

Supporting Information for

Photoirradiation generated radicals in two-component supramolecular gel for polymerization

Hejin Jiang^{a,b,†}, Qingxian Jin^{c,†}, Jing Li^c, Shuyu Chen^c, Li Zhang^{*a}, Minghua Liu^{*a,b,d,e}

^aBeijing National Laboratory for Molecular Science, CAS Key Laboratory of Colloid, Interface and Chemical Thermodynamics, Institute of Chemistry, Chinese Academy of Sciences, Zhongguancun North First Street 2, 100190, Beijing, China. E-mail: zhangli@iccas.ac.cn, liumh@iccas.ac.cn.

^bUniversity of Chinese Academy of Sciences, Beijing 100049, China.

^cHenan Provincial Key Laboratory of Surface & Interface Science, Zhengzhou University of Light Industry, Zhengzhou, Henan 450002, China.

^dNational Center for Nanoscience and Technology, Beijing, 100190, China.

^eA Collaborative Innovation Center of Chemical Science and Engineering, Tianjin 300072, China.

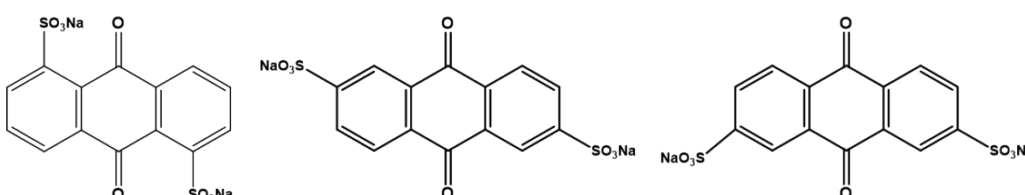
				
	AQ15	AQ26	AQ27	
PUDG:AQ= 2: 1 H₂O: EtOH (1ML)	AQ15	AQ26	AQ27	
5:5	white suspension	gel 	gel 	
6:4	white suspension	gel 	gel 	

Fig. S1 Chemical structures of AQ15, AQ26, AQ27 and the gelation situation of PUDG with AQ15, AQ26, AQ27 (molar ratio 2:1) at different ethanol/H₂O mixing solvents (5:5, 4:6).

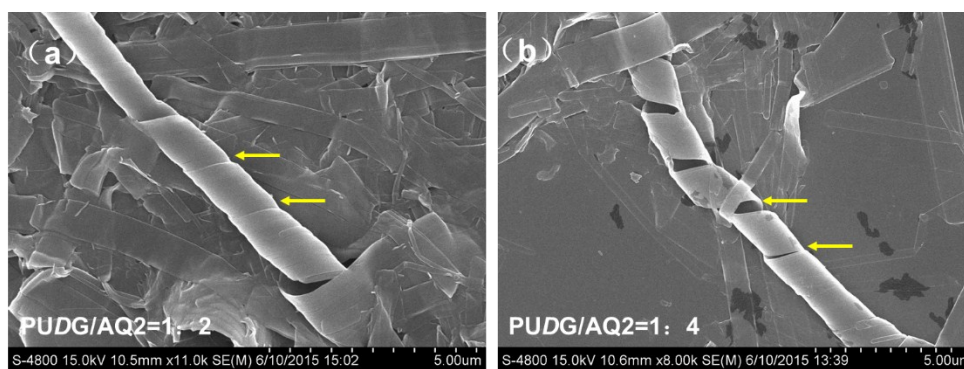


Fig. S2 (a) SEM image of PUDG/AQ2=1:2 and PUDG/AQ2=1:4 in EtOH/H₂O (5:5) mixed solution.

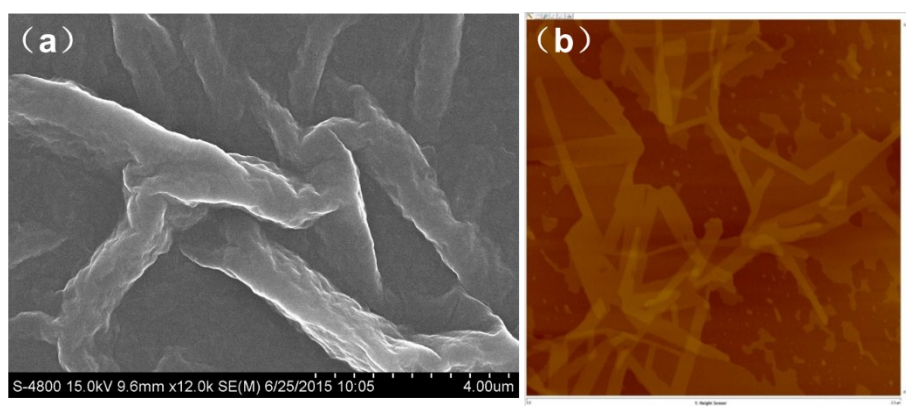


Fig. S3 (a) SEM image of PUDG/AQ1; (b) AFM image of PUDG/AQ1 in EtOH/H₂O (5:5) mixed solution.

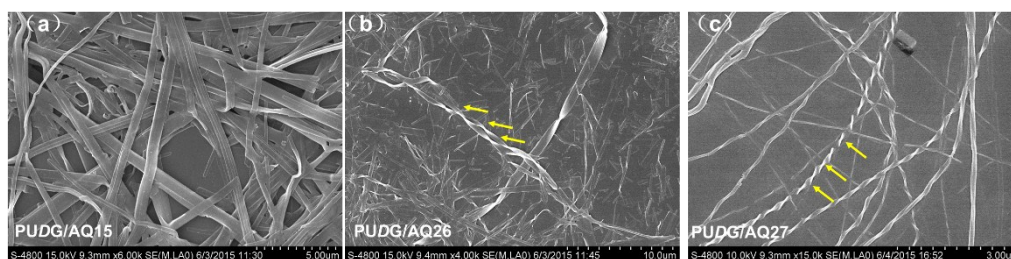
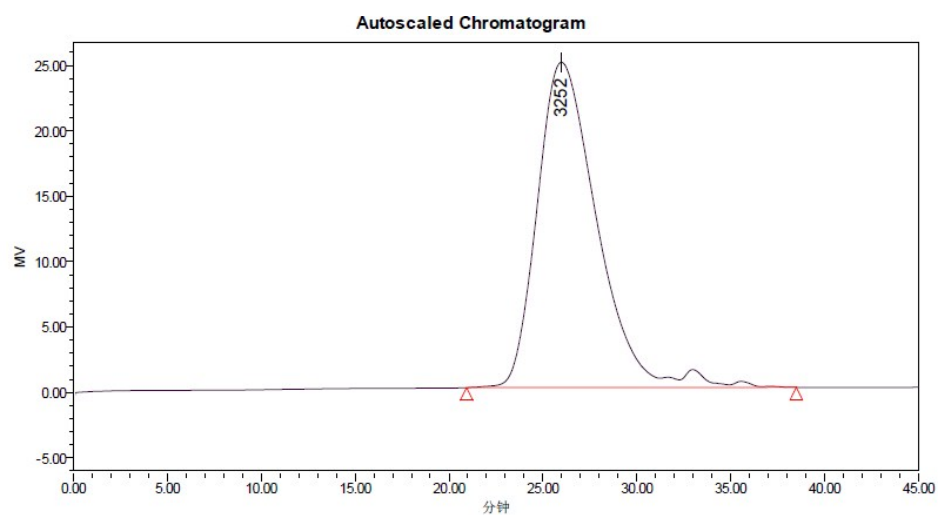


Fig. S4 (a) SEM image of PUDG/AQ15; (b) PUDG/AQ26 and (c) PUDG/AQ27 in EtOH/H₂O (5:5) mixed solution.



GPC Results

	Dist Name	Elution Volume (ml)	Retention Time (min)	Adjusted RT (min)	Mn	Mw	MP	Mz	Mz+1	Mz/Mw	Mz+1/Mw
1		25.990	25.990	25.990	1977	3534	3252	5893	9790	1.667198	2.769912
2			31.381								
3			32.900								

Fig. S5 GPC (Gel Permeation Chromatography) result of as-prepared PAA.