

Supplementary Materials

Hexagonal Co₃O₄ anchored reduced graphene oxide sheets for high-performance supercapacitors and non-enzymatic glucose sensing[†]

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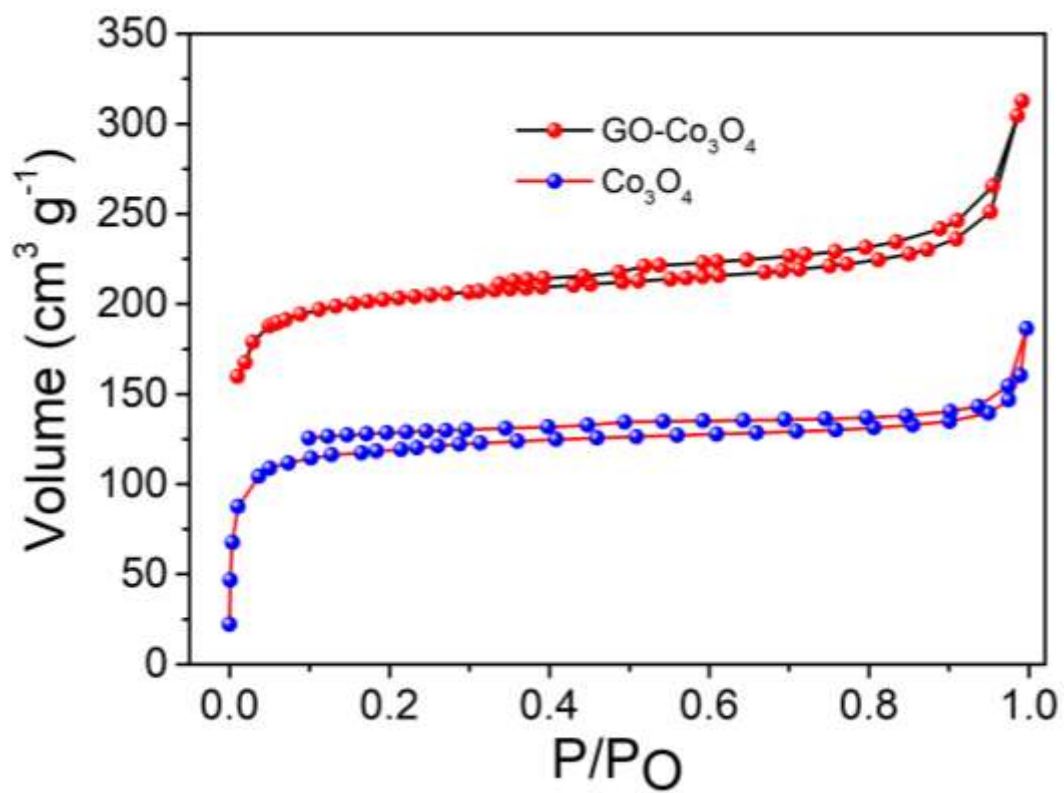


Fig. S1. Adsorption-desorption isotherms of GO-Co₃O₄ and Co₃O₄.

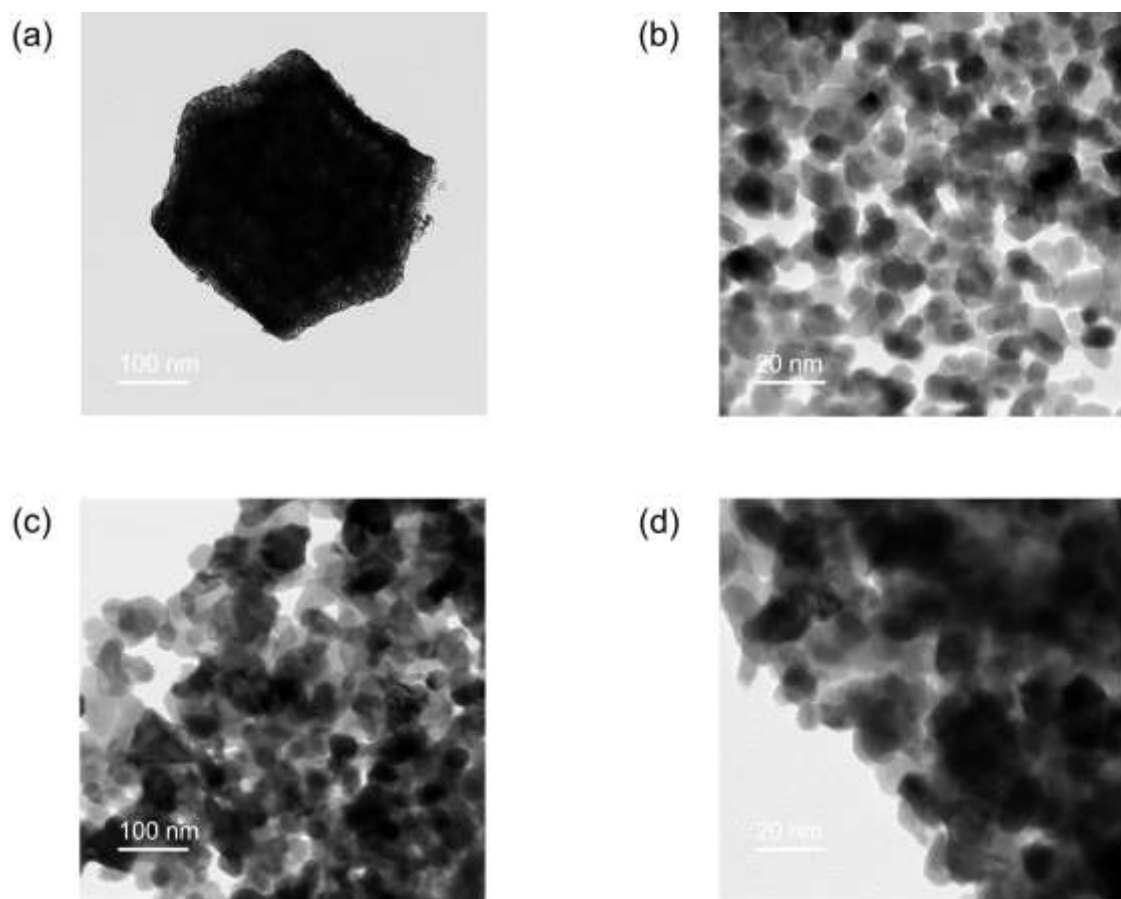


Fig. S2. TEM images of (a) Co_3O_4 and (b) RGO- Co_3O_4 sheets at different magnifications.

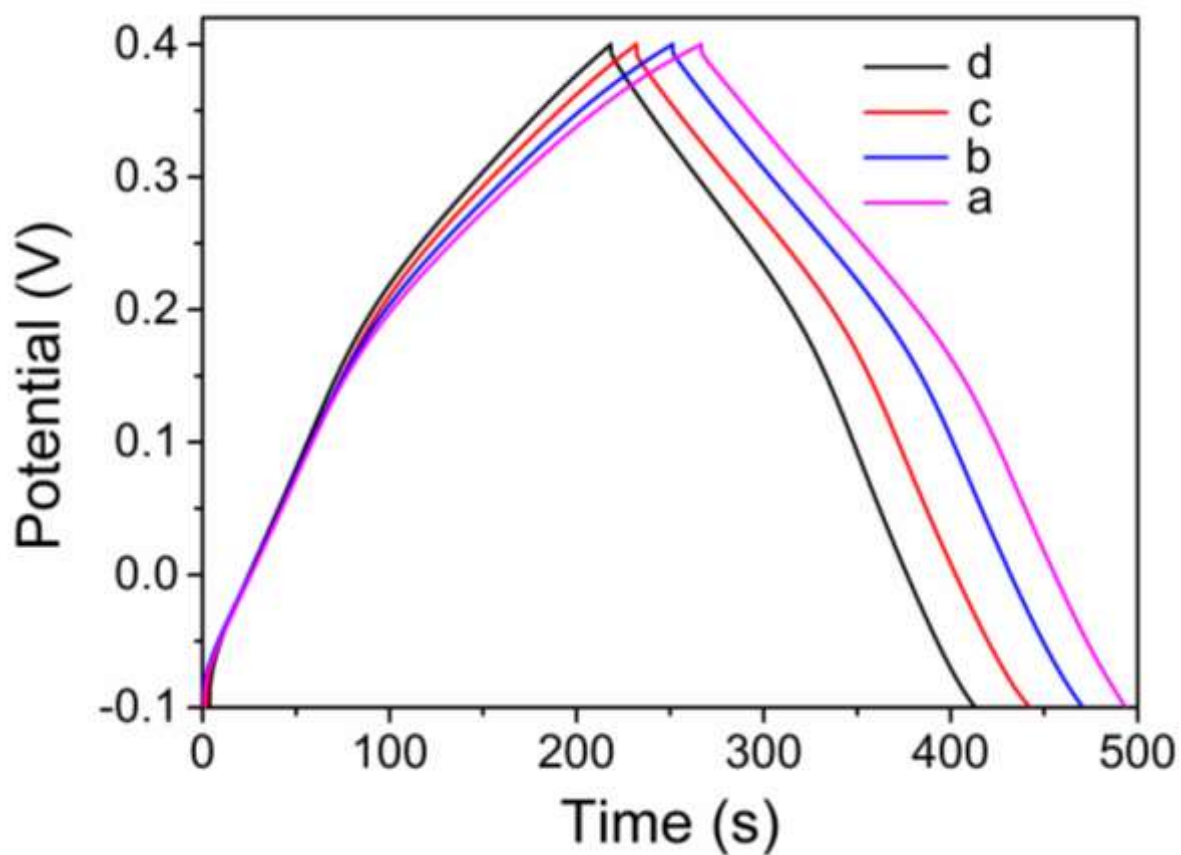


Fig. S3. (a) Galvanostatic charge-discharge curves of RGO-Co₃O₄ at a various mass loadings (a-d; 0.2, 0.6, 1.3 mg cm⁻²) at 4 A g⁻¹.

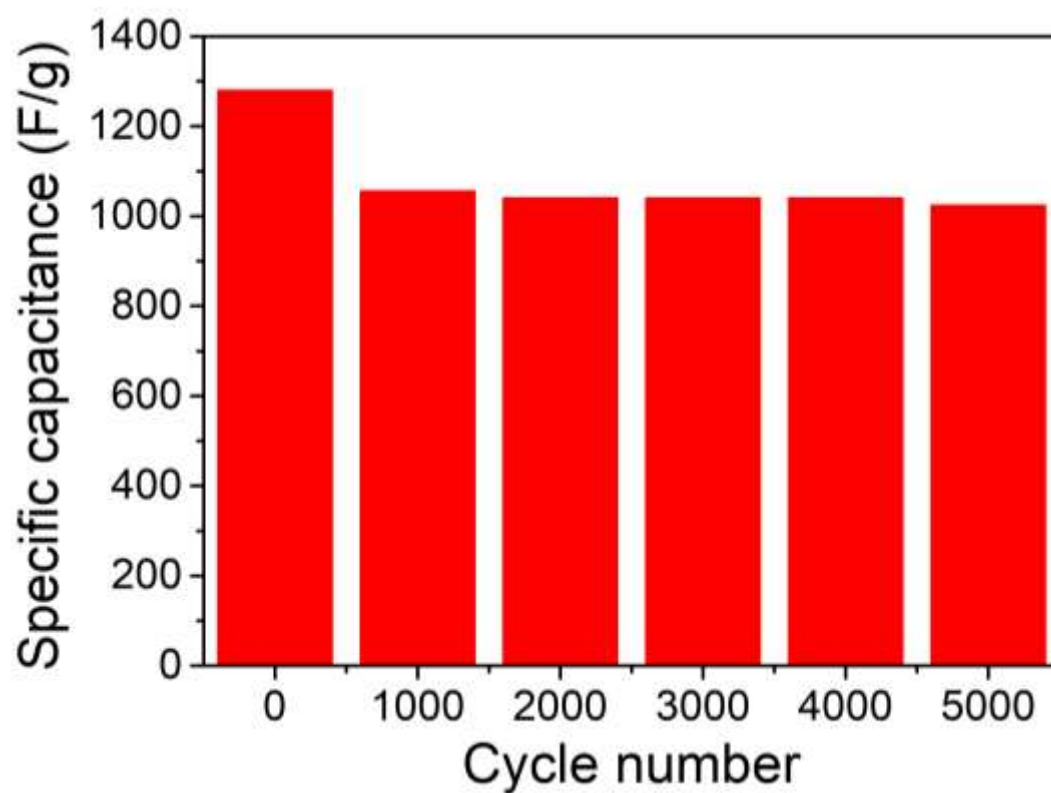


Fig. S4. A Plot between specific capacitance and respective cycles of RGO-Co₃O₄ at an applied current density of 4A/g.

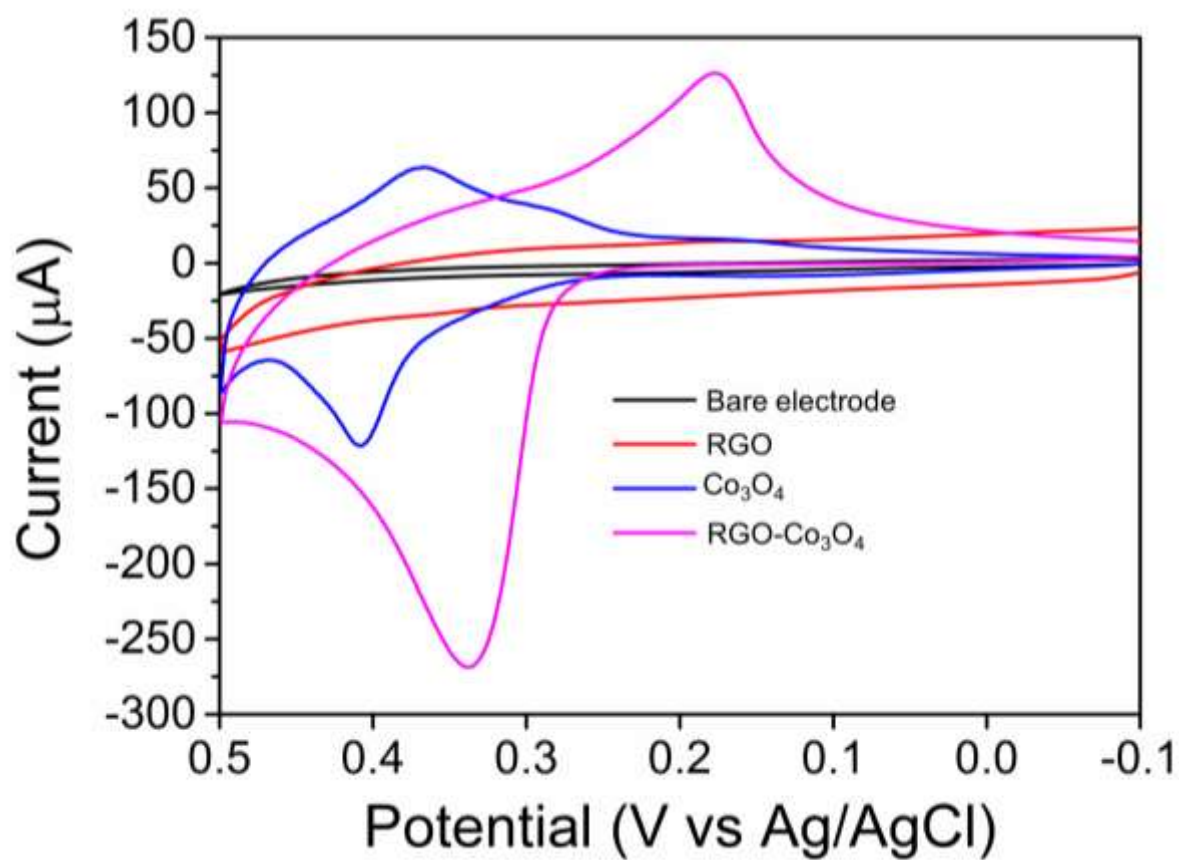


Fig. S5. CVs obtained for 4 mM glucose at different electrodes in 0.1 M KOH at a scan rate of 50 mV s^{-1} .

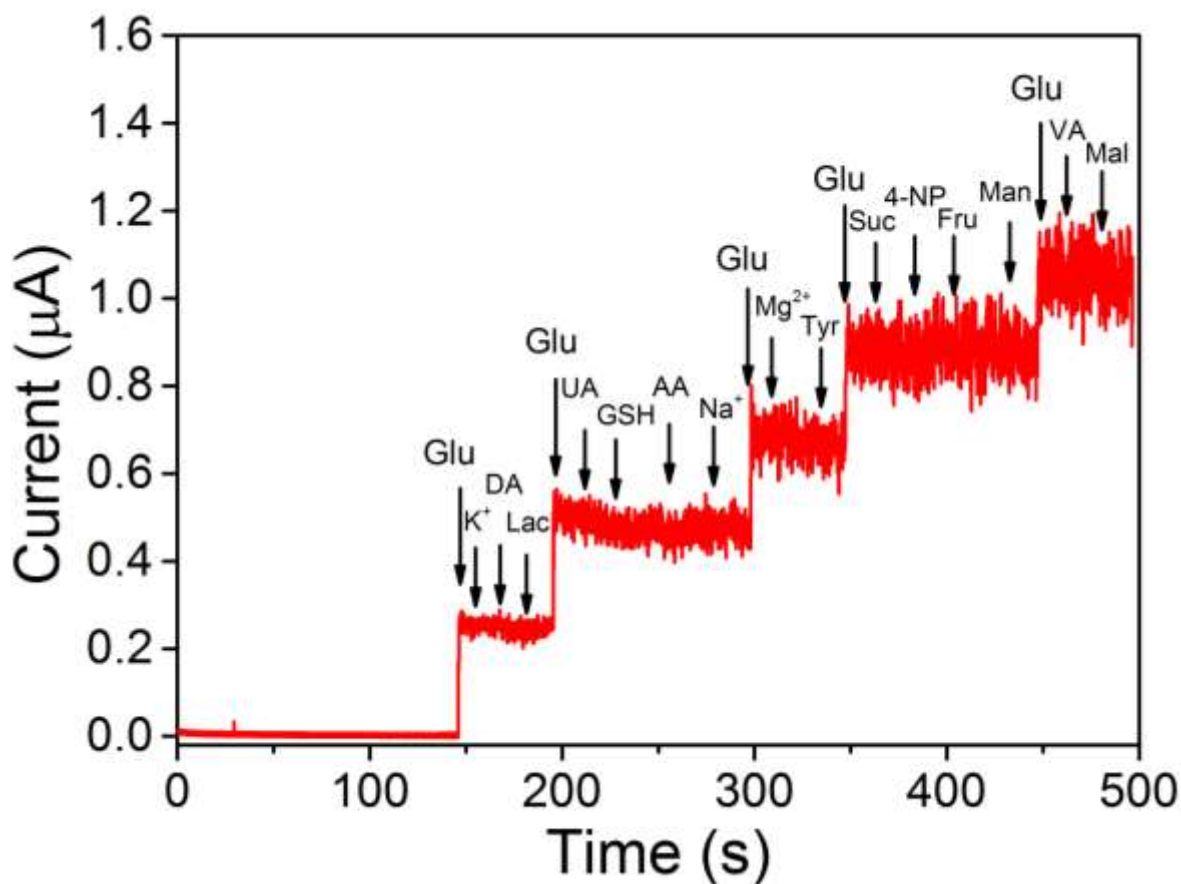


Fig. S6. Amperometric i-t response at RGO-Co₃O₄ modified electrode upon addition of 1 μM of glucose, Sucrose, Fructose, D-Mannose, glutathione, ascorbic acid, uric acid, D-maltose, dopamine, 4-nitrophenol, D-Lactose, vanillin, tyrosine and 500 μM of K⁺, Na⁺ and Mg²⁺ to glucose aqueous solutions into continuously stirred N₂ saturated KOH solution (Applied potential: +0.35 V).

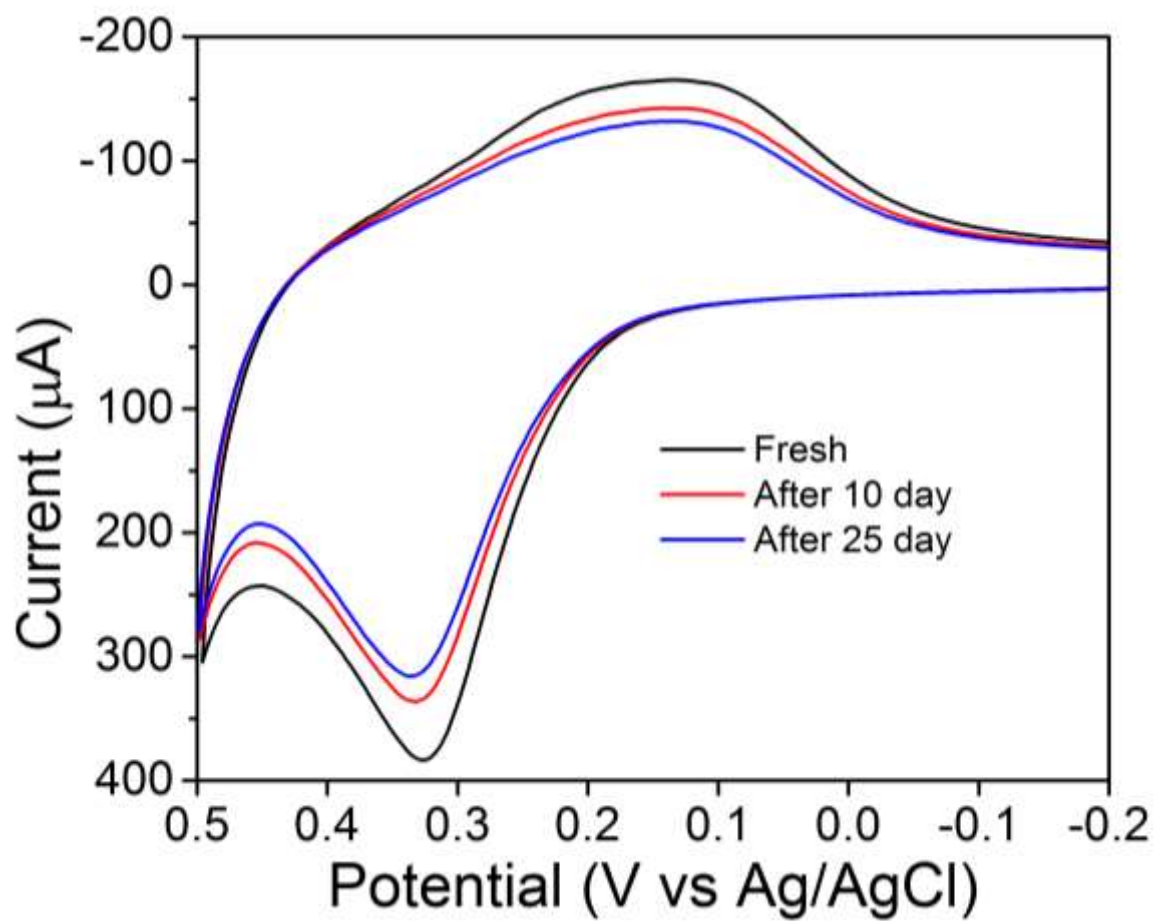


Fig. S7. Long-term stability of RGO-Co₃O₄ sensor measured in 1 mM glucose solution for 25 days.

Table S1 Comparisons of the proposed RGO-Co₃O₄ performance with the previous report non-enzymatic glucose sensors.

Electrode Material	Linear range (μM)	LOD (μM)	Reference
CuCo ₂ O ₄	up to 320	5	1
NiCo ₂ S ₄	1-664	1.2	2
MnCo ₂ O ₄	20-100	3.2	3
NiOHSs-RGO-NF	0.6246-10,500	0.03	4
NiO/Pt/ERGO	50-5660	0.2	5
RGO-Co ₃ O ₄	1-500	0.4	This work

References

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