

Defect Modulation on $\text{CaZn}_{1-x}\text{Ag}_{1-y}\text{Sb}$ ($0 < x < 1$; $0 < y < 1$) Zintl phases and Enhanced Thermoelectric Properties with High zT Plateaus

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1. The calculation formula of Lorenz number.

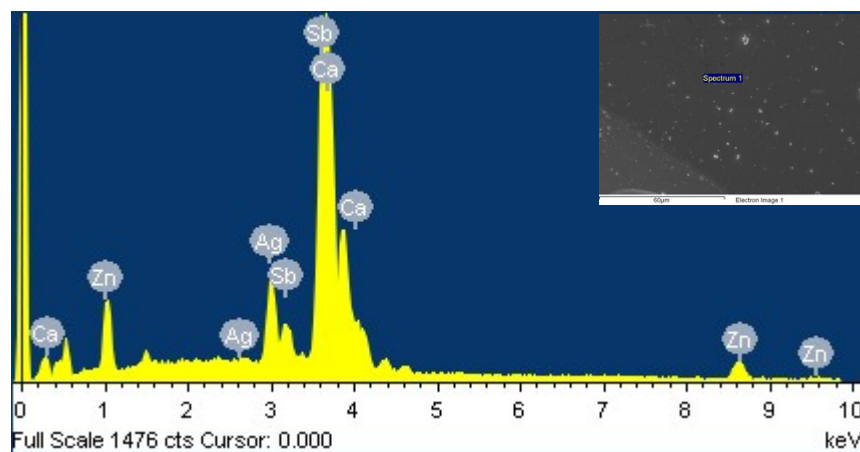
For most thermoelectric materials, the actual Lorenz number depends on the reduced Fermi energy $\eta = E_f/k_B T$ and the scattering parameter r , and is determined by Eq (1), in which $F_n(\eta)$ is the Fermi integration, k_B is the Boltzmann constant, e is the electron charge, and E_f is the Fermi energy. Eq (2) shows the value of L_0 for free electrons. The reduced Fermi energy η can be derived from the Seebeck coefficient by Eq (3).

$$L = \left(\frac{k_B}{e} \right)^2 \left(\frac{(r+7/2)F_{r+5/2}(\eta)}{(r+3/2)F_{r+1/2}(\eta)} - \left[\frac{(r+5/2)F_{r+3/2}(\eta)}{(r+3/2)F_{r+1/2}(\eta)} \right] \right) \quad (1)$$

$$L_0 = \frac{\pi^2}{3} \left(\frac{k_B}{e} \right)^2 \cong 2.45 \times 10^{-8} W \cdot \Omega \cdot K^{-2} \quad (2)$$

$$S = \pm \frac{k_B}{e} \left(\frac{(r+5/2)F_{r+3/2}(\eta)}{(r+3/2)F_{r+1/2}(\eta)} - \eta \right) \quad (3)$$

2. EDS analysis on the single crystals of $\text{CaZn}_{0.40(2)}\text{Ag}_{0.20(2)}\text{Sb}$.



Sample 1

Element	Weight%	Atomic%	Composition
Ca K	16.36	34.41	1.00
Zn L	11.37	14.66	0.42
Ag L	10.00	7.81	0.22
Sb L	62.27	43.11	1.25

Sample 2

Element	Weight%	Atomic%	Composition
Ca K	17.51	36.17	1.00
Zn L	11.57	14.66	0.40
Ag L	10.69	8.21	0.23
Sb L	60.23	40.96	1.12

Sample 3

Element	Weight%	Atomic%	Composition
Ca K	17.69	36.36	1.00
Zn L	12.04	15.17	0.41
Ag L	10.74	8.20	0.23
Sb L	59.53	40.27	1.11

Sample 4

Element	Weight%	Atomic%	Composition
Ca K	17.88	36.80	1.00
Zn L	11.22	14.16	0.38
Ag L	11.28	8.63	0.24
Sb L	59.62	40.40	1.09

Averaged Composition: $\text{CaZn}_{0.40}\text{Ag}_{0.23}\text{Sb}_{1.17}$

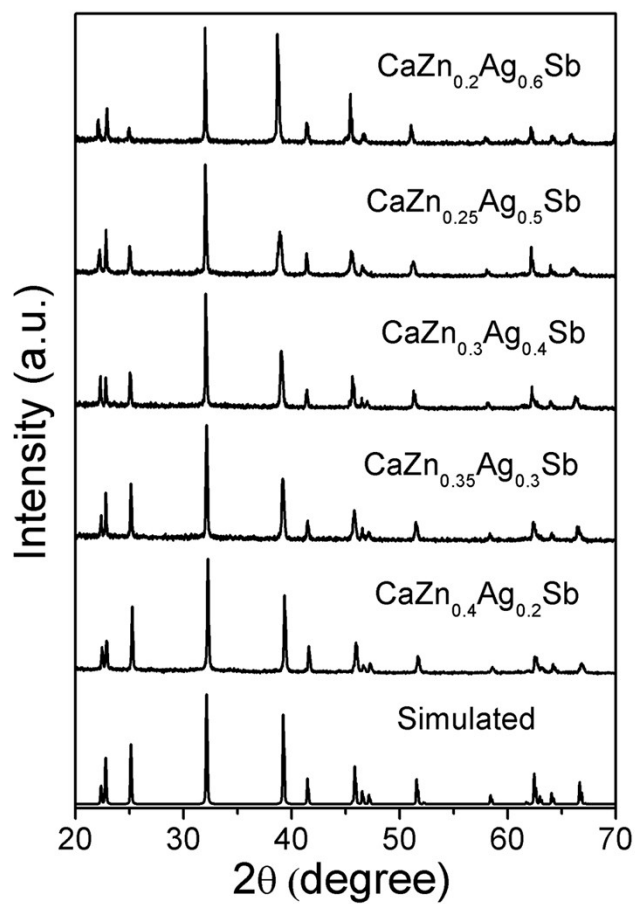


Figure S1. Powder X-ray diffraction patterns for materials with nominal compositions of $\text{CaZn}_{1-x}\text{Ag}_y\text{Sb}$ ($0 < x < 1$; $0 < y < 1$). The simulated pattern of $\text{CaZn}_{0.40(2)}\text{Ag}_{0.20(2)}\text{Sb}$ is provided for compared.

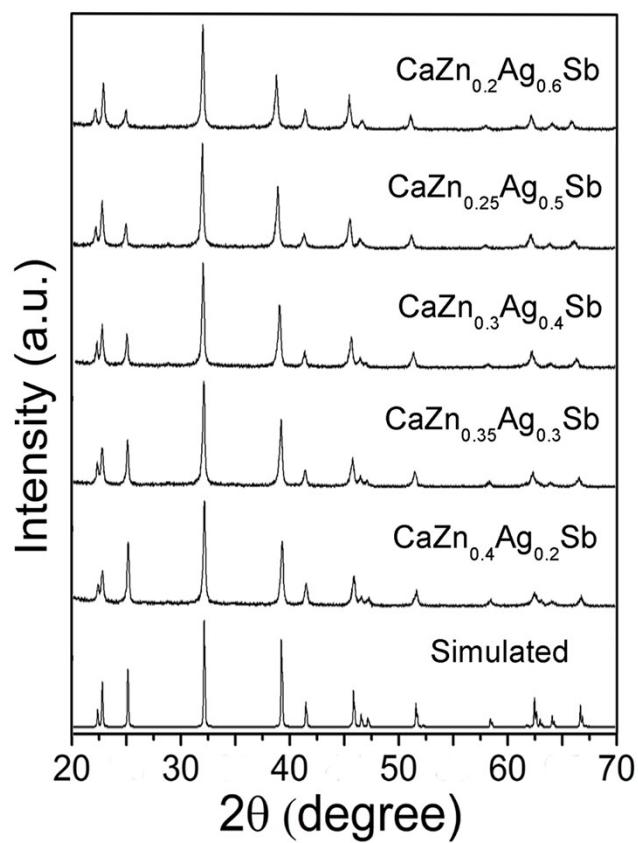


Figure S2. Powder X-ray diffraction patterns for materials with nominal compositions of $\text{CaZn}_{1-x}\text{Ag}_{1-y}\text{Sb}$ ($0 < x < 1$; $0 < y < 1$) after SPS experiments.

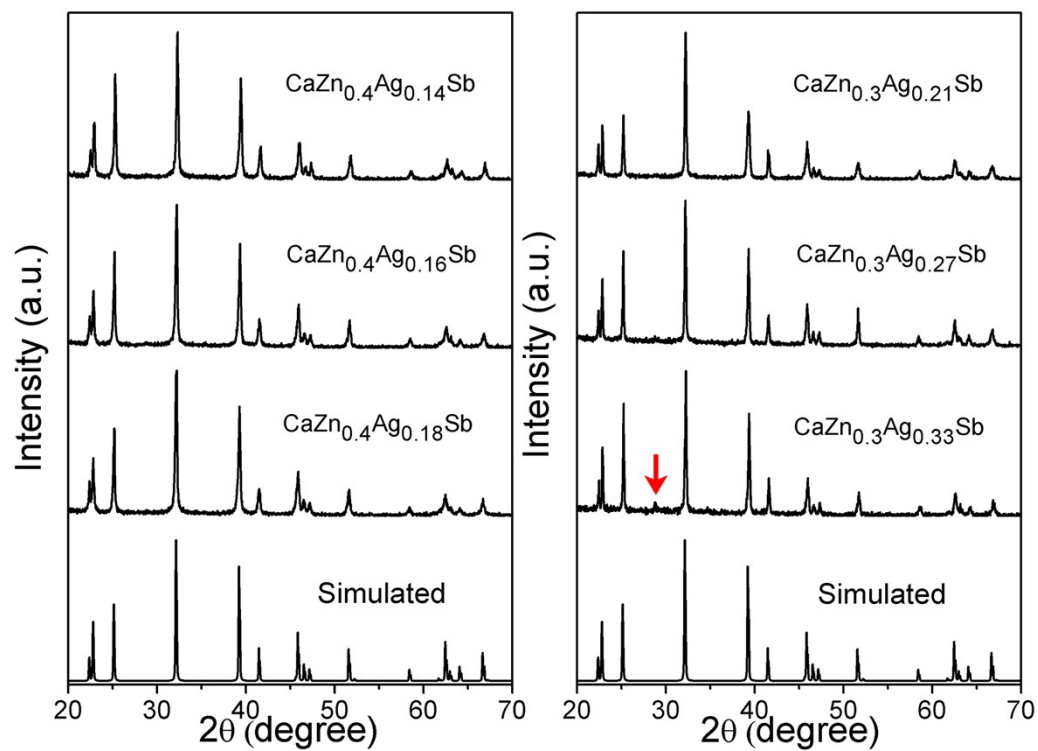


Figure S3. Powder X-ray diffraction patterns for materials with nominal compositions of $\text{CaZn}_{0.4}\text{Ag}_{1-y}\text{Sb}$ ($0 < y < 1$) and $\text{CaZn}_{0.3}\text{Ag}_{1-y}\text{Sb}$ ($0 < y < 1$). The peak of suspicious impurities is indicated by a red arrow and the simulated pattern of $\text{CaZn}_{0.40(2)}\text{Ag}_{0.20(2)}\text{Sb}$ is provided for compared.

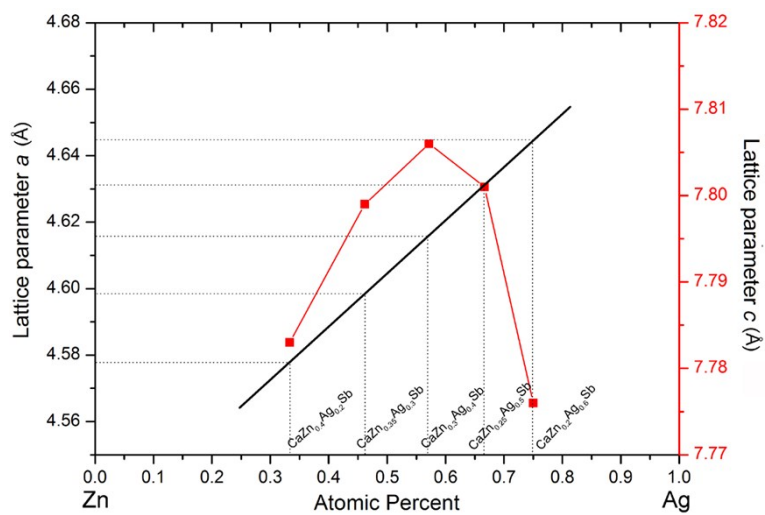


Figure S4. Plot of composition vs. lattice parameter of $\text{CaZn}_{(1-x)/2}\text{Ag}_x\text{Sb}$ ($x = 0.2, 0.3, 0.4, 0.5, 0.6$).

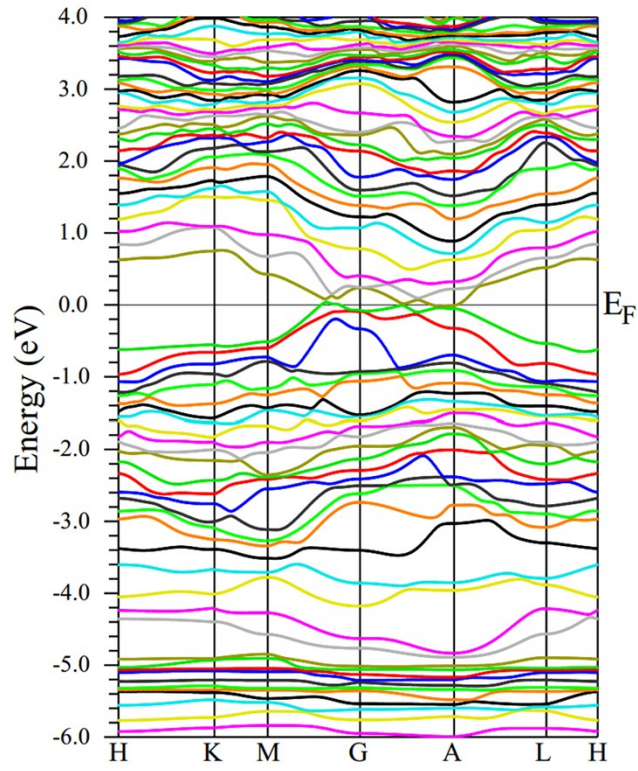


Figure S5. Calculated electronic band structures of the $\text{CaZn}_{0.40(2)}\text{Ag}_{0.20(2)}\text{Sb}$. Fermi level is chosen as the energy reference.

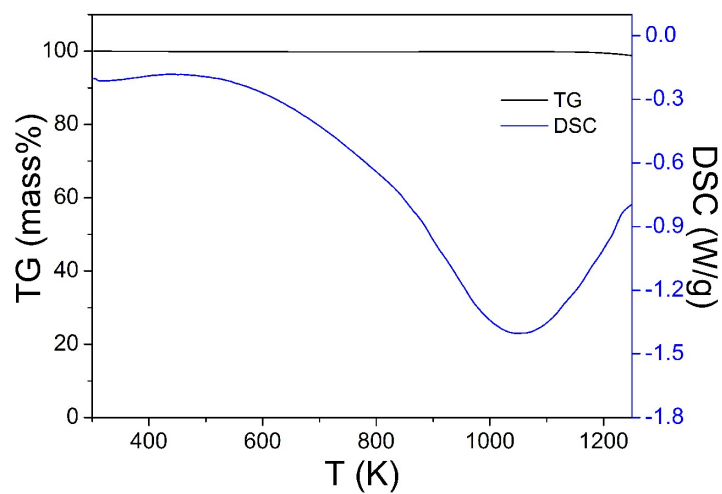


Figure S6. TG-DSC measurements on $\text{CaZn}_{0.40(2)}\text{Ag}_{0.20(2)}\text{Sb}$, conducted in inert argon atmosphere over the temperature range of 300–1273 K.

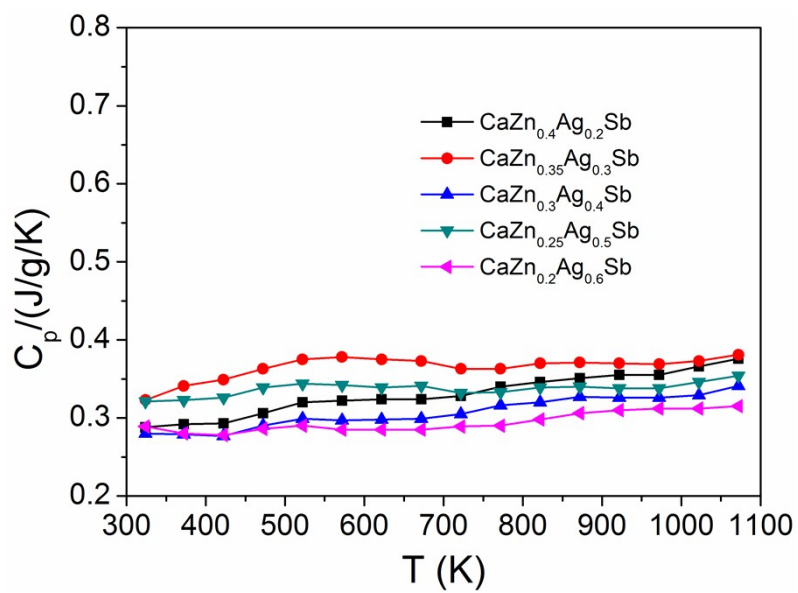


Figure S7. Temperature dependence of the heat capacity of materials $\text{CaZn}_{1-x}\text{Ag}_y\text{Sb}$ ($0 < x < 1$; $0 < y < 1$).

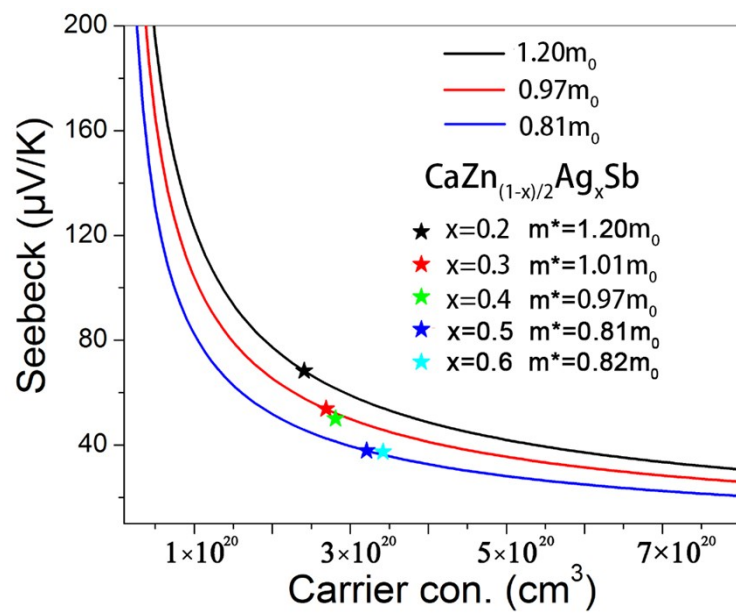


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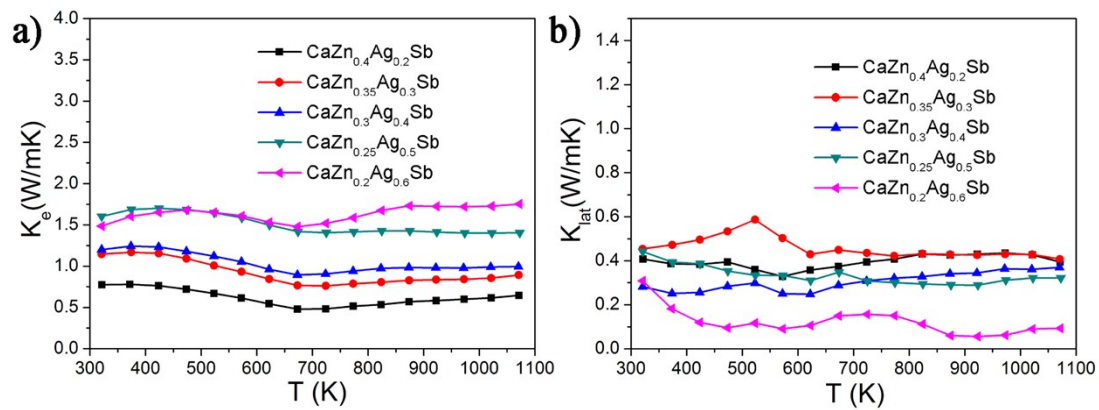


Figure S9. Temperature dependence of (a) electronic thermal conductivity and (b) lattice thermal conductivity of $\text{CaZn}_{(1-x)/2}\text{Ag}_x\text{Sb}$ ($x = 0.2, 0.3, 0.4, 0.5, 0.6$) samples.

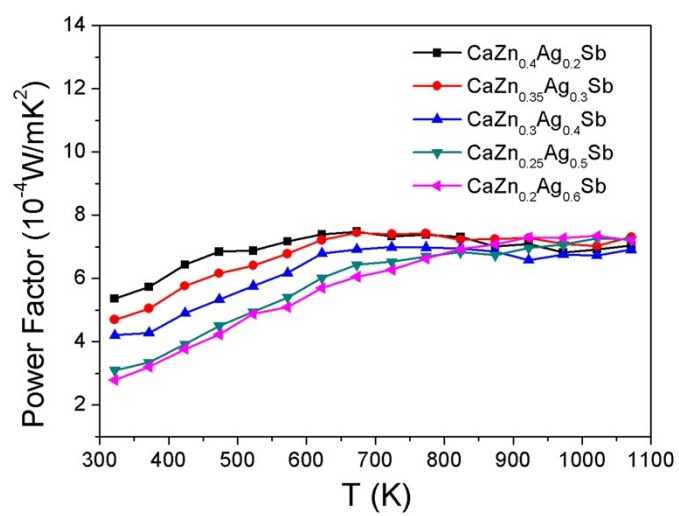


Figure S10. Temperature dependence of power factor $S^2\sigma$ for $\text{CaZn}_{1-x}\text{Ag}_y\text{Sb}$ ($0 < x < 1$; $0 < y < 1$).

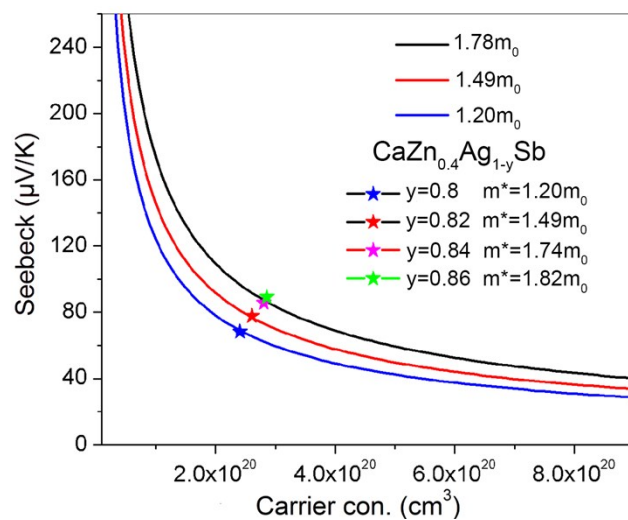


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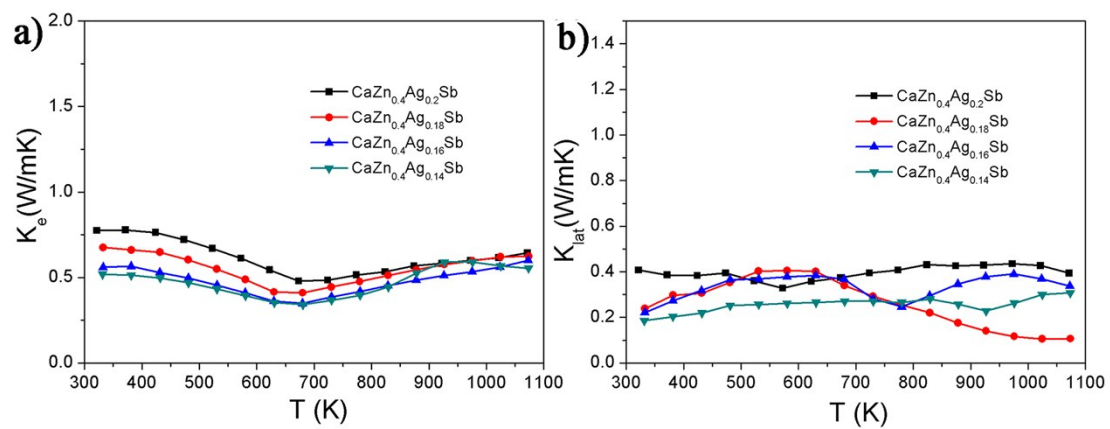


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