

Supporting Information

FeS₂ Microspheres Wrapped by N-doped rGO from a Fe-Based Ionic Liquid Precursor for Rechargeable Lithium Ion Batteries

Xueda Ding^{ab}, Chengfeng Du^{*c}, Jianrong Li^{*a} and Xiaoying Huang^a

^aState Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences 350002 Fuzhou (P.R. China)

^bCollege of Chemistry, Fuzhou University, Fuzhou, Fujian 350108, PR China

^cState Key Laboratory of Solidification Processing, Center of Advanced Lubrication and Seal Materials, Northwestern Polytechnical University, Xi'an, Shaanxi 710072, P.R. China

E-mail: jrli@fjirsm.ac.cn (J. Li) and cfdu@nwpu.edu.cn (C. Du)

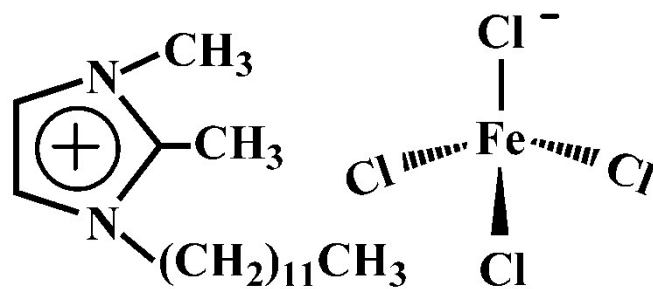


Fig. S1 Structural view of the Fe-based ionic liquid [C₁₂MMim]FeCl₄.

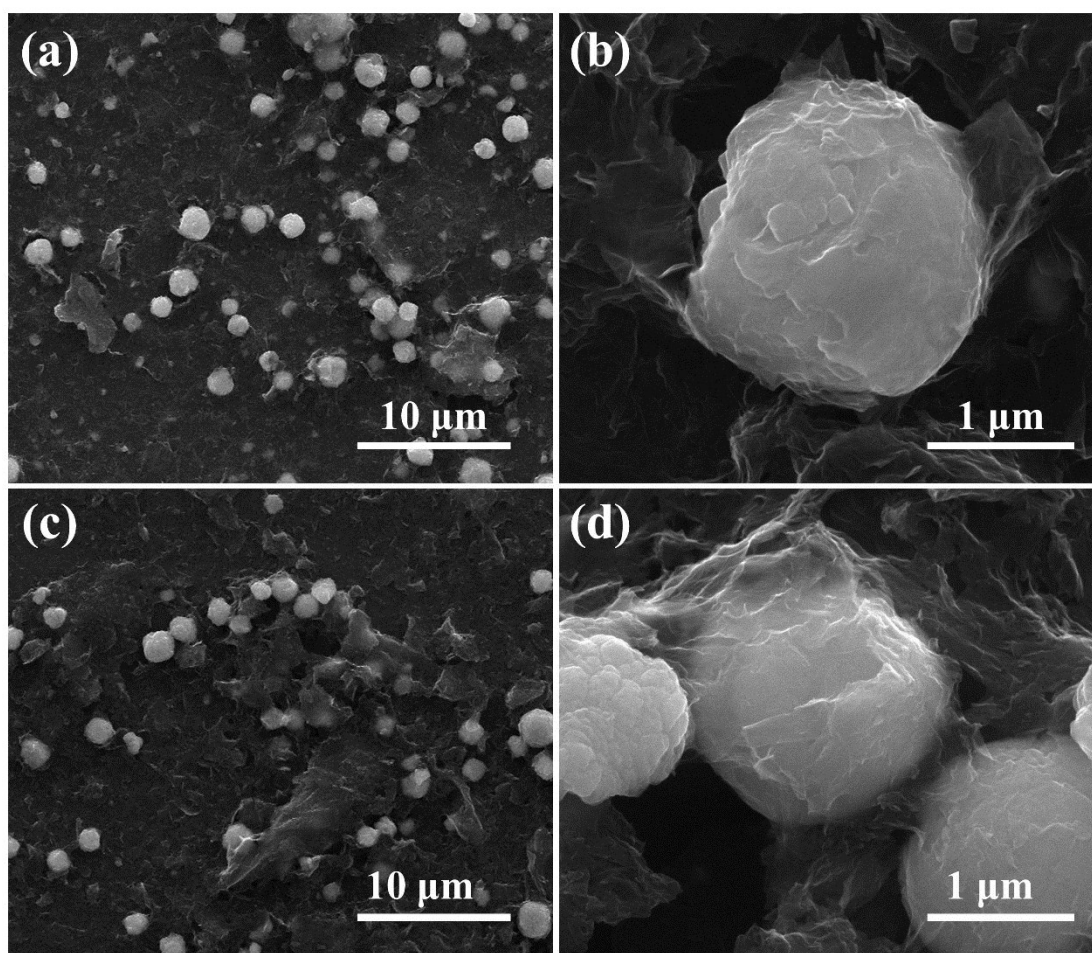


Fig. S2 SEM images of (a~b) FeS₂/N-rGO-0.5 and (c~d) FeS₂/N-rGO-2.

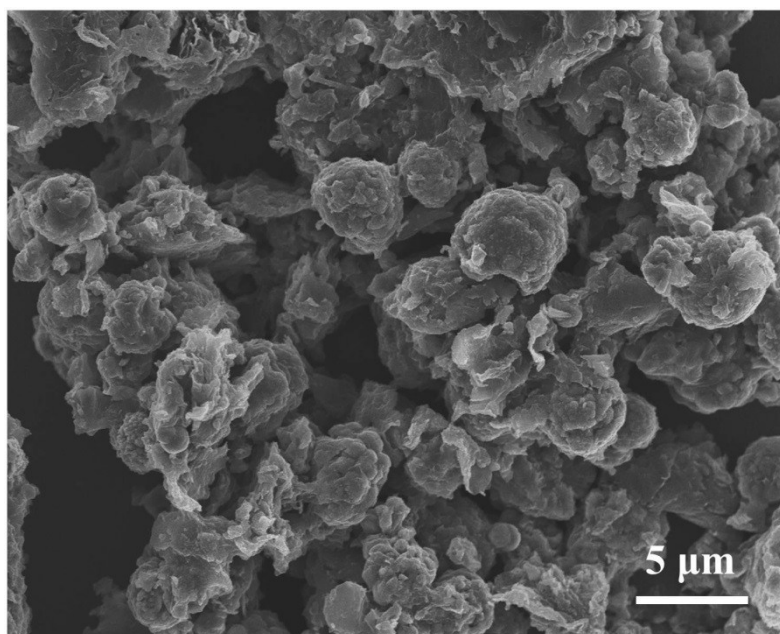


Fig. S3 SEM image of FeS₂/rGO.

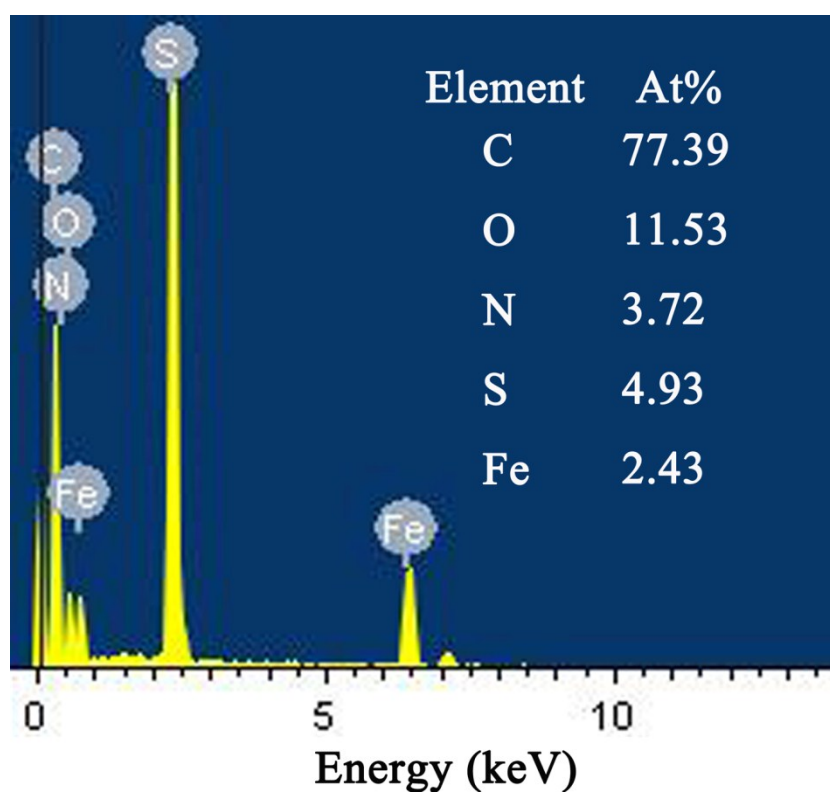


Fig. S4 EDS of FeS₂/N-rGO-1.

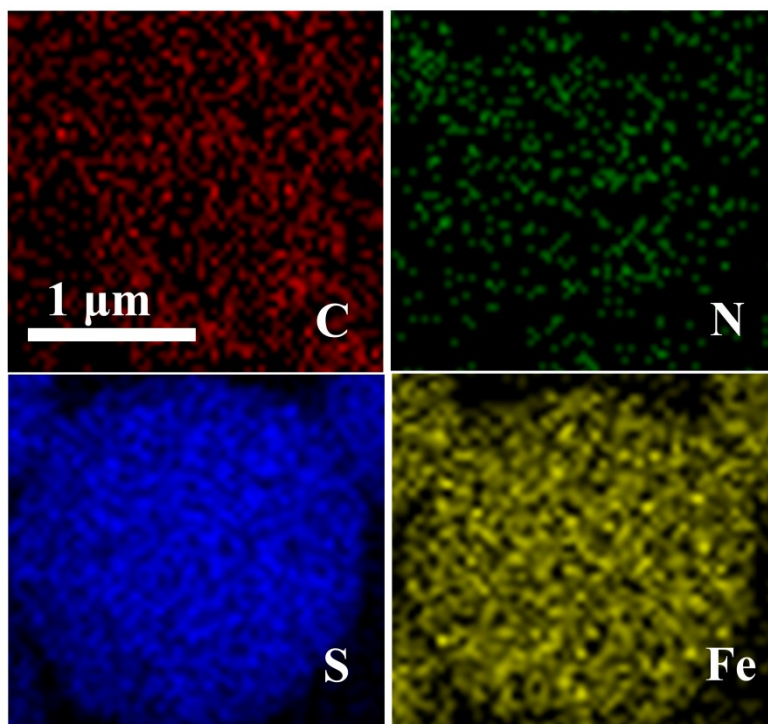


Fig. S5 SEM-EDS elemental mapping of C, N, S, and Fe elements for FeS₂/N-rGO-1.

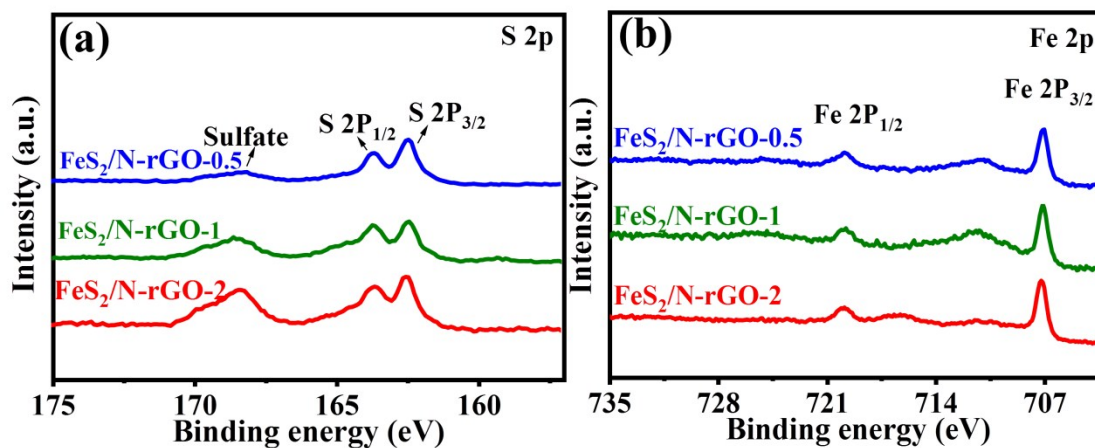


Fig. S6 High-resolution XPS spectra (a) S 2p (b) Fe 2p of FeS₂/N-rGO-0.5, FeS₂/N-rGO-1, and FeS₂/N-rGO-2.

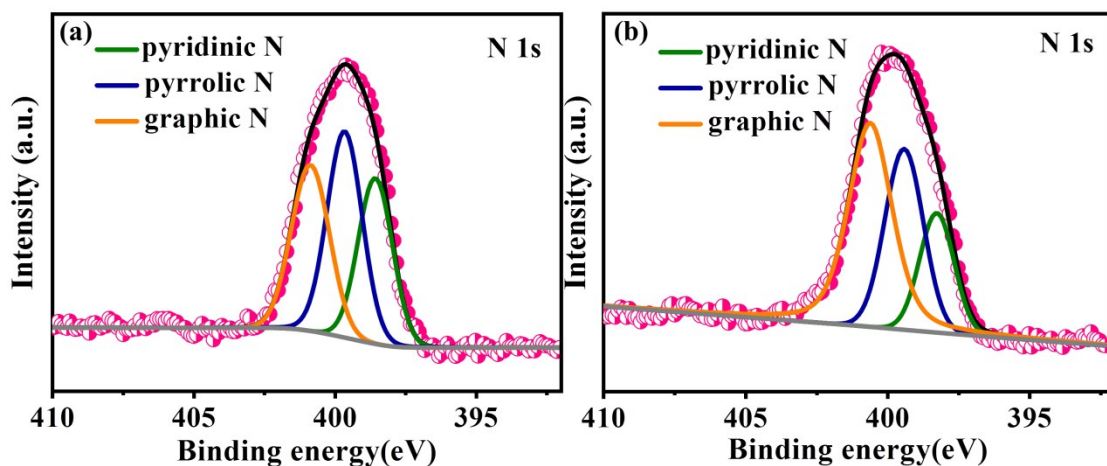


Fig. S7 High-resolution N 1s XPS spectra of (a) FeS₂/N-rGO-0.5 and (b) FeS₂/N-rGO-2.

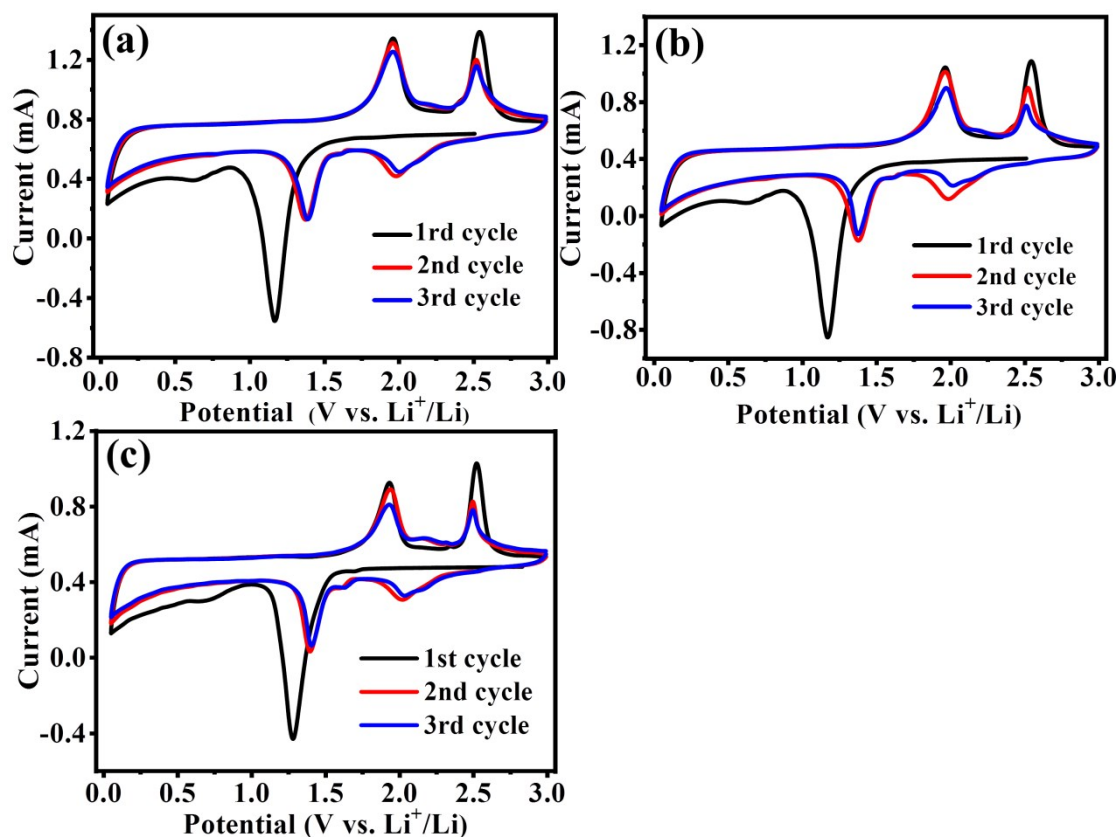


Fig. S8 CV curves of (a) FeS₂/rGO, (b) FeS₂/N-rGO-0.5, and (c) FeS₂/N-rGO-2 at a scan rate of 0.1 mV s⁻¹ between 0.05 and 3.0 V.

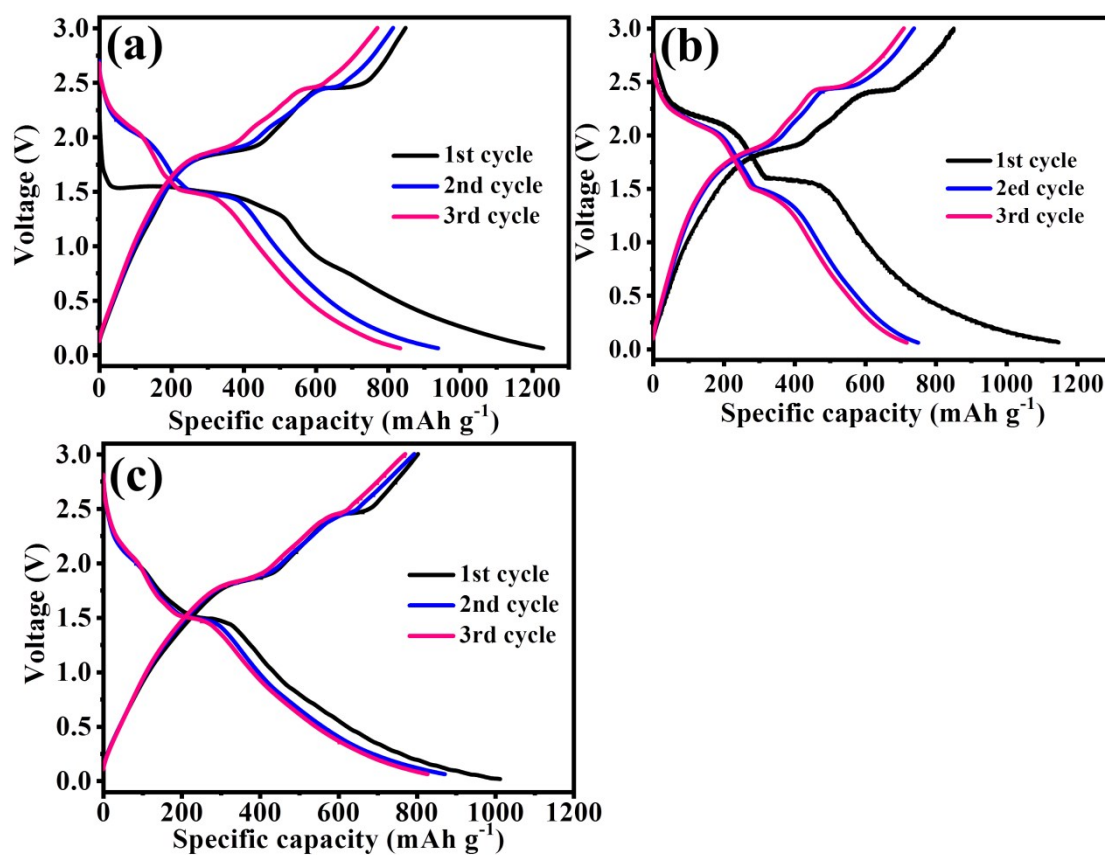


Fig. S9 Charge/discharge curves of (a) FeS₂/rGO, (b) FeS₂/N-rGO-0.5, and (c) FeS₂/N-rGO-2 at the current density of 150 mA g⁻¹, respectively.

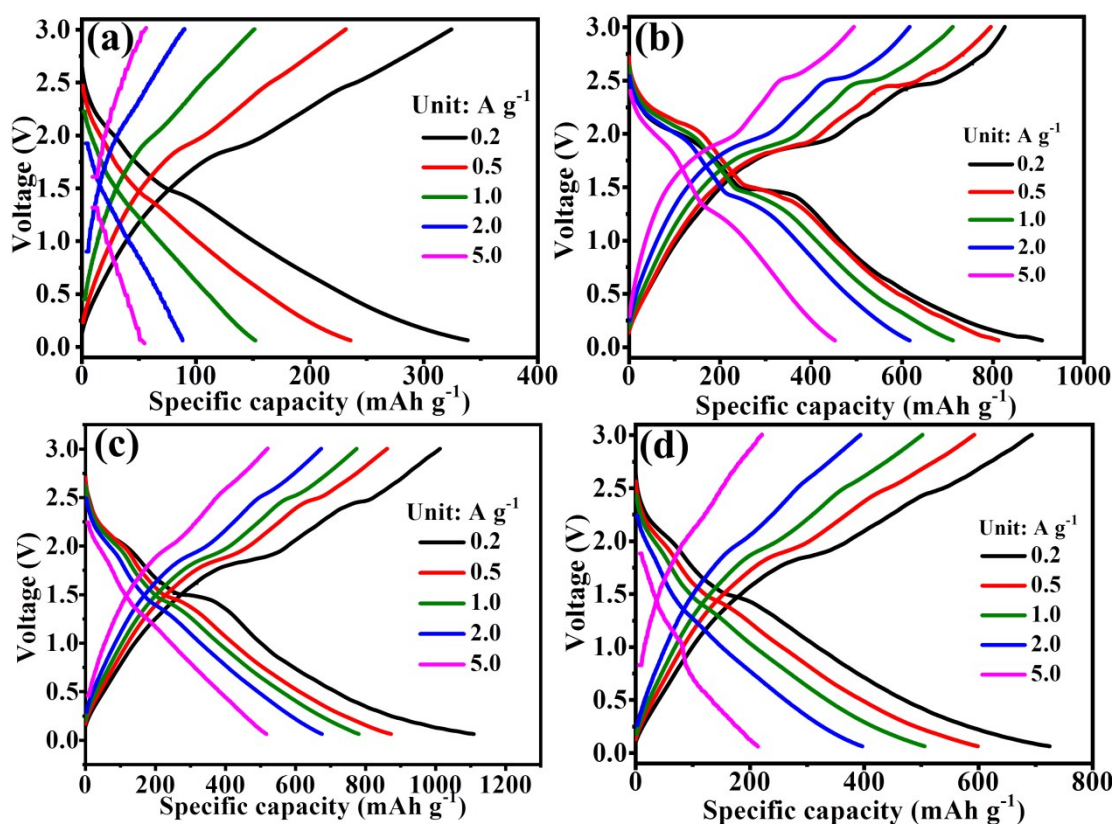


Fig. S10 The charge/discharge profiles of (a) FeS₂/rGO, (b) FeS₂/N-rGO-0.5, (c) FeS₂/N-rGO-1, and (d) FeS₂/N-rGO-2 at various current densities.

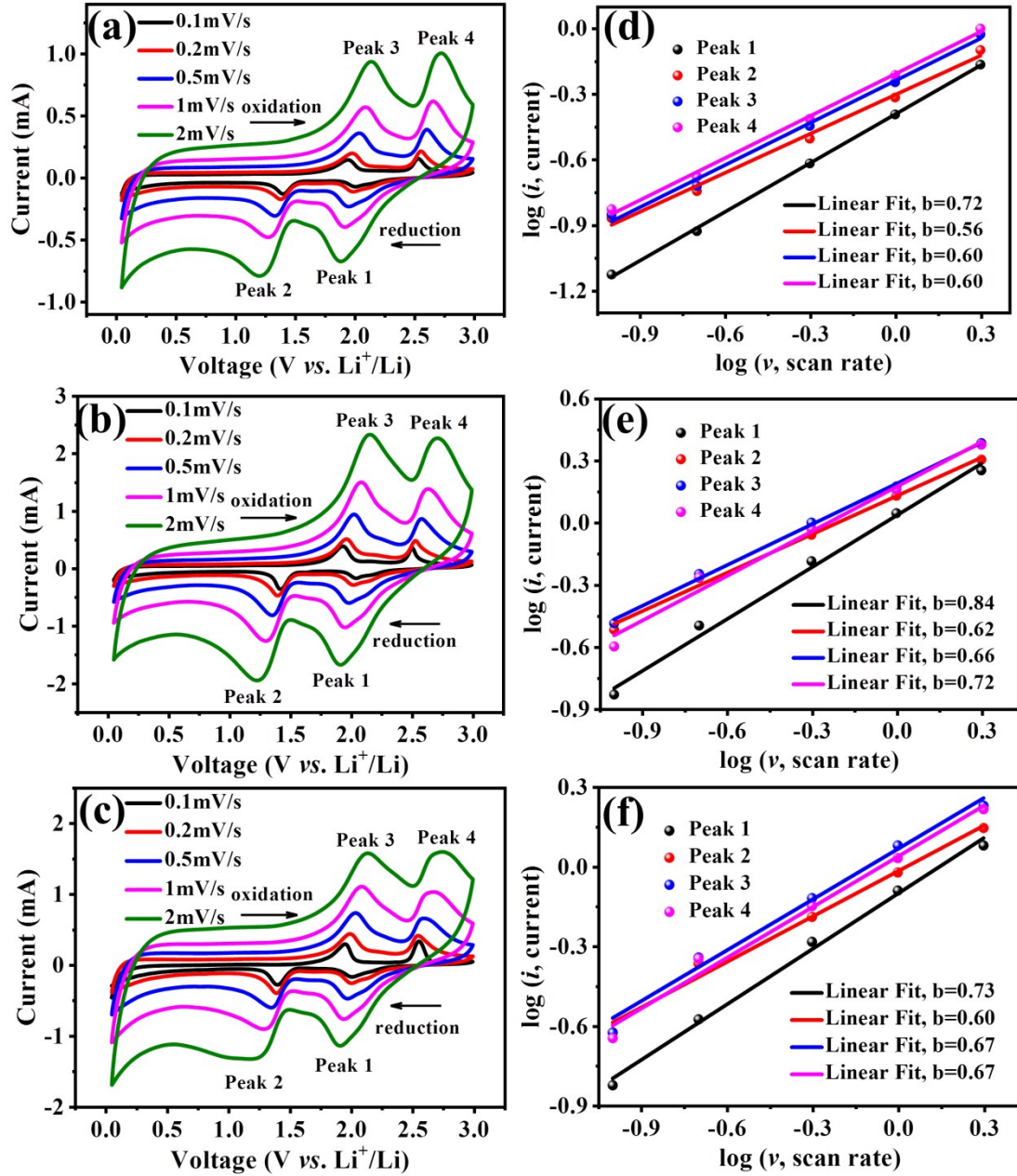


Fig. S11 Kinetic analysis of different cathodes: CV curves with scan rates from 0.1 to 2 mV s⁻¹ for (a) FeS₂/rGO, (b) FeS₂/N-rGO-0.5, and (c) FeS₂/N-rGO-2. CV peak current (I_p) logarithmically potted as a function of the sweep rates (v) to give the slopes (b) for (d) FeS₂/rGO, (e) FeS₂/N-rGO-0.5, and (f) FeS₂/N-rGO-2.

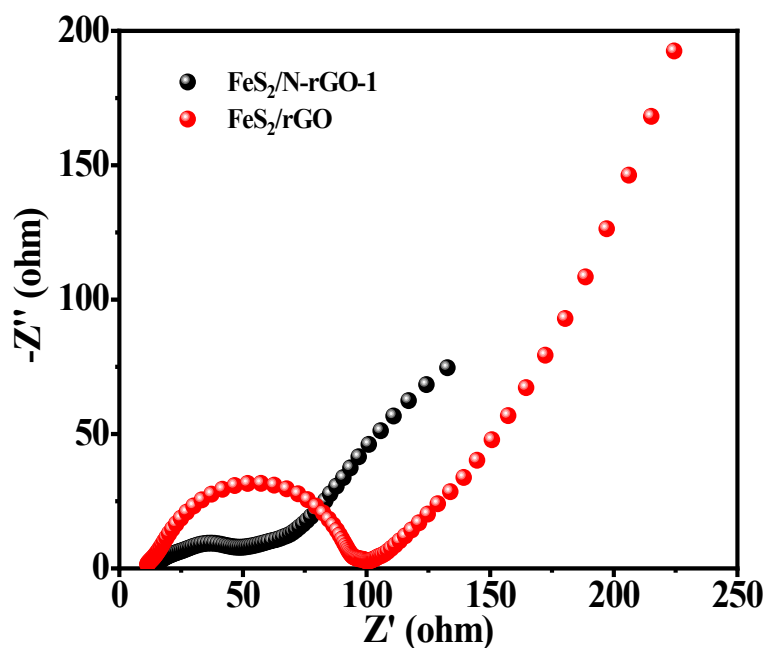


Fig. S12 Electrochemical impedance spectroscopy (EIS) spectra at the first cycle in the frequency ranging from 100 kHz to 0.01 Hz of FeS₂/rGO and FeS₂/N-rGO-1.

Table S1. Comparison of different FeS₂-based materials for LIBs.

Materials	Current density (mA g ⁻¹)	Reversible Capacity (mAh g ⁻¹)/cycle	Current density (A g ⁻¹)/Rate Capability (mAh g ⁻¹)	References
FeS ₂ /N-rGO-1 (N-doped microsphere)	150	950 / 140	5.0 / 510	This work
FeS ₂ /rGO microparticle	100	1001.41 / 60	1.0 / 500	[1]
Resilient carbon encapsulated FeS ₂	89.4	110 / 50	-	[2]
Al ₂ O ₃ -coated FeS ₂ @carbon fiber	200	530 / 100	0.2 / 275	[3]
FeS ₂ @N-graphene	500	401.7 / 400	5.0 / 285	[4]
FeS ₂ /C nanotubes arrays	223.5	689 / 100	1.34 / 482	[5]
FeS ₂ @3-D graphene foam	179	1080.3 / 100	5.0 / ~350	[6]
FeS ₂ @C nanospheres	300	614 / 100	5.0 / ~450	[7]
FeS ₂ /CNT hybrids	200	800 / 200	5.0 / 345	[8]
Biomass-carbon@FeS ₂	445	850 / 80	4.47 / 725	[9]
FeS ₂ micro/nano-structured	100	180.9 / 350	-	[10]

References:

- 1 X. Wen, X. L. Wei, L. W. Yang and P. K. Shen, *J. Mater. Chem. A*, 2015, **3**, 2090-2096.
- 2 T. S. Yoder, M. Tussing, J. E. Cloud and Y. A. Yang, *J. Power Sources*, 2015, **274**, 685-692.
- 3 Y. J. Zhu, X. L. Fan, L. M. Suo, C. Luo, T. Gao and C. S. Wang, *ACS Nano*, 2016, **10**, 1529-1538.
- 4 R. Tan, J. L. Yang, J. T. Hu, K. Wang, Y. Zhao and F. Pan, *Chem. Commun.*, 2016, **52**, 986-989.
- 5 G. X. Pan, F. Cao, X. H. Xia and Y. J. Zhang, *J. Power Sources*, 2016, **332**, 383-388.
- 6 J. R. He, Q. Li, Y. F. Chen, C. Xu, K. R. Zhou, X. Q. Wang, W. L. Zhang and Y. R. Li, *Carbon*, 2017, **114**, 111-116.
- 7 X. J. Xu, J. Liu, Z. B. Liu, J. D. Shen, R. Z. Hu, J. W. Liu, L. Z. Ouyang, L. Zhang and M. Zhu, *ACS Nano*, 2017, **11**, 9033-9040.
- 8 L. Xu, Y. J. Hu, H. X. Zhang, H. Jiang and C. Z. Li, *Acs Sustainable Chem. Eng.*, 2016, **4**, 4251-4255.
- 9 X. Xu, Z. Meng, X. L. Zhu, S. L. Zhang and W. Q. Han, *J. Power Sources*, 2018, **380**, 12-17.
- 10 W. Q. Ma, X. Z. Liu, X. F. Lei, Z. H. Yuan and Y. Ding, *Chem. Eng. J.*, 2018, **334**, 725-731.