Electronic supplementary information

Switchable synthesis of p- and n-type Cu-In-S grooved pyramid-like

microcrystals for unassisted photoelectrochemical water splitting

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Table S1 SEM-EDX results of the Cu-In-S films prepared with different amount of thiourea.

Sample	Cu element	In element	S element
Cu15S0	58.10%	8.07%	33.83%
Cu15S10	21.31%	24.99%	53.70%
Cu15S15	17.48%	27.17%	55.35%
Cu15S30	8.70%	34.87%	56.43%
Cu15S45	16.61%	27.82%	55.57%

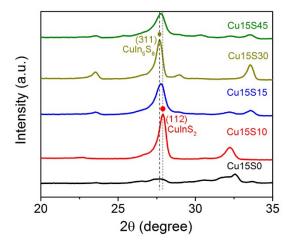


Fig. S1 The magnified XRD patterns of as-prepared samples.

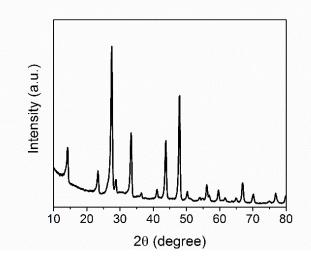


Fig. S2 XRD pattern of In_2S_3 film.

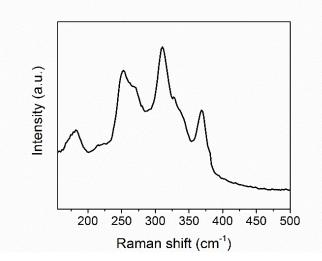


Fig. S3 Raman spectrum of In₂S₃ film.

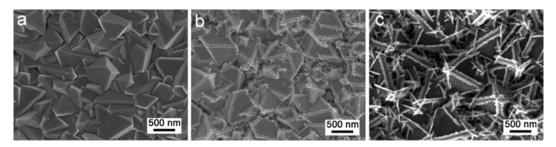


Fig. S4 Top-view SEM images of (a) In_2S_3 film; (b) Cu15S10 sample solvothermally treated for 3 h; and (c) Cu15S10 sample solvothermally treated for 12 h.

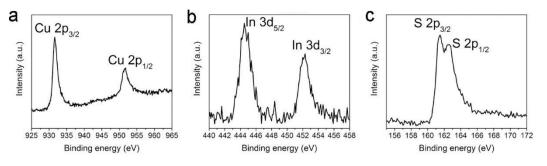


Fig. S5 XPS spectra of powder extracted from the solution after the solvothermal reaction for preparing Cu15S10 sample.

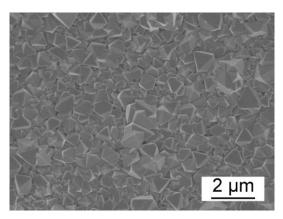


Fig. S6 Top-view SEM images of Cu0S10 film prepared without adding copper source.

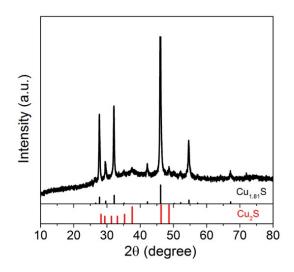


Fig. S7 XRD patterns of powder extracted from the solution after the solvothermal reaction for preparing Cu15S10 sample.

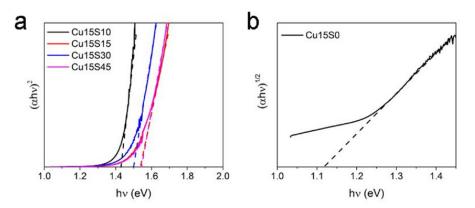


Fig. S8 (a, b) plots of $(\alpha h \upsilon)^n$ vs h υ of Cu-In-S films prepared with different amount of thiourea.

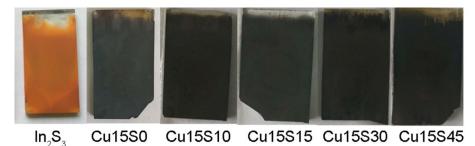


Fig. S9 Photograph of In_2S_3 film and Cu-In-S films prepared with different amounts of thiourea.

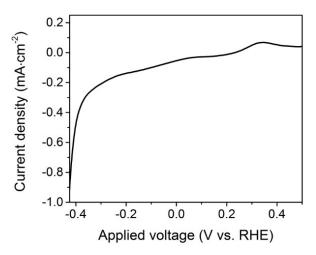


Fig. S10 Linear sweep voltammetry of Cu15S0 film was measured at $0.1 \text{ M Na}_2\text{SO}_4$ (pH adjusted to 3 by H_2SO_4) under chopped irradiation light.

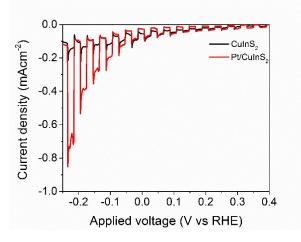


Fig. S11 Current-potential curves of CuInS₂ and Pt/CuInS₂.

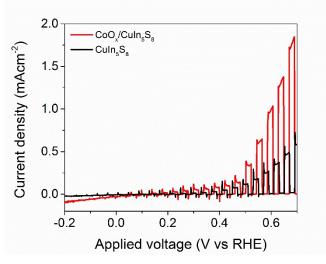


Fig. S12 Current-potential curves of CuIn₅S₈ and CoO_x/CuIn₅S₈.

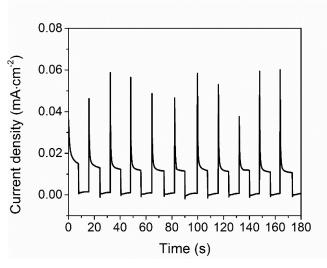


Fig. S13 Current-time curve for the tandem PEC cell using $CoO_x/CuIn_5S_8$ photoanode and Pt/CuInS₂ photocathode under light-chopping conditions. The electrolyte is 0.1 M Na₂SO₄ aqueous solution (pH adjusted to 3 by H₂SO₄).