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Supporting information

Robust FeCo Nanoparticles Embedded in N-doped Porous Carbon Framework for High Oxygen Conversion Catalytic Activity in Alkaline and Acidic Media

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Figure S1. Scanning electron microscope (SEM) image of bare PDA sub-microspheres.



Figure S2. (a) XRD patterns of M@N-C (Co@N/C, Fe@N/C, and FeCo@N/C) before acid leaching. (b) High resolution XRD patterns in the range of 40-60° for Co-N/C and Fe-N/C.



Figure S3. Raman spectra of bare N/C, Co-N/C, Fe-N/C, and FeCo-N/C.



Figure S4. Nitrogen adsorption/desorption isotherms and corresponding pore size distributions of Fe@N/C, Co@N/C, and FeCo@N/C.



Figure S5. TEM images of FeCo-N/C (a), Fe-N/C (b), and Co-N/C (c).

| | Surface area (m ² /g) | Pore volume | BJH pore diameter |
|----------|----------------------------------|--------------|-------------------|
| | | (cm^{3}/g) | (nm) |
| N/C | 214.8 | 0.17 | 2.1 |
| Fe@N/C | 220.6 | 0.21 | 4.2 |
| Co@N/C | 266.6 | 0.34 | 4.5 |
| FeCo@N/C | 268.9 | 0.31 | 5.0 |
| Fe-N/C | 384.5 | 0.43 | 8.0 |
| Co-N/C | 417.7 | 0.46 | 15.4 |
| FeCo-N/C | 442.6 | 0.47 | 19.0 |

Table S1. The average pore size and pore volume of the bare N/C, M@N/C (Fe@N/C, Co@N/C, and FeCo@N/C), and M-N/C (Fe-N/C, Co-N/C, and FeCo-N/C) according to N_2 adsorption/desorption measurements.



Figure S6. (a) TEM image and (b-e) selected-area HR-EDS mapping for Co and Fe for the indivial nanoparticle embedded in carbon layer.



Figure S7. XPS survey scans of bare N/C, Fe-N/C, Co-N/C, and FeCo-N/C.

Table S2. Elemental compositions of bare C/N, Fe-N/C, Co-N/C, and FeCo-N/C, as determined by XPS.

| | C atom % | N atom % | O atom % | NPM atom % |
|----------|----------|----------|----------|------------|
| N/C | 92.4 | 4.7 | 2.9 | |
| Co-N/C | 91.5 | 5.9 | 2.2 | 0.4 |
| Fe-N/C | 91.6 | 6.0 | 1.8 | 0.6 |
| FeCo-N/C | 91.8 | 5.4 | 2.2 | Co: 0.3 |
| | | | | Fe: 0.4 |

Table S3.Contents of metals in the M-N/C obtained by ICP-OES.

| | Fe-N/C | Co-N/C | FeCo-N/C |
|-------------|--------|--------|----------|
| Fe (atom.%) | 0.71 | - | 0.42 |
| Co (atom.%) | - | 0.59 | 0.36 |

| | Pyridinic-N (%) | Pyrrolic-N | Graphitic-N | Pyridinic- N ⁺ -O ⁻ |
|----------|--------------------|------------|-------------|--|
| N/C | 26.1 | 27.1 | 32.5 | 14.3 |
| Co-N/C | 49.2 | 3.4 | 37.8 | 9.6 |
| Fe-N/C | 50.6 | 6.0 | 33.4 | 10.2 |
| FeCo-N/C | 42.8 | 6.5 | 39.4 | 11.3 |

Table S4. Summary of quantitative analysis of N 1s XPS spectra for bare N/C, Co-N/C, Fe-N/C, and FeCo-N/C.



Figure S8. LSV curves

of various catalysts in O₂-saturated 0.1 M KOH electrolyte at different rotation rates and corresponding Koutecky-Levich plots.



Figure S9. LSV curves of various catalysts in O_2 -saturated 0.5 M H_2SO_4 electrolyte at different rotation rates and corresponding Koutecky-Levich plots.



Figure S10. OER polarization curves of FeCo-N/C and Pt/C catalysts. The potential at 10 mA cm⁻² for FeCo-N/C and Pt/C are 1.617 V and 1.737 V, respectively, suggesting the high OER catalytic activity of FeCo-N/C.