

Electronic Supplementary Information for

## Boosting supercapacitive performance of ultrathin mesoporous NiCo<sub>2</sub>O<sub>4</sub> nanosheet arrays by surface sulfation

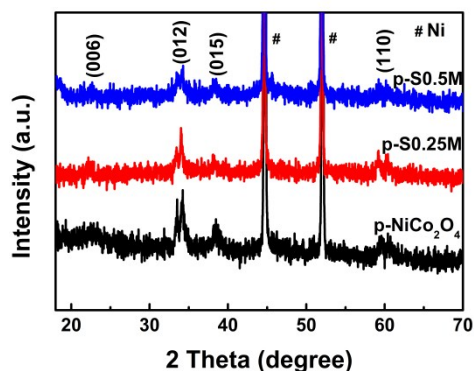


Figure S1 XRD spectrum for the as-deposited sample on Ni foam

The as-deposited sample is denoted as p-NiCo<sub>2</sub>O<sub>4</sub>, p-S0.25M and P-S0.5M, respectively. The four main peaks at 22.4°, 34.0°, 38.5°, 59.2° correspond to a rhombohedral structure of Co–Ni layered double hydroxides.<sup>1</sup>

The average crystalline sizes of all samples are calculated by Scherrer Equation<sup>2</sup> and combined with XRD data (1) as following:

$$D_{hkl} = \frac{K\lambda}{B_{hkl}\cos\theta}$$

$D_{hkl}$  is the crystallite size in nanometers, hkl are the Miller indices of the lattice planes being analyzed.  $\lambda$  is the wavelength of X-ray, taken as 0.154056 nm.  $B_{hkl}$  is the peak width of the X-ray diffraction peak profile at half-maximum height in radians and K is the numerical crystallite-shape factor, normally taken as 0.9 and  $\theta$  is the Bragg angle. Full Width Half Maximum (abbreviated as FWHM) is the  $B_{hkl}$  in degrees. In order to acquire relatively accurate crystallite size, the (311) plane with most prominent diffraction peak is chosen for calculation.

Table S1. The crystalline sizes of the samples

Sample Name	2 $\theta$ (degree)	FWHM (degree)	D <sub>311</sub> (nm)
Pristine NiCo <sub>2</sub> O <sub>4</sub>	36.894	0.655	12.02
S0.25M	36.668	0.723	11.45
S0.5M	36.688	0.719	11.51

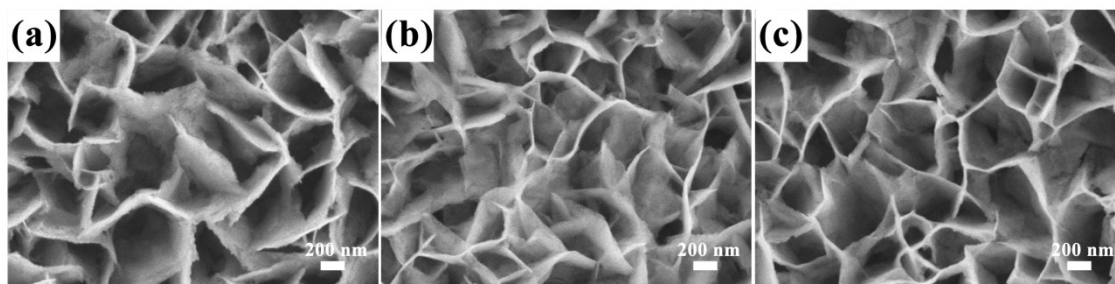


Figure S2 High-magnified SEM images of (a) NiCo<sub>2</sub>O<sub>4</sub>, (b) S0.25M and (c) S0.5M after 5000 cycling tests.

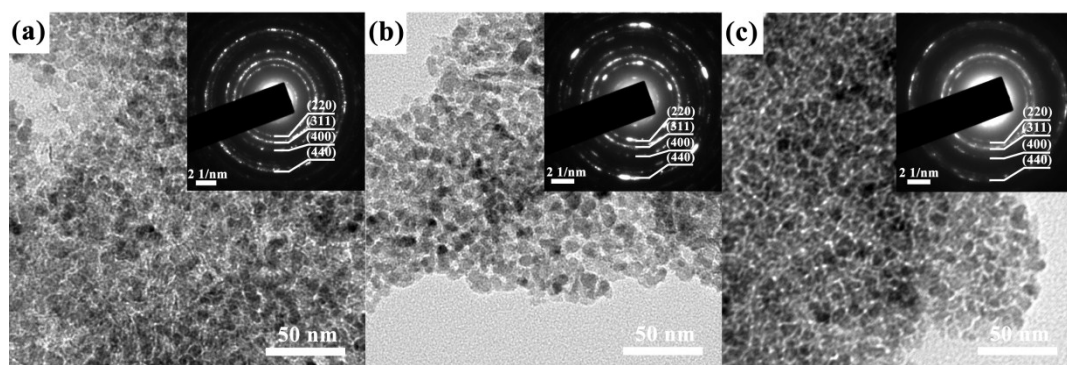


Figure S3 TEM images and the inset SAED pattern of (a) NiCo<sub>2</sub>O<sub>4</sub>, (b) S0.25M and (c) S0.5M after 5000 cycling tests.

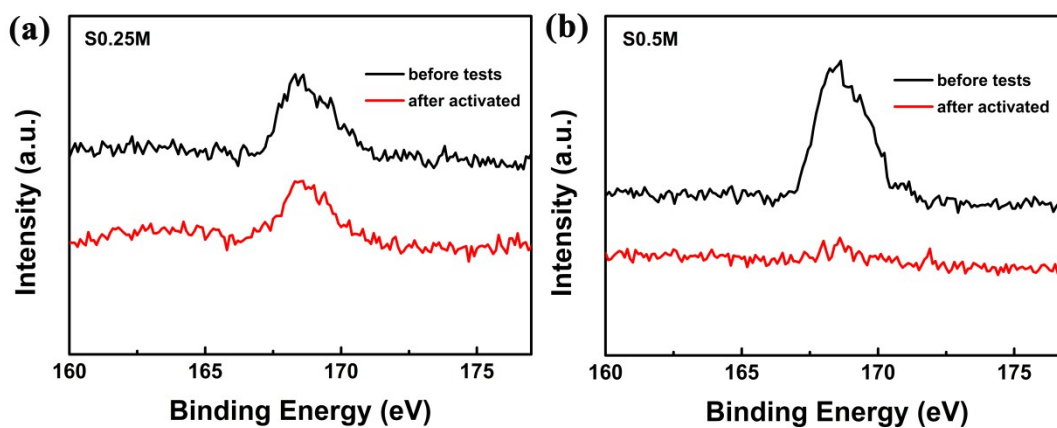


Figure S4 The core-level S 2p spectra of (a) S0.25M and (b) S0.5M at different stages including before cycling tests, after fully activated.

## References

1. L. Qian, L. Gu, L. Yang, H. Yuan and D. Xiao, *Nanoscale*, 2013, **5**, 7388-7396.
2. U. Holzwarth and N. Gibson, *Nature nanotechnology*, 2011, **6**, 534-534.