
Supporting Information

Towards zero bias photoelectrochemical water splitting: onset potential improvement of Mg:GaN-modified Ta₃N₅ photoanode

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Abstract: Tantalum nitride (Ta₃N₅) based photoanodes were overlaid with magnesium-doped gallium nitride (Mg:GaN) thin films by plasma-enhanced chemical vapor deposition (PCVD) technique, and subjected to photoelectrochemical activity tests, aiming for a negative shift of onset potential for O₂ evolution. A remarkable negative shift of the onset potential was observed after annealing Mg:GaN in N₂ gas, reaching 0 V vs RHE, despite a lower photocurrent than that on bare Ta₃N₅. Mg:GaN annealed in NH₃ exhibited an improvement of the photocurrent. A detailed study of the photoelectrochemical performance for various samples and a thorough characterization have revealed the effects of N₂/NH₃ post annealing over Mg activation/Ta₃N₅ damage recovery, controlling the onset potential shift and the current density improvement. N₂ post annealing shifted the onset potential to 0 V vs RHE but decreased the current density. On the other hand, NH₃ post annealing slightly shifted the onset potential and increased the current density largely. Albeit the current density loss, this onset potential shift unlocks the prospect of unassisted photoelectrochemical water splitting on Ta₃N₅.

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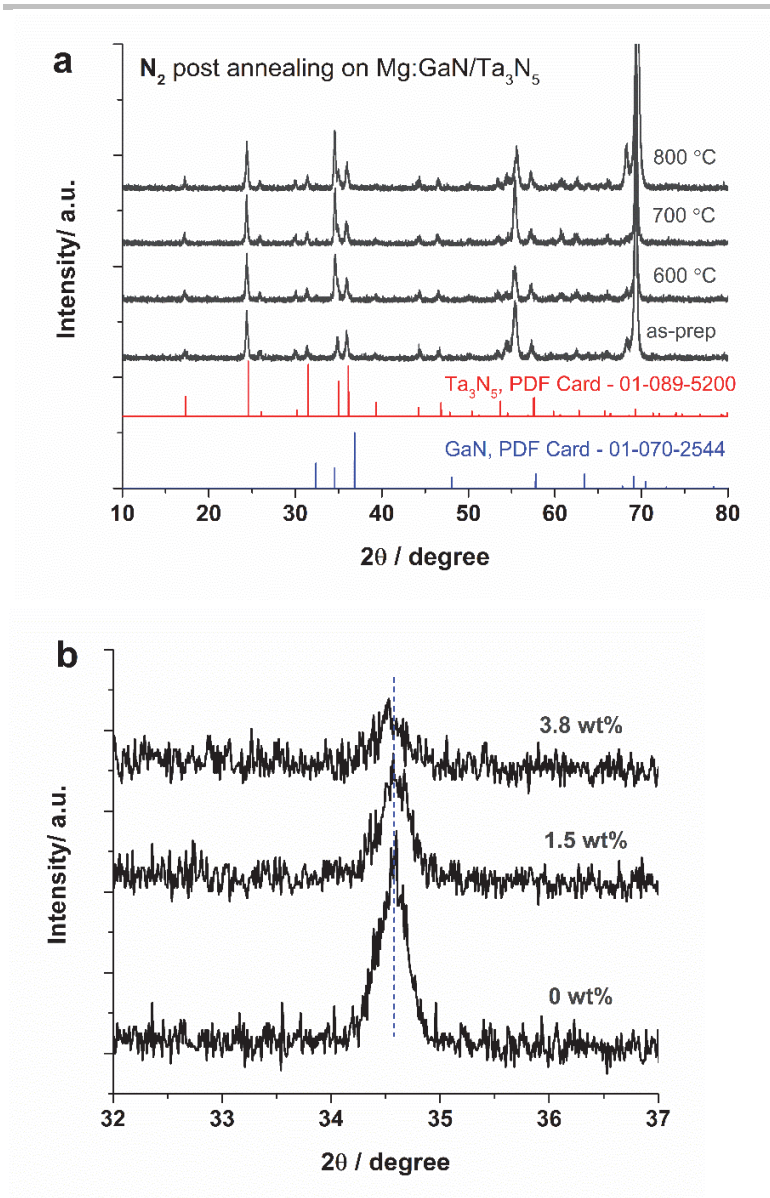
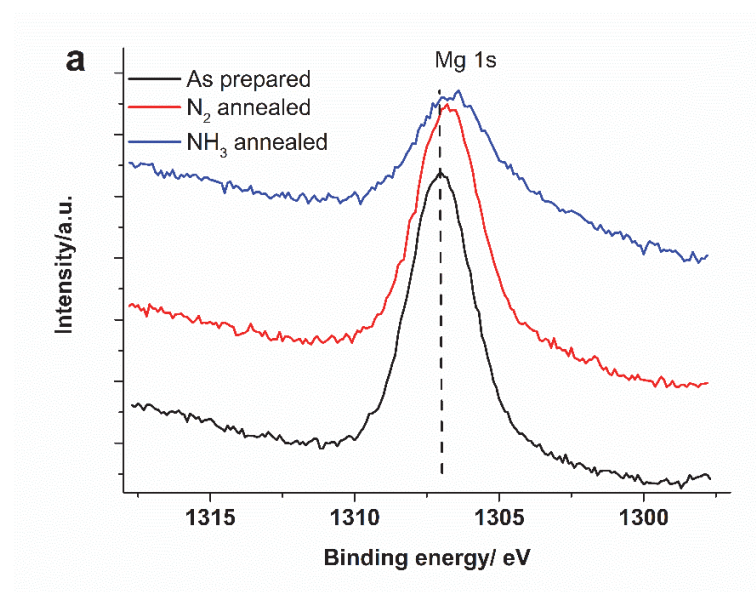
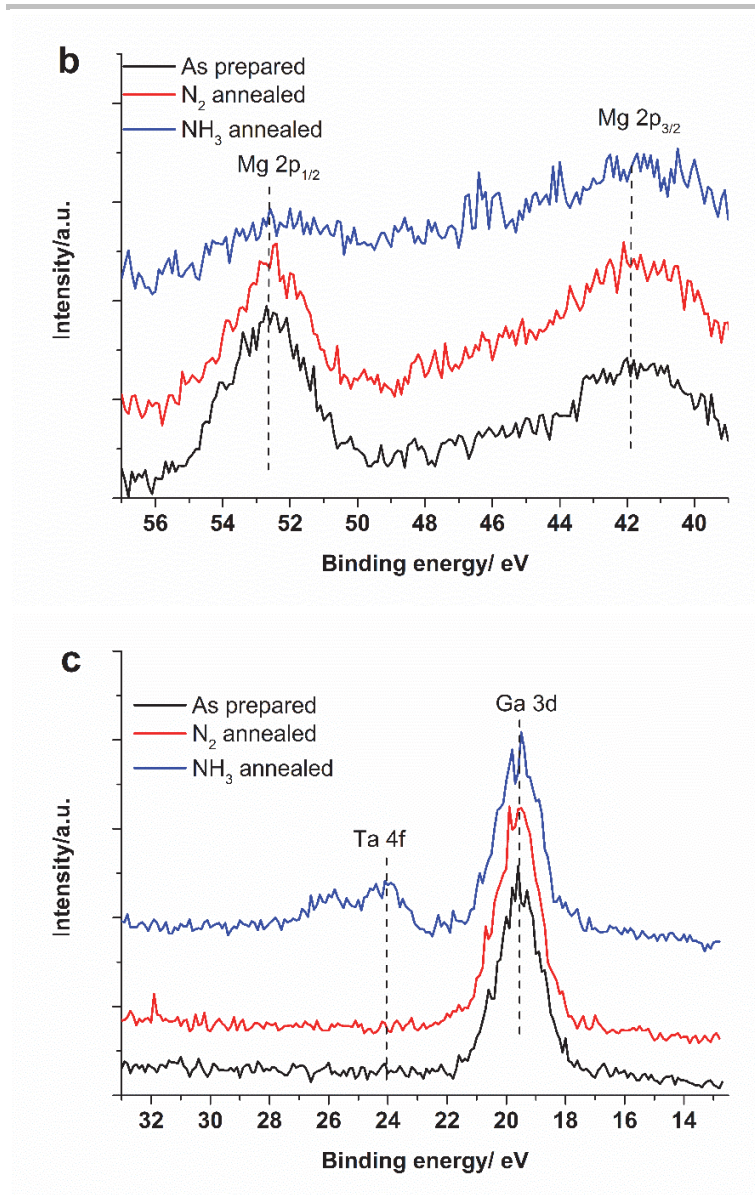


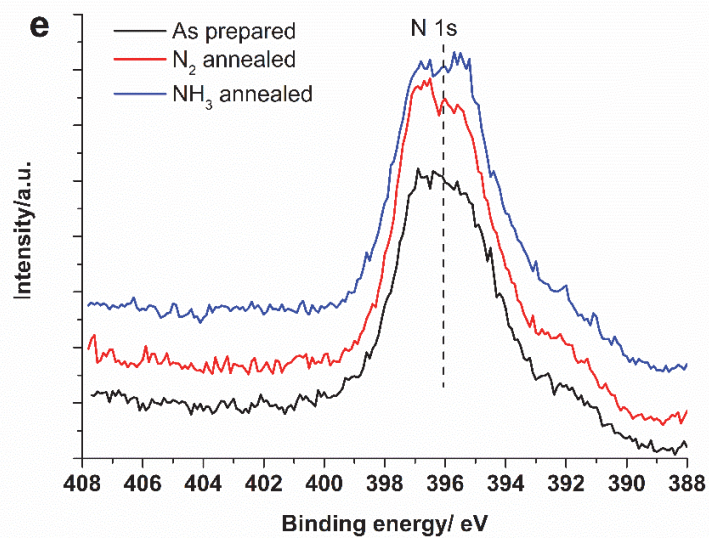
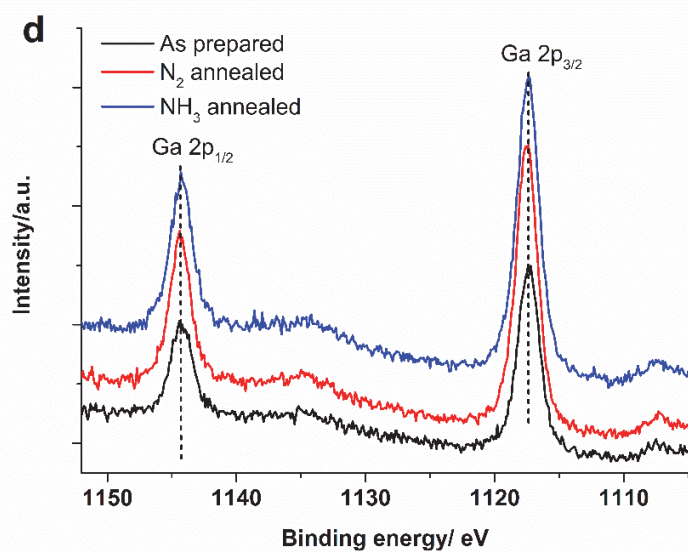
Figure S1. XRD patterns of a) $\text{Mg:GaN}/\text{Ta}_3\text{N}_5$ with different Mg:GaN thickness, b) $\text{Mg:GaN}/\text{Pyrex}$ glass with different Mg concentration.

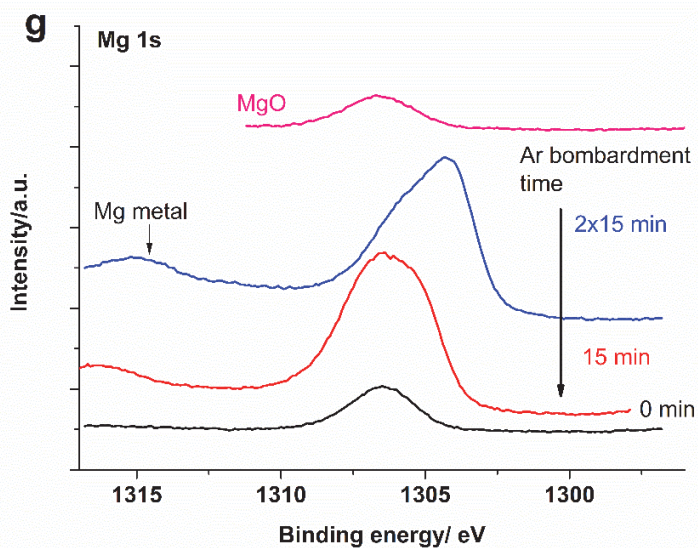
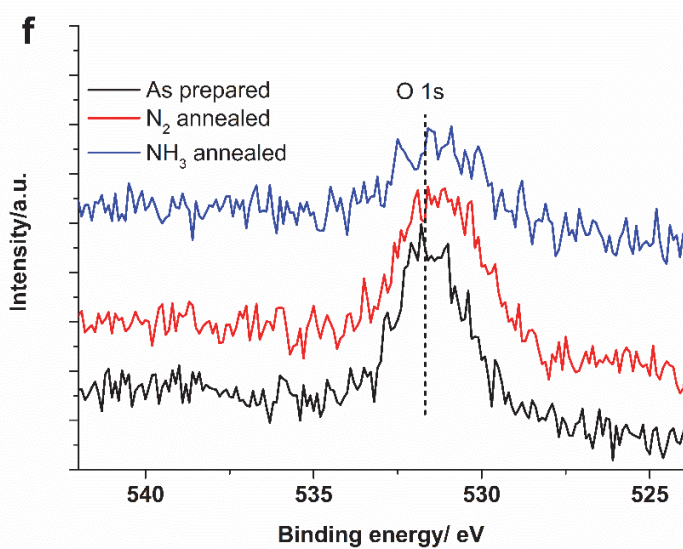
Table S1. EDS elemental analysis of Mg:Ga₂N/Ta₃N₅ sample before and after post annealing in N₂ or NH₃.

Elements	Concentration, wt%		
	As-prepared	N ₂ -post annealed	NH ₃ -post annealed
Mg	2.2	2.0	0.7
Ga	76.2	76.6	70.4
N	14.5	14.9	12.9
O	3.9	3.2	1.5
Ta	3.2	3.5	14.5
Total	100.0	100.0	100.0









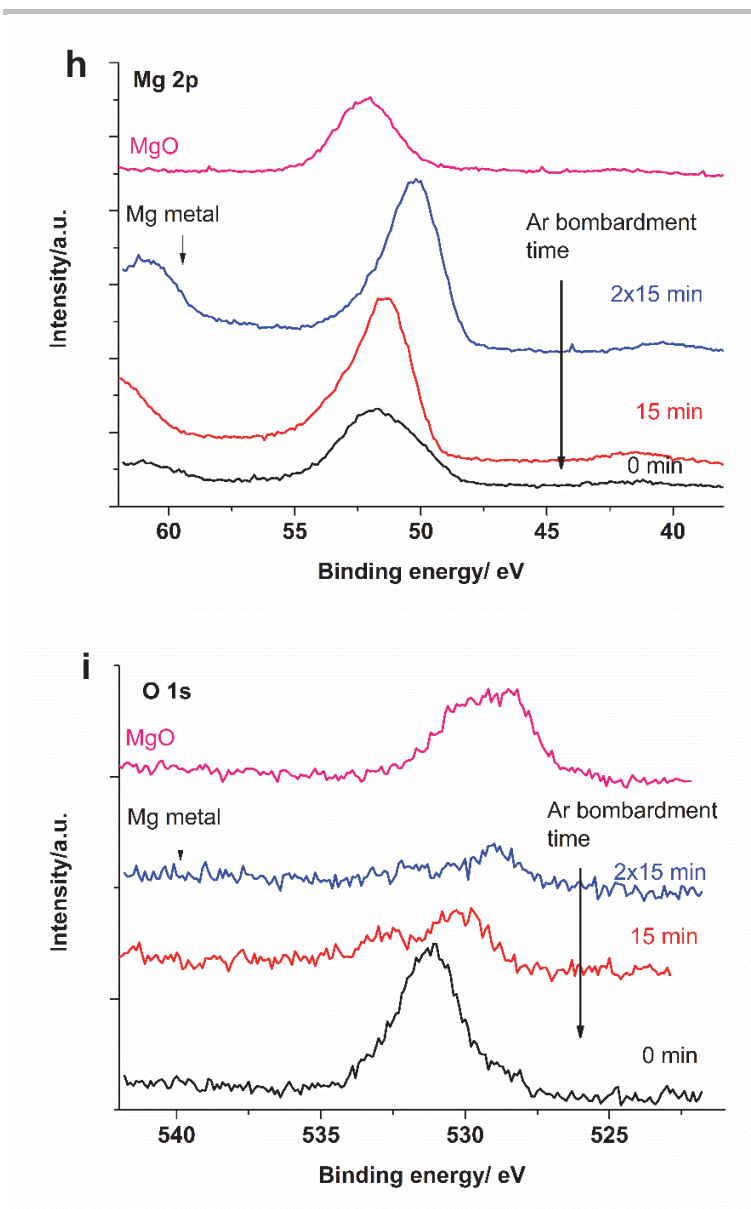


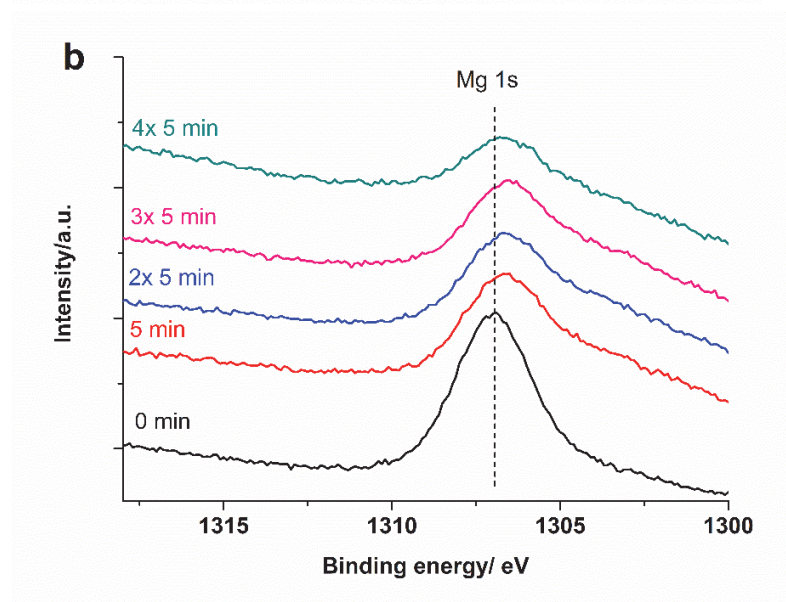
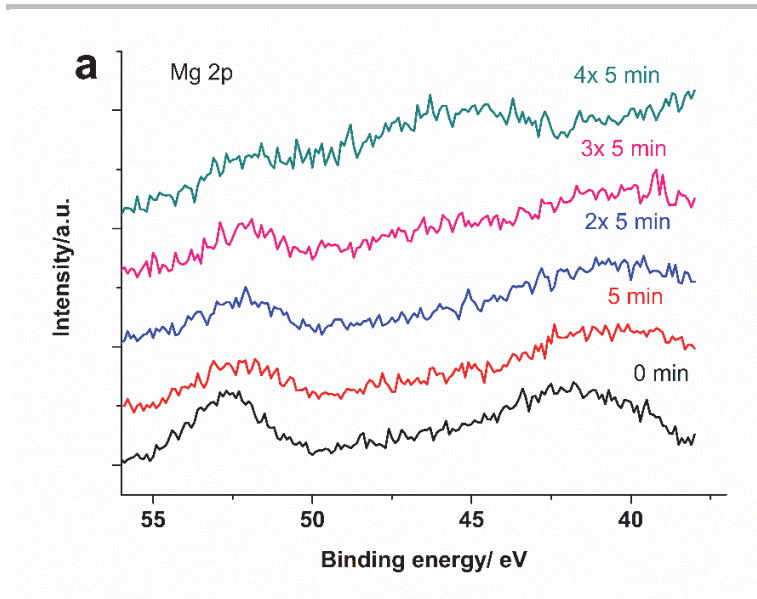
Figure S2.

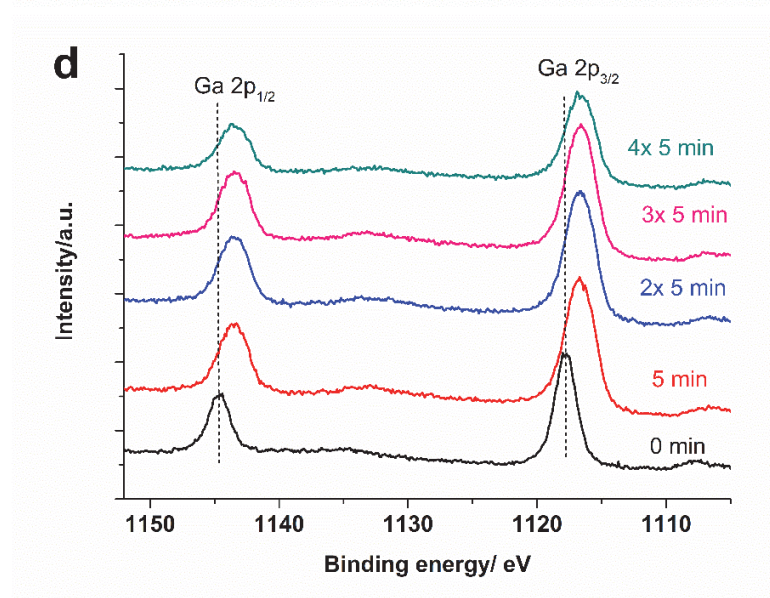
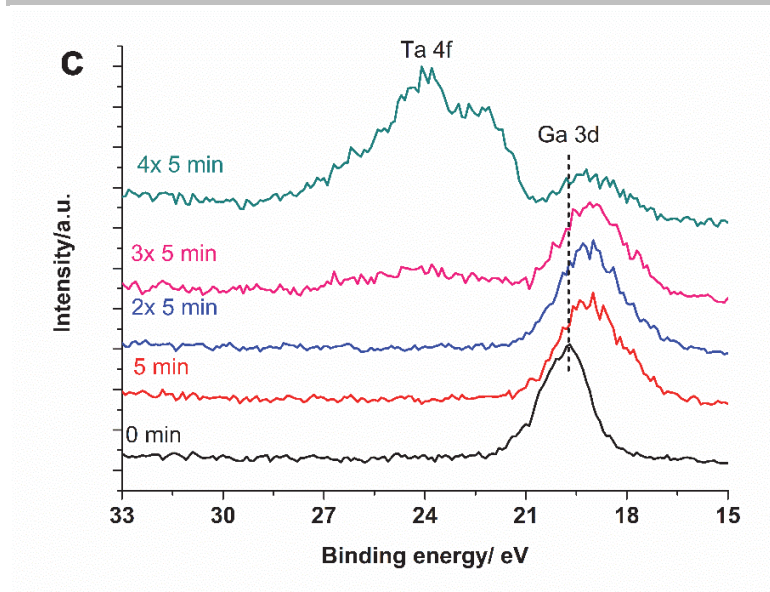
a – f : XPS elemental peaks of Mg:GaN/Ta₃N₅ sample surfaces as prepared, post-annealed in N₂, and post-annealed in NH₃.

g – i : XPS elemental peaks for standard Mg materials. Mg 1s, Mg 2p and O 1s signals on MgO and Mg metal (Ar-bombarded for 15 min and 30 min for cleaning).

The Ga 2p_{3/2} peaks in d are positioned at 1117.5 eV of the binding energy, closely to that of typical GaN films grown by thermal CVD.²³

Mg 1s peaks in a and b are positioned at that of oxidized Mg²⁺ (compared to g and h).





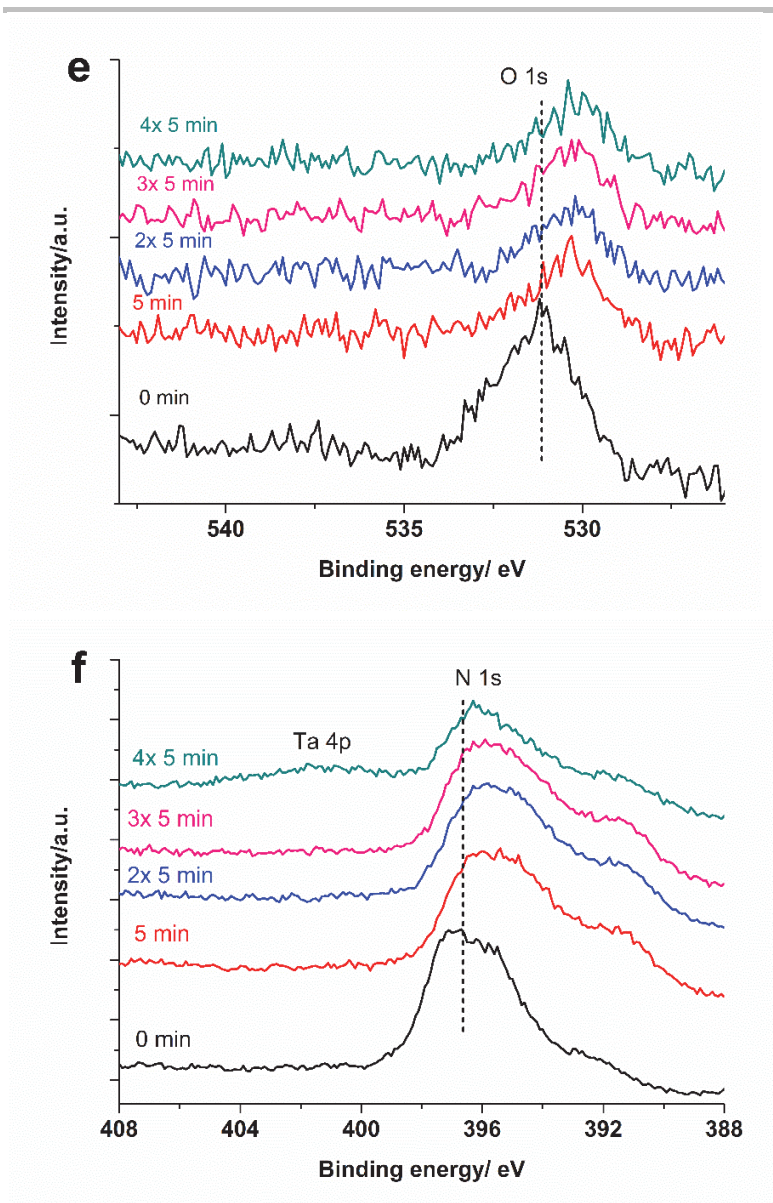


Figure S3. XPS of a) Mg 2p, b) Mg 1s, c) Ga 3d, d) Ga 2p, e) O 1s, f) N 1s on depth profiling of N₂ post annealed-Mg:GaN/Ta₃N₅.

Table S2. EDS elemental analysis of Mg:GaN area in Mg:GaN/Ta₃N₅ sample after N₂ post annealing.

Elements	Concentration, wt%
Mg	1.3
Ga	71.6
N	0.2
O	18.2
Ta	8.7
Total	100.0

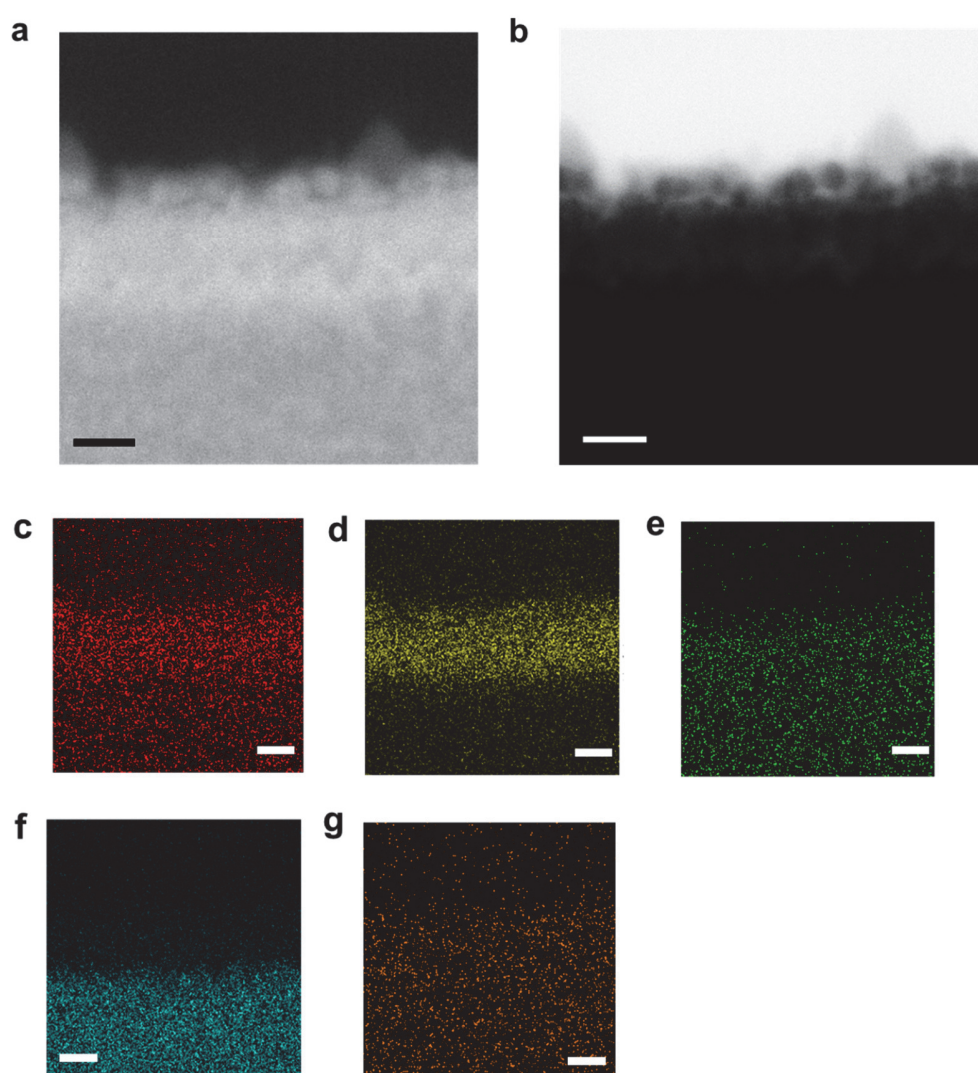


Figure S4. a) Dark- and b) bright-field cross-sectional STEM images of as-prepared Mg:GaN/Ta₃N₅ and STEM-EDS mapping of c) Mg, d) Ga, e) N, f) Ta, and g) O for the same observing area. Scale bar = 50 nm.

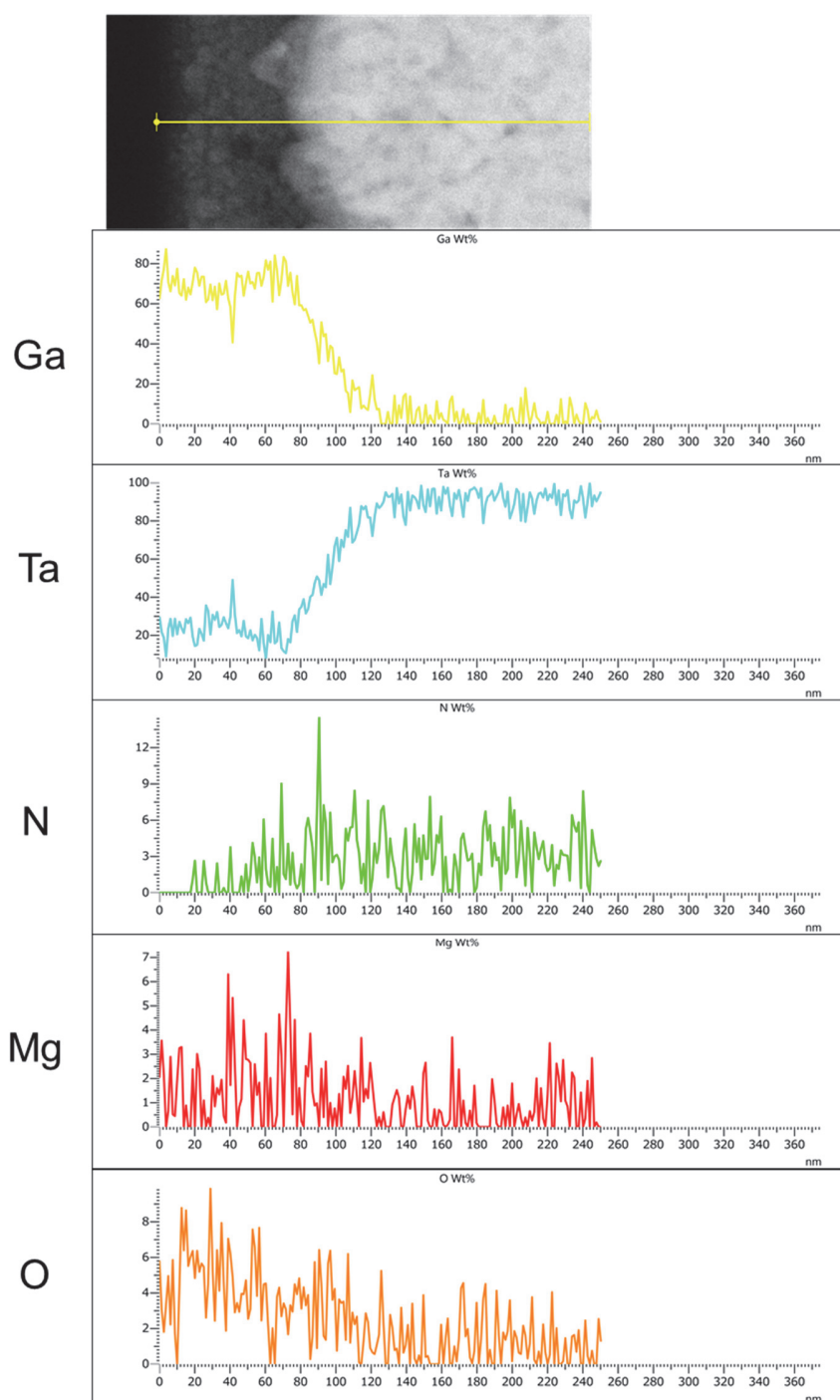


Figure S5. EDS line scan of the Mg:GaN/Ta₃N₅ after N₂ post annealing. The abscissa of each viewgraph corresponds to the position on the yellow line in the SEM image. The coordinate of each viewgraph indicates EDS counts (in an arbitrary unit).

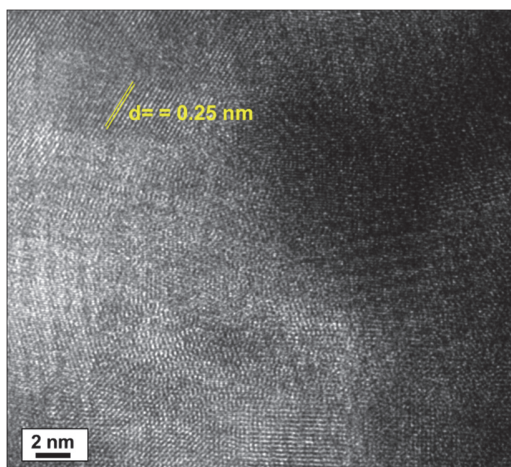
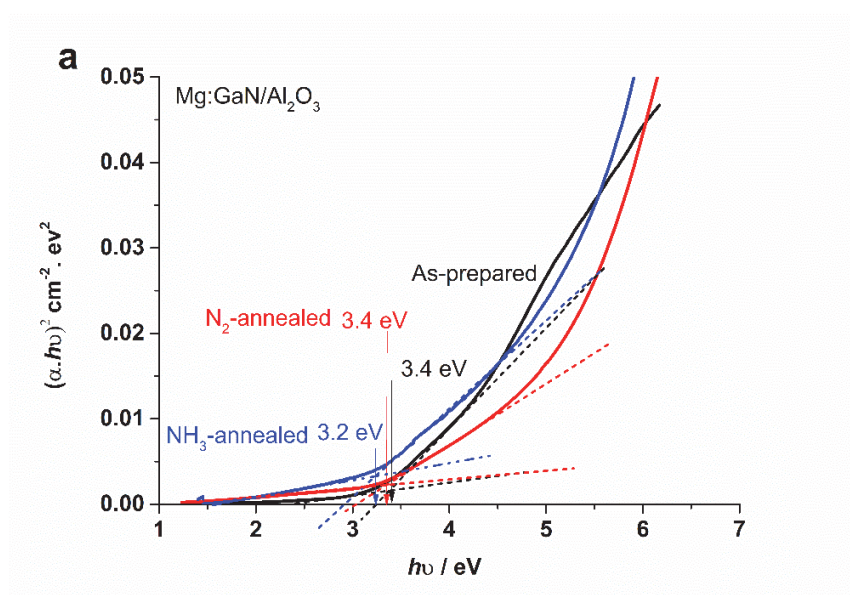


Figure S6. HRTEM image at interface of Ta_3N_5 (dark, right side) and Mg:GaN (bright, left side) after N_2 post annealing. The spacing of 0.25 nm in the striped pattern in the GaN side matches the (002) plane spacing of wurtzite GaN. This part of GaN agrees with the XRD pattern (Fig. 2) involving the intense peak at $2\theta = 34.5^\circ$.



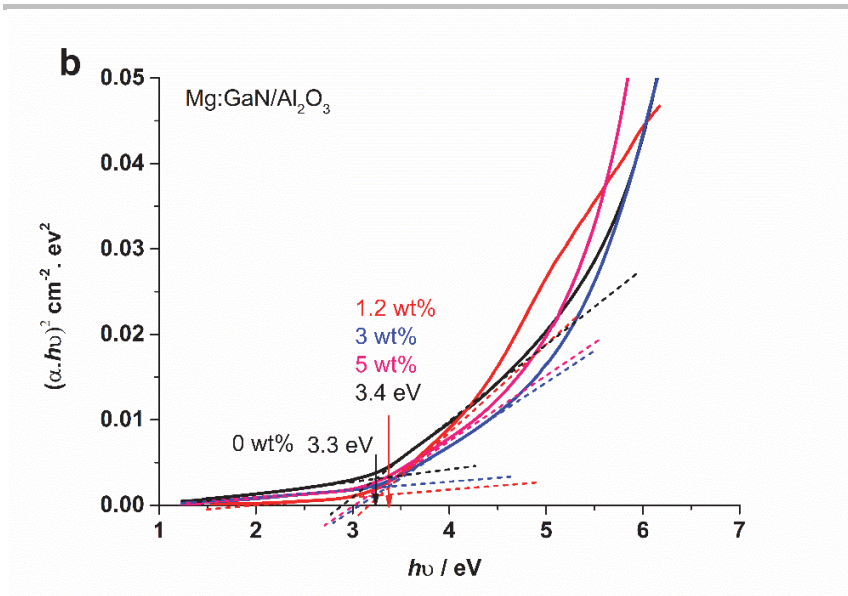
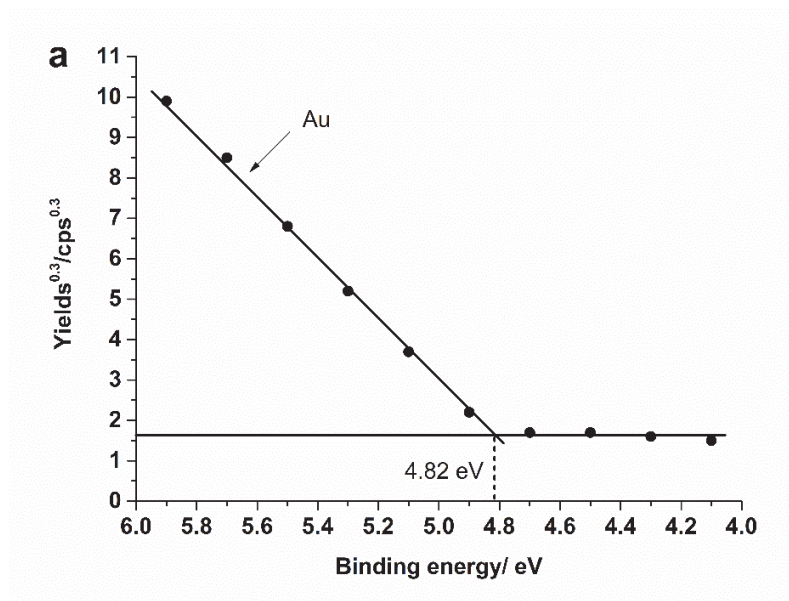
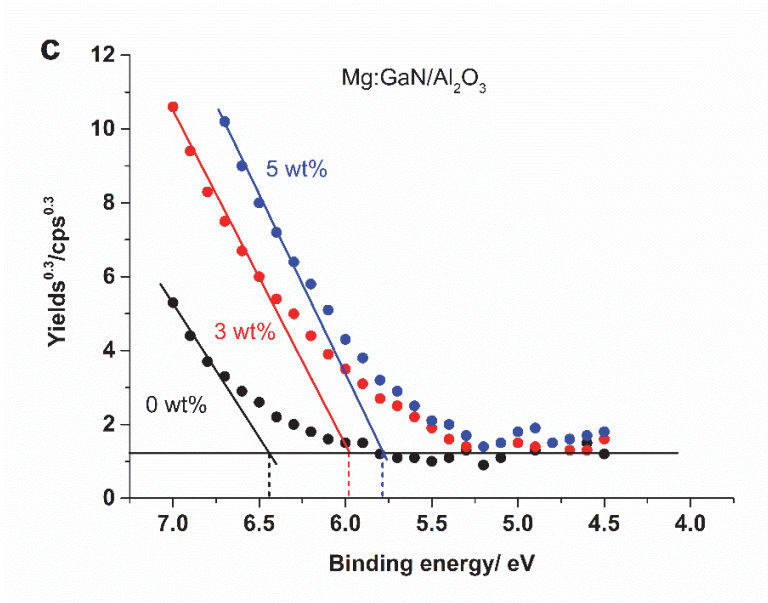
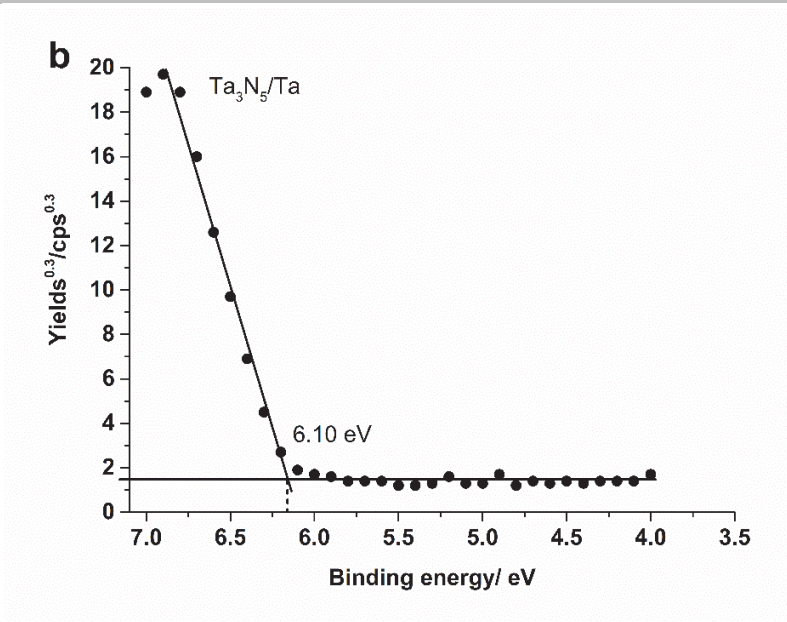


Figure S7. Tauc's plot of Mg:GaN/Al₂O₃ a) before and after post annealing, and b) with different Mg concentration. The Tauc's plot are used to determine the bandgap energy of the corresponding samples.





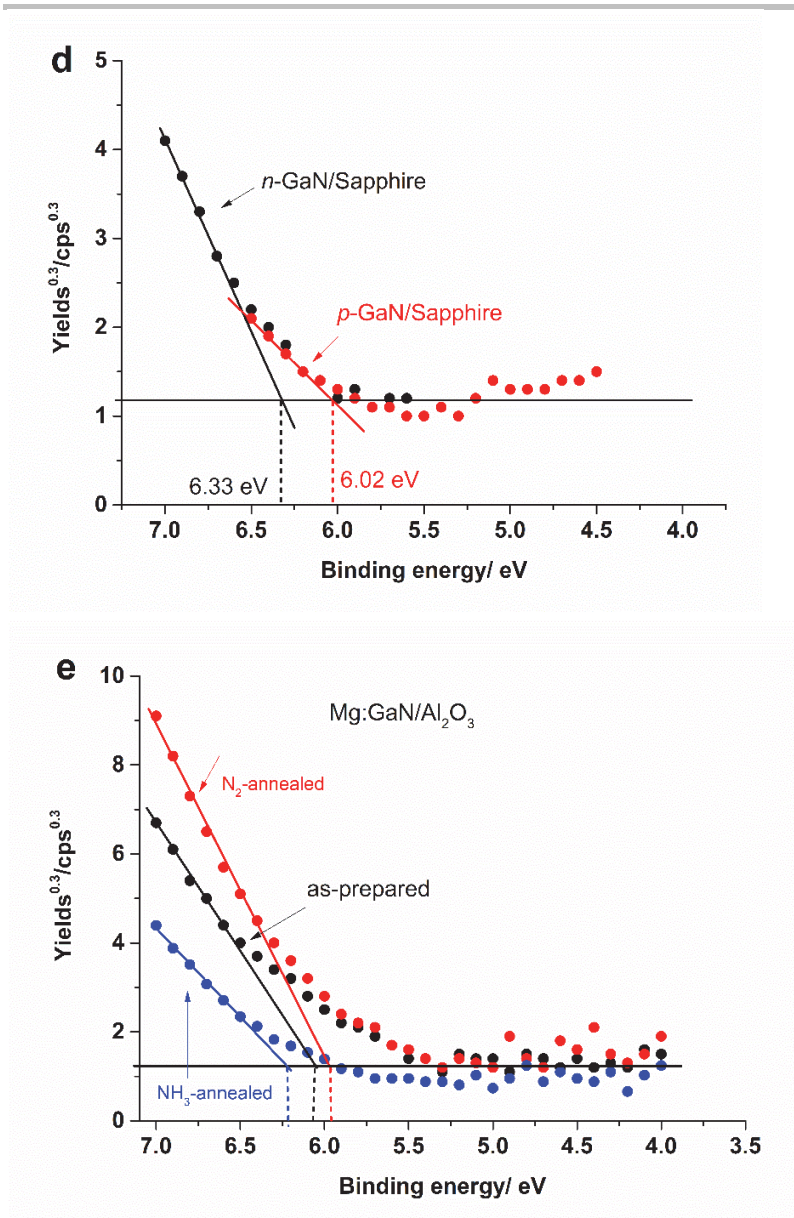
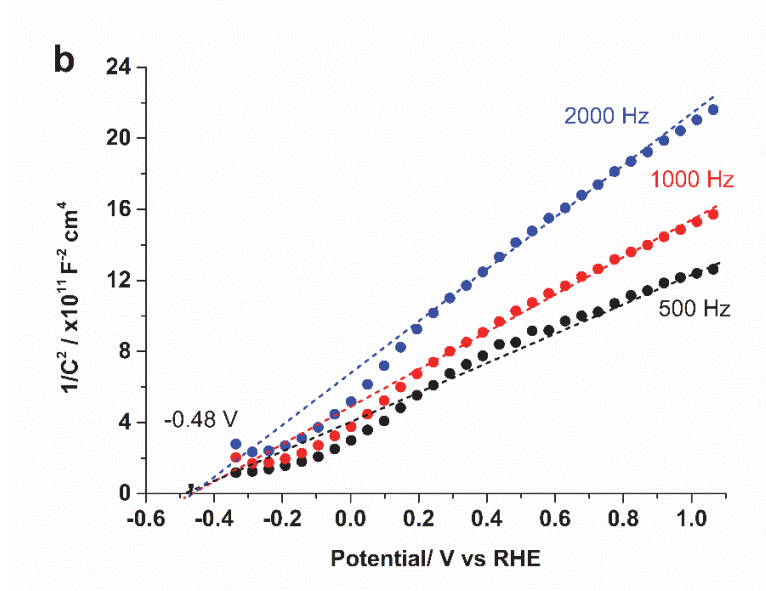
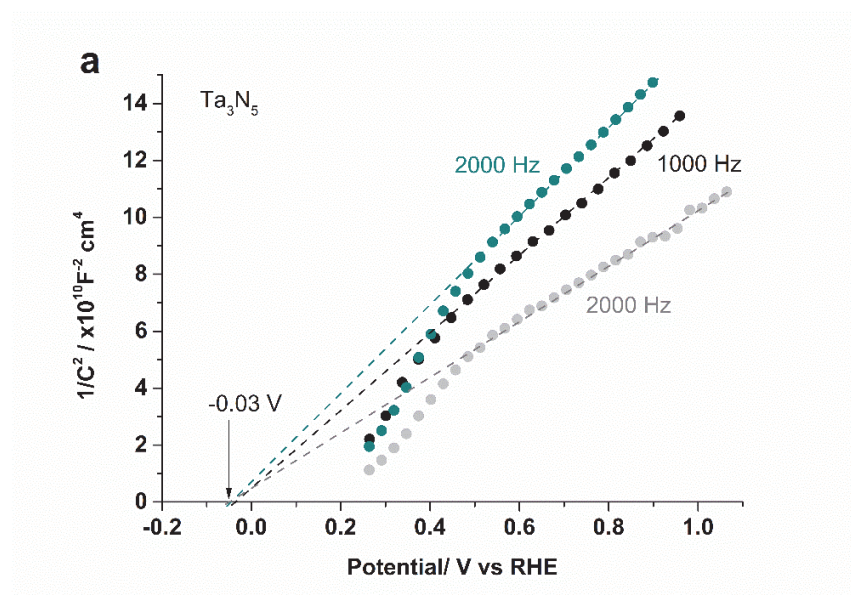
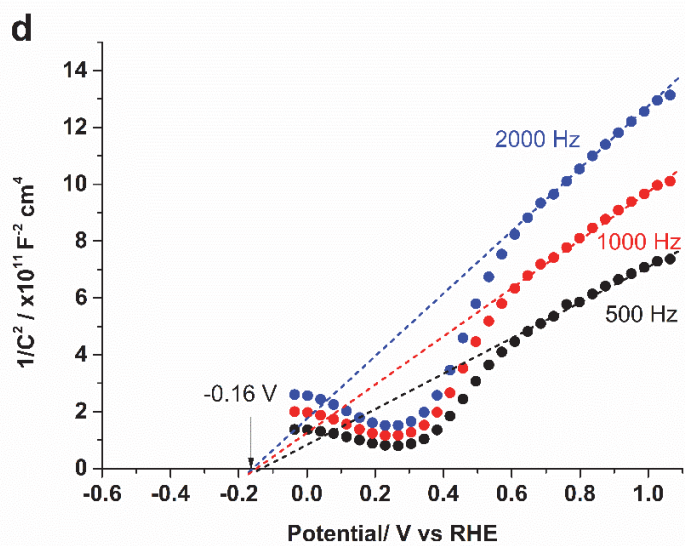
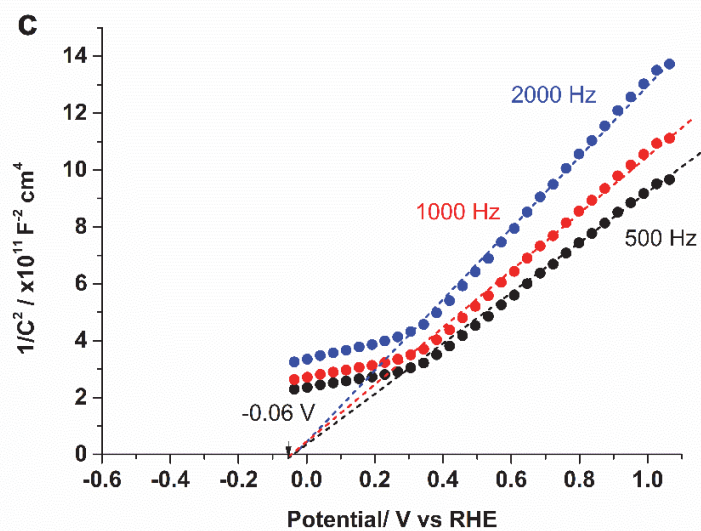


Figure S8. PES spectra of a) Au, b) Ta₃N₅, c) Mg:GaN/Al₂O₃ with different Mg concentration, d) commercial n -GaN/sapphire and p -GaN/sapphire, and e) Mg:GaN/Al₂O₃ before and after post annealing.





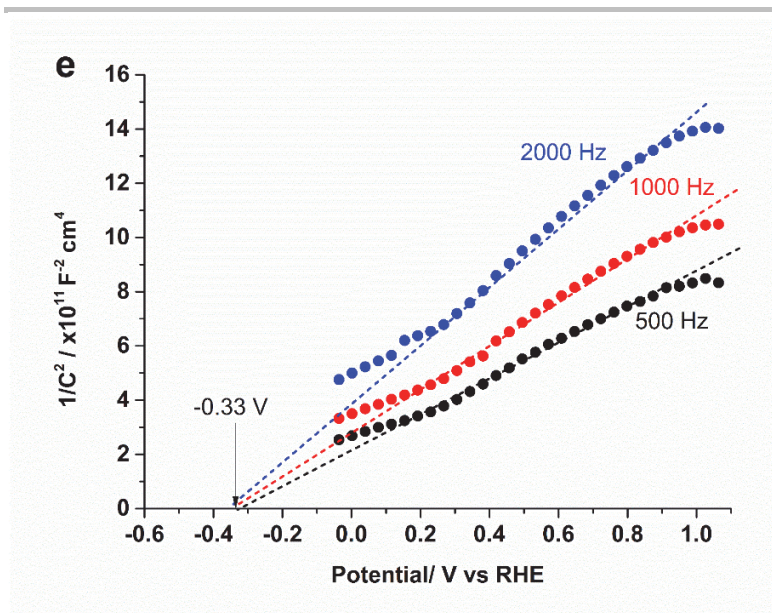


Figure S9. MS plot of a) Ta_3N_5 , b) $\text{GaN}/\text{Al}_2\text{O}_3$, c) as-prepared $\text{Mg:GaN}/\text{Al}_2\text{O}_3$, d) N_2 post annealed- $\text{Mg:GaN}/\text{Al}_2\text{O}_3$, e) NH_3 post annealed- $\text{Mg:GaN}/\text{Al}_2\text{O}_3$.

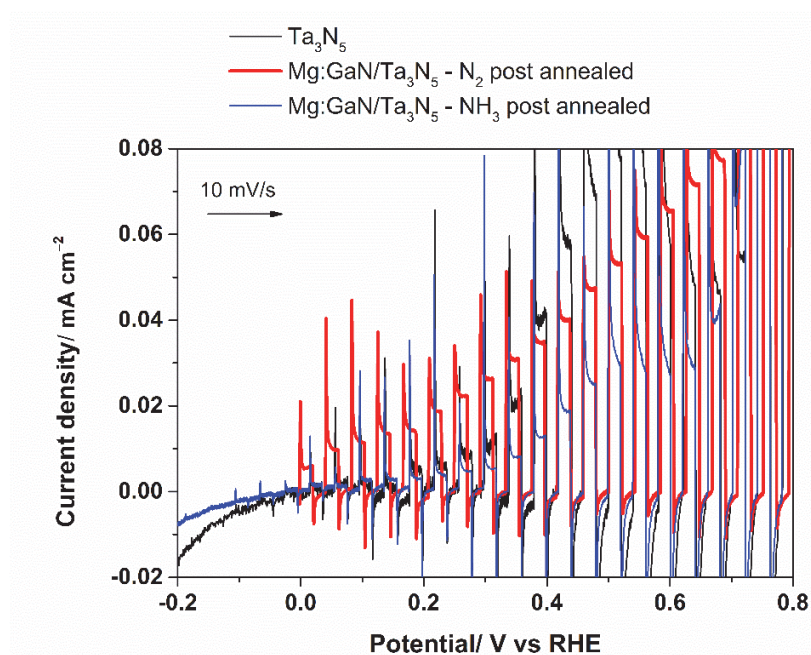


Figure S10. Magnification of background-subtracted LSV curve in Figure 5a, showing the onset potentials of Ta_3N_5 and $\text{Mg:GaN}/\text{Ta}_3\text{N}_5$ after N_2 and NH_3 post annealing.

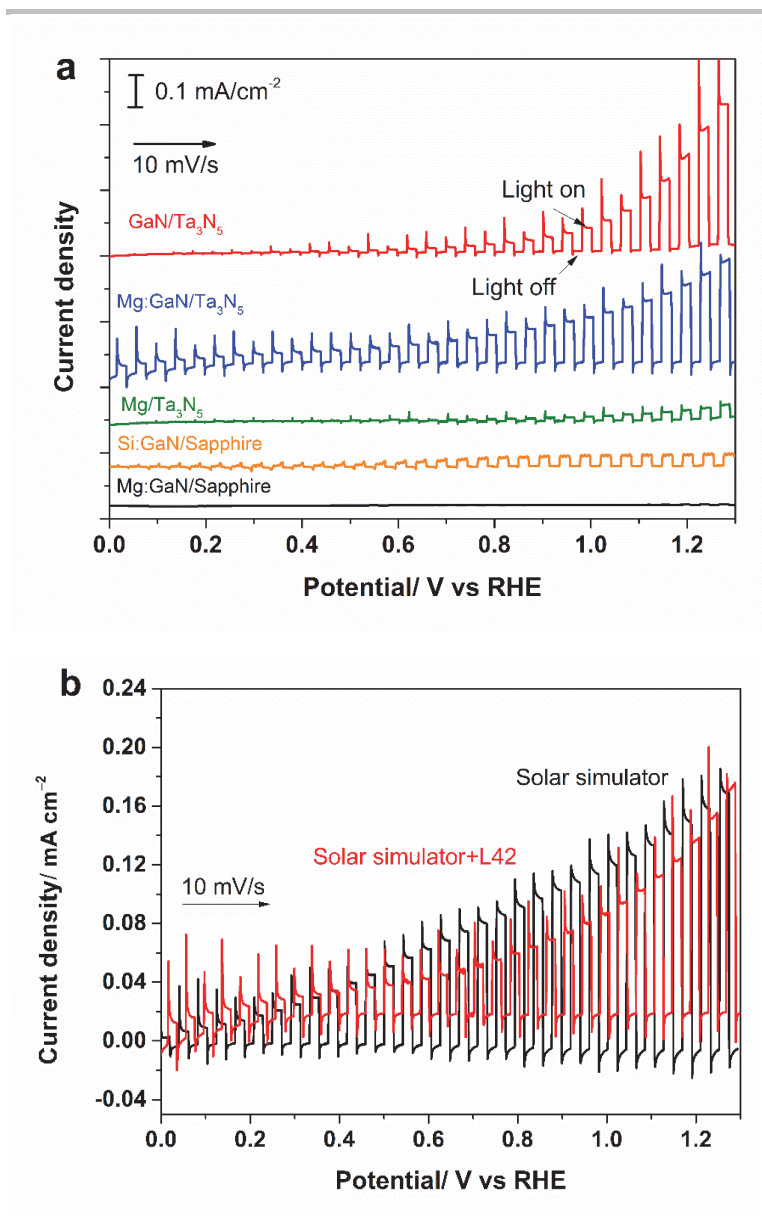
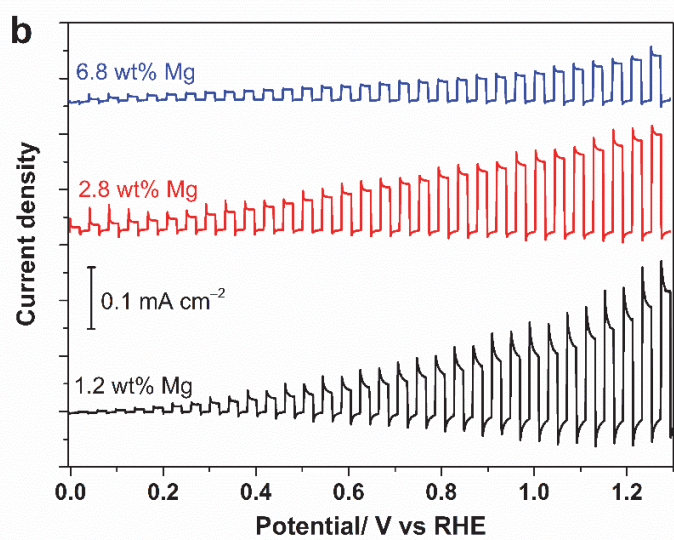
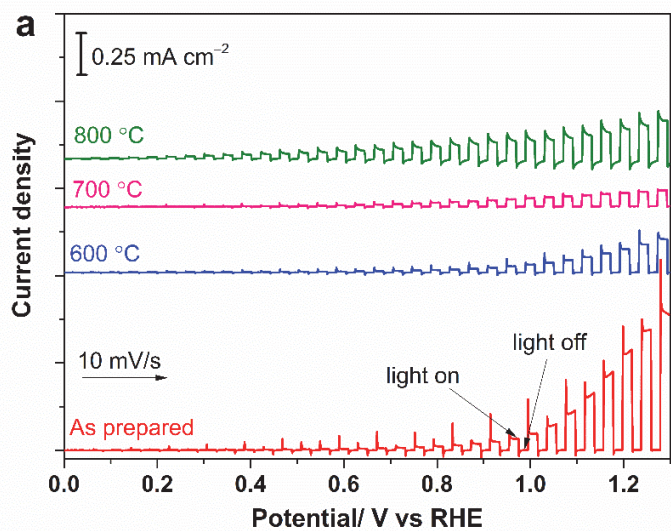
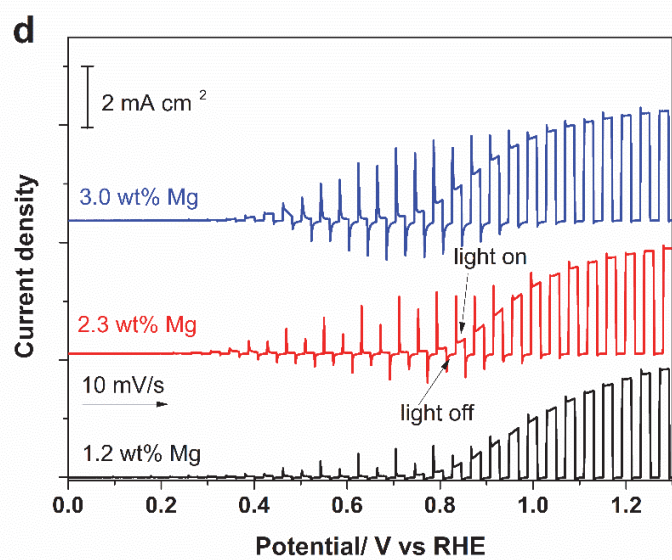
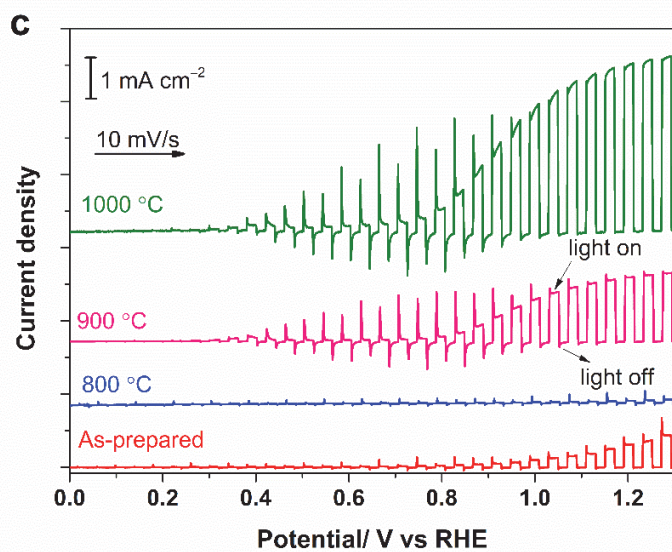


Figure S11. LSV of a) Si:GaN/sapphire, GaN/Ta₃N₅, Mg:GaN/Ta₃N₅ and Mg/Ta₃N₅, under full-light irradiation by solar simulator, b) Mg:GaN/Ta₃N₅ in solar simulator full light and solar simulator+L42 (visible light, $\lambda > 420$ nm). All the experiments were done using 100 mL of 0.5 M K₂HPO₄ solution at pH 13 (KOH adjusted) electrolyte, with CoPi cocatalyst addition.





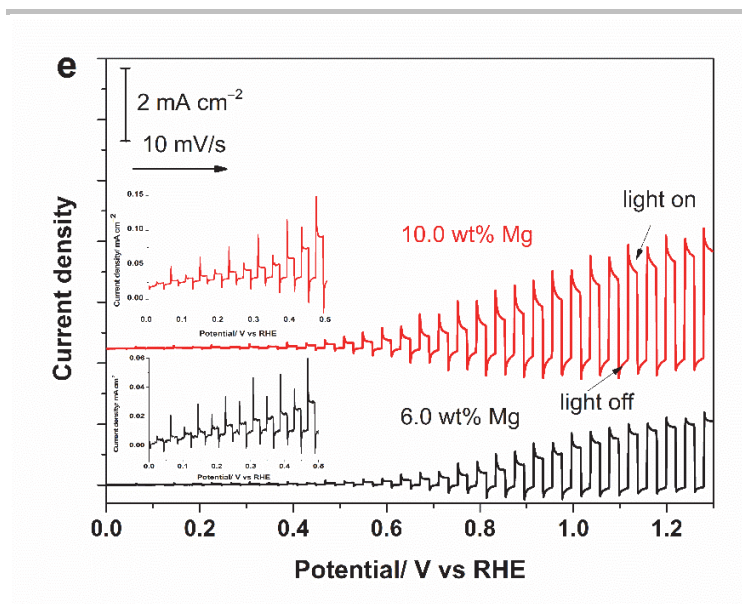


Figure S12. LSV of 100 nm Mg:GaN/Ta₃N₅ a) after N₂ post annealing with different temperatures, b) after N₂ post annealing at 800 °C with different Mg concentration, c) after NH₃ post annealing with different temperatures, d) after NH₃ post annealing at 1000 °C with different Mg concentration, e) 300 nm Mg:GaN/Ta₃N₅ after NH₃ post annealing.

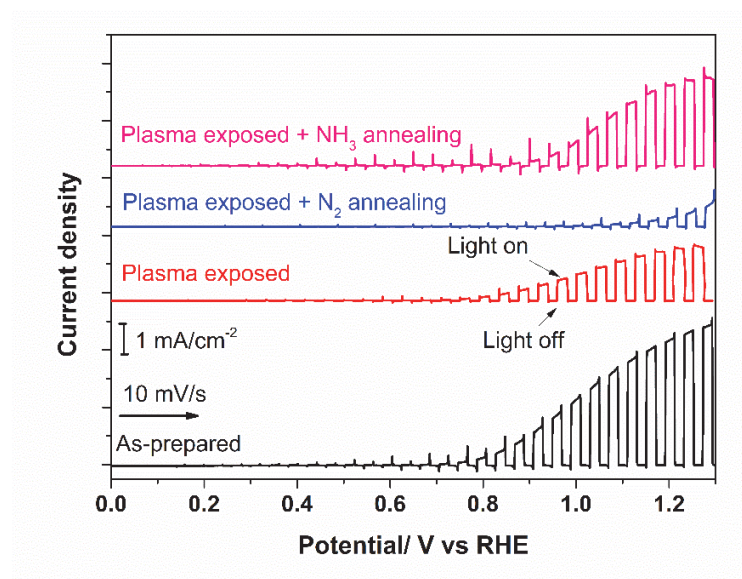
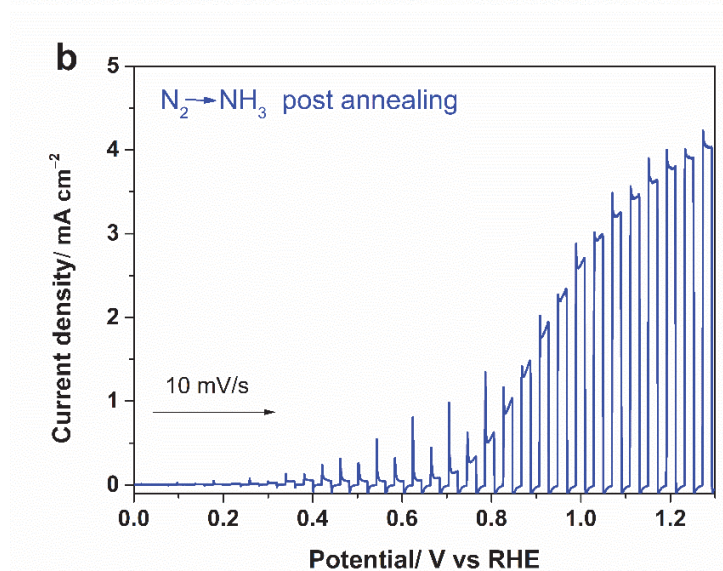
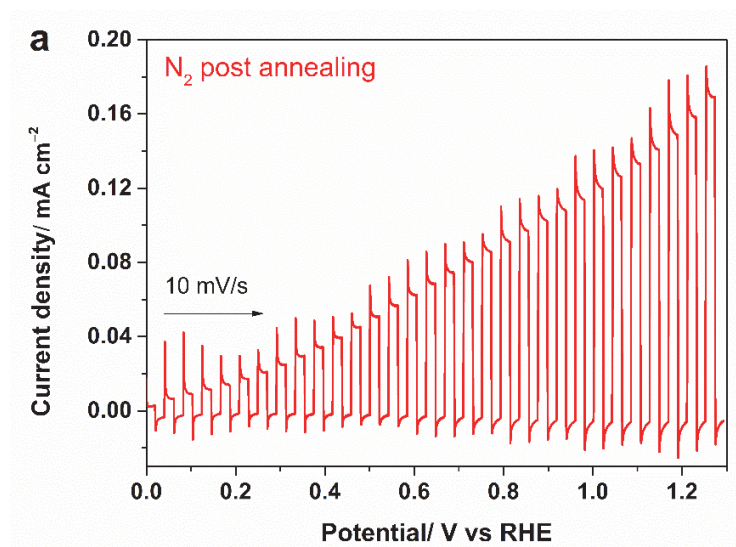


Figure S13. LSV of bare Ta₃N₅ exposed to PCVD chamber followed by N₂ or NH₃ post annealing.



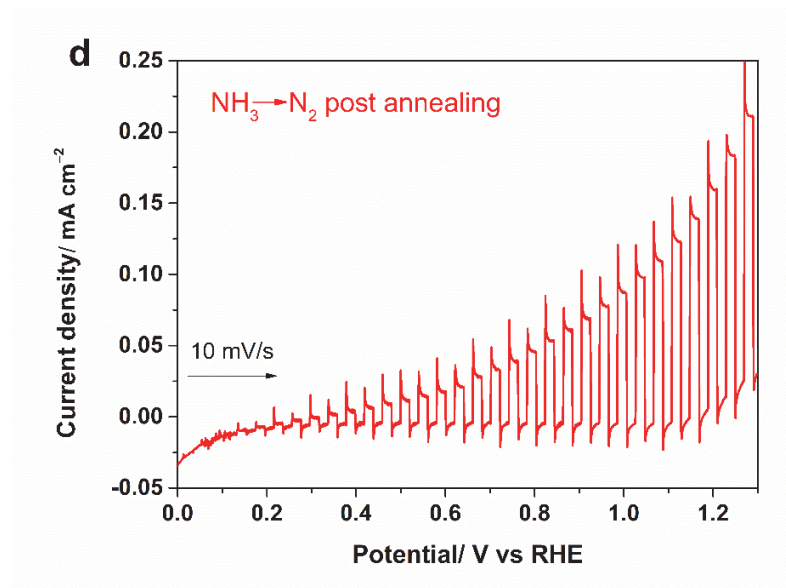
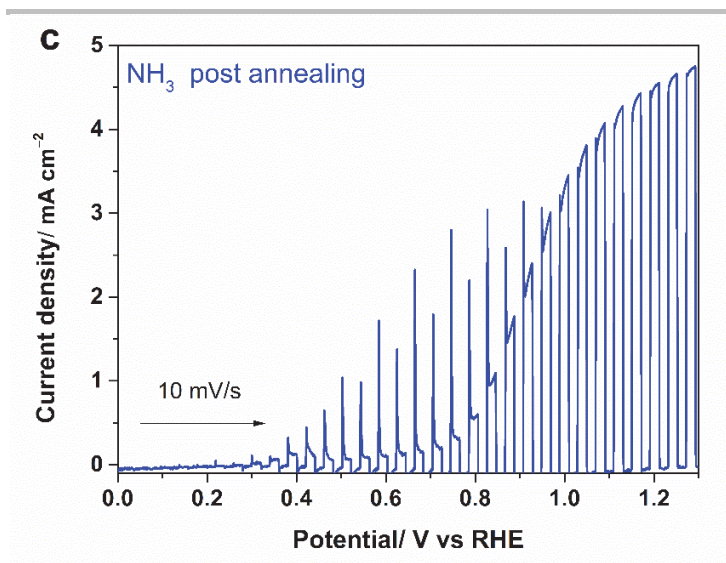


Figure S14. LSV of $\text{Mg}:\text{GaN}/\text{Ta}_3\text{N}_5$ subjected to a) N_2 post annealing, b) N_2 post annealing followed by NH_3 post annealing, c) NH_3 post annealing, d) NH_3 post annealing followed by N_2 post annealing.

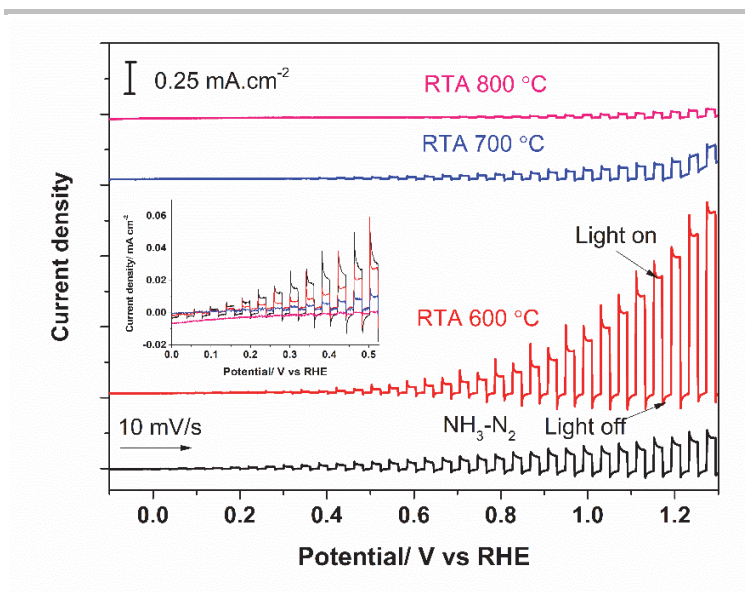


Figure S15. LSV of NH_3 post annealed-Mg:GaN/ Ta_3N_5 sample followed by rapid thermal annealing (RTA) at different temperature in static N_2 .