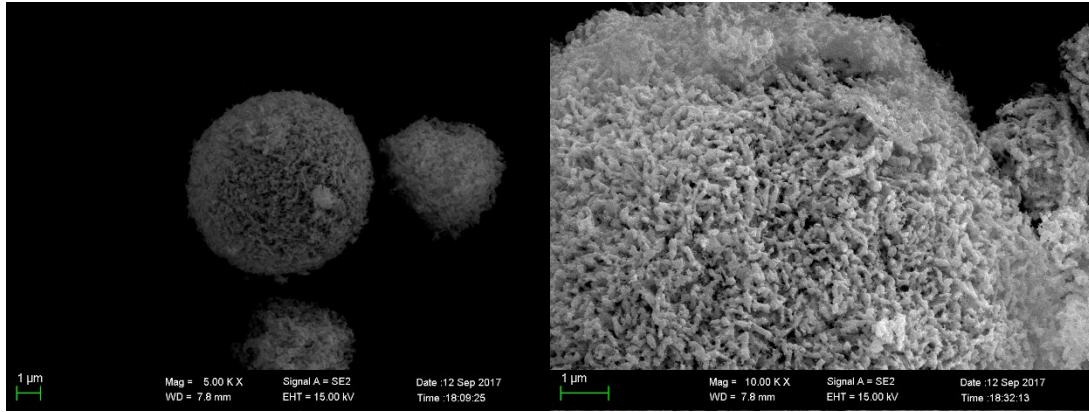


Fig. S1. Sphere structure of SD-Zr₂₀.

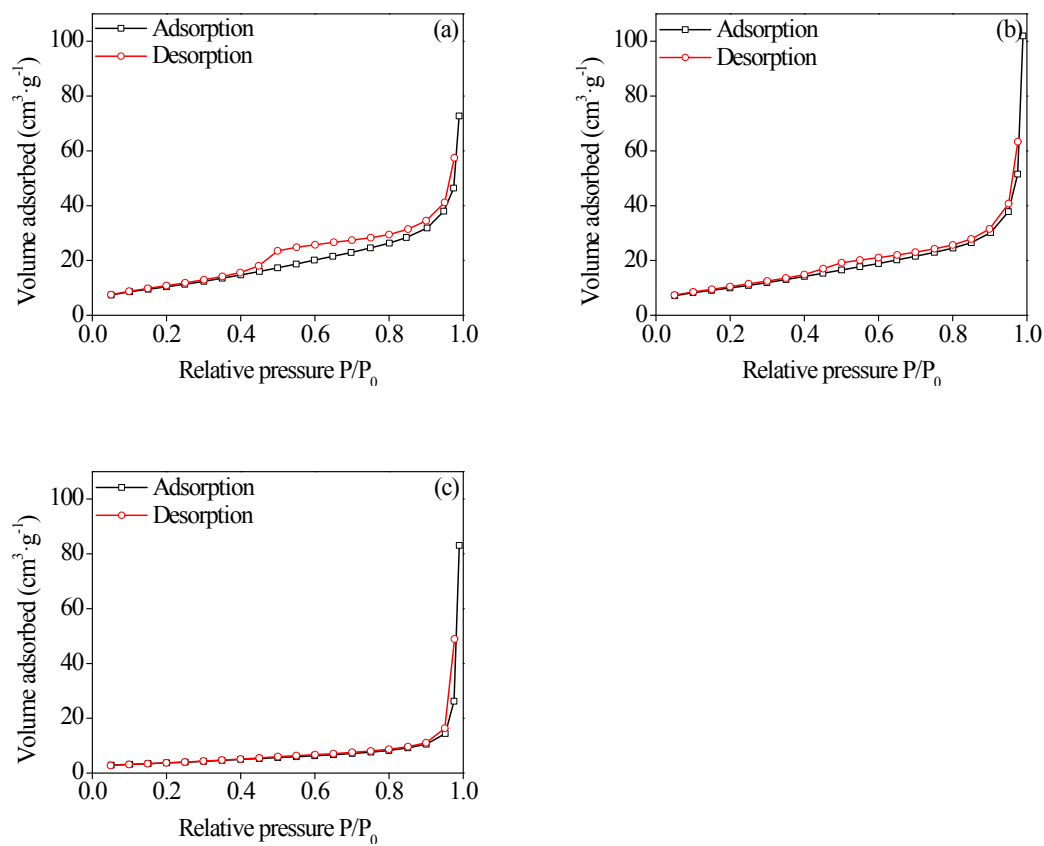


Fig. S2. The N₂ physi-sorption isotherms of (a) FD-Zr₂₀, (b) SD-Zr₂₀ and (c) HD-Zr₂₀.

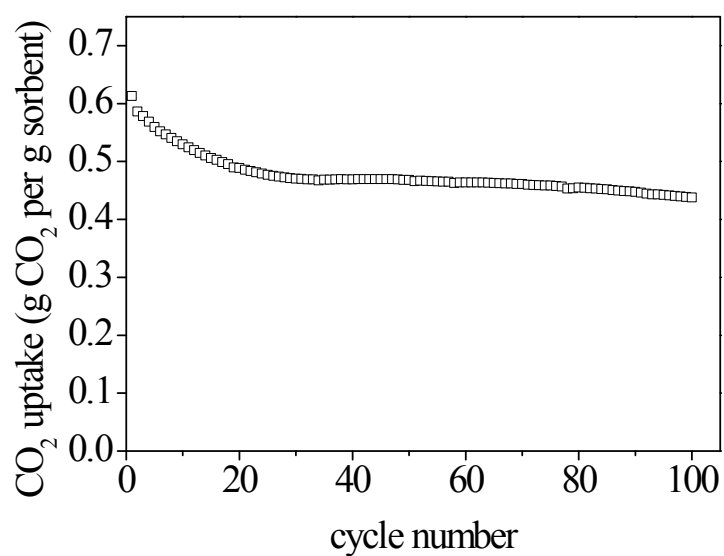


Fig. S3. The CO₂ uptake of SD-Zr20 over the 100 cycles under severe conditions (carbonation in 90 vol.% CO₂ at 650 °C for 10 min and calcination in 90 vol.% CO₂ at 950 °C with no dwell time).

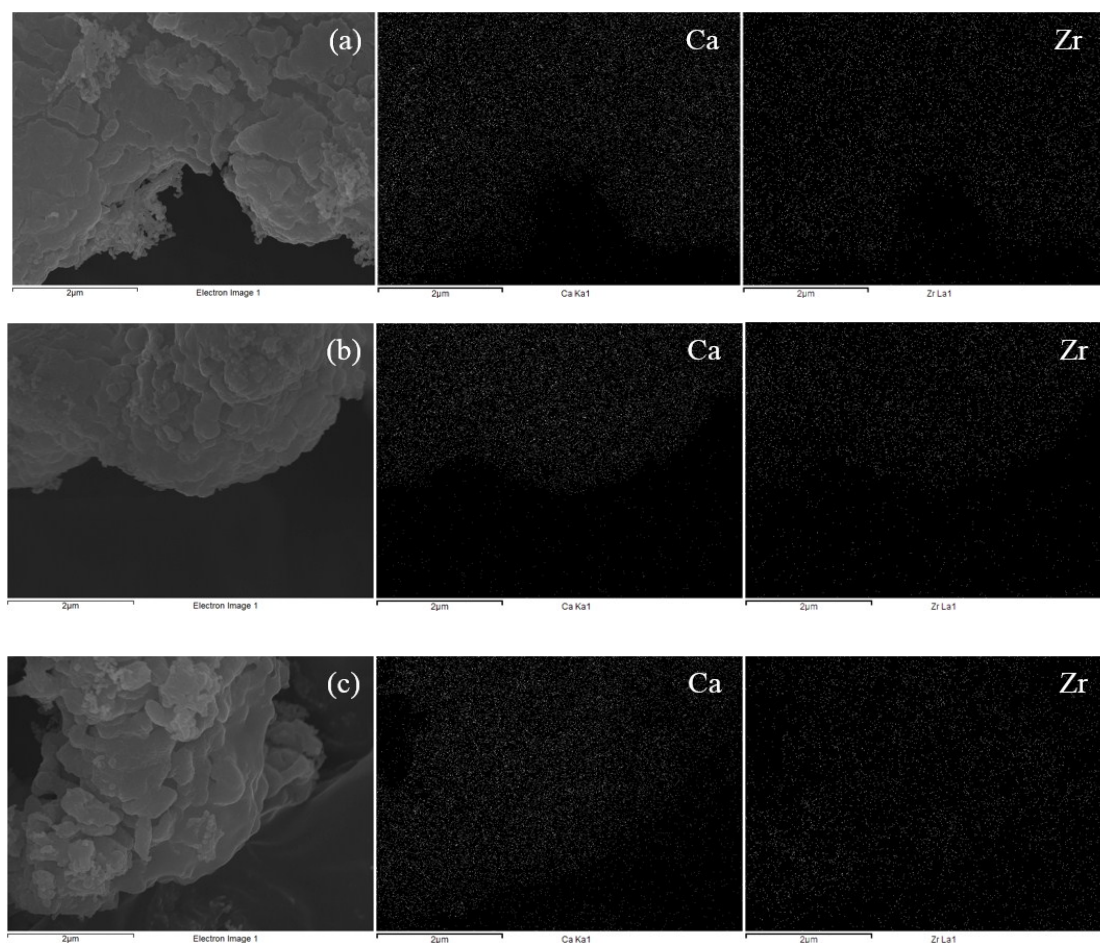


Fig. S4. EDS figures of FD-Zr20 (a), SD-Zr20 (b) and HD-Zr20 (c) after cycles under severe conditions.