

Supporting information

Uniformly Core-shell Nanobiscuits of Fe₇S₈@C for Lithium-ion and Sodium-ion Batteries with Excellent Performance

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Table S1. A comparison of electrochemical properties of metal sulfides for LIBs and SIBs.

Types of materials	Types of battery	Voltage range(V)	Cycling performance	Reference
CoS ₂ Nanobubble Hollow Prisms	LIBs	0.05-3	737 mAh g ⁻¹ after 200 cycles at 1 A g ⁻¹	1
CoS/C	SIBs	0.01-3	542 mAh g ⁻¹ after 2000 cycles at 1 A g ⁻¹	2
Fe ₇ S ₈ @C NCs	SIBs	0.08-3	447 mAh g ⁻¹ after 1000 cycles at	3

			180 mA g ⁻¹	
Hollow nanocubes PBC1-1S	SIBs	0.01-3	87 mAh g ⁻¹ after 150 cycles at 500 mA g ⁻¹	4
NiS nanoplates	LIB	0.005-3	468 mAh g ⁻¹ after 100 cycles at 1 A g ⁻¹	5
	SIBs	0.005-3	166 mAh g ⁻¹ after 100 cycles at 1 A g ⁻¹	
CoS ₂ /NCNTF electrode	LIBs	0.05-3	973 mAh g ⁻¹ after 160 cycles at 1A g ⁻¹	6
Fe ₃ O ₄ /Fe _{1-x} S@C@MoS ₂ nanosheets	LIBs	0.01-3	1142 mAh g ⁻¹ after 700 cycles at 1A g ⁻¹	7
	SIBs	0.01-3	402 mAh g ⁻¹ after 1000 cycles at 1A g ⁻¹	
FeS ₂ @C yolk-shell nanoboxes	SIBs	0.1-2	330 mAh g ⁻¹ after 800 cycles at 2A g ⁻¹	8
Mesoporous NiS ₂ Nanospheres	SIBs	0.01-2.9	319 mAh g ⁻¹ after 1000 cycles at 500 mA g ⁻¹	9
MoS ₂ /SnS ₂ -GS	LIBs	0.01-3	772 mAh g ⁻¹ after 200cycles at 750 mA g ⁻¹	10
	SIBs	0.01-3	546 mAh g ⁻¹ after 150 cycles at 1.5 A g ⁻¹	
TiO ₂ @NC@MoS ₂	LIBs	0.01-3	590 mAh g ⁻¹ after 200 cycles at 1 A g ⁻¹	11
EG-MoS ₂	LIBs	0.01-3	1250 mAh g ⁻¹ after 150 cycles at 1 A g ⁻¹	12
	SIBs	0.01-3	509 mAh g ⁻¹ after 250 cycles at 1 A g ⁻¹	
spongy CoS ₂ /C	LIBs	0.01-3	610 mAh g ⁻¹ after 120 cycles at 500 mA g ⁻¹	13

	SIBs	0.01-3	358.7 mAh g ⁻¹ after 60 cycles at 500 mA g ⁻¹	
SnS@C nanotubes	SIBs	0.01-3	440.3 mAh g ⁻¹ after 100 cycles at 200 mA g ⁻¹	14
ce-V ₅ S ₈ -C	SIBs	0.01-3	496 mAh g ⁻¹ after 500 cycles at 1 A g ⁻¹	15
Fe _{1-x} S@CNTs	SIBs	0.01-2.3	449.2 mAh g ⁻¹ after 200 cycles at 500 mA g ⁻¹	16
Fe ₇ S ₈ @C NCs	SIBs	0.08-3	447 mAh g ⁻¹ after 1000 cycles at 180 mA g ⁻¹	17
FeS@CNS	LIBs	0.01-3	703 mAh g ⁻¹ after 150 cycles at 1 A g ⁻¹	18
G@FeS-GNRs	LIBs	0.01-3	536 mAh g ⁻¹ after 100 cycles at 400 mA g ⁻¹	19
FeS ₂ /CNT	SIBs	0.8-3	394 mAh g ⁻¹ after 400 cycles at 200 mA g ⁻¹ ; 309 mAh g ⁻¹ after 1800 cycles at 1 A g ⁻¹	20
Fe₇S₈@C	LIBs	0.01-3	781 mAh g⁻¹ after 500 cycles at 1 A g ⁻¹ , 547.3 mAh g⁻¹ after 600 cycles at 5 A g ⁻¹	This work
	SIBs	0.01-3	596 mAh g⁻¹ after 500 cycles at 1 A g ⁻¹ , 530.8 mAh g⁻¹ after 1000 cycles at 5 A g ⁻¹	This work

Table S2. Values of R_{ct} obtained by fitting date.

	Li ⁺ /Rct(Ω)	Na ⁺ /Rct(Ω)
Before cycle	52.25	10.05
After cycle	18.62	7.645

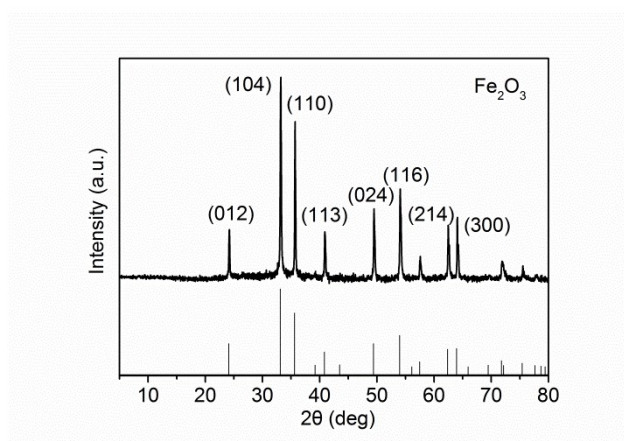


Fig. S1. XRD spectra of the pure Fe₂O₃.

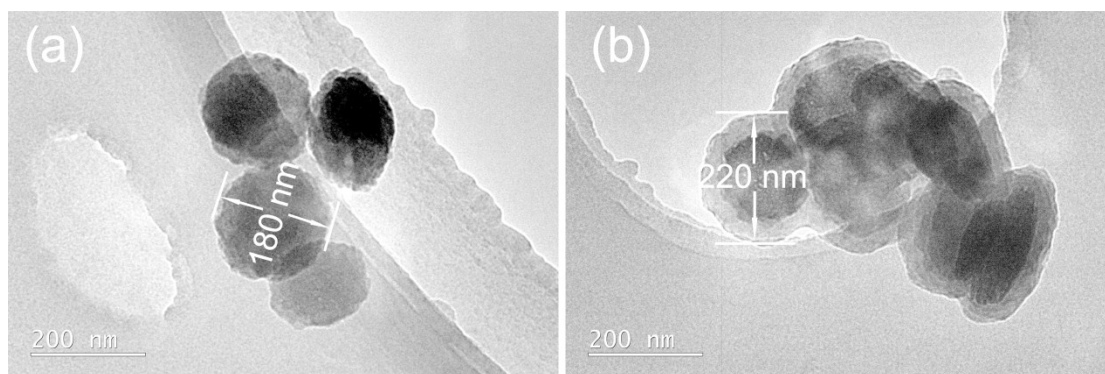


Fig. S2. TEM images of (a) Fe₂O₃ and (b) Fe₂O₃@PDA.

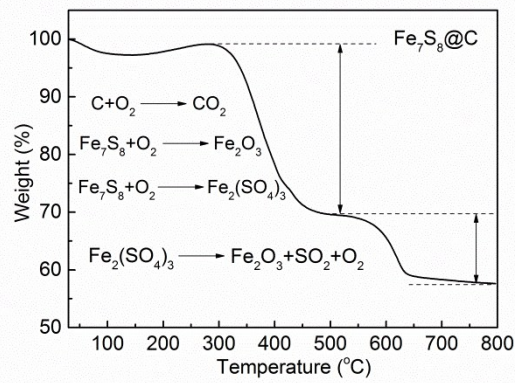


Fig. S3. TGA curves of Fe₇S₈@C.

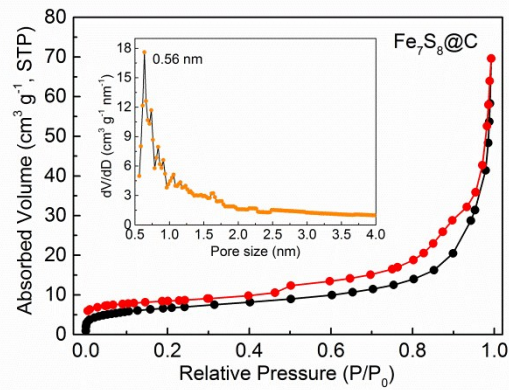


Fig. S4. The nitrogen adsorption/desorption isotherm plots. Micropore (inset) size distribution of Fe₇S₈@C.

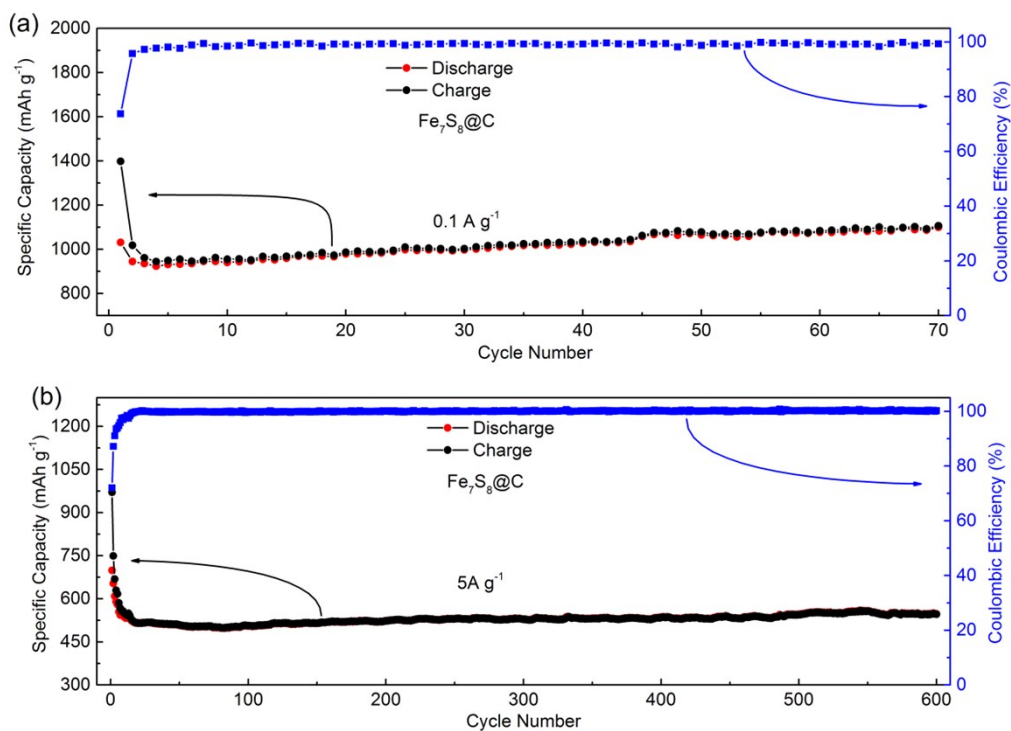


Fig. S5. Long-term cyclic performance at current densities of 0.1 A g^{-1} (a) and 5 A g^{-1} (b) for LIBs.

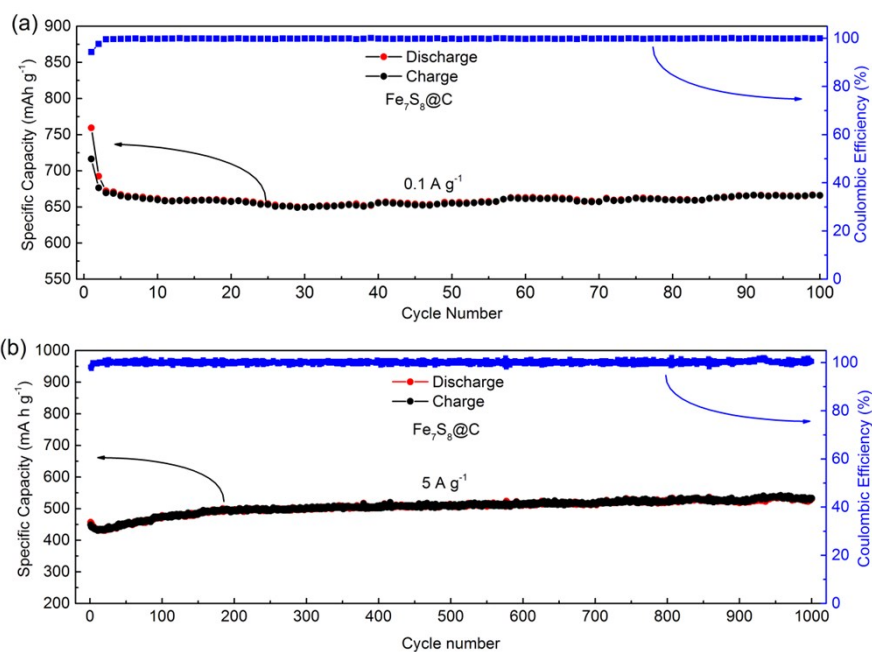


Fig. S6. Long-term cyclic performance at current densities of 0.1 A g^{-1} (a) and 5 A g^{-1} (b) for SIBs.

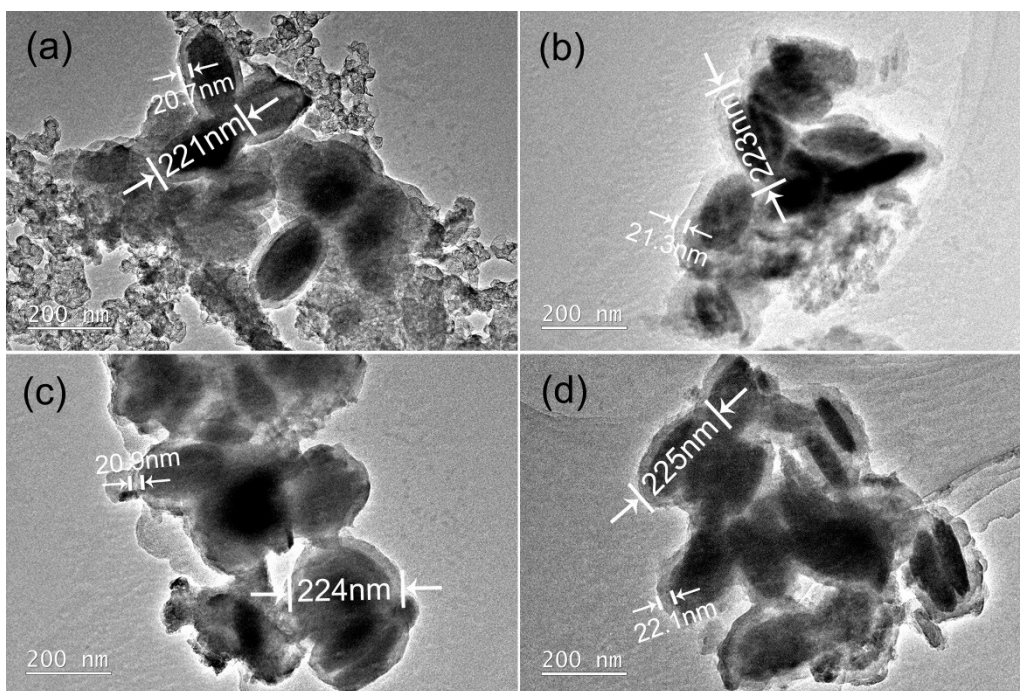


Fig. S7. TEM images of $\text{Fe}_7\text{S}_8@\text{C}$ nanobiscuits electrodes after 1st cycle (a, c) and 500th cycle (b, d) at the current density of 1 A g^{-1} for (a, b) LIBs and (c, d) SIBs.

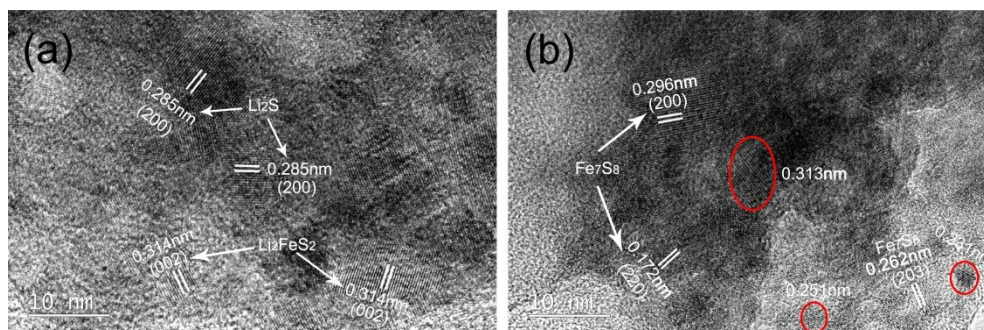


Fig. S8. Ex-situ HRTEM for LIBs (a) and SIBs (b) after first cycle, both batteries charge to 2 V.

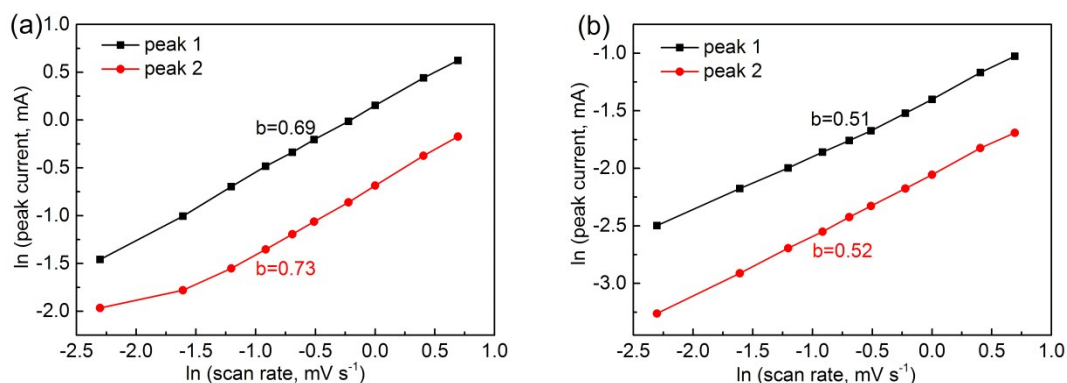


Fig. S9. The fitted lines and \ln (peak current) versus \ln (scan rate) plots at different oxidation (black) and reduction (red) state of (a) LIBs and (b) SIBs.

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