

Supporting Information

Construction of a stable lithium sulfide membrane to greatly confine polysulfides for high performance lithium–sulfur batteries

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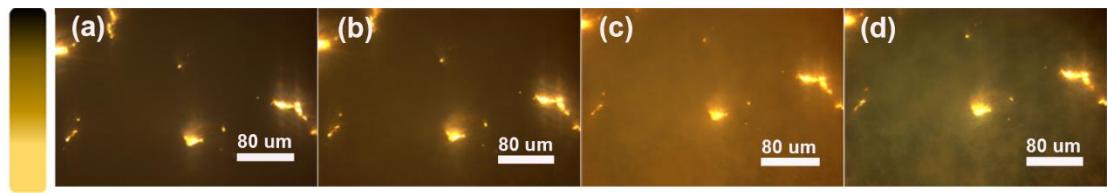


Fig. S1 The in-situ optical microscopy images of the different stages in an electrochemical charge/discharge process at 0.1 mV s⁻¹ scan rate and 1.7-2.7 V voltage window. (a), A initial state. (b), discharge to 2.2 V. (c), discharge to 1.7 V and (d), re-charge to 2.7 V. (The shape guide shows the color variation of cathode surface from black to yellow).

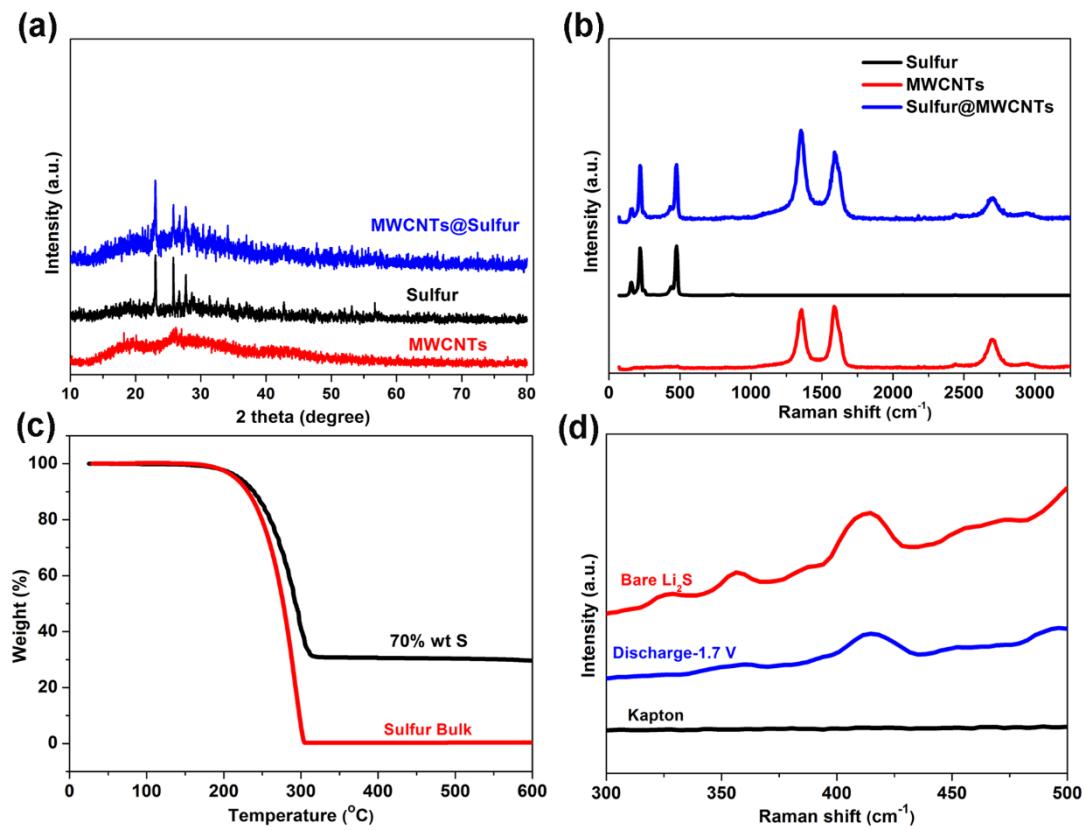


Fig. S2 (a) XRD and (b) Raman spectrum includes MWCNTs@sulfur, sulfur and MWCNTs. (c) TGA curve shows 70% wt. sulfur in composites.(d) Raman spectrum of commercial Li_2S refer to electrode (discharge to 1.7 V), all of them had been protected by kapton.

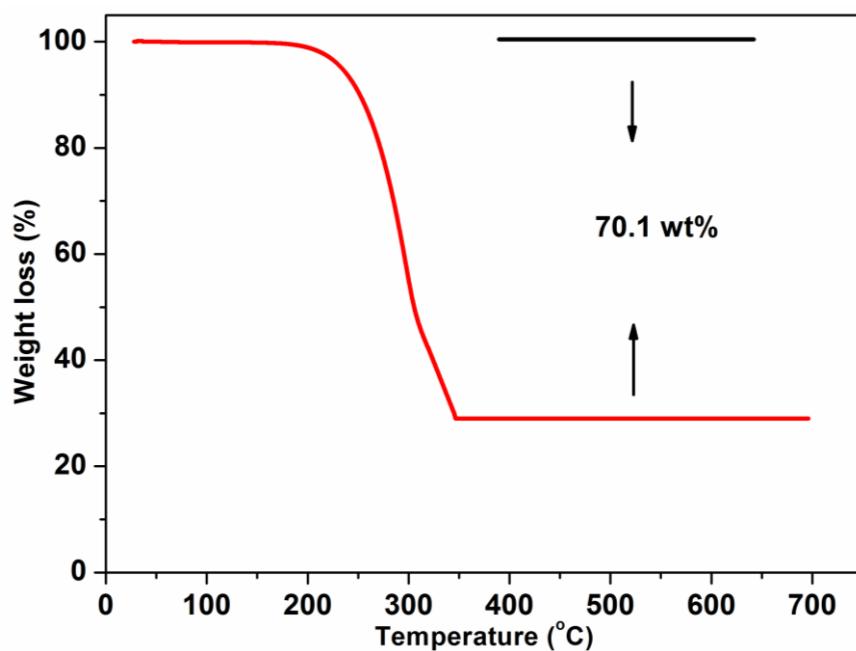


Fig. S3 TGA curve shows 70.1% wt. sulfur in KCNTs.

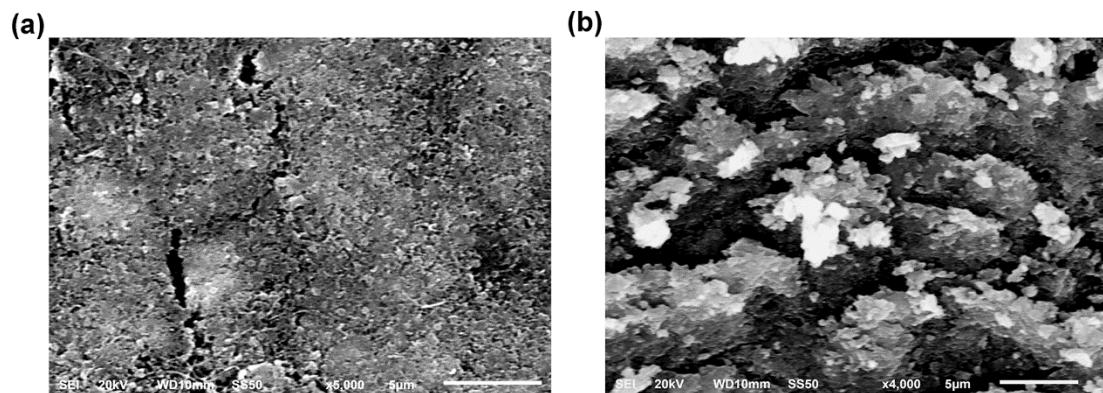


Fig.S4 (a) the SEM image of KCNTs and (b) Ketjen black and super P with sulfur as the electrode materials after the cycling.

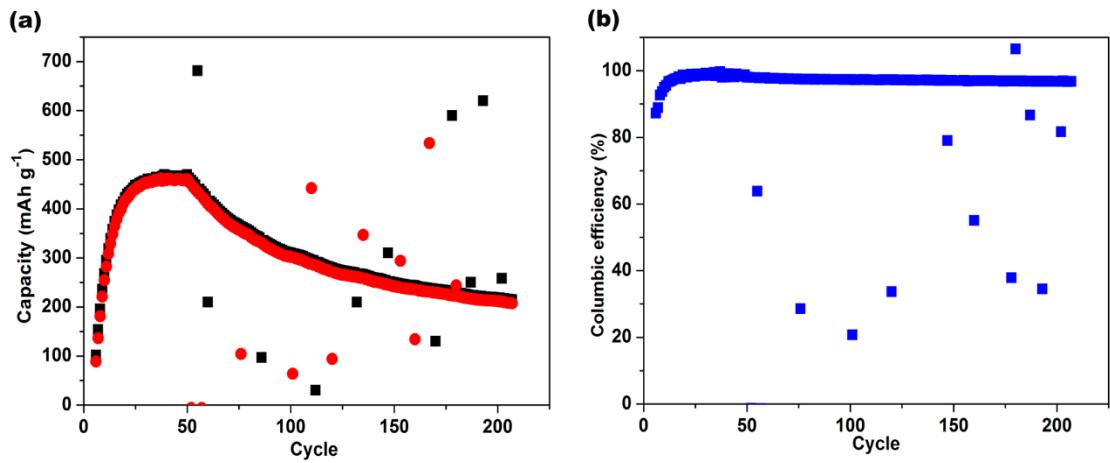


Fig.S5 The KCNTs@S electrode without LS membrane has cycling performance (a) and columbic efficiency (b) at 1C

current density.

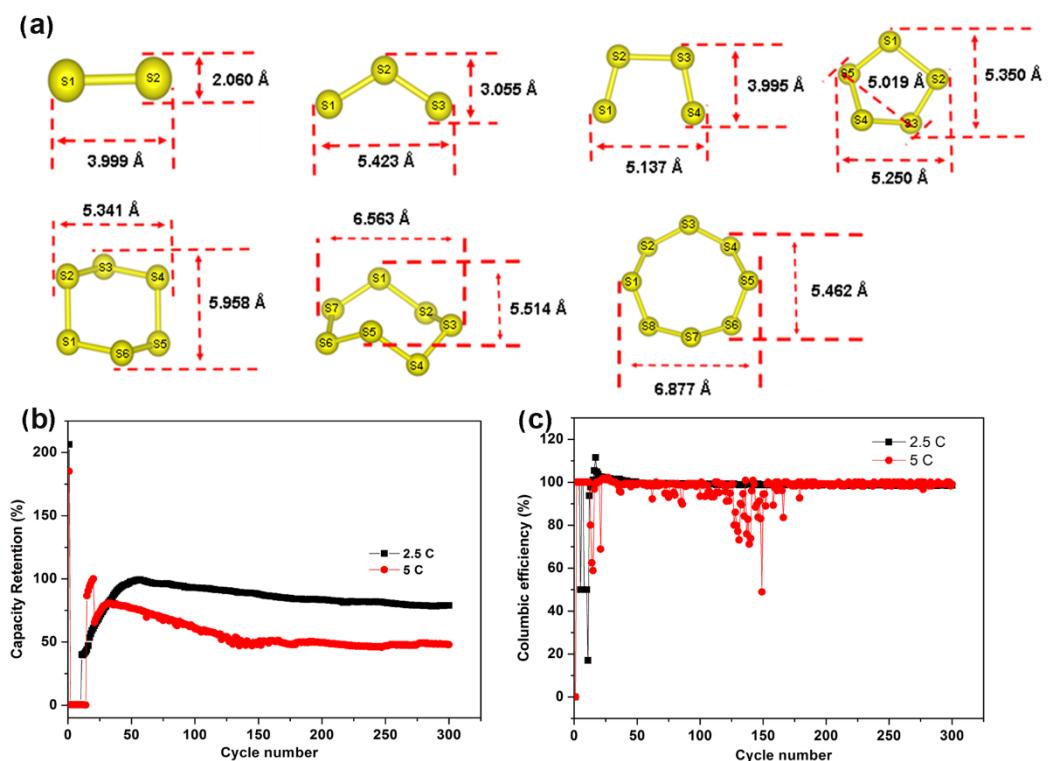


Fig. S6 (a) the theoretical calculation of various polysulfide intermediates space size. (b) The cycling performance of LS membrane improved cathode (70% wt. S) at 2.5 C and 5 C. (c) columbic efficiency of LS membrane improved cathode (70% wt. S) at 2.5 C and 5 C.

Table 1 | Summary the performance of MWCNTs/Sulfur as cathode for Li–S battery systems.

Approach	Voltage window (V)	Rate performance	Sulphur loading in electrode	Total cycle number	Degradation rate per cycle	Reference
nano-sulfur/MWCNTs composite	1.5~3	N/A	48%	30	1.1% (300)	1
S-coated-MWCNTs cathode	1.5~2.5	N/A	68%	60	N/A	2
polyacrylonitrile-sulfur@MWCNT composite	1~3	450 (4 C)	48%	50	0.3% (C/10)	3
Dual core–shell structured sulfur cathode composite	1.5~2.8	665 (1.2 C)	47.8%	200	0.177% (1500)	4
hierarchical S/MWCNT nanomicrosphere	1.7~2.7	600 (1 C)	45.6%	200	0.27% (1000)	5

Self-weaving sulfur–carbon composite	1.5~2.8	1012 (4 C)	40%	100	0.323% (1 C)	6
Aligned carbon nanotube/sulfur composite	1.6~3	280 (1 C)	76.5%	80	~0.43% (0.1 C)	7
Graphene/CNT @Porous Carbon	1.5~3	970 & 617 (5 C)	45% & 69.3%	150	0.12% & 0.18%(1 C)	8
Nitrogen-Doped Aligned Carbon Nanotube/Graphene	1.6~3	770 (5 C)	44.7%	80	0.3% (1 C)	9
Sulfur Nanocrystals Confined in Carbon Nanotube Network	1.9~2.6	934 (5 C)	50%	100	0.164% (1 C)	10
This work	1.7~2.7	560 (5 C)	56%	210	0.031% (1 C)	

Table 2 | Characteristic of various carbon-sulfur composites with high sulfur loading on the electrode (>60 wt%)

Approach	Sulfur loading density (mg cm ⁻²)	LiNO ₃ additive	Cycling performance (mAh g ⁻¹)	Capacity retention rate	Reference
Electrocatalysis of polysulfide conversion by sulfur-deficient MoS ₂ nanoflakes for lithium–sulfur batteries	0.9	2 wt%	1159.9~628 (0.5 C, 600 cycles) 1159.9~819.9 (0.5 C, 150 cycles)	54% 70%	11
A Sulfur-Rich Copolymer@CNT Hybrid Cathode with	1.2~1.58	0.2 m	818~647 (1 C,~450 cycles)	79%	12

Dual-Confinement of Polysulfides					
Graphitized porous carbon materials with high sulfur loading for lithium-sulfur batteries	4	0.5 m	908~739 (0.1C, 100 cycles)	81%	13
Sole Chemical Confinement of Polysulfides on Nonporous Nitrogen/Oxygen Dual-Doped Carbon at the Kilogram Scale for Lithium-Sulfur Batteries	1.2~1.5	2 wt%	587~529 (1C ,180 cycles)	90%	14
This work	3	2 wt%	756~685 (1 C, 330 cycles)	90%	

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