

## Electronic supplementary information

### **Three-dimensional self-assembled SnS<sub>2</sub>-Nano-Dots@Graphene Hybrid Aerogel as an Efficient Polysulfide Reservoir for High-performance Lithium-sulfur Batteries**

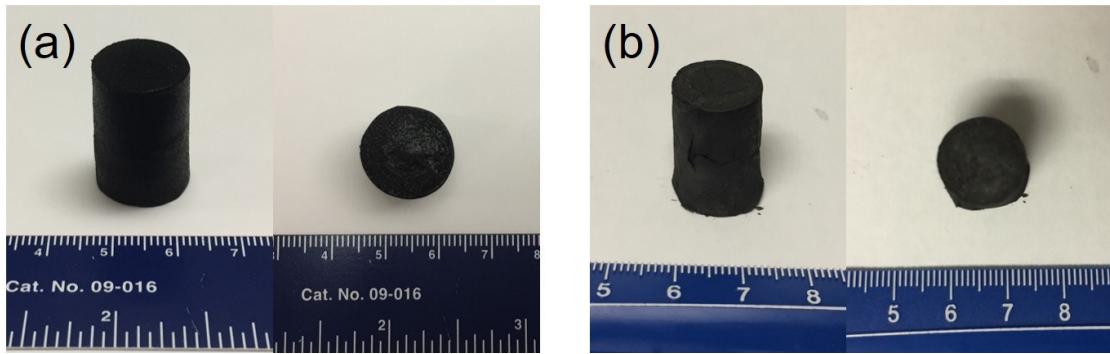
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McKetta Department of Chemical Engineering & Texas Materials Institute

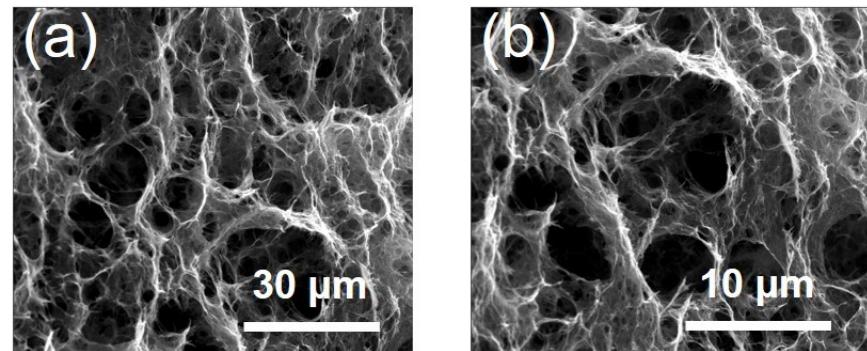
The University of Texas at Austin, Austin, TX 78712, USA

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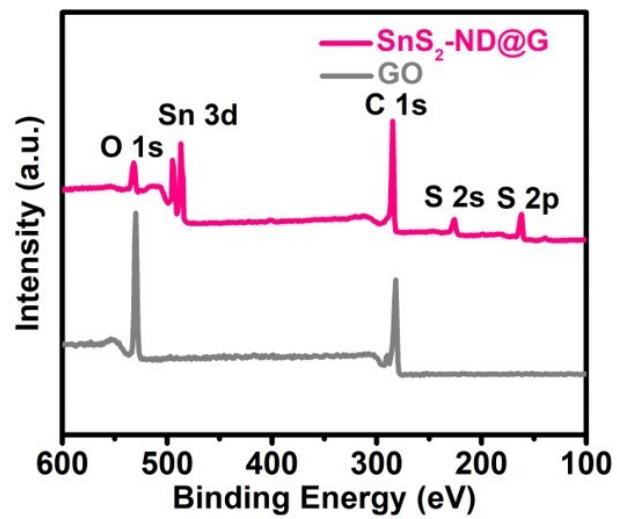
## Supporting figures



**Fig. S1** Digital photos of the as-prepared (a) SnS<sub>2</sub>-ND@G and (b) graphene aerogels with front and top views.



**Fig. S2** SEM images of graphene aerogel under (a) high and (b) low magnifications.



**Fig. S3** XPS survey spectra of GO (gray line) and  $\text{SnS}_2\text{-ND@G}$  aerogels (pink line).



**Fig. S4** Digital picture of the cycled lithium metal anode retrieved from the cell employing the  $\text{SnS}_2\text{-ND@G}$  cathode.

**Table S1** Comparative summary of the Li-S cell parameters of the cell parameters in this work and other reported studies involving metal-sulfide/carbon and other metal compounds as sulfur cathode materials

Cathode material	Preparation method	Metal sulfide content in the composite (wt.%)	Sulfur content in the cathode (wt.%)	Sulfur loading (mg cm <sup>-2</sup> )	Peak capacity (mA h g <sup>-1</sup> )	Remaining capacity (mA h g <sup>-1</sup> )	Peak areal capacity (mA h cm <sup>-2</sup> )	Cycle number	Capacity retention (%)	Capacity-fate rate (% per cycle)	E/S ratio	Condition	Reference
MoS <sub>2</sub> + rGO	Physical mixing	22	60	2 – 4	1200	628	3.6	600	52	0.083	13 – 25	1.8 – 2.6 V (C/2)	45
CoS <sub>2</sub> +G	Physical mixing	15 – 30	60	0.4 – 2.9	1368	750	3.2	250	55	0.18	4 – 30	1.7 – 2.8 V (C/2)	18
VS <sub>2</sub> @G/CNT	Ball-milling	50	56	0.5 – 0.6	830	701	0.5	300	84	0.052	33 – 40	1.5 – 2.8 V (C/2)	23
CoS <sub>2</sub> @G/CNT	Ball-milling	50	56	0.5 – 0.6	684	581	0.41	300	85	0.049	33 – 40	1.5 – 2.8 V (C/2)	23
TiS <sub>2</sub> @G/CNT	Ball-milling	50	56	0.5 – 0.6	700	546	0.42	300	78	0.073	33 – 40	1.5 – 2.8 V (C/2)	23
FeS@G/CNT	Ball-milling	50	56	0.5 – 0.6	705	334	0.42	300	47.4	0.175	33 – 40	1.5 – 2.8 V (C/2)	23
SnS <sub>2</sub> @G/CNT	Ball-milling	50	56	0.5 – 0.6	610	191	0.37	300	31.3	0.23	33 – 40	1.5 – 2.8 V (C/2)	23
Ni <sub>3</sub> S <sub>2</sub> @G/CNT	Ball-milling	50	56	0.5 – 0.6	526	153	0.32	300	29.1	0.24	33 – 40	1.5 – 2.8 V (C/2)	23
CuS + carbon aerogel	Physical mixing	--	48	1.3 – 3.0	1137	840	2.4	500	74	0.05	--	1.8 – 3.0 V (C/2)	21
TiO <sub>2</sub> + G	Hydrothermal	41	52	1.1 – 1.3	1065	663	1.3	100	62	0.38	--	1.7 – 2.8 V (C/10)	8
Co <sub>3</sub> S <sub>4</sub> + S	Hydrothermal	21 – 34	53	2 – 4	975	760	2.1	200	78	0.11	10 – 20	1.6 – 2.6 V (C/2)	39
TiS <sub>2</sub> + S	Physical mixing	45	45	--	875	810	--	30	93	0.25	--	1.7 – 2.8 V (C/10)	20

$\text{Co}_3\text{S}_8 + \text{S}$	Melt-diffusion	25	60	1.5	857	643	4.3	400	75	0.06	--	1.8 – 3.0 V (2C)	46
$\text{TiO} + \text{carbon hollow sphere}$	Coating and annealing	57	56	1.5	1190	750	3.5	500	63	0.08	36	1.9 – 2.6 V (C/5)	13
$\text{MgO} + \text{S}$	Ball-milling	10	54	1.8 – 2.0	860	700	1.6	100	81	0.19	--	1.5 – 3.0 V (C/5)	16
$\text{MnO}_2 + \text{S}$	Physical mixing	25	56	0.7 – 1.0	1120	1030	1.12	200	92	0.04	--	1.8 – 3.0 V (C/5)	15
$\text{Ti}_4\text{O}_7 + \text{S}$	Melt-diffusion	30 – 40	56	1.5 – 1.8	850	600	1.53	500	88	0.06	28 – 33	1.7 – 3.0 V (2C)	14
$\text{SnS}_2\text{-ND}@G$	Hydrothermal process	16	75	2.5 – 10	1234	1016	11	300	82	0.06	10	1.7 – 2.8 V (C/5)	This work