

Electronic Supplementary Information

Two Anthracene-based Metal-Organic Frameworks for Highly Effective Photodegradation and Luminescent Detection in Water

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SI. Supporting Figures

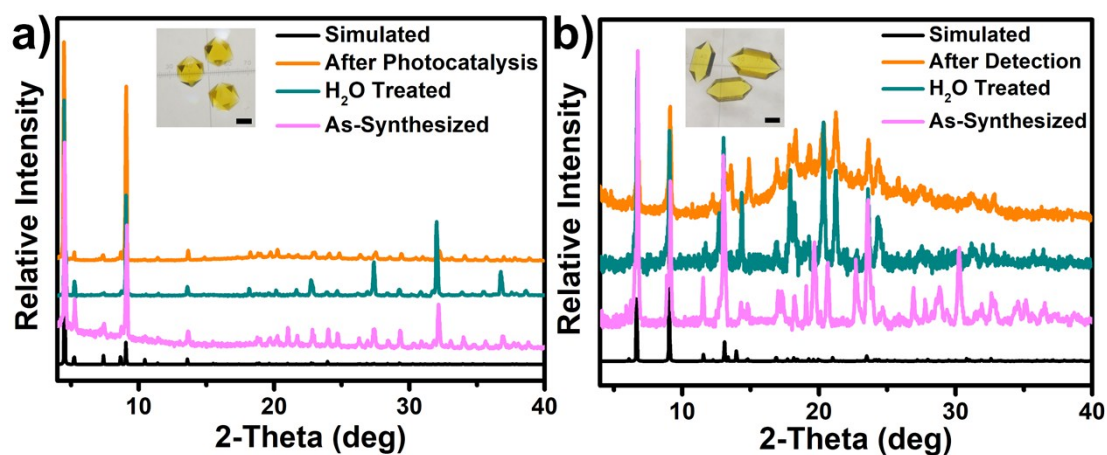


Fig. S1 Powder X-ray diffraction (PXRD) patterns of a) Eu-ADBA and b) In-ADBA. Inset: the single crystal image, scale bar: 200 μm.

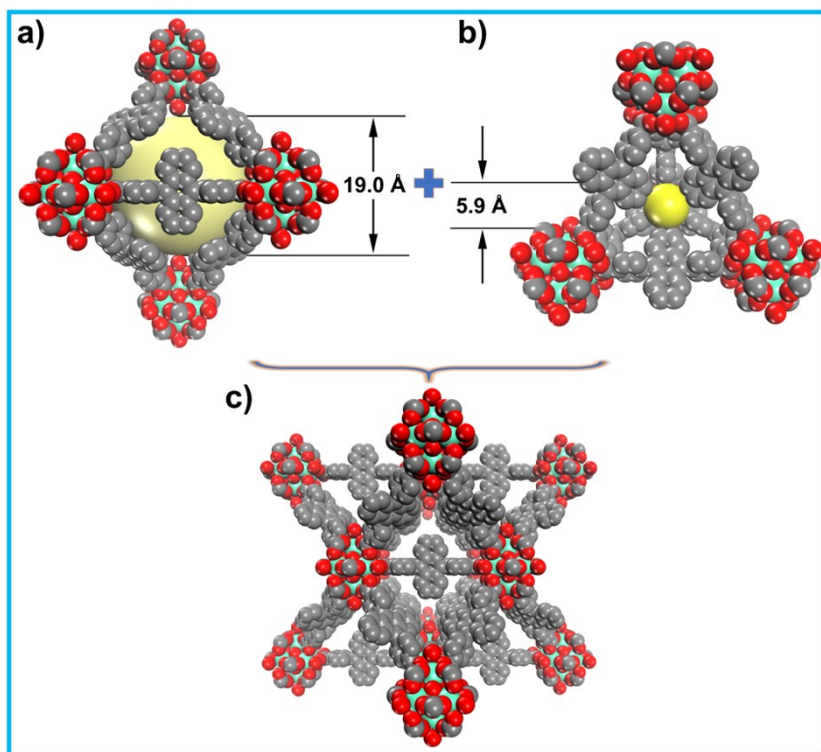


Fig. S2 CPK views of structure of **Eu-ADBA**: a) octahedral cage with diameter of 19.0 Å, b) tetrahedral cage with diameter of 5.9 Å, c) unit cell.

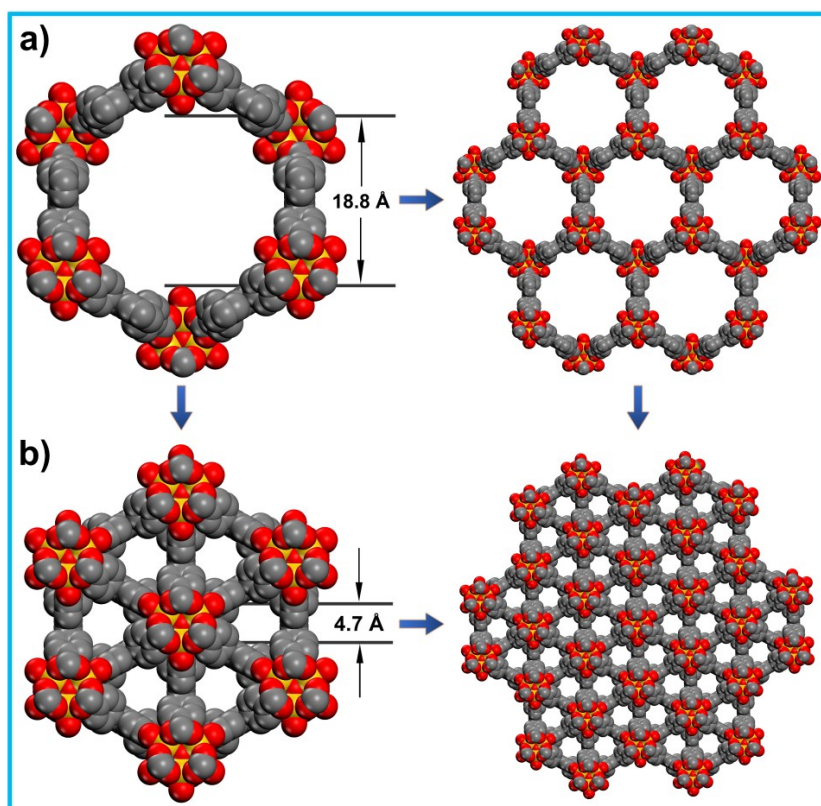


Fig. S3 CPK views of structure of **In-ADBA**: a) non-folded network with hexagonal channel with diameter of 18.8 Å, b) 3-folded network with triangular channel with diameter of 4.7 Å.

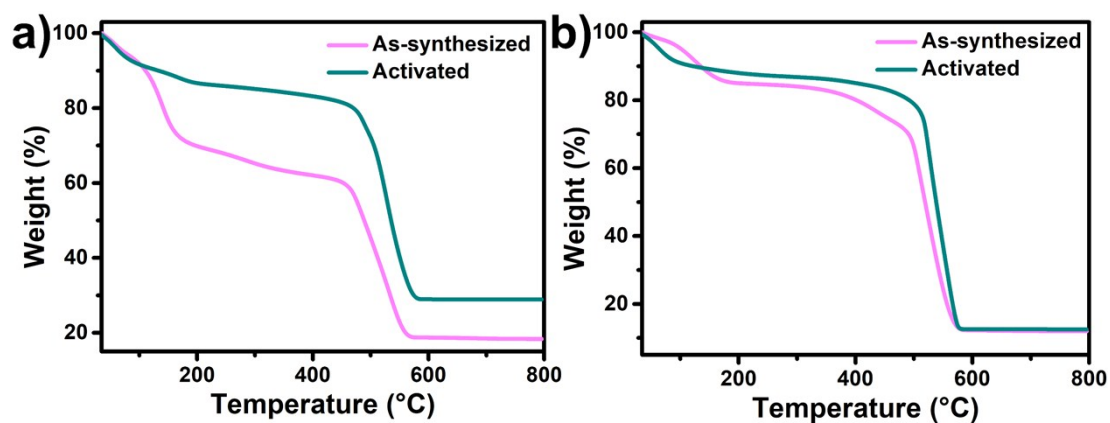


Fig. S4 TGA curves of a) Eu-ADBA and b) In-ADBA.

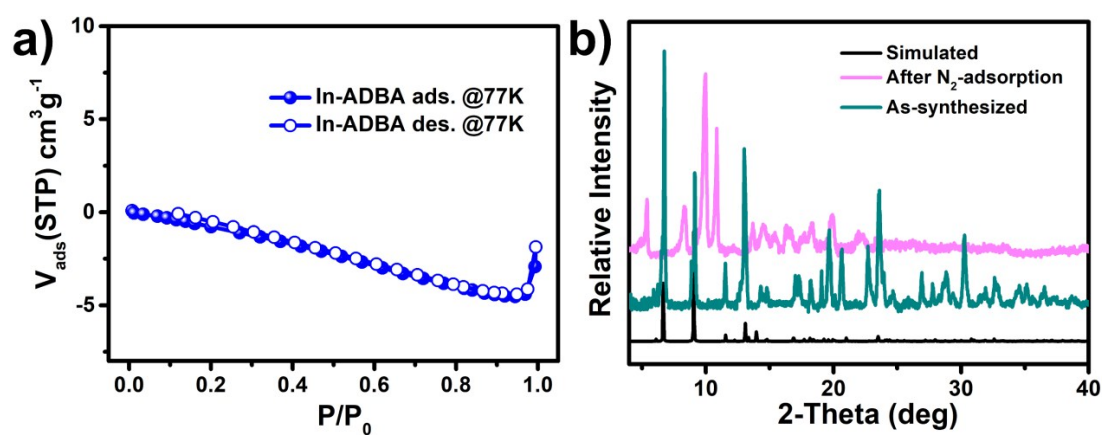


Fig. S5 a) N_2 sorption isotherm at 77 K under 1 bar of In-ADBA, b) PXRD patterns of In-ADBA.

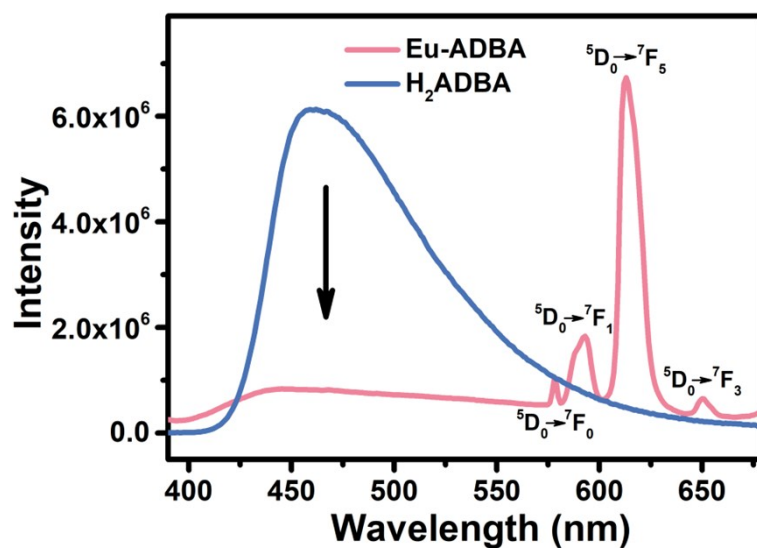


Fig. S6 Solid-state photoluminescent spectra of Eu-ADBA and free H_2ADBA ligand excited at 350 nm at room temperature.

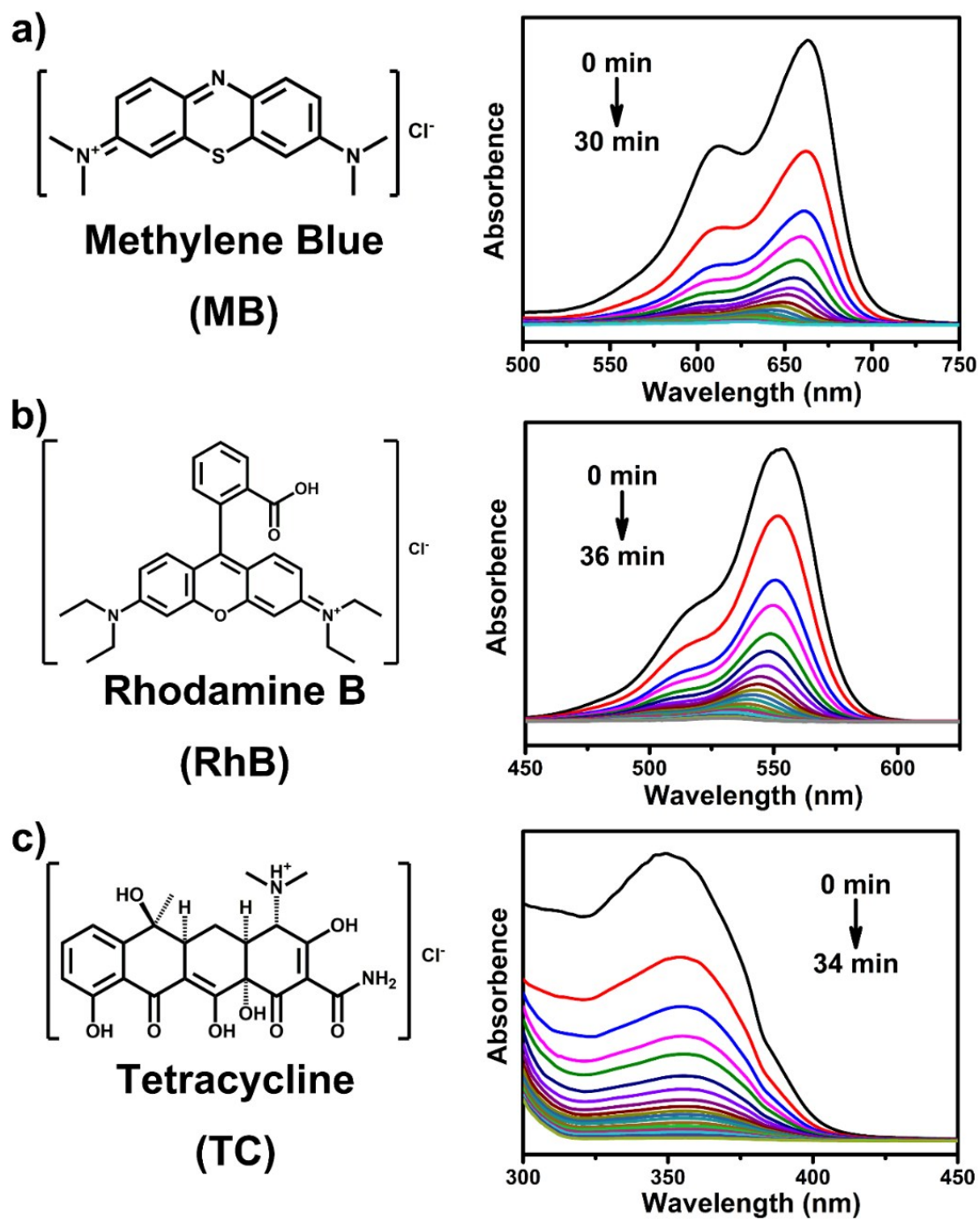


Fig. S7 Structural formula and time-dependent liquid-state UV-vis spectra of photodegradation of a) MB, b) RhB and c) TC using **Eu-ADBA**.

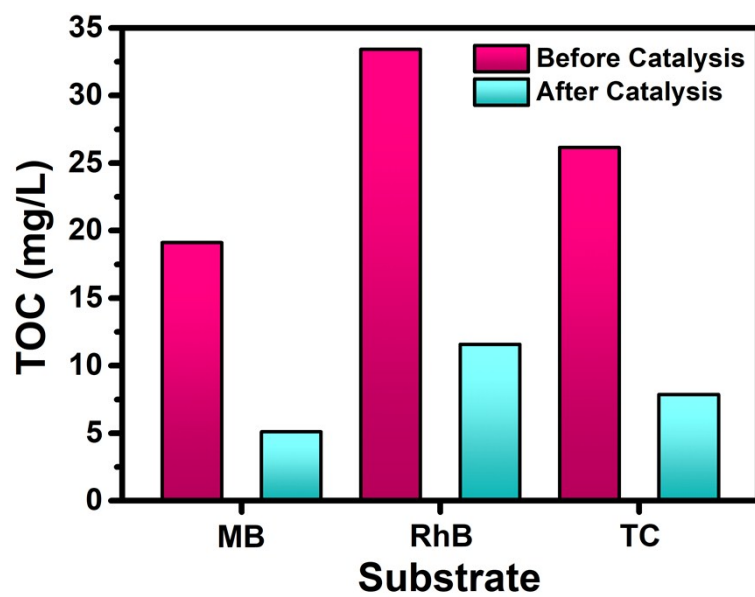


Fig. S8 TOC test of photodegradation using Eu-ADBA.

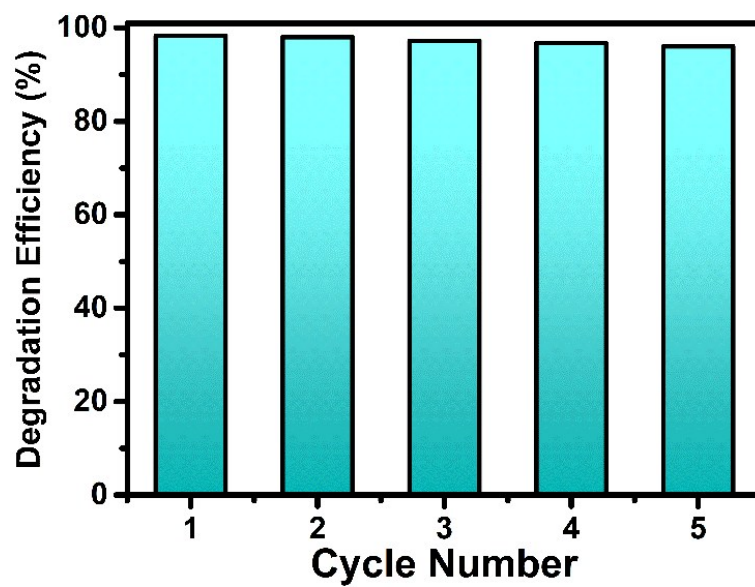


Fig. S9 Cyclic test of photocatalytic degradation of RhB using Eu-ADBA.

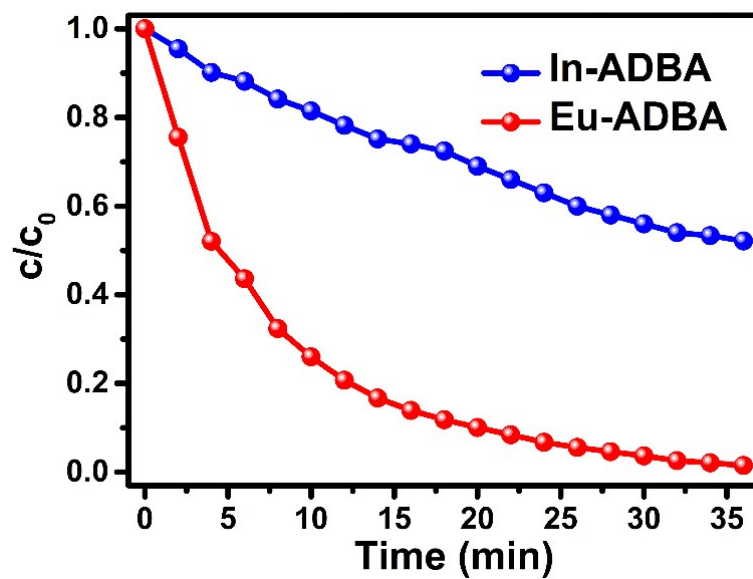


Fig. S10 Comparison of photocatalytic activities between **Eu-ADBA** and **In-ADBA**.

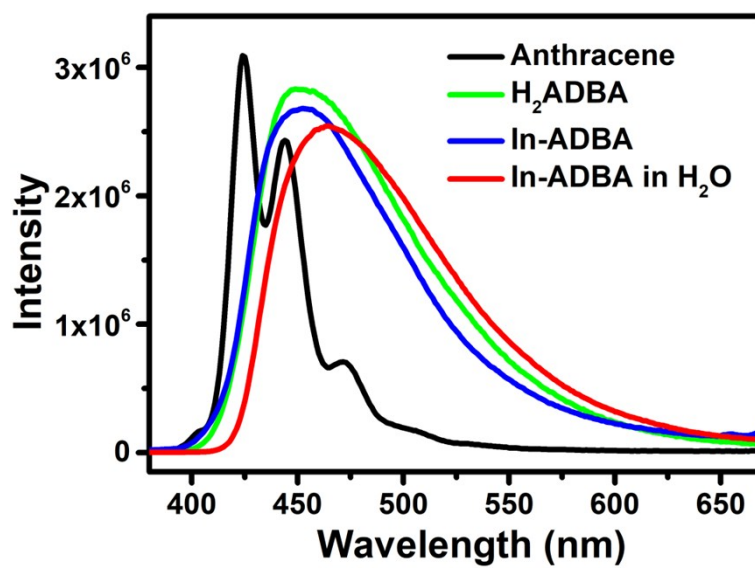


Fig. S11 Photoluminescent spectra of free anthracene molecules, free H₂ADBA ligand, **In-ADBA** and **In-ADBA** dispersed in water excited at 350nm.

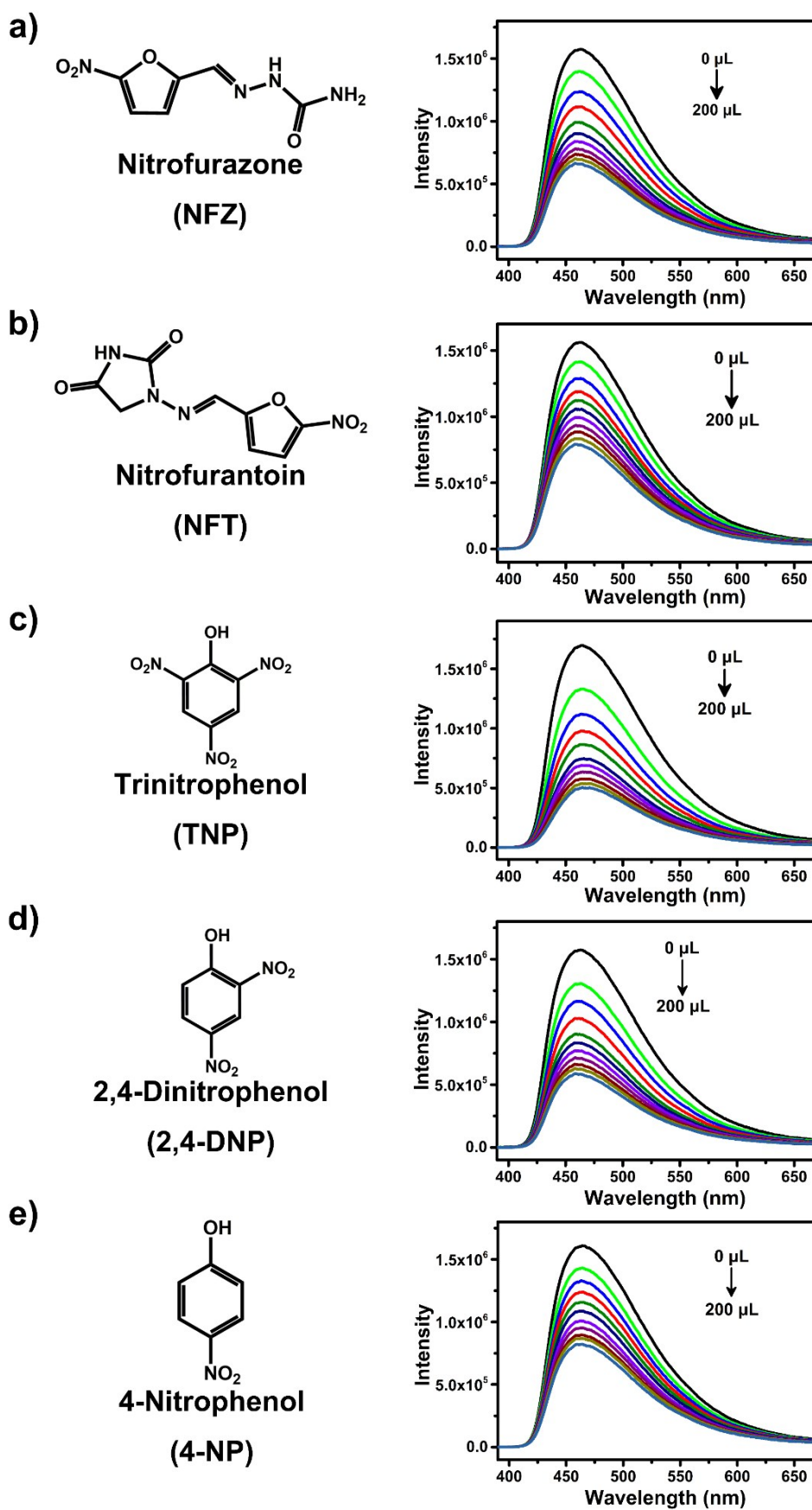


Fig. S12 Structural formula and luminescence spectra of low-concentration titrations of a) NFZ, b) NFT, c) TNP, d) 2,4-DNP and e) 4-NP using In-ADBA.

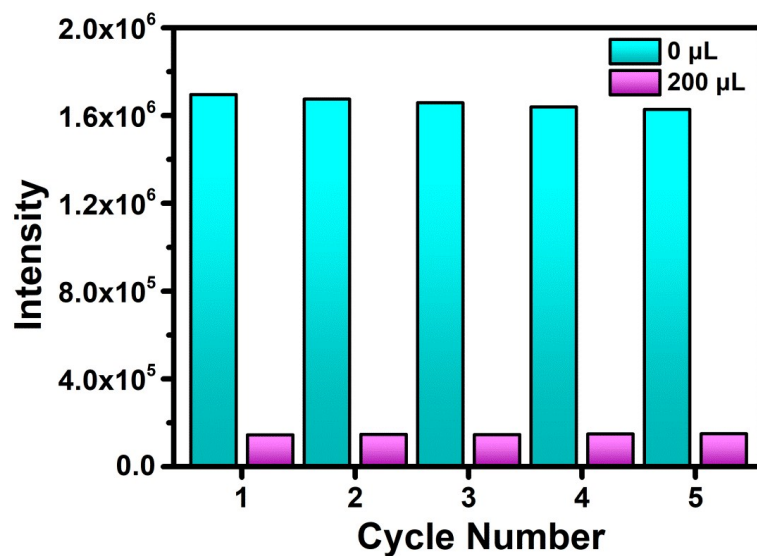


Fig. S13 Cyclic detection of TNP using In-ADBA.

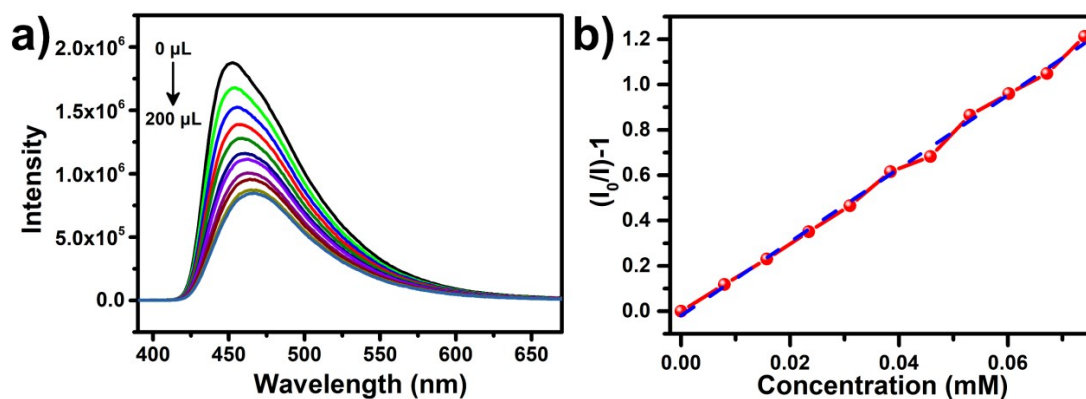


Fig. S14 a) Luminescence spectra of fluorescence-quenching titrations of TNP using H₂ADBA, b) K_{SV} plot.

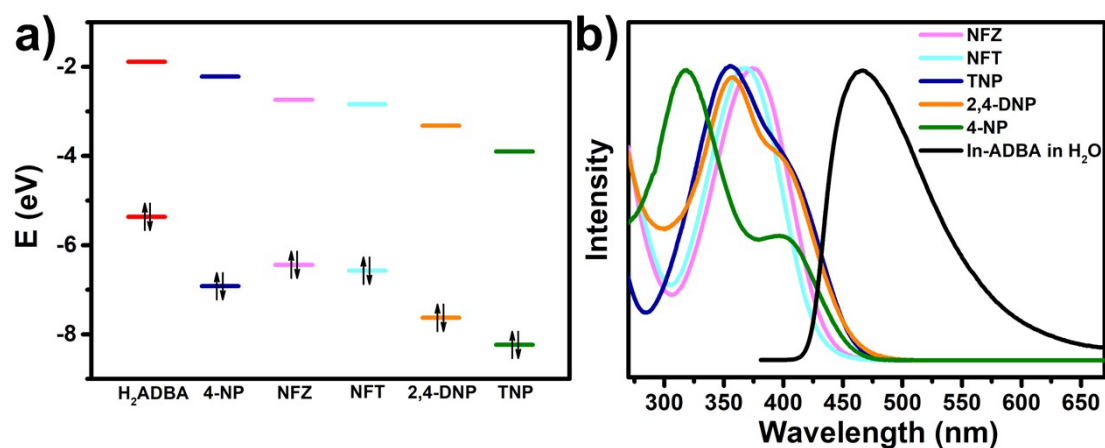


Fig. S15 a) HOMO and LUMO energies of analytes and H₂ADBA ligand, b) Spectral overlap between normalized absorption spectra of analytes and the normalized emission spectra of In-ADBA.

SII. Supporting Tables

Table S1. Crystal data and structure refinement for **Eu-ADBA** and **In-ADBA**.

formula	C ₂₁₇ H ₂₃₇ O ₅₃ N ₁₉ Eu ₆	C ₈₄ H ₆₆ O ₂₅ NIn ₃
formula Weight	4871.51	1833.84
temperature	293(2) K	293(2) K
wavelength	0.71073 Å	0.71073 Å
crystal system, space group	Cubic, $Fm\bar{3}m$	Hexagonal, $P6_3/mcm$
a (Å)	33.816(4)	15.306(2)
b (Å)	33.816(4)	15.306(2)
c (Å)	33.816(4)	28.885(6)
V (Å ³)	38670(8)	5860.1(17)
Z , D_c (Mg/M ³)	220, 3.674	2, 1.039
F (000)	37400	1848
θ range (deg)	1.04~25.06	3.01~25.74
reflns collected/unique	63145/1759	30185/2043
R_{int}	0.0508	0.0420
data/restraints/params	1759/46/53	2043/61/100
GOF on F^2	1.096	1.150
R_I , wR_2 ($I > 2 \sigma(I)$)	$R_I=0.0541$, $wR_2=0.1576$	$R_I=0.0953$, $wR_2=0.2334$
R_I , wR_2 (all data)	$R_I=0.0591$, $wR_2=0.1623$	$R_I=0.0987$, $wR_2=0.2350$

Table S2. Photocatalytic degradation of RhB, MB and TC in water.

Catalyte	Eg (eV)	Irradiation	C ₀ (mg/L)	C _{Catalyst} (mg/L)	C _{H₂O₂} (M)	T (min)	Efficiency (%)	k (min ⁻¹)	Ref.
RhB									
NNU-15(Ce)	2.11	Vis	10	600	0.6	12	99	0.2397	1
MIL-53	2.88	Vis	10	400	0.04	50	98	0.0794	2
[Co ₃ (BPT) ₂ (DMF)(bpy)]·DMF	2.1	Vis	23.9	1000	5 drops	120	90	0.0192	3
NNU-36	2.28	Vis	10	375	1	70	96.2	0.0468	4
ST-MOF235	1.98	Vis	19.2	200	0.02	40	100	0.0039	5
NTU-9	1.72	Vis	47.9	500	5 drops	80	100	-	6
[(Co(L) ₂ (H ₂ O) ₆ ·H ₂ O)] _n	1.86	Vis	9.6	3333	0.083	110	83	-	7
[UO ₂ Cd(bipy)(mal) ₂]·H ₂ O	-	UV	20	1000	-	270	94.6	-	8
Ni(1,4-bib) _{1.5} (TPA-Cl ₂)·H ₂ O	3.08	Vis	10	400	-	120	85.7	-	9
g-C ₃ N ₄ /MIL-125(Ti)	3.24	Vis	50	400	-	60	95.2	0.0624	10
Eu-ADBA	2.35	Vis	48	200	0.03	36	98	0.1079	This work
MB									
{[(CH ₃) ₂ NH ₂] ₃ (In ₃ L ₄)}·(solvent) _x	2.89	Vis	18.02	125	-	100	99	0.0438	11
NNU-36	2.28	Vis	10	375	1	80	94.2	0.0315	4
MIL-88A	2.05	Vis	37.4	400	0.004	50	100	0.01	12
[Cu ₂ (mpTZ) ₂ Br ₂]·H ₂ O	3.1	Vis	10	300	0.02	150	99	-	13
[(Co(L) ₂ (H ₂ O) ₆ ·H ₂ O)] _n	1.86	Vis	7.4	3333	0.083	120	99	-	7
{[Ni ^{II} (SalImCy)] ₂ (Cu ^I CN) ₉ }] _n	1.6	Vis	12	750	0.0625	22	99	-	14
NTU-9	1.72	Vis	31.9	500	5 drops	20	100	-	6
Fe ₃ O ₄ @MIL-100(Fe)	-	UV-Vis	40	100	0.01	200	99	0.1042	15
TiO ₂ NS@MIL-100(Fe)	2.87	Vis	50	400	0.02	60	98	0.045	16
ZIF-8/rGO	-	UV	10	500	-	120	51.8	0.00432	17
Eu-ADBA	2.35	Vis	32	200	0.03	30	99	0.1318	This work
TC									
C ₅₂ H ₃₄ N ₈ O ₃₄ U ₄	2.25	Vis	40	1000	-	300	75.8	0.00485	18
C ₂₉ H ₂₂ N ₂ O ₁₂ U ₂	2.64	Vis	40	1000	-	180	88.82	-	19
C ₃ N ₄ -ZIF-8	2.86	Vis	88.8	100	-	60	96	-	20
In ₂ S ₃ @MIL-125(Ti)	2.28	Vis	46	300	-	60	63.3	-	21
Eu-ADBA	2.35	Vis	48	200	0.03	34	98	0.1123	This work

“-” : The article does not list the data.

Table S3. Luminescent detection of nitroaromatics and antibiotics by MOFs.

MOFs	Solvent	Efficiency (%)	K_{SV} (M^{-1})	Ref.
TNP				
BUT-13	H ₂ O	96	5.1×10^5	22
BUT-12	H ₂ O	98	3.1×10^5	22
$\{[Cd_{1.5}(TPO)(bipy)_{1.5}] \cdot 3H_2O\}_{2n}$	H ₂ O	92	1.4×10^5	23
$\{[Cd_4(L)_2(L_2)_3(H_2O)_2](8DMF)(8H_2O)\}_n$	C ₂ H ₅ OH	94.28	3.89×10^4	24
$[Cd(NDC)_{0.5}(PCA)] \cdot G_x$	CH ₃ CN	78	3.5×10^4	25
$\{Zn_5Na_2(BPTC)_4\}_n$	H ₂ O	96.7	3.2×10^4	26
$\{[Zn_2(L)_2(azp)] \cdot (DMF)_2 \cdot (H_2O)\}_n$	H ₂ O	89.8	3.11×10^4	27
bio-MOF-1	H ₂ O	93	4.6×10^4	28
UiO-67@N	H ₂ O	73	2.9×10^4	29
UiO-68-mtpdc/etpdc	CH ₃ OH	92	2.8×10^4	30
$[Cu(L)(I)]_{2n} \cdot 2nDMF \cdot nMeCN$	CH ₃ CN	65	2.9×10^4	31
$[Tb_2(H_2L)_3(H_2O)_2] \cdot 21H_2O$	H ₂ O	89.4	1.5×10^4	32
In-ADBA	H ₂ O	92	1.282×10^5	This work
2,4-DNP				
UiO-68-mtpdc/etpdc	CH ₃ OH	76	2.3×10^4	30
In-ADBA	H ₂ O	91	8.994×10^4	This work
4-NP				
UPC-21	DMSO	98	3.097×10^6	33
BUT-13	H ₂ O	95	4.7×10^4	22
BUT-12	H ₂ O	97	4.2×10^4	22
$[Zn_2(TPOM)(NH_2-BDC)_2] \cdot 4H_2O$	DMF	90	2.17×10^4	34
UPC-17	THF	75	1.26×10^4	24
$[Zn(L)(H_2O)] \cdot H_2O$	H ₂ O	-	1.25×10^4	35
UiO-68-mtpdc/etpdc	CH ₃ OH	42	7.2×10^3	30
In-ADBA	H ₂ O	90	5.122×10^4	This work
NFZ				
CTGU-8	H ₂ O	95	1.83×10^6	36
BUT-13	H ₂ O	95	7.5×10^4	22
BUT-12	H ₂ O	92	1.1×10^5	22
$[Cd_2Na(L)(BDC)_{2.5}] \cdot 9H_2O$	DMF	93	5.06×10^4	37
$[Cd_2(L)(2,6-NDC)_2] \cdot DMF \cdot 5H_2O$	DMF	99	1.04×10^5	37
$[Cd_2(L)(BPDC)_2] \cdot DMF \cdot 9H_2O$	DMF	98	1.33×10^5	37
In-ADBA	H ₂ O	89	7.605×10^4	This work

NFT				
CTGU-8	H ₂ O	95	9.25×10 ⁵	36
BUT-13	H ₂ O	94	6.0×10 ⁴	22
BUT-12	H ₂ O	91	3.8×10 ⁴	22
[Cd ₂ Na(L)(BDC) _{2.5}]·9H ₂ O	DMF	83	3.57×10 ⁴	37
[Cd ₂ (L)(2,6-NDC) ₂]·DMF·5H ₂ O	DMF	96	7.19×10 ⁴	37
[Cd ₂ (L)(BPDC) ₂]·DMF·9H ₂ O	DMF	96	6.96×10 ⁴	37
In-ADBA	H ₂ O	87	5.142 × 10 ⁴	This work

“-” : The article do not list the data.

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