

Supporting Information

Highly Efficient Quasi-solid-state Dye-sensitized Solar cell using Polyethylene oxide (PEO) and Poly(methyl methacrylate) (PMMA)-based Printable Electrolytes

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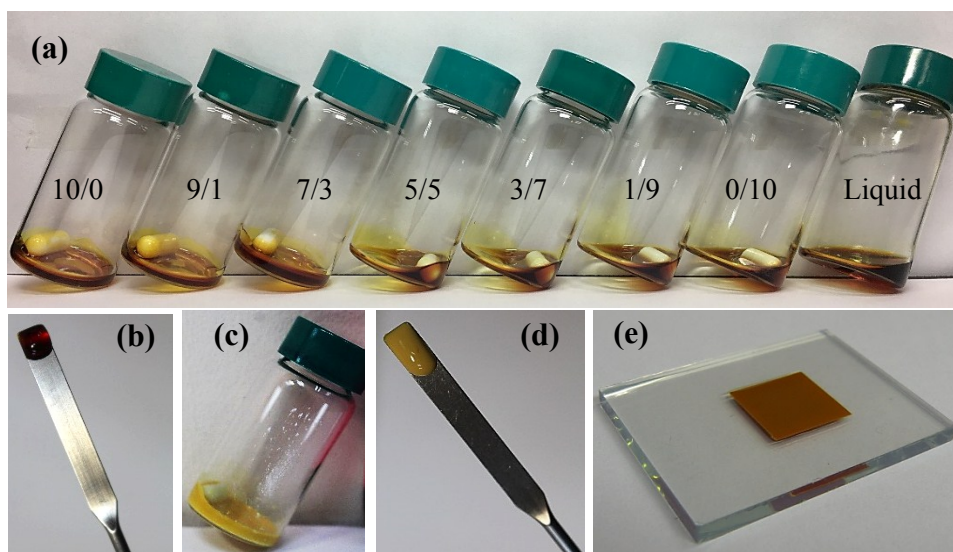


Fig. S1. (a) Preparation of printable electrolytes at different stages. The electrolytes contain 9 wt. % polymer with various compositions of PEO/PMMA in MPN-liquid electrolyte, (b) a printable electrolyte paste prepared by using PEO/PMMA (7/3), (c-d) PEO/PMMA (7/3)/TiO₂ NFs (10 wt. %) based PE paste and (d) printed electrolyte on the photoanode surface.

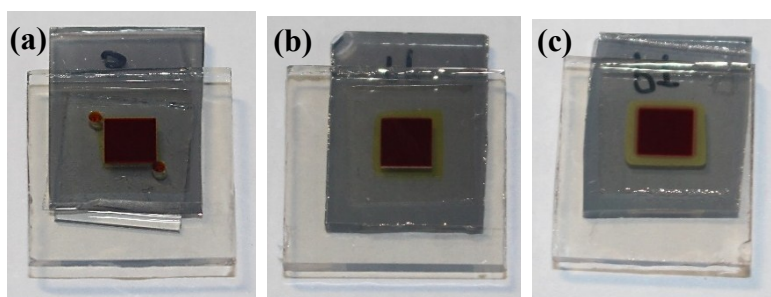


Fig. S2. DSSCs assembled with (a) liquid electrolyte, (b) PEO/PMMA (7/3) PE and (C) PEO/PMMA/TiO₂ PE.

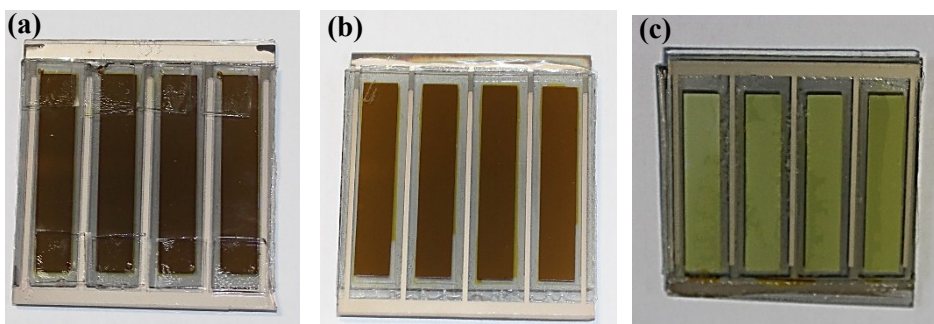


Fig. S3. Sub-module DSSCs assembled with (a) liquid electrolyte, (b) PEO/PMMA (7/3) PE and (C) PEO/PMMA/TiO₂ PE.

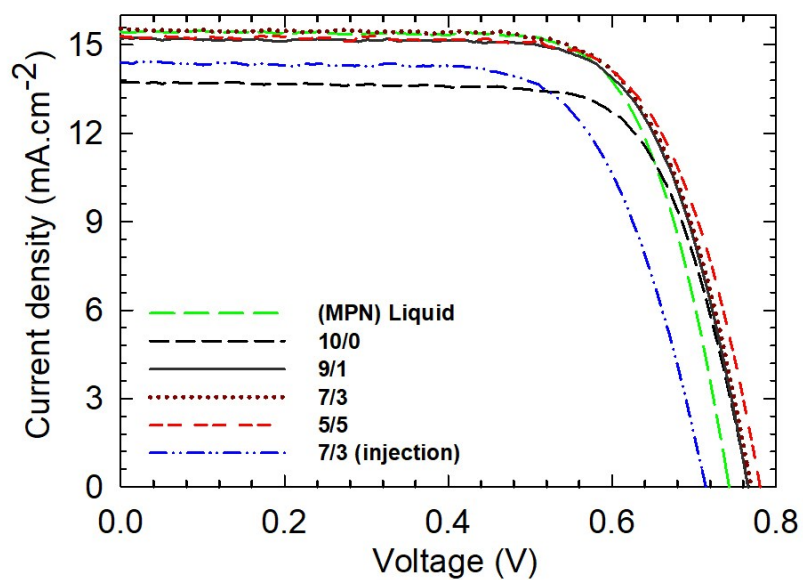


Fig. S4. J-V curves measured for the DSSCs using liquid electrolyte and printable electrolyte systems. The PEs contains 9 wt. % polymer and the PEO/PMMA is 7/3. These curves were measured under a one-sun condition with the intensity of 100 mW cm⁻².

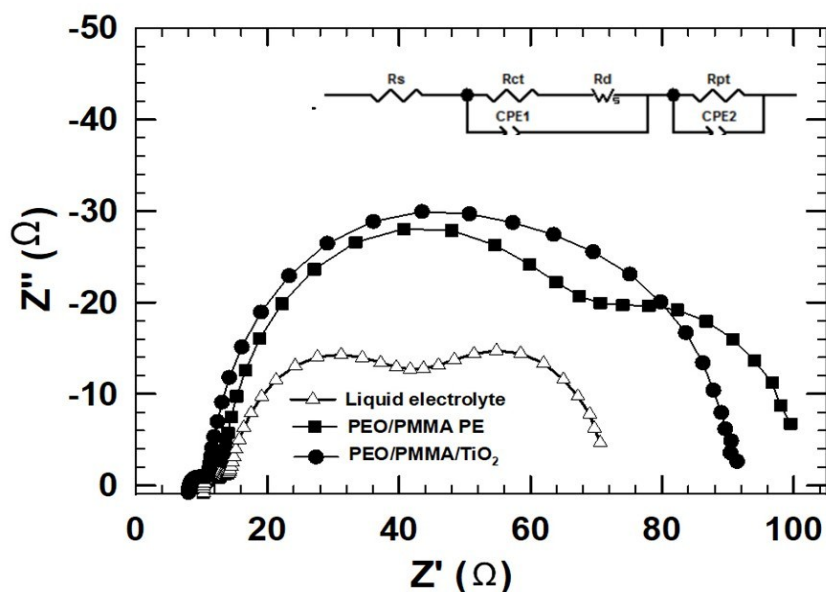


Fig. S5. Nyquist plots of the EIS spectra measured at 0.75 V under dark conditions for the liquid-state cell and QS-DSSCs with and without 10 wt. % TiO₂ NFs.

Description: The EIS spectra in Fig S5 show the three semi-circles for the cells using liquid-electrolyte and PEO/PMMA PE. These circles are corresponding to the R_{pt} (high frequency region), the R_{ct} (middle frequency region), and Nernstian diffusion (R_w) (low frequency region) within the electrolytes. For the cell using PEO/PMMA/TiO₂ NFs, only two semicircles are observed because of overlaps occurred between middle and low frequency circles.

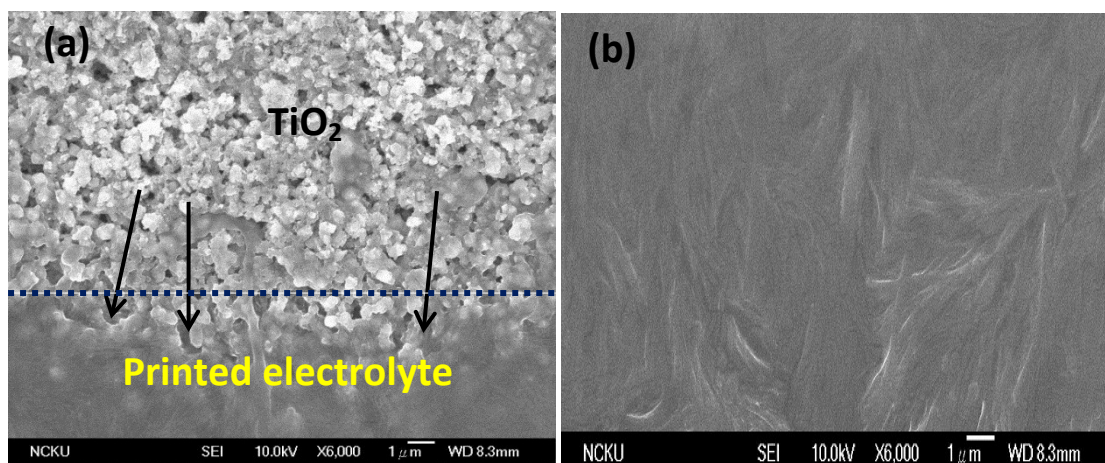


Fig. S6. (a) SEM image of the partially printed electrolyte on the photoanode. The arrow marks on the image indicate a strong penetration of the PE into the nanopores of the TiO₂. (b) The surface of the photoelectrode after completely covered with printed electrolyte.

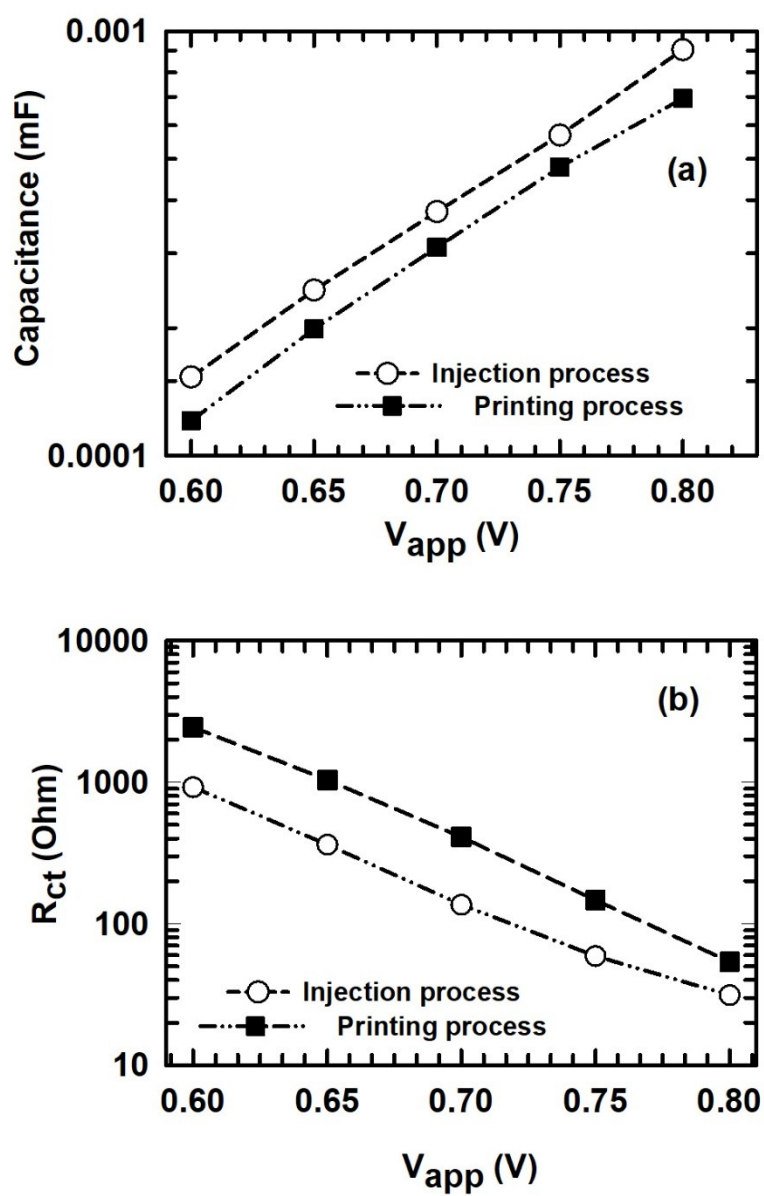


Fig. S7. Comparison between the injection and printing process for the PEO/PMMA electrolyte based on the (a) capacitance and (b) R_{ct} .

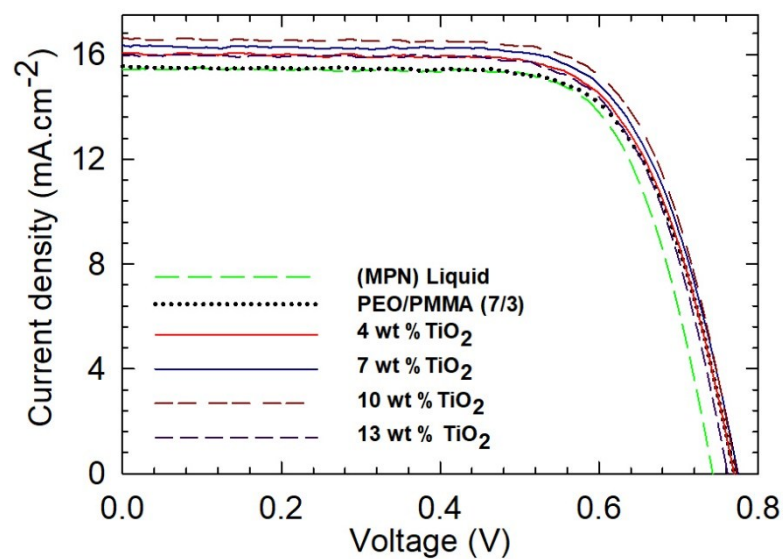


Fig. S8. J-V curves measured for the QS-DSSCs using the printable electrolyte with TiO₂ NFs. The compositions of TiO₂ NFs are 4-13 wt.%. These curves were measured under a one-sun condition with the intensity of 100 mW cm⁻².

Table S1. Comparison among the characteristics of PEO/PMMA-based electrolytes (PGEs and PEs) and corresponding performance of QS-DSSCs reported in the literature.

References	Polymer utilized	Electrolyte	$\sigma \times 10^{-3}$ (S cm ⁻¹)	R _{pt}	R _{ct}	J _{sc} (mA cm ⁻²)	V _{oc} (V)	FF	η (%)
Shi et al 2009 (ref. 36)	PMA/PEG (65/35, 27 wt %)	EC/PC (8/2)-liquid							
		PGE (PMA)	1.90			7.5	0.640	0.66	3.17
		PGE (PMA/PEG)	2.10			9.3	0.687	0.59	4.71
Wang et al 2013 (ref. 25)	[P(VA-co-MMA)] (20 wt %)	MPN-liquid	3.40		41.01	17.40	0.74	0.71	9.14
		PE	2.60		36.31	17.32	0.71	0.70	8.61
		PE/TiO ₂	2.62		38.09	17.58	0.73	0.70	8.98
Theerthagiri et al 2015 (ref. 37)	PMMA-PEO	ACN-liquid							
		PGE (PEO-PMMA)	4.18			5.54	0.69	0.59	2.20
		PGE (PEO-PMMA-2-MCP)	15.5			6.06	0.73	0.69	3.00
Aram et al 2015 (ref. 35)	PEO	EC/PC/DMC (4/2/1)-liquid				12.86	0.68	0.59	5.15
	PEO	PGE (PEO)	2.20	High		5.89	0.74	0.472	2.06
	PEO/PMMA (10/90)	PGE (PEO/PMMA)	7.00	low		11.11	0.76	0.58	4.90
Mohan et al 2017 (ref. 38)	PMMA	ACN/NMP (8/2) liquid		1.47	7.35	12.87	0.734	0.630	5.96
		PGE (PMMA)	0.987	7.73	29.34	4.15	0.794	0.841	2.77
		PGE/CB (0.57 wt %)	3.32	3.48	12.99	12.43	0.766	0.580	5.52
		MPN	1.95	1.31	Low	15.47	0.743	0.726	8.34
Present Study		PE (PEO)	1.28	2.54		13.75	0.767	0.725	7.65
	PEO/PMMA (7/3, 9 wt%)	PE (PEO/PMMA)	1.40	1.84	Middle	15.56	0.770	0.708	8.50
		PE/TiO ₂	1.29	1.20	high	16.64	0.773	0.709	9.12