

**APTES-functionalized Thin-walled Porous WO₃ Nanotubes for
Highly Selective Sensing of NO₂ in a Polluted Environment**

Wei Liu, Lin Xu, Kuang Sheng, Cong Chen, Xiangyu Zhou, Biao Dong, Xue Bai,
Shuang Zhang, Geyu Lu, Hongwei Song**

State Key Laboratory of Integrated Optoelectronics, College of Electronic Science
and Engineering, Jilin University, Changchun, 130012, People's Republic of China

*Email address: linxu@jlu.edu.cn, songhw@jlu.edu.cn;

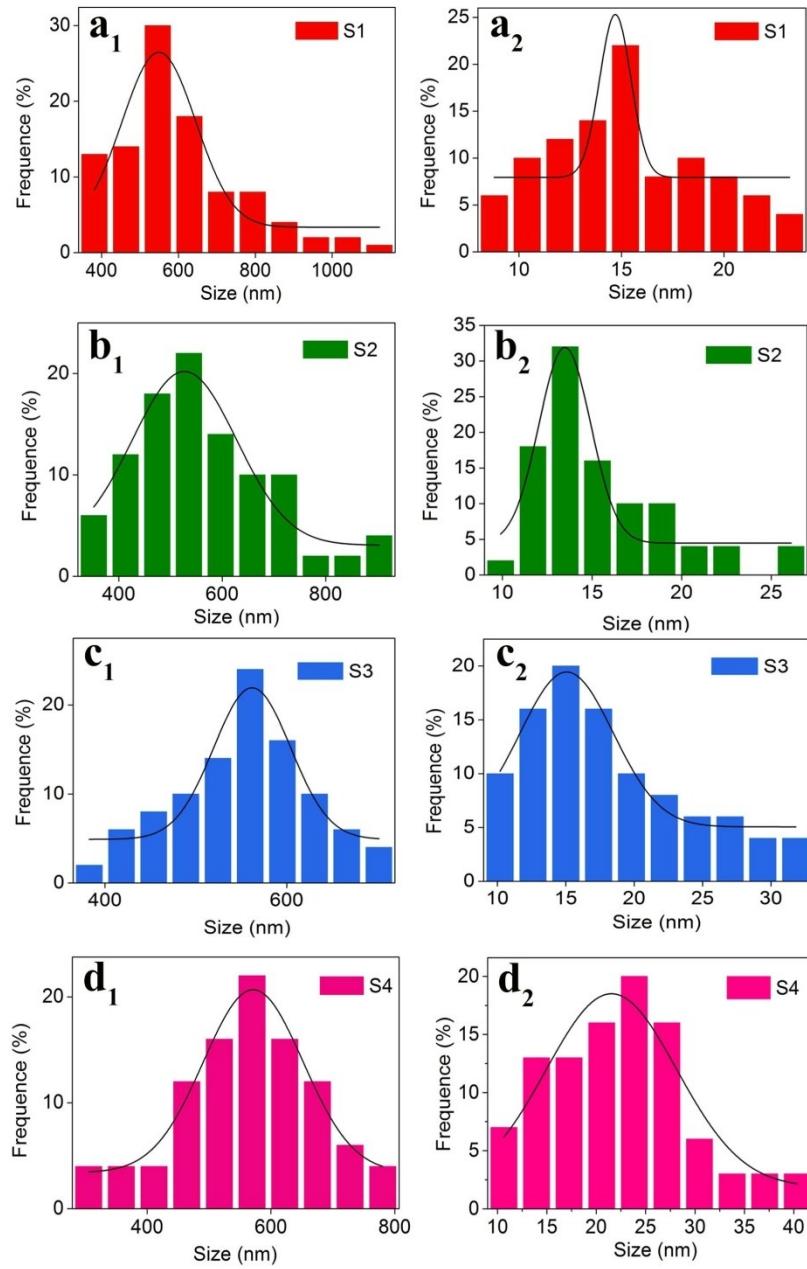


Fig. S1 The diameters and wall thickness distribution of (a₁) and (a₂) SiO₂/WO₃ NTs, (b₁) and (b₂) porous WO₃ NTs with 5 mol% SiO₂ removed, (c₁) and (c₂) porous WO₃ NTs with 10 mol% SiO₂ removed and (d₁) and (d₂) APTES functionalized porous WO₃ NTs based on the one with 10 mol% SiO₂ removed. The size distributions were obtained by measured the corresponding sizes from at least 100 NTs.

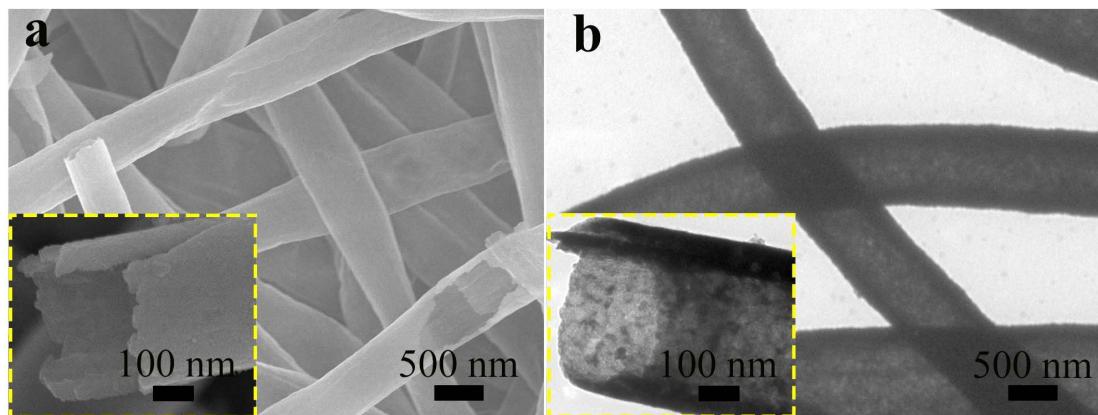


Fig. S2 (a) SEM image and (b) TEM image of pristine WO_3 NTs. The insert images are the enlarged SEM and TEM image of pristine WO_3 NTs.

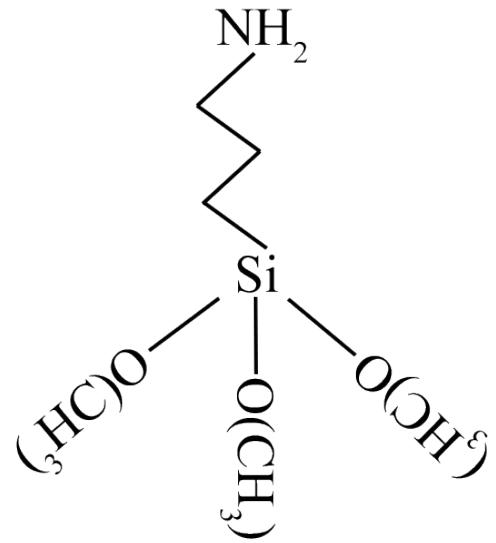


Fig. S3 The molecular structure of APTES.

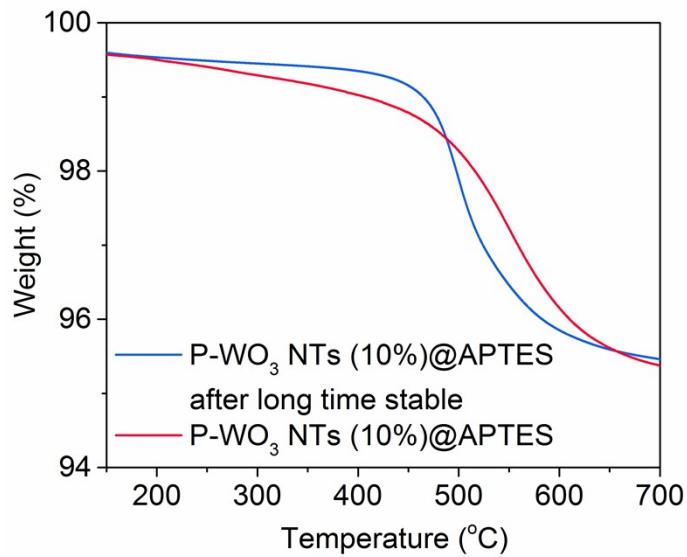


Fig. S4 The TGA of the P-WO₃ NTs (10%)@APTES and P-WO₃ NTs (10%)@APTES after long time stable.

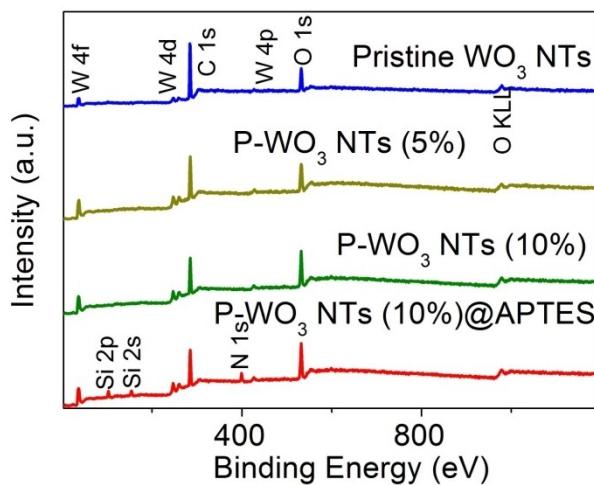


Fig. S5 The survey of X-ray photoelectron spectra of the pristine WO₃ NTs, P-WO₃ NTs (5%), P-WO₃ NTs (10%) and P-WO₃ NTs (10%)@APTES.

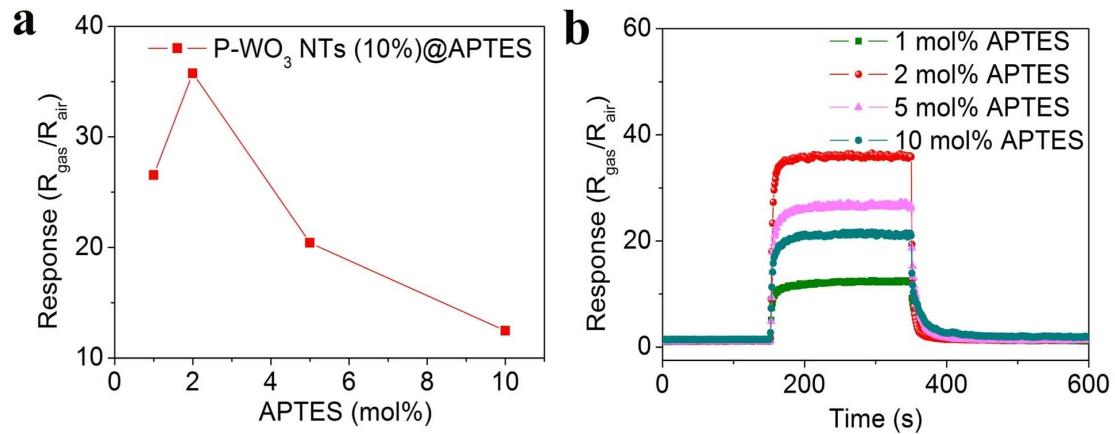


Fig. S6 (a) The response of 1 ppm NO₂ concentration with different amounts of APTES, (b) the dynamic response curve of P-WO₃ NTs (10%) modified with different amounts of APTES exposed to 1 ppm NO₂.

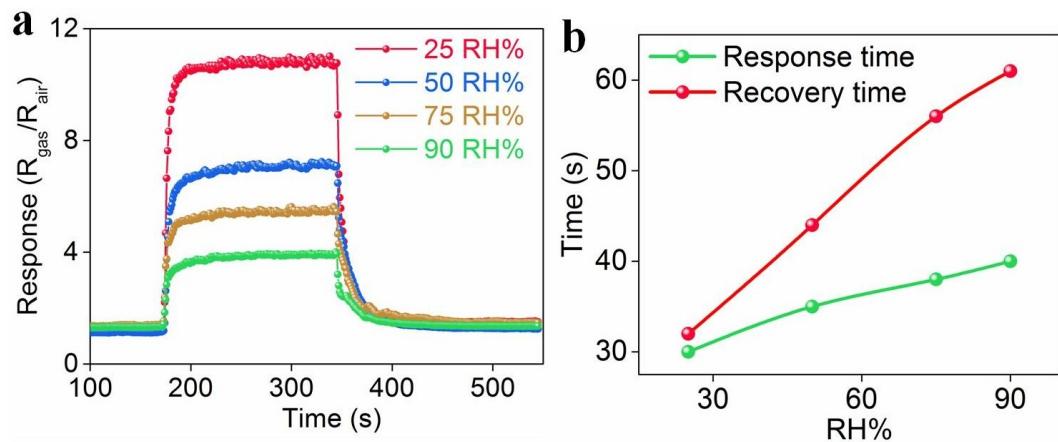


Fig. S7 (a) The dynamic curves and of P-WO₃ NTs (10%) sensor under different RH% (25~90%) for 1 ppm NO₂. (b) The corresponding response and recovery times of P-WO₃ NTs (10%) sensor under different RH% (25~90%) for 1 ppm NO₂.

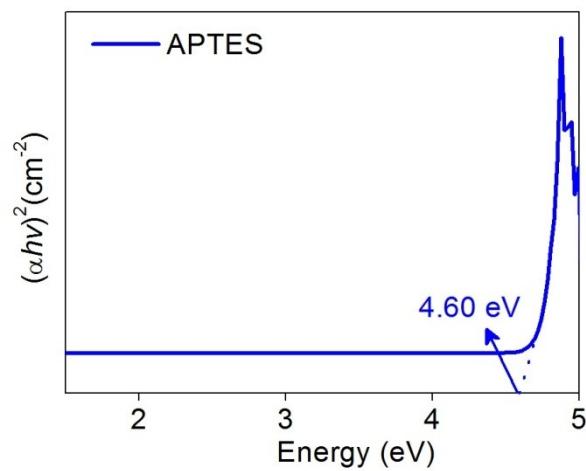


Fig. S8 Plot of $(\alpha h\nu)^2$ versus $h\nu$ of APTES.

Table S1. A comparison of WO_3 based NO_2 gas sensors in the literature to P- WO_3 NTs (10%)@APTES sensor in this work.

| Sensing material | Sensing performance | Response in this work | Temperatur e | Detection limit | Response/Recovery time | Reference |
|----------------------------------------------|---------------------|-----------------------|------------------|-----------------|------------------------|-----------|
| WO_3 NRs/graphene | 61/1 ppm | 45 | 300°C | 25 ppb | - | 1 |
| WO_3 nanorods | 209/10 ppm | 184 | 200°C | 700 ppb | - | 2 |
| WO_3 nanoparticles | 48/1 ppm | 45 | 50°C | - | -/~10 min | 3 |
| NiO/ WO_3 nanocomposites | 4.8/10 ppm | 184 | room temperature | - | 2.5/1.1 s | 4 |
| WO_3 -rGO porous nanocomposite | 4.3/10 ppm | 184 | 90°C | - | 5.8/8.7 s | 5 |
| WO_3 thin film | 20/10 ppm | 184 | 200°C | - | 3/151 s | 6 |
| WO_3 thin film | 511/10 ppm | 184 | 200°C | 500 ppb | 5.9/8.0 min | 7 |
| Ag-loaded mesoporous WO_3 | 44/1 ppm | 45 | 75°C | 100 ppb | 5.05/2.46 min | 8 |
| WO_3 nanowires | 4.49/1 ppm | 45 | 250°C | 250 ppb | 9/7 s | 9 |
| Au-functionalized WO_3 microspheres | 16/5 ppm | 112 | 50°C | - | 75/9 s | 10 |
| WO_3 nanowires/porous silicon | 6.75/5 ppm | 184 | 100°C | - | 175/44 s | 11 |
| WO_3 NPs/porous silicon | 3.4/2 ppm | 56 | 150°C | 50 ppb | 2/20 min | 12 |
| Graphene oxide/ WO_3 | 7.69/5 ppm | 112 | room temperature | - | 10/18 min | 13 |
| Porous silicon/ WO_3 nanorods | 3.38/1 ppm | 45 | room temperature | 100 ppb | 92/398 s | 14 |
| WO_3 thin films | 4.1/2 ppm | 56 | 300°C | 200 ppb | 4.5/4.5 min | 15 |
| WO_3 nanorod bundles | 111/5 ppm | 112 | 250°C | - | 230/42 s | 16 |
| Graphene oxide-Fe doped WO_3 | 4.5/1 ppm | 45 | 25°C | - | 4.9/5.47 min | 17 |
| WO_3 - TiO_2 thick film | 2.5/500 ppm | - | 600°C | - | - | 18 |
| WO_3 thin-film | 5/1 ppm | 45 | 370°C | 100 ppb | - | 19 |
| P- WO_3 NTs (10%)@APTES | 45/1 ppm | 45 | 340°C | 10 ppb | 10/11 s | This work |

Table S2. A comparison of some typical semiconductor oxide based NO₂ gas sensors with that of P-WO₃ NTs (10%)@APTES sensor in this work.

| Sensing material | Sensing performance | Response in this work | Temperature | Detection limit | Response/Recovery time | Reference |
|------------------------------------------------------------|---------------------|-----------------------|------------------|-----------------|------------------------|---------------|
| rGO-Cu ₂ O | 1.68/2 ppm | 56 | room temperature | 64 | - | ²⁰ |
| Co ₃ O ₄ -SnO ₂ nanowires | 3.47/10 ppm | 184 | 350°C | 2 ppm | >100/100 s | ²¹ |
| ZnO-rGO | 1.25/5 ppm | 112 | room temperature | - | 165/499 s | ²² |
| SnS ₂ materials | 14.9/10 ppm | 184 | 250°C | 500 ppb | 6/40 s | ²³ |
| ZnO microspheres | 4/1 ppm | 45 | 400°C | 500 ppb | - | ²⁴ |
| In ₂ O ₃ microspheres | 1.5/5 ppm | 112 | 250°C | 5 ppm | 5/20 s | ²⁵ |
| Ag-SnO ₂ microrods | 24/50 ppm | 541 | 300°C | 100 ppb | - | ²⁶ |
| CuO/p-porous silicon | 7.8/1 ppm | 45 | 300°C | 125 ppb | 257/374 s | ²⁷ |
| NiO nanosheets | 1.5/10 ppm | 184 | 250°C | 1 ppm | - | ²⁸ |
| Cr-doped TiO ₂ -NT | 2.9/50 ppm | 541 | 500°C | - | 3/6 min | ²⁹ |
| P-WO ₃ NTs (10%)@APTES | 45/1 ppm | 45 | 340°C | 10 ppb | 10/11 s | This work |

Table S3. The anti-interference test of P-WO₃ NTs (10%)@APTES sensor to 10 ppm interfering gases in the existence of 1 ppm NO₂.

| Gas species | Response of P-WO ₃ NTs (10%)@APTES sensor |
|-------------------------------------------------|------------------------------------------------------|
| 1 ppm NO ₂ | 45.5 |
| 1 ppm NO ₂ + 10 ppm NH ₃ | 44.5 |
| 1 ppm NO ₂ + 10 ppm H ₂ S | 43.7 |
| 1 ppm NO ₂ + 10 ppm ethanol | 45.3 |
| 1 ppm NO ₂ + 10 ppm toluene | 44.9 |
| 1 ppm NO ₂ + 10 ppm methanol | 45.2 |
| 1 ppm NO ₂ + 10 ppm acetone | 44.2 |
| 1 ppm NO ₂ + 10 ppm CO | 43.4 |
| 1 ppm NO ₂ + 10 ppm O ₃ | 44.8 |
| 1 ppm NO ₂ + 10 ppm NO | 43.9 |
| 1 ppm NO ₂ + 10 ppm SO ₂ | 45.1 |

Table S4. The pollution data came from the real-time air quality information of Changchun environmental protection bureau.

| Air pollution species | Satisfied | Light pollution | Moderate pollution | Heavy pollution |
|-----------------------|-----------|-----------------|--------------------|-----------------|
| SO ₂ (ppb) | 9 | 11 | 17 | 19 |
| PM 10 (ppb) | 75 | 89 | 92 | 152 |
| CO (ppb) | 15 | 17 | 18 | 22 |
| O ₃ (ppb) | 5 | 10 | 13 | 41 |

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