Supplementary Information for

Porous N-doped Hard-Carbon Derived from Waste Separators for Efficient Lithium-Ion and Sodium-Ion Batteries

Yong Wang,^a Yong Li,^{b,a} Samuel S. Mao,^d Daixin Ye,^{e*} Wen Liu,^a Rui Guo,^a Zhenhe Feng,^a Jilie Kong,^b Jingying Xie^{a,c*}



Supplementary Figure 1: TG curve of the precursor.



Supplementary Figure 2: SEM images of HC-800 (a); HC-1000 (b); HC-1100 (c) and HC-1300(d)

Sample	Surface Area (m ² g ⁻¹)	Pore Volume (cm ³ g ⁻¹)
HC-800	405.4966	0.195817
HC-900	398.3485	0.193779
HC-1000	213.2200	0.130613
HC-1100	8.1924	0.057399
HC-1300	1.3459	0.015570
Commercial HC	0.1762	0.002274

$$\label{eq:supplementary} \begin{split} \text{Supplementary Table 1: BET surface area and pore volume of the} \\ \text{samples tested by N_2 adsorption/desorption} \end{split}$$



Supplementary Figure 3: TEM images of HC-800 (a); HC-1000 (b); HC-1100 (c) and HC-1300(d)



Supplementary Figure 4: N 1s spectra of HC-800 (a); HC-1000 (b); HC-1100 (c) and HC-1300(d)



Supplementary Figure 5: N 1s spectrum of commercial hard carbon.



Supplementary Figure 6: Charge-discharge curves of first 10 cycles of LiNi $_{1/3}Co_{1/3}Mn_{1/3}O_2//Li$ half cell



Supplementary Figure 7: Charge-discharge curves of (a) first 2 cycles of Na_{2- δ}MnHCF//Na half cell and (b) first 10 cycles of Na_{2- δ}MnHCF//HC-900 full cell.