

Supporting information

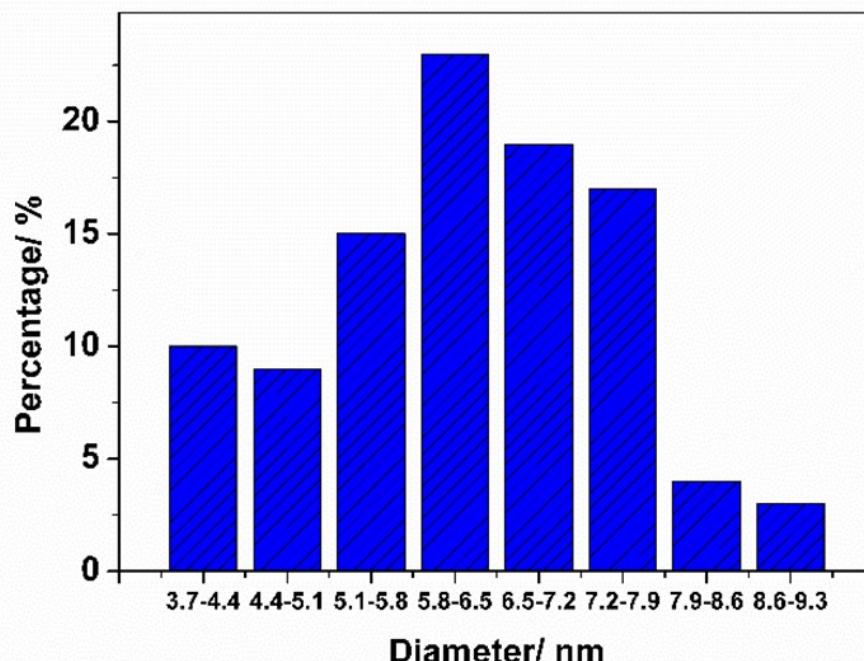


Figure S1. Size distribution of the PdNPs biosynthesized on *S. oneidensis* MR-1. Data from TEM images (Fig. 3D).

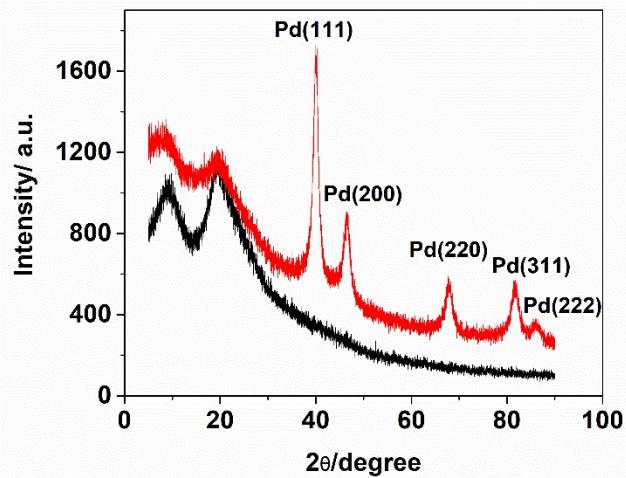
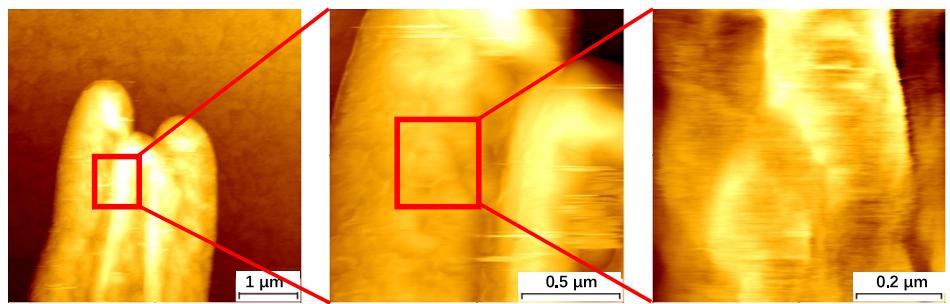


Figure S2. XRD spectra of *S. oneidensis* MR-1 before (black line) and after (red line) the exposure to 1 mM $\text{Na}_2[\text{PdCl}_4]$. Each sample was scanned in the θ -range of 5-90°.

Topography



Deflection

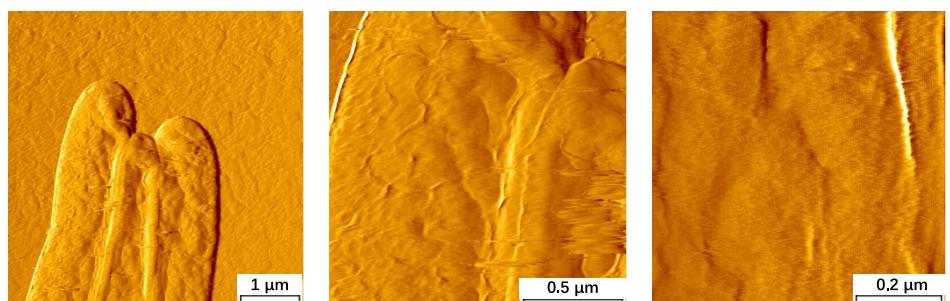


Figure S3. Tapping mode AFM images of *S. oneidensis* MR-1 cells with PdNPs on Pt sheet ($1.0 \times 1.0 \text{ cm}^2$).

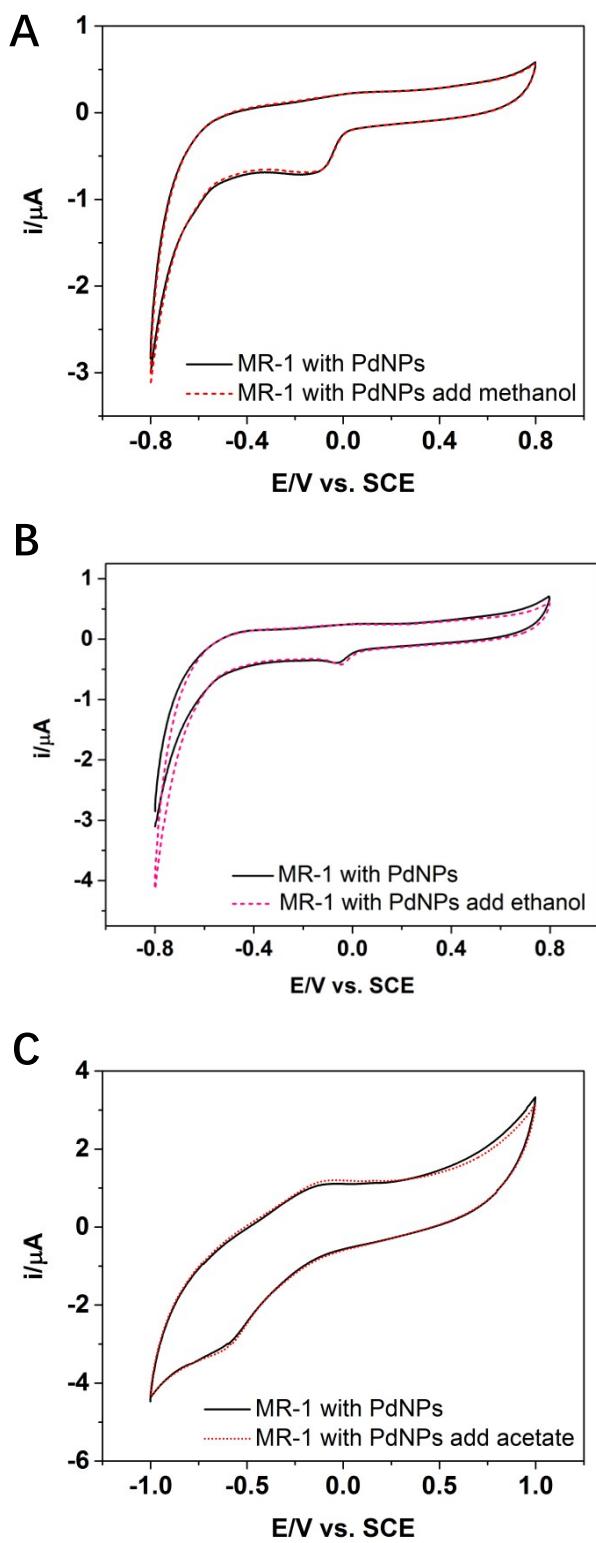


Figure S4. CVs of *S. oneidensis* MR-1 with bio-PdNPs modified GC electrode in 0.10 M pH 7.0 phosphate buffer (solid lines) and in the presence of 100 mM (A) methanol; (B) ethanol and (C) acetate (dotted lines), at a scan rate of 10 mV/s.

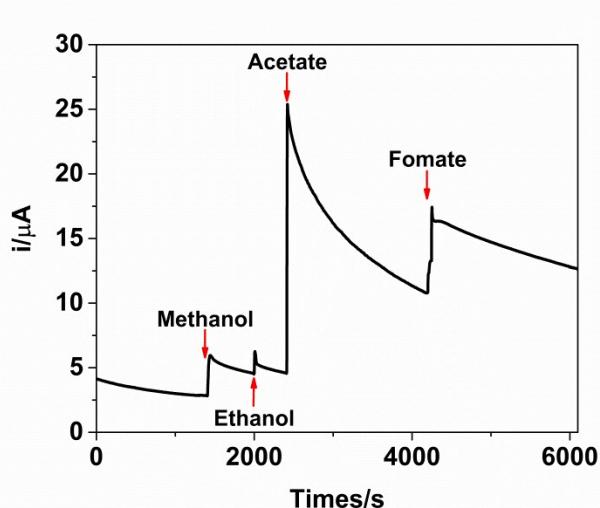


Figure S5. Chronoamperometric curve of methanol, ethanol, acetate, and formate oxidation at electro-deposited Pd modified GC electrode. The potential held at 0.10 V in 0.10 M pH 7.0 phosphate buffer solution. Final concentrations of methanol, ethanol, acetate, and formate are 0.10 M.

Reference

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4. Babauta, J. T.; Nguyen, H. D.; Beyenal, H., Redox and pH Microenvironments within Shewanella oneidensis MR-1 Biofilms Reveal an Electron Transfer Mechanism. *Environmental Science & Technology* **2011**, 45 (15), 6654-6660.
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