

Supporting Information

Investigation of the exceptional charge performances of $\text{Li}_{4-x}\text{Mn}_2\text{O}_5$ as Li-ion Battery electrode material

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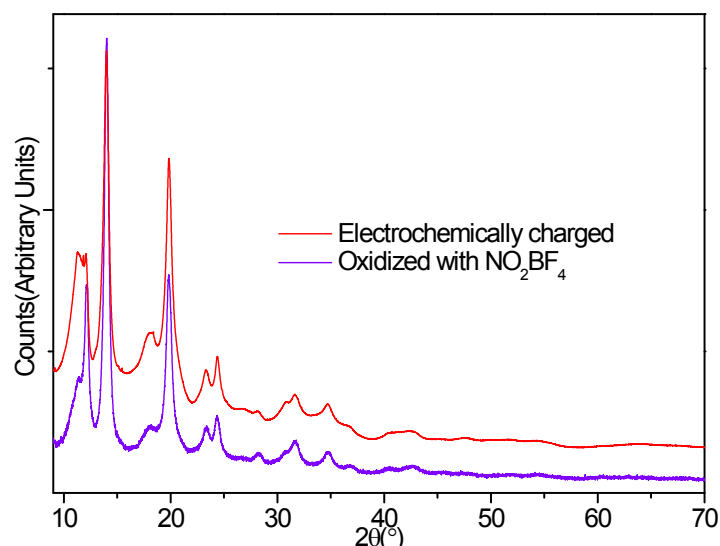


Figure SI-1. X-ray diffraction data for electrochemically charged (red) and chemically oxidized (violet) $\text{Li}_{4-x}\text{Mn}_2\text{O}_5$.

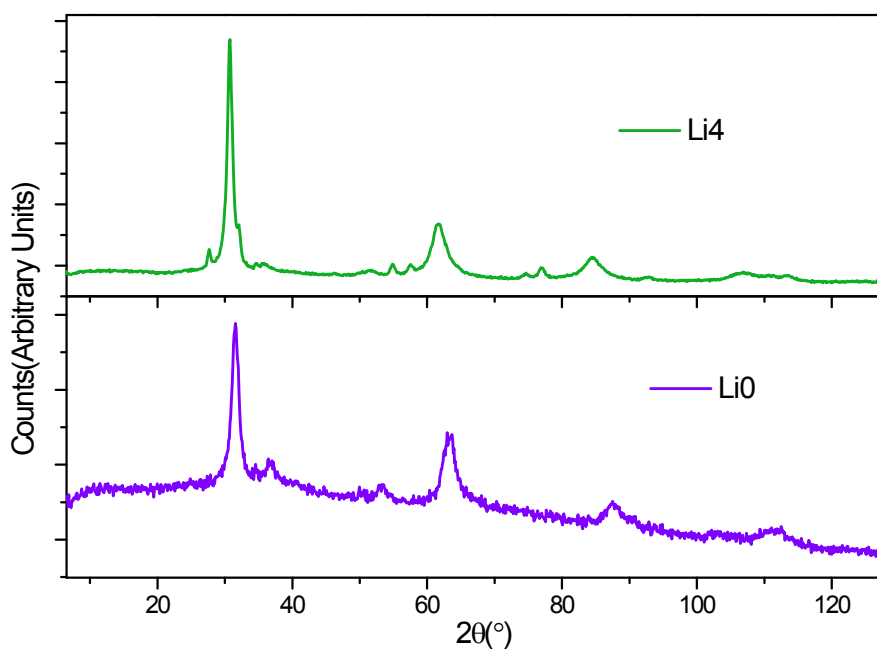


Figure SI-2. Acquired NPD patterns. The 2θ region of the Rietveld refinements was reduced to the range given in Figure 5 of the main text where peaks are observed.

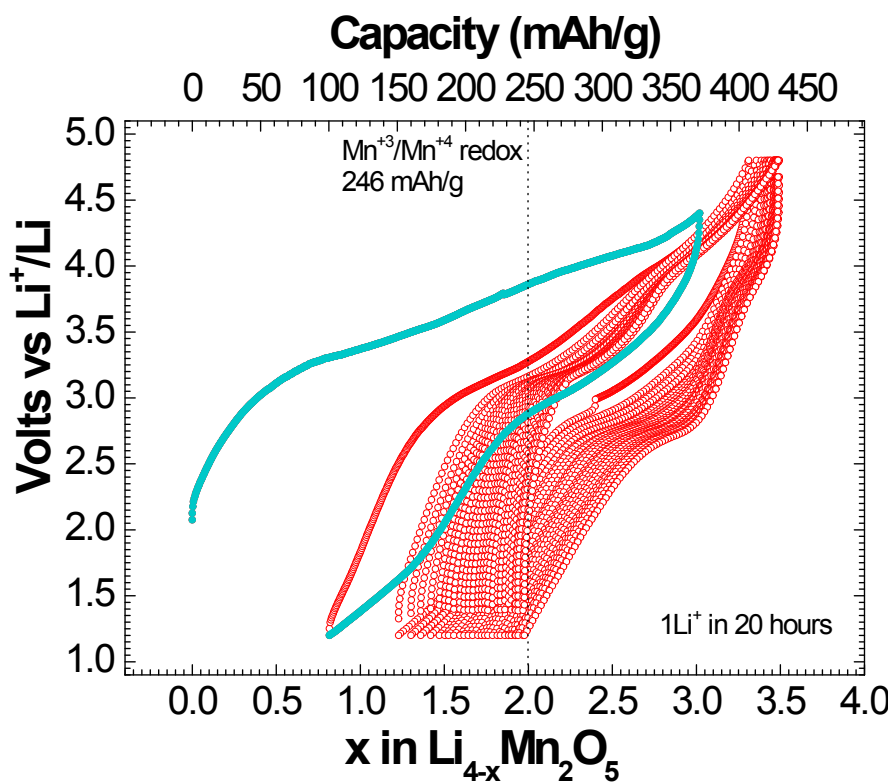


Figure SI-3. Typical voltage-composition profile of $\text{Li}_4\text{Mn}_2\text{O}_5$ obtained at C/80, in coin cell.

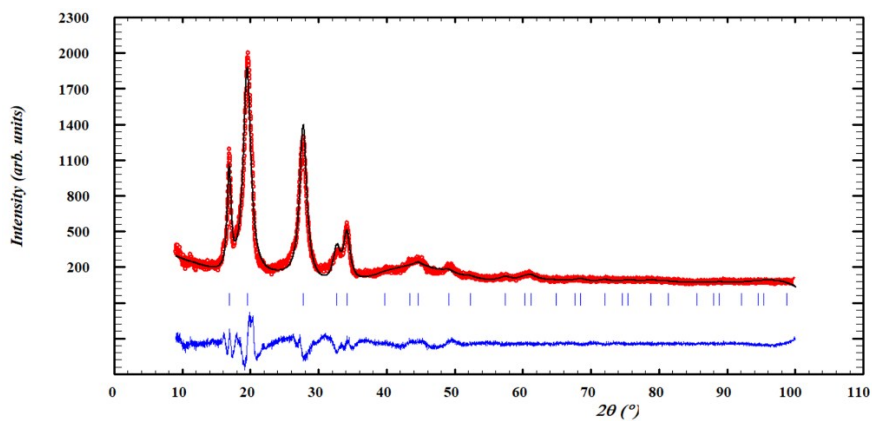


Figure SI-4. Le Bail profile fitting of the X-ray diffraction pattern of $\text{Li}_{3.6}\text{Mn}_{2.4}\text{O}_{5.4}$ ($a = 4.17 \text{ \AA}$) without any trace of Li_2O . The data was collected with a Bruker D8 Advance diffractometer equipped with a Mo source $K_{\alpha,1} = 0.7093 \text{ \AA}$.

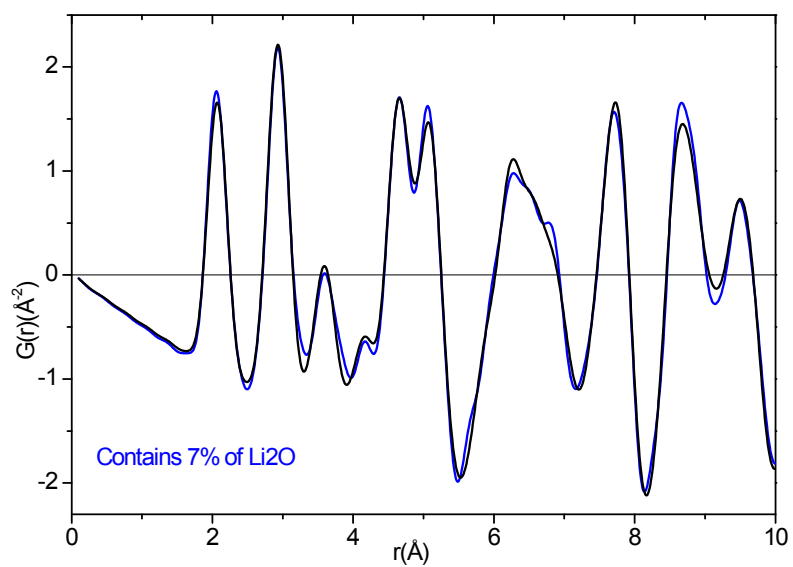


Figure SI-5. PDF simulation of a main Li_4 phase with an MnO structure with (blue) and without (black) 7 wt% of Li_2O .

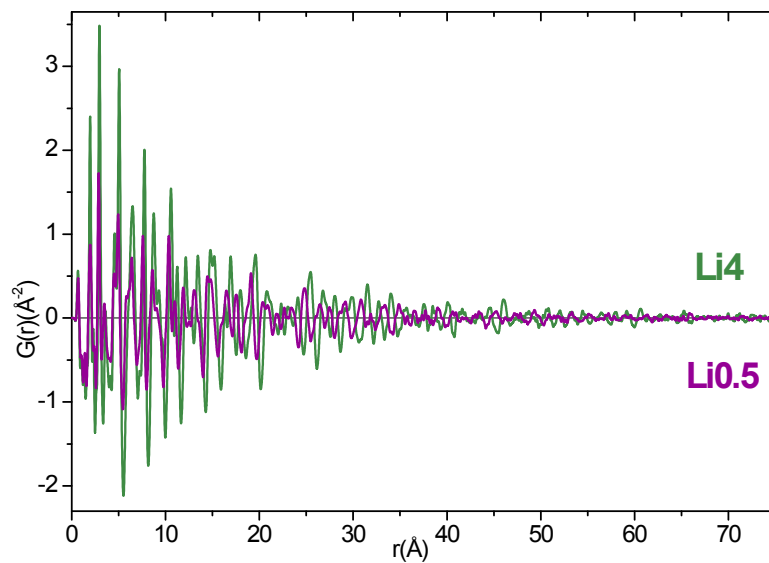


Figure SI-6. PDF data showing a coherent size domain of around 6 nm.

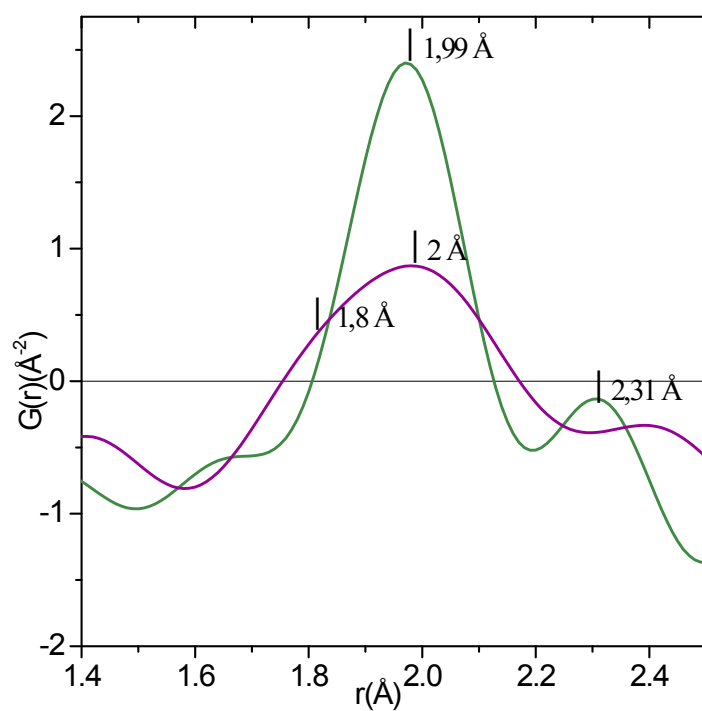


Figure SI-6. First PDF peaks for Li4 (green) and Li0.