

B-site doping effects of $\text{NdBa}_{0.75}\text{Ca}_{0.25}\text{Co}_2\text{O}_{5+\delta}$ double perovskite catalysts for oxygen evolution and reduction reactions

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Supplementary Figures

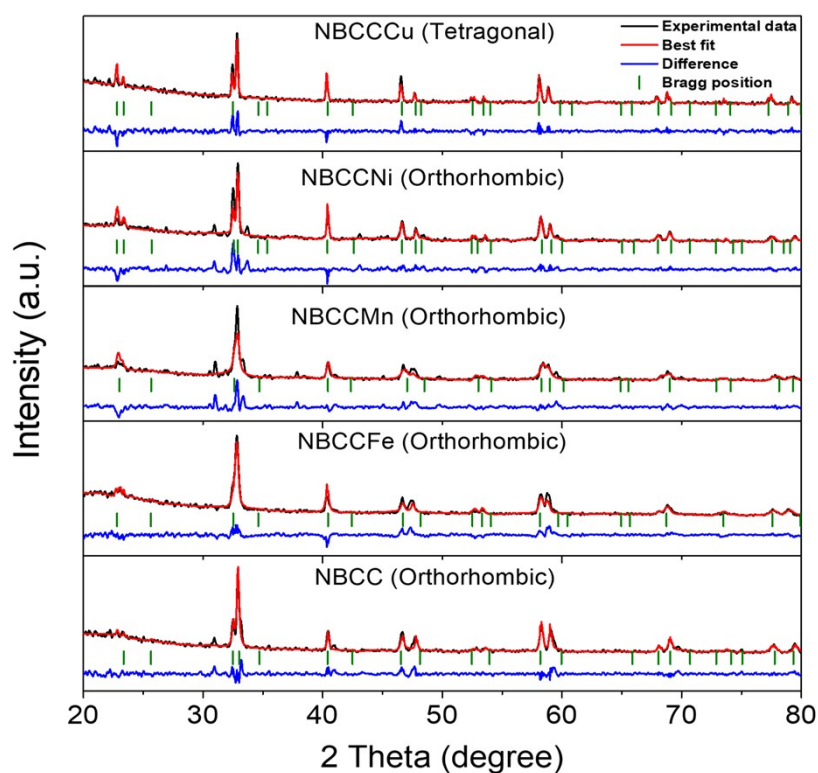


Fig. S1. Rietveld refinements of NBCC-based catalysts processed by the Fullprof program.

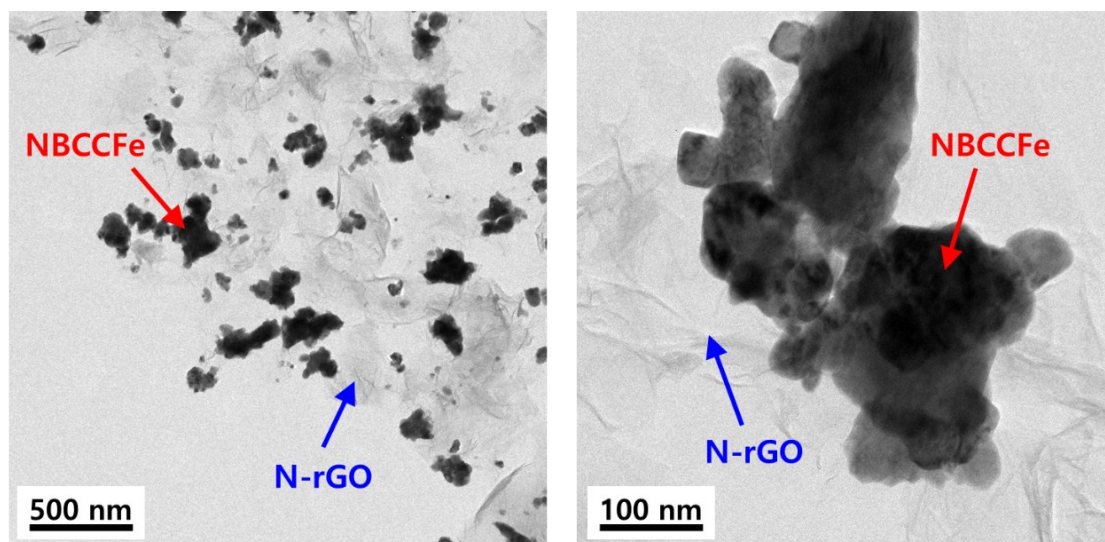


Fig. S2. Morphological characteristics of NBCCFe/N-rGO hybrid catalyst analyzed by HRTEM.

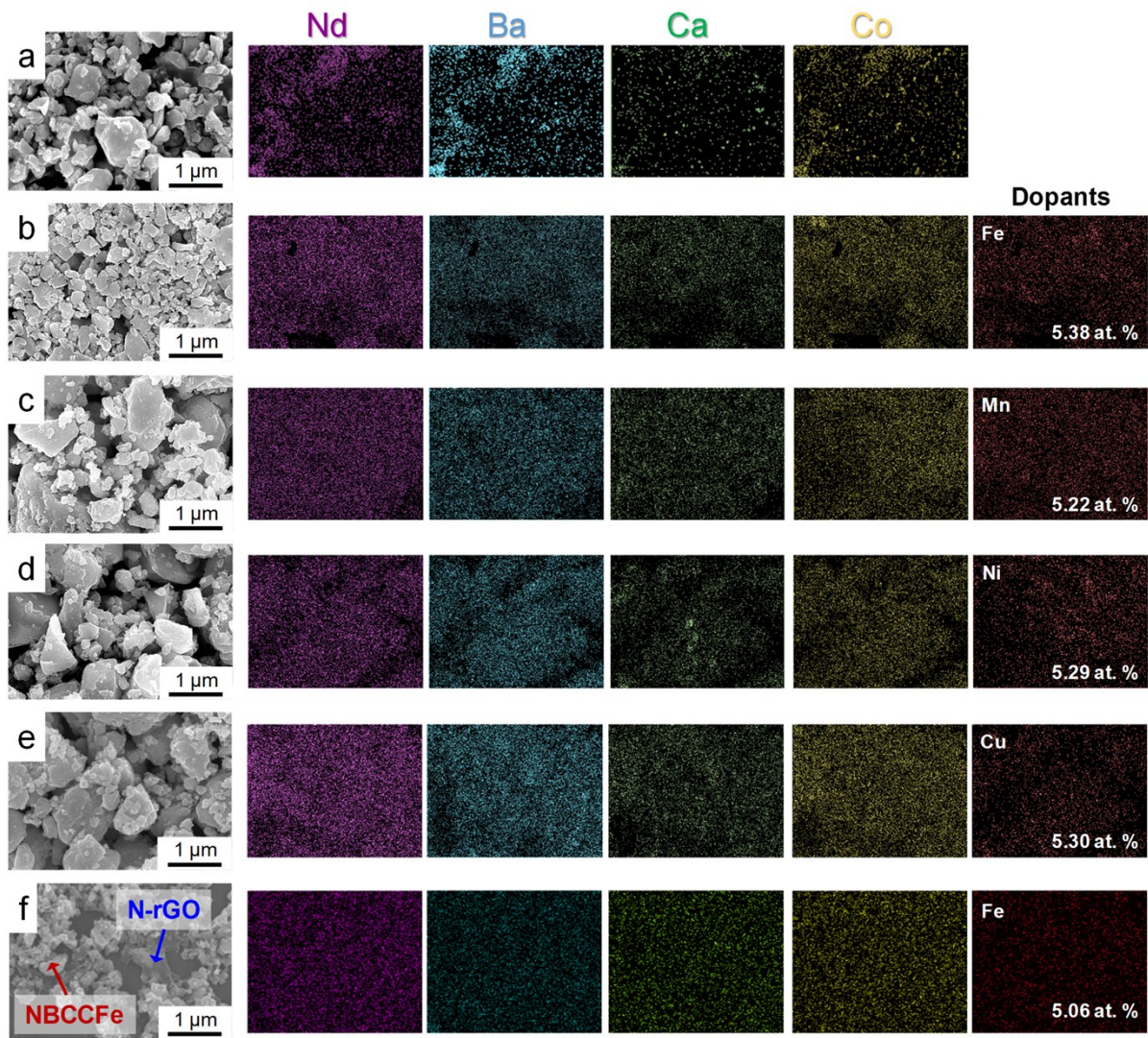


Fig. S3. Surface morphology and distribution of elemental compositions of catalysts by FE-SEM and EDX mapping: (a) NBCC (b) NBCCFe (c) NBCCMn (d) NBCCNi, (e) NBCCCu, and (f) NBCCFe/N-rGO.

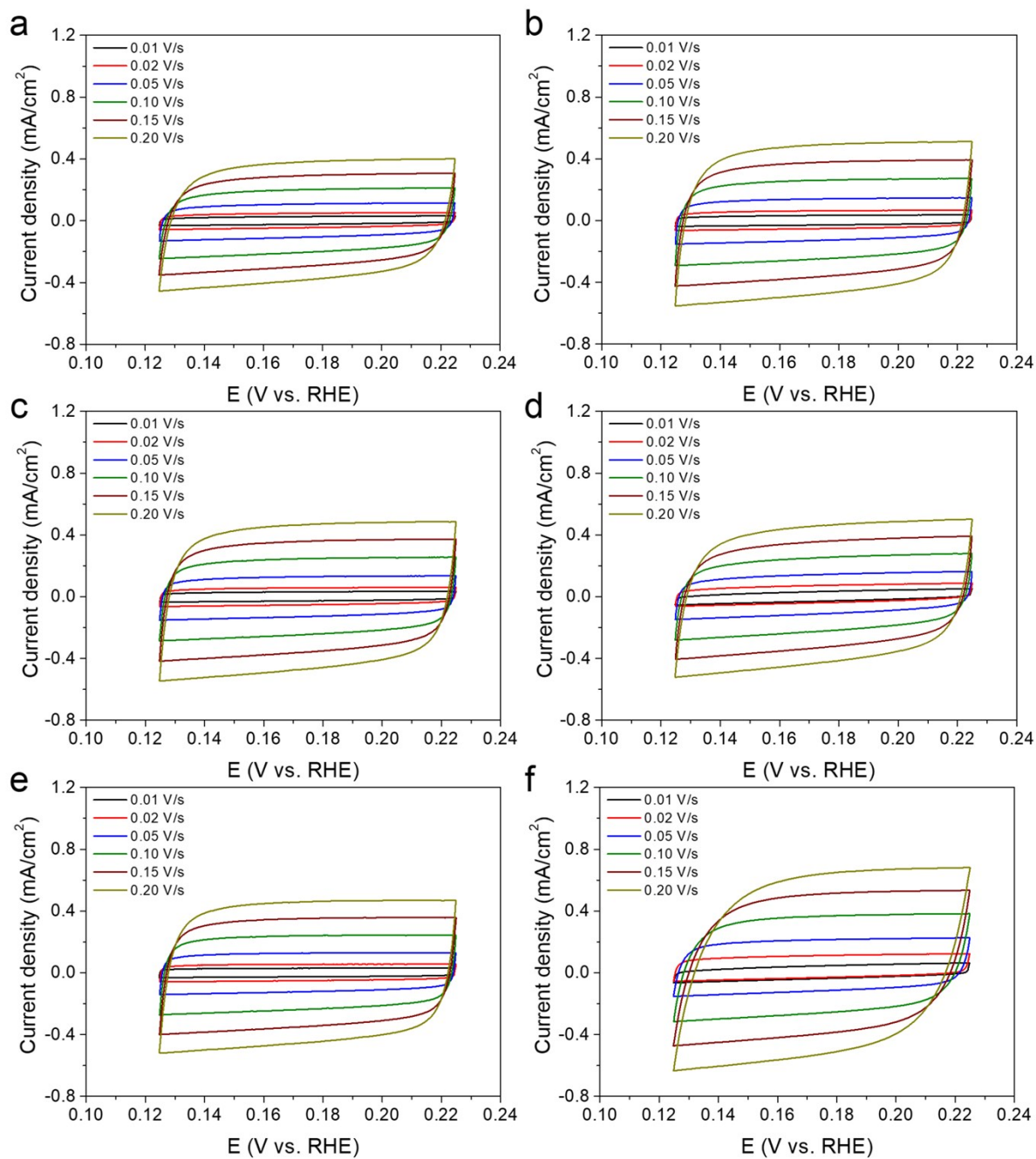


Fig. S4. Cyclic voltammogram curves of the catalysts with different scan rates (10-200 $\text{mV}\cdot\text{s}^{-1}$). (a) NBCC, (b) NBCCFe, (c) NBCCMn, (d) NBCCNi, (e) NBCCCu, and (f) NBCCFe/N-rGO.

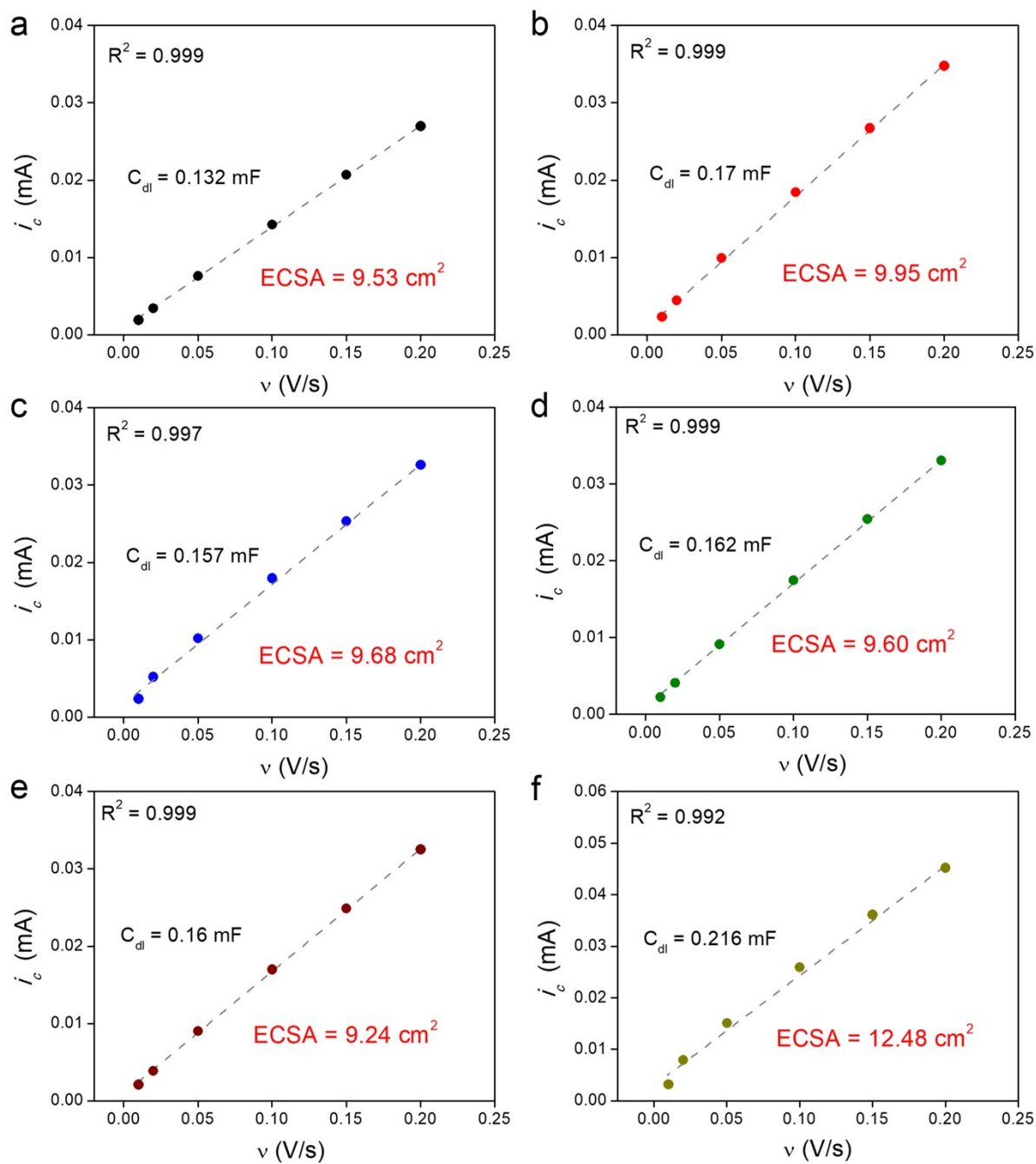


Fig. S5. Linear plots for the measured charging current (i_c) versus scan rate for calculation of the ECSA. (a) NBCC, (b) NBCCFe, (c) NBCCMn, (d) NBCCNi, (e) NBCCCu, and (f) NBCCFe/N-rGO. Double layer capacitances of the materials were determined from the linear plots.

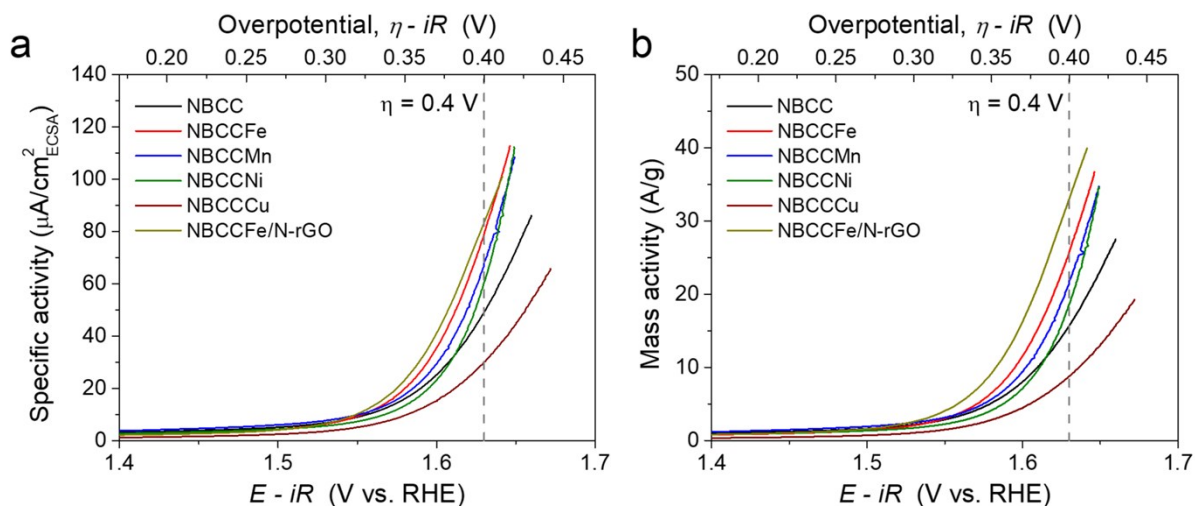


Fig. S6. (a) Specific activity ($\mu\text{A}\cdot\text{cm}^{-2}_{\text{ECSA}}$) and (b) mass activity ($\text{A}\cdot\text{g}^{-1}$) of catalysts in OER polarization curves normalized by the ECSA at $\eta = 0.4\text{ V}$ in 0.1 M KOH solution.

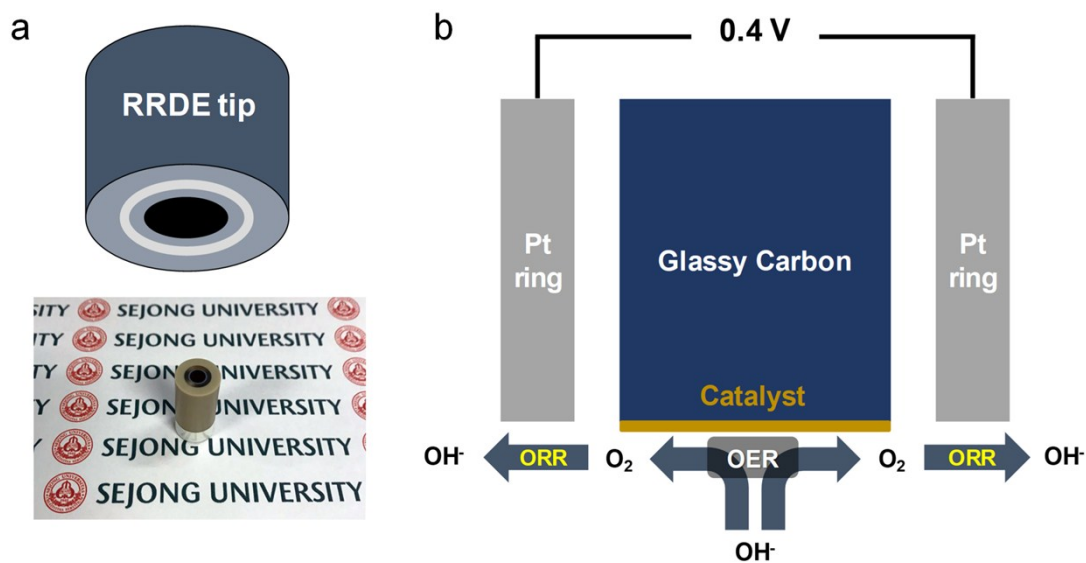


Fig. S7. (a) Schematic illustration (top) and Actual photograph (bottom) of the RRDE tip. (b) The continuous OER (disk) to ORR (ring) process initiated on a RRDE (N_2 -saturated 0.1 M KOH , ring potential: 0.40 V).

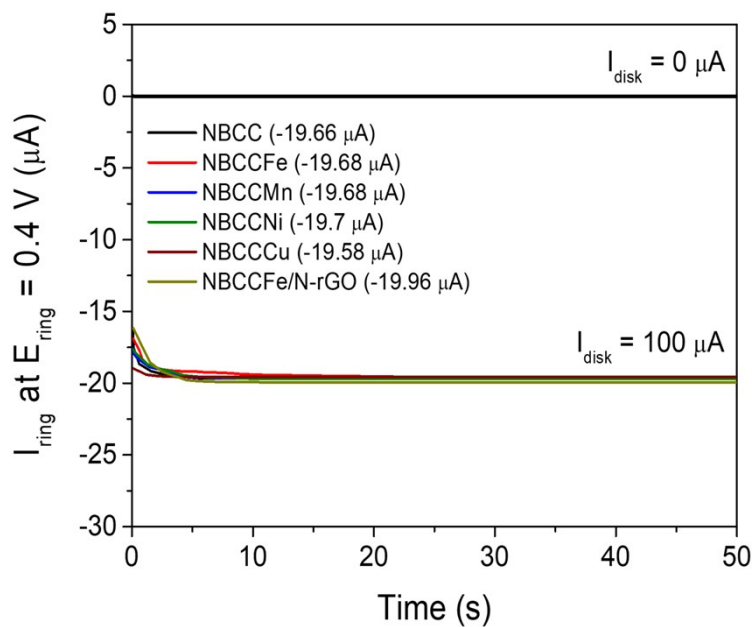


Fig. S8. The ring currents of NBCC-based catalysts on a RRDE (1,600 rpm) in 0.1 M KOH solution (ring potential: 0.40 V).

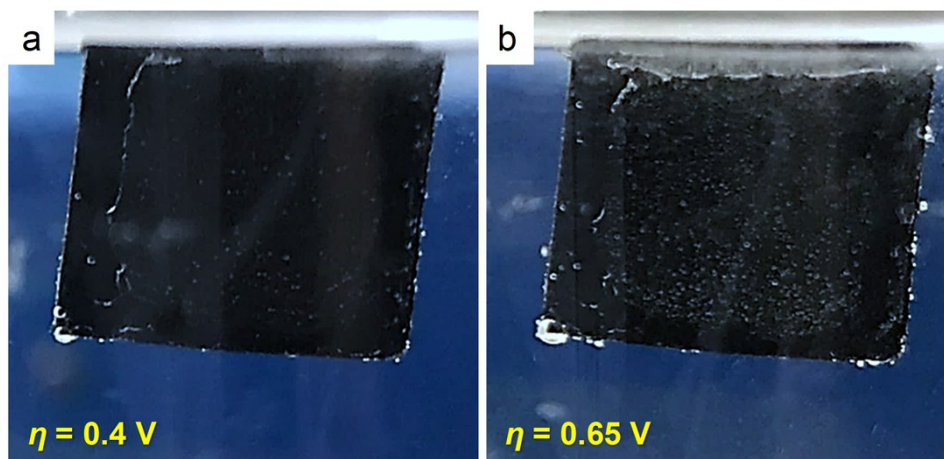


Fig. S9. Images of oxygen bubble generation in a large-scale water electrolysis half-cell (1 cm²) using the NBCCFe catalyst at $\eta = 0.4$ V (left) and $\eta = 0.65$ V (right).

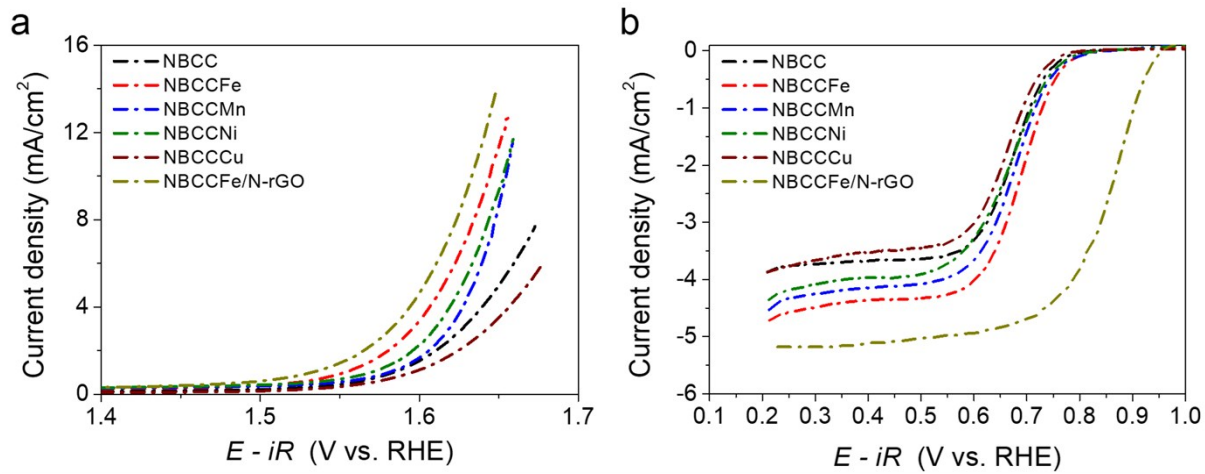


Fig. S10. 1,500th OER polarization curves and 10,000th ORR polarization curves of NBCC and doped-NBCC catalysts.

2. Supplementary Tables

Table S1. Comparison of oxygen activity of NBCCFe and NBCCFe/N-rGO with reported perovskite-based electrocatalysts.

Catalysts	E_{ORR} (V) at $j = -3$ mA cm^{-2}	E_{OER} (V) at $j = 10$ mA cm^{-2}	Oxygen activity ^a (V)	Reference
NBCCFe/N-rGO	0.848	1.609	0.761	This Work
NBCCFe	0.679	1.627	0.948	This Work
Core-Corona LaNiO ₃ /N-CNT	0.64	1.67	1.03	Nano Lett. 2012, 12, 1946 (Ref. 1)
La _{0.3} (Ba _{0.5} Sr _{0.5}) _{0.7} Co _{0.8} Fe _{0.2} O _{3-δ} /KB	0.58	1.59	1.01	Angew. Chem. Int. Ed. 2014, 53, 4582 (Ref. 2).
porous CaMnO ₃	0.8	1.81	1.01	Adv. Mater. 2014, 26, 2047 (Ref. 3).
Ar treated Ba _{0.5} Sr _{0.5} Co _{0.8} Fe _{0.2} O ₃ /KB	0.62	1.77	1.05	Adv. Energy Mater. 2015, 5, 1501560 (Ref. 4)
LaTi _{0.65} Fe _{0.35} O _{3-δ} /N-CNR	0.72	1.77	1.05	Nano Energy 2015, 15, 92 (Ref. 5).
La _{0.5} Sr _{0.5} Co _{0.8} Fe _{0.2} O ₃ -PR/N-rGO	0.81	1.73	0.92	Nano Energy 2014, 10, 192 (Ref. 6).
La _{0.58} Sr _{0.4} Fe _{0.2} Co _{0.8} O ₃ /N-CNT	0.7	1.63	0.93	Electrochimica Acta 2016, 208, 25 (Ref. 7).
Multi-shelled La _{0.9} Sr _{0.1} CoO ₃	0.64	1.76	1.12	J. Mater. Chem. A, 2015, 3, 22448 (Ref. 8).
NdBa _{0.5} Sr _{0.5} Co _{1.5} Fe _{0.5} O _{5+δ} /CB	0.651	1.624	0.973	J. Mater. Chem. A, 2017, 5, 13019 (Ref. 9).
PrBa _{0.25} Sr _{0.75} Co ₂ O _{5+δ} /AB	0.68	1.65	0.97	Journal of Power Sources 2016, 334, 86 ((Ref. 10).
Mn-oxide film	0.73	1.77	1.04	J. Am. Chem. Soc. 2010, 132, 13612 (Ref. 11).
Co ₃ O ₄ /N,S-doped carbon	0.82	1.61	0.79	Adv. Funct. Mater. 2014, 24, 7655 (Ref. 12).

MnO _x on stainless steel mesh	0.82	1.62	0.8	Energy Environ. Sci. 2014, 7, 2017 (Ref. 13).
Mn _x O _y /N-doped Carbon	1.68	0.81	0.87	Angew. Chem. Int. Ed. 2014, 53, 8508 (Ref. 14).
CNT@NCNT	0.65	1.73	1.08	Adv. Funct. Mater. 2014, 24, 5956 (Ref. 15)
N-doped graphene/CNT	0.64	1.65	0.96	Small 2014, 10, 2251 (Ref. 16).
N,S,Fe-doped Carbon	0.87	1.78	0.91	J. Am. Chem. Soc. 2014, 136, 14486 (Ref. 17).
N-doped Carbon	0.77	1.61	0.84	Nat. Commun. 2013, 4, 2390 (Ref. 18).

References

1. Z. Chen, A. Yu, D. Higgins, H. Li, H. Wang, Z. Chen, Highly Active and Durable Core–Corona Structured Bifunctional Catalyst for Rechargeable Metal–Air Battery Application, *Nano Lett.* 12 (2012) 1946-1952.
2. J.-I Jung, H.Y. Jeong, J.-S. Lee, M.G. Kim, J. Cho, A Bifunctional Perovskite Catalyst for Oxygen Reduction and Evolution, *Angew. Chem. Int. Ed.* 53 (2014) 4582-4586.
3. X. Han, F. Cheng, T. Zhang, J. Yang, Y. Hu, J. Chen, Hydrogenated Uniform Pt Clusters Supported on Porous CaMnO₃ as a Bifunctional Electrocatalyst for Enhanced Oxygen Reduction and Evolution, *Adv. Mater.* 26 (2014) 2047-2051.
4. J.-I. Jung, S. Park, M-G. Kim, J. Cho, Tunable Internal and Surface Structures of the Bifunctional Oxygen Perovskite Catalysts, *Adv. Energy Mater.* 5 (2015) 1501560.
5. M. Prabu, P. Ramakrishnan, P. Ganesan, A. Manthiram, S. Shanmugam, LaTi_{0.65}Fe_{0.35}O_{3-δ} nanoparticle-decorated nitrogen-doped carbon nanorods as an advanced hierarchical air electrode for rechargeable metal-air batteries, *Nano Energy* 15 (2015) 92-103.

6. H.W. Park, D.U. Lee, P. Zamani, M.H. Seo, L.F. Nazar, Z. Chen, Electrospun porous nanorod perovskite oxide/nitrogen-doped graphene composite as a bi-functional catalyst for metal air batteries, *Nano Energy* 10 (2014) 192-200.
7. K. Elumeeva, J. Masa, J. Sierau, F. Tietz, M. Muhler, W. Schuhmann, Perovskite-based bifunctional electrocatalysts for oxygen evolution and oxygen reduction in alkaline electrolytes, *Electrochim. Acta* 208 (2016) 25-32.
8. S. Bie, Y. Zhu, J. Su, C. Jin, S. Liu, R. Yang, J. Wu, One-pot fabrication of yolk-shell structured $\text{La}_{0.9}\text{Sr}_{0.1}\text{CoO}_3$ perovskite microspheres with enhanced catalytic activities for oxygen reduction and evolution reactions, *J. Mater. Chem. A* 3 (2015) 22448-22453.
9. N.-I. Kim, R.A. Afzal, S.R. Choi, S.W. Lee, D. Ahn, S. Bhattacharjee, S.-C. Lee, J.H. Kim, J.-Y. Park, Highly active and durable nitrogen doped-reduced graphene oxide/double perovskite bifunctional hybrid catalysts, *J. Mater. Chem. A* 5 (2017) 13019-13031.
10. Z. Wu, L.-P. Sun, T. Xia, L.-H. Huo, H. Zhao, A. Rougier, J.C. Grenier, Effect of Sr doping on the electrochemical properties of bi-functional oxygen electrode $\text{PrBa}_{1-x}\text{Sr}_x\text{Co}_2\text{O}_{5+\delta}$, *J. Power Sources* 334 (2016) 86-93.
11. Y. Gorlin, T.F. Jaramillo, A Bifunctional Nonprecious Metal Catalyst for Oxygen Reduction and Water Oxidation, *J. Am. Chem. Soc.* 132 (2010) 13612-13614.
12. C. Zhang, M. Antonietti, T.-P. Fellingner, Blood Ties: Co_3O_4 Decorated Blood Derived Carbon as a Superior Bifunctional Electrocatalyst, *Adv. Funct. Mater.* 24 (2014) 7655-7665.
13. J.W.D. Ng, M. Tang, T.F. Jaramillo, A carbon-free, precious-metal-free, high-performance O_2 electrode for regenerative fuel cells and metal-air batteries, *Energy Environ. Sci.* 7 (2014) 2017-2024.

14. J. Masa, W. Xia, I. Sinev, A. Zhao, Z. Sun, S. Grützke, P. Weide, M. Muhler, W. Schuhmann, Mn_xO_y/NC and Co_xO_y/NC Nanoparticles Embedded in a Nitrogen-Doped Carbon Matrix for High-Performance Bifunctional Oxygen Electrodes, *Angew. Chem. Int. Ed.* 53 (2014) 8508-8512.
15. G.-L. Tian, Q. Zhang, B. Zhang, Y.-G. Jin, J.-Q. Huang, D.S. Su, Toward Full Exposure of “Active Sites”: Nanocarbon Electrocatalyst with Surface Enriched Nitrogen for Superior Oxygen Reduction and Evolution Reactivity, *Adv. Funct. Mater.* 24 (2014) 5956-5961.
16. G.-L. Tian, M.-Q. Zhao, D. Yu, X.-Y. Kong, J.-Q. Q. Zhang, F. Wei, Nitrogen-Doped Graphene/Carbon Nanotube Hybrids: In Situ Formation on Bifunctional Catalysts and Their Superior Electrocatalytic Activity for Oxygen Evolution/Reduction Reaction, *small* 10 (2014) 2251-2259.
17. N.R. Ashraie, J.P. Paraknowitsch, C. Göbel, A. Thomas, P. Strasser, Noble-Metal-Free Electrocatalysts with Enhanced ORR Performance by Task-Specific Functionalization of Carbon using Ionic Liquid Precursor Systems, *J. Am. Chem. Soc.* 136 (2014) 14486-14497.
18. Y. Zhao, R. Nakamura, K. Kamiya, S. Nakanishi, K. Hashimoto, Nitrogen-doped carbon nanomaterials as non-metal electrocatalysts for water oxidation, *Nat. Commun.*, 4 (2013) 2390.