Electronic supplementary information

Light-induced water oxidation by polymorphs of Zn-Co-Ni Oxide spinel

catalyst: A comparative study

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Figure

S1: (a) XRD pattern

of 3D highly porous $Zn_{0.2}Ni_{0.8}Co_2O_4$ microspheres. FE-SEM image of $Zn_{0.2}Ni_{0.8}Co_2O_4$ microspheres as the inset figure. (b) XRD pattern of mesoporous $Zn_{0.2}Ni_{0.8}Co_2O_4$ 1D rods. TEM image of $Zn_{0.2}Ni_{0.8}Co_2O_4$ rods as the inset figure.





Figure S2: (a) N₂ adsorption–desorption isotherms and the corresponding pore size distribution (inset) of 3D highly porous $Zn_{0.2}Ni_{0.8}Co_2O_4$ microspheres. (b) N₂ adsorption–desorption isotherms and the corresponding pore size distribution (inset) of mesoporous $Zn_{0.2}Ni_{0.8}Co_2O_4$ rods



Fig. S3

Figure S3: Oxygen evolution in the presence of **MPS** catalyst (3 mg in 2 mL) in different pH range; $\text{Ru}(\text{bpy})_3^{2+}= 1 \text{ mM}$ and $\text{Na}_2\text{S}_2\text{O}_8 = 10 \text{ mM}$; Phosphate buffer (10 mM, pH 6.5); Phosphate buffer (10 mM, pH 7.4) and Phosphate buffer (10 mM, pH 8)

Fig. S4(a-b)



Figure S4: (a) Evolved oxygen in the gas phase with **MPS** (1 mg in 2 mL Phosphate buffer (10 mM, pH 7); $Ru(bpy)_3^{2+}= 1$ mM and $Na_2S_2O_8 = 10$ mM). (b) Evolved oxygen in the gas phase with **MPR** (3 mg in 2 mL Phosphate buffer (10 mM, pH 7); $Ru(bpy)_3^{2+}= 1$ mM and $Na_2S_2O_8 = 10$ mM).



Figure S5: (a) Co $2p_{3/2}$, Ni $2p_{3/2}$, and Zn $2p_{3/2}$ XP spectra of MPS. (b) (a) Co $2p_{3/2}$, Ni $2p_{3/2}$, and Zn $2p_{3/2}$ XP spectra of MPR.



Figure S6: (a) Raman spectra of **MPS** before and after the water oxidation reaction (after the 1st and 5th catalytic cycle). (b) Raman spectra of **MPR** before and after the water oxidation reaction (after the 1st and 5th catalytic cycle).