

## Supporting Information for

### **Ultrafast lithium energy storage enabled by interfacial constructing interlayer-expanded MoS<sub>2</sub>/N-doped carbon nanowires**

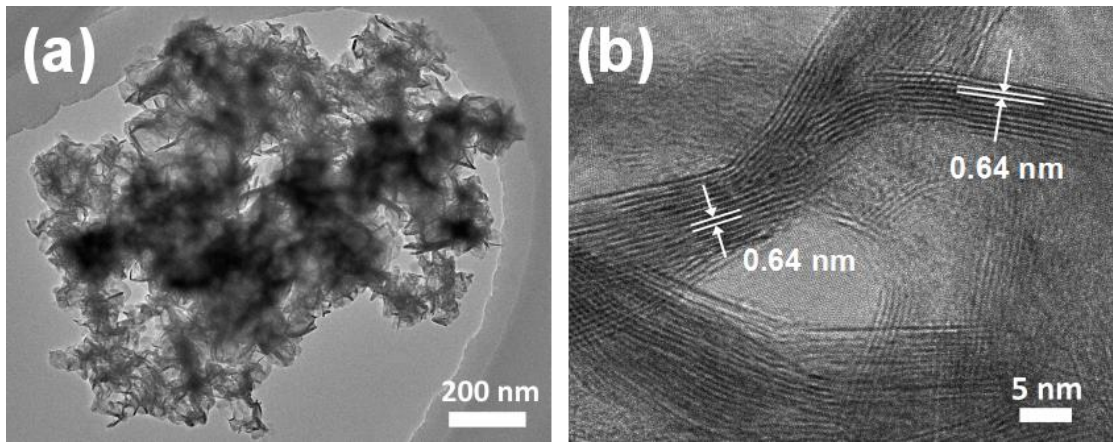
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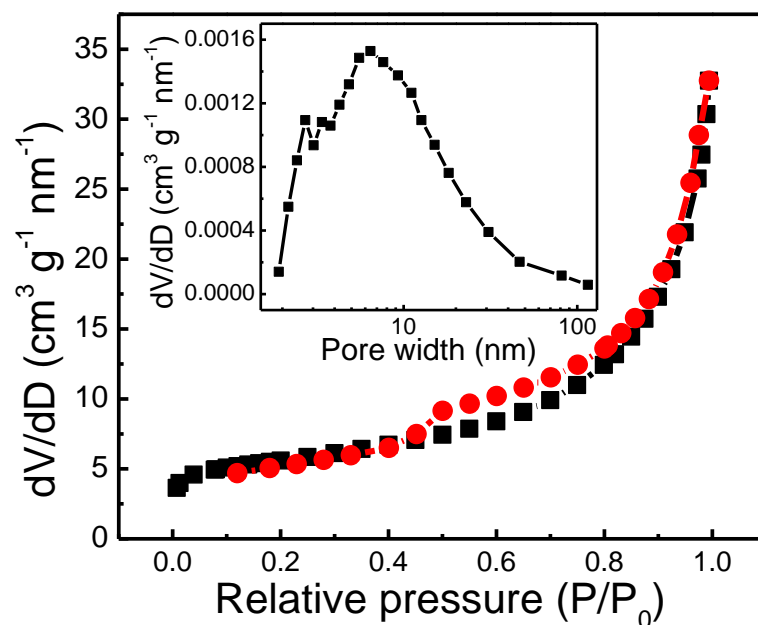
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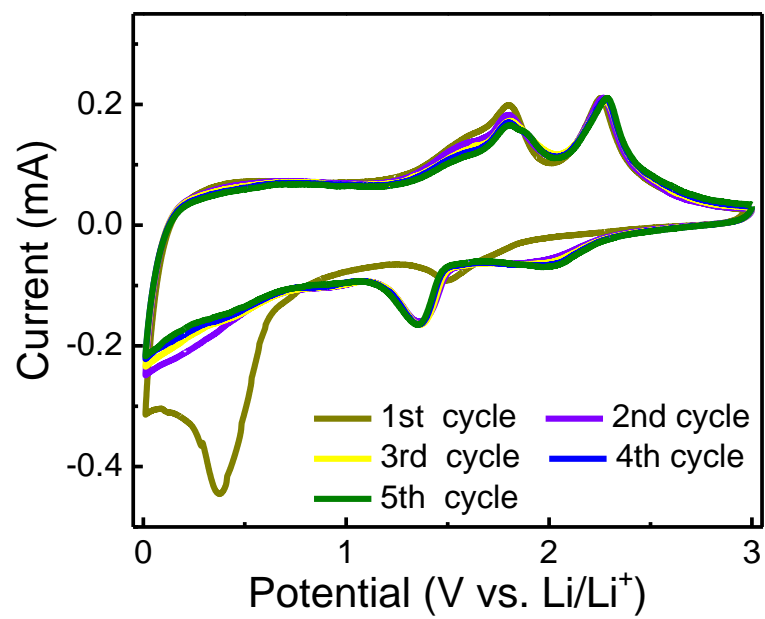
wangjiangan@nwpu.edu.cn (J.-G. Wang)



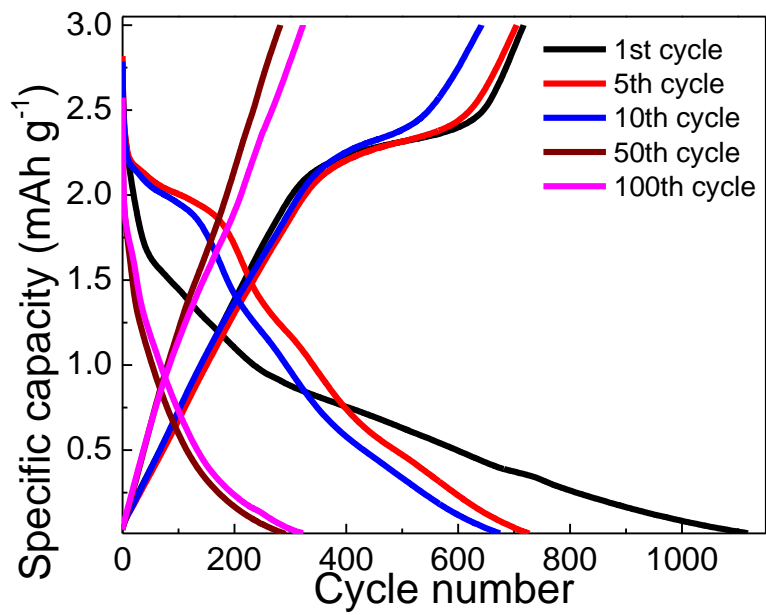
**Fig. S1.** TEM and HRTEM images of pure MoS<sub>2</sub>.



**Fig. S2.** N<sub>2</sub> adsorption-desorption isotherms and pore size distribution (the inset) of pure MoS<sub>2</sub>.

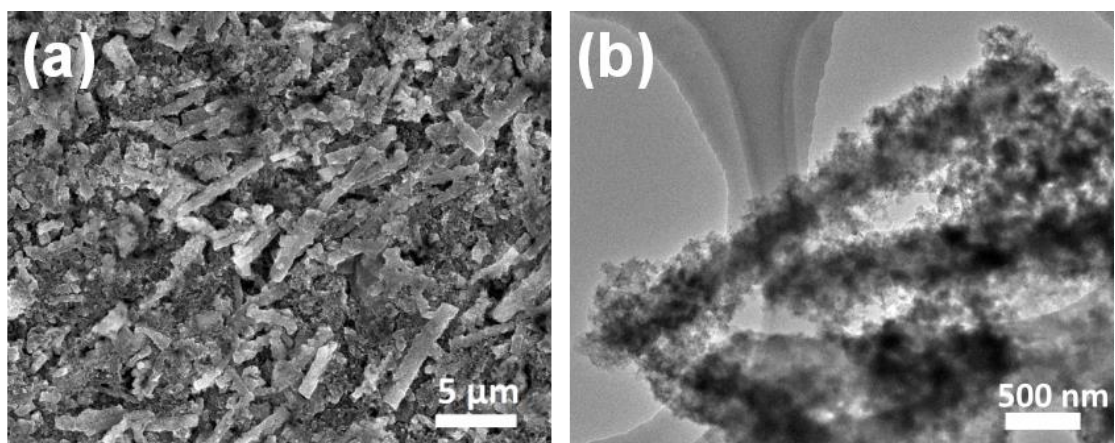


**Fig. S3.** CV curves of pure MoS<sub>2</sub> electrode in the first five cycles.



**Fig. S4.** Galvanostatic charge/discharge profiles of pure MoS<sub>2</sub> electrode at different cycles.

μ



**Fig. S5.** (a) SEM and (b) TEM images of images of MoS<sub>2</sub>/N-C NWs after 100 cycles.

**Table S1.** Li-ion storage comparison of MoS<sub>2</sub>-based materials.

Sample	Specific Capacity at 0.1 A g <sup>-1</sup> (mAh g <sup>-1</sup> )	Rate capability at different current densities relative to the initial value				Ref.
		1 A g <sup>-1</sup>	2 A g <sup>-1</sup>	5 A g <sup>-1</sup>	10 A g <sup>-1</sup>	
TiO <sub>2</sub> @Carbon@MoS <sub>2</sub>	925	81.7 %	72.4 %	-	-	1
MoS <sub>2</sub> /CMK-3	893	79.8%	66.2%	-	-	2
MoS <sub>2</sub> /N-Carbon	1299	46.7 %	42.7 %	38 %	33.1 %	3
MoS <sub>2</sub> /C	820	62.2 %	55.7 %	-	-	4
MoS <sub>2</sub> /C	944	80.7%	75.3%	61%	-	5
MoS <sub>2</sub> /N-rGO film	1109	78 %	70.8 %	64 %	-	6
rGO/MoS <sub>2</sub> /N-rGO	770	76.5 %	68.6 %	-	-	7
MoS <sub>2</sub> / rGO	1077	82.6 %	-	-	-	8
MoS <sub>2</sub> @carbon	1110	65.3 %	54.1 %	40.5 %	-	9
MoS <sub>2</sub> /graphene	825	69.2%	-	-	-	10
MoS <sub>2</sub> /C	1127	78 %	62.9 %	42.6 %	22.2 %	11
MoS <sub>2</sub> /graphene	854	73%	-	51.1%	38.8%	12
MoS <sub>2</sub> /TiO <sub>2</sub> nanosheet	990	70.7 %	-	57.1 %	-	13
MoS <sub>2</sub> /Carbon	900	66.7 %	50.5 %	-	-	14
MoS <sub>2</sub> /C	900	77.8%	-	-	-	15
MoS <sub>2</sub> /C	653	78.5 %	67.3 %	51.2 %	-	16
MoS <sub>2</sub> @MoO <sub>3</sub>	929	41.3 %	-	-	-	17
MoS <sub>2</sub> /N-C	840	83.7 %	76.7 %	71.4 %	53.9 %	This work

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