

Supporting Information

Improved cycling stability of NiS₂ cathode through designing “kiwano” hollow structure

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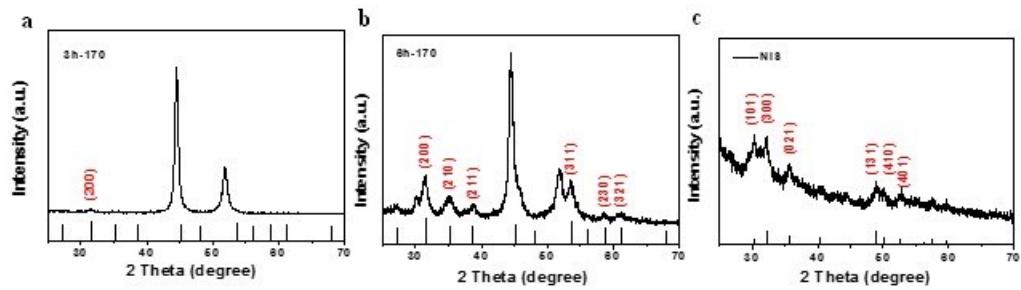


Fig. S1 XRD patterns of NiS_2 samples at 170 °C with the different reaction time: (a) 3 h and (b) 6 h. (c) XRD pattern of a phase-converted NiS from the initial NiS_2 after long cycles.

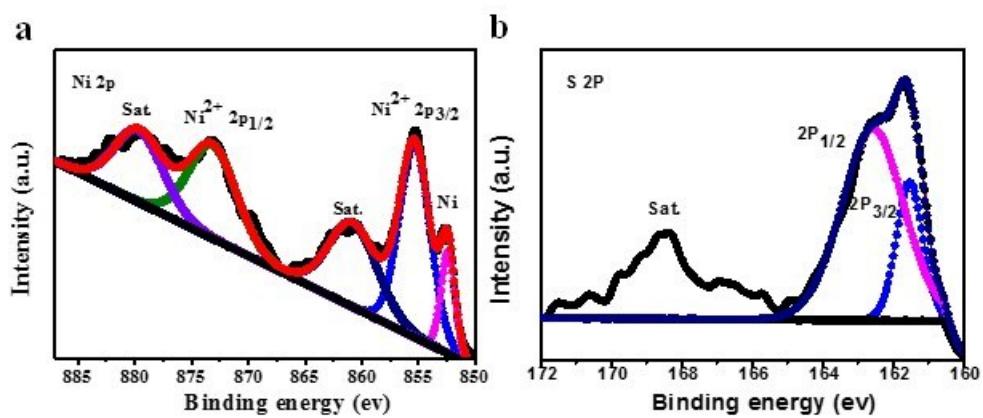


Fig. S2 (a-b) XPS profiles of Ni 2p and S 2p for NiS₂.

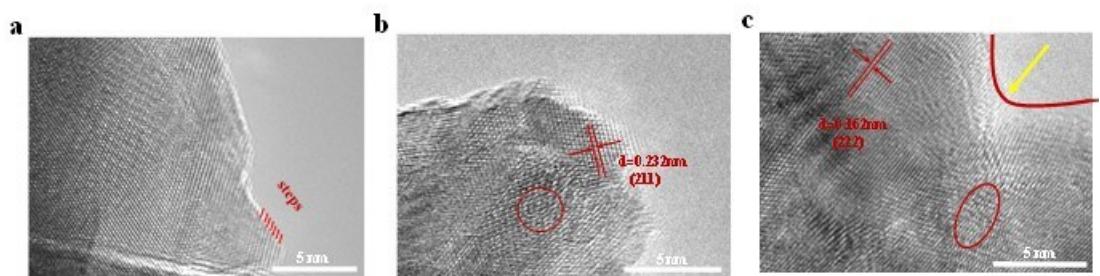


Fig. S3 (a-c) HRTEM images of the NiS₂ hollow spheres.

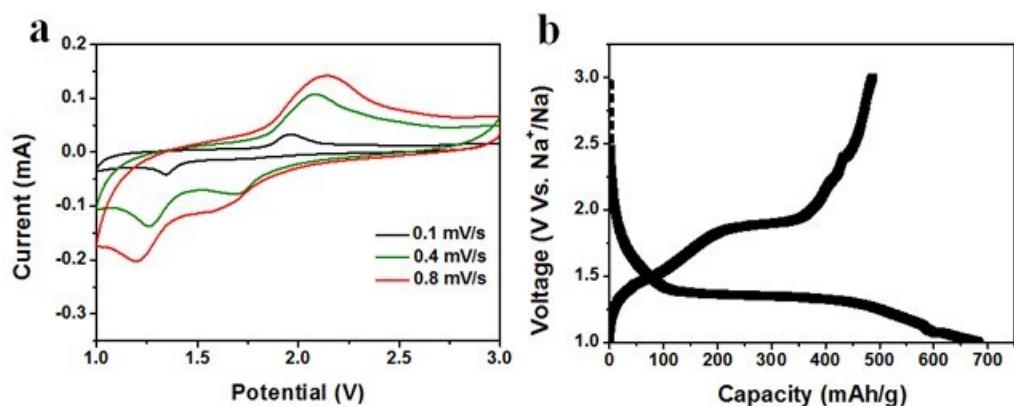


Fig. S4 (a) CV curves of a NiS_2 electrode in the voltage window of 1.0–3.0 V at a scan rate of 0.1 mV/s, 0.4 mV/s and 0.8 mV/s after long cycles. (b) The galvanostatic charge/discharge profiles of a NiS_2 electrode in the voltage window of 1.0–3.0 V at 50 mA/g.

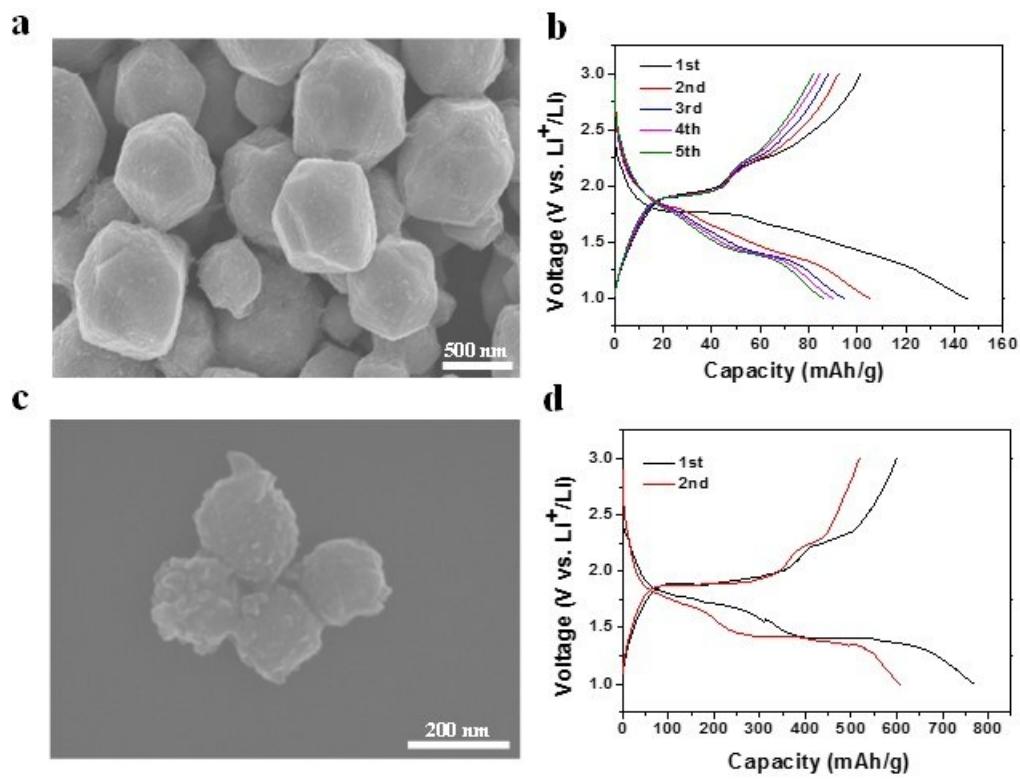


Fig. S5 The SEM images and the galvanostatic charge/discharge profiles of NiS₂ samples prepared at 170 °C with the different reaction time: (a-b) 3 h and (c-d) 6 h.

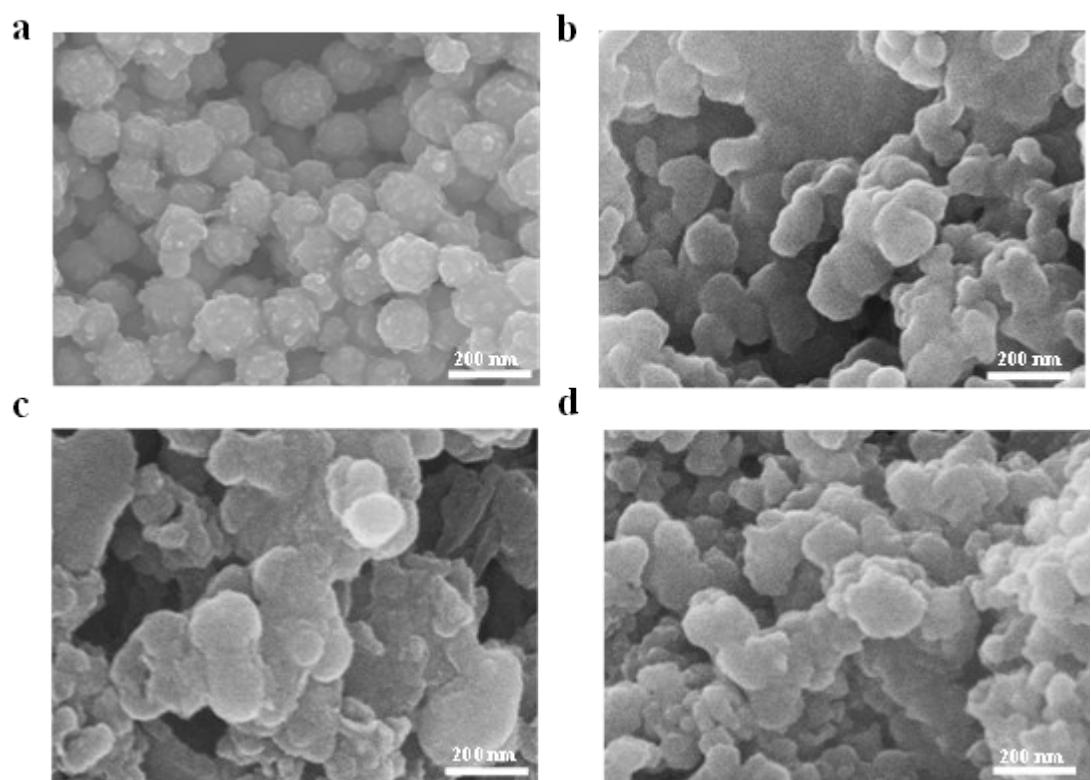


Fig. S6 (a) SEM image of the NiS_2 hollow spheres before cycles. (b-d) SEM images of a NiS_2 electrode material after 10, 50, 100 cycles.

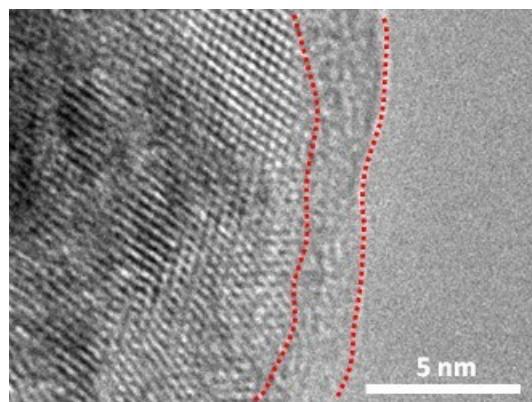


Fig. S7 HRTEM image of the NiS₂ surface after first lithiation.

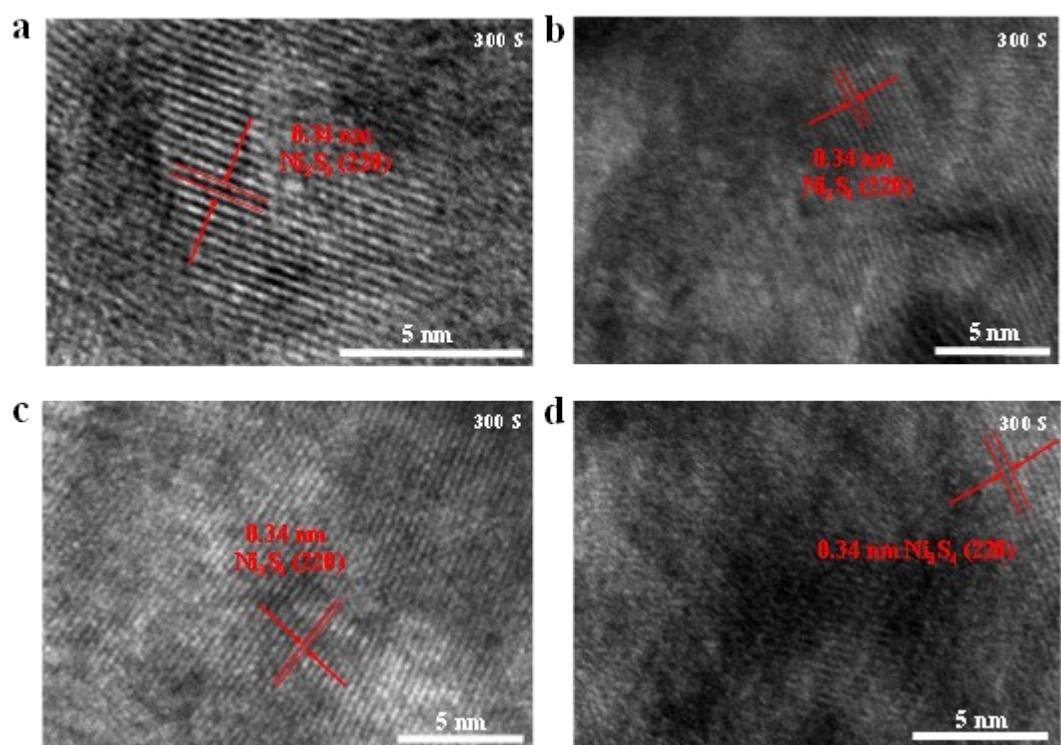


Fig. S8 HRTEM images of the different lithiated NiS₂ particles after the first lithiation

for 300 s as observed via *in situ* TEM.

Notes and references

1. Q. N. Chen, W. X. Chen, J. B. Ye, Z. Wang and J. Y. Lee, *J. Power Sources*, 2015, **294**, 51-58
2. K. Liang, K. Marcus, S. Zhang, L. Zhou, Y. Li, S. T. De Oliveira, N. Orlovskaia, Y.-H. Sohn and Y. Yang, *Adv. Energy Mater.*, 2017, **7**, 1701309.
3. G. J. An, C. G. Liu, Y. D. Hou, X. L. Zhang and Y. Q. Liu, *Mater. Lett.*, 2008, **62**, 2643-2646.
4. X. Wang, Q. H. Weng, X. Z. Liu, X. B. Wang, D. M. Tang, W. Tian, C. Zhang, W. Yi, D. Q. Liu, Y. Bando and D. Golberg, *Nano Lett.*, 2014, **14**, 1164-1171.
5. Y. Yang, D.-M. Tang, C. Zhang, Y. Zhang, Q. Liang, S. Chen, Q. Weng, M. Zhou, Y. Xue, J. Liu, J. Wu, Q. H. Cui, C. Lian, G. Hou, F. Yuan, Y. Bando, D. Golberg and X. Wang, *Energy Environ. Sci.*, 2017, **10**, 979-986.
6. J. H. Wang, Z. Cheng, J. L. Bredas and M. L. Liu, *J. Chem. Phys.*, 2007, **127**, 8.