Supporting Information

Insights into CO₂ Adsorption and Chemical Fixation Properties of VPI-100 Metal-Organic Frameworks

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Section 1. Structure determination and refinement of Hf-VPI-100 (Cu) and Hf-VPI-100 (Ni)

XRD indexing and Rietveld analysis was performed with TOPAS version 5. Figure 2 shows the XRD profiles of Hf-VPI-100 (Cu) and Hf-VPI-100 (Ni) at room temperature after activation at 100 °C under helium flow. The high similarity of the diffraction patterns suggested the Cu and Ni analogues are isostructural. Indexing of the Hf-VPI-100 (Ni) patterns suggested a body-centered tetragonal lattice with unit cell parameters: $a \sim 16.6$ Å and $c \sim 18.6$ Å. For Rietveld refinement, our previously published Zr-VPI-100 structure was used as the initial model, in which Zr atoms were substituted by Hf. Rigid body constraint was applied to the cyclam ligand and the Hf₆ cluster. The final CIFs are included with this submission and the refinement plots are shown as Figure S1 and S2.



Figure S1. Refinement plots of Hf-VPI-100 (Ni) as synthesized under room temperature. The bottom panel is a zoomed-in view of the low intensity peaks in the top panel. The blue line is the observed curve, the red line is the calculated curve, and the grey line oscillating around 0 is the difference between the observed and the calculated.



Figure S2. Refinement plots of Hf-VPI-100 (Cu) as synthesized under room temperature. The bottom panel is a zoomed-in view of the low intensity peaks in the top panel. The blue line is the observed curve, the red line is the calculated curve, and the grey line oscillating around 0 is the difference between the observed and the calculated.

Table S1. Crystallographic data for Hf-VPI-100 (Cu) and Hf-VPI-100 (Ni) and refinement statistics

	Hf-VPI-100 (Cu)	Hf-VPI-100 (Ni)
Chemical Formula*	$C_{24} H_{44} Cl_{2.68} Cu_2 Hf_3 N_8 O_{16}$	$C_{24} H_{44} Cl_{1.64} Ni_2 Hf_3 N_8 O_{16}$
Space group	<i>I</i> 4/m	<i>I</i> 4/m
a (Å)	16.793(1)	16.594(1)
<i>c</i> (Å)	19.020 (3)	18.661(2)
V (Å ³)	5364(1)	5139(1)
Temperature (K)	298	298
Wavelength (Å)	0.45260(2)	0.45260(2)
2θ range (°)	1.5-16.4	1.5-17.6
R_p	0.0529	0.0537

*Charge balancing chloride and solvent in the MOF pores are not included in the chemical formula

Section 2. Characterization of Hf-VPI-100 (Cu) and Hf-VPI-100 (Ni)



Figure S3. SEM image of the microcrystalline powder of Hf-VPI-100 (Cu) (left) and Hf-VPI-100 (Ni) (right).



Figure S4. PXRD patterns of as synthesized Hf-VPI-100 and Zr-VPI-100 powders measured at room temperature.



Figure S5. XPS spectrum of Hf-VPI-100 (Cu) powder displaying (left to right) Hf4 $f_{7/2}$, N1s and Cu2 $p_{3/2}$ binding energy regions.



Figure S6. XPS spectrum of Hf-VPI-100 (Ni) powder displaying (left to right) Hf4 $f_{7/2}$, N1s and Ni2 $p_{3/2}$ binding energy regions.



Figure S7. FTIR spectra of metal-cyclam linkers, Zr-VPI-100 MOFs and Hf-VPI-100 MOFs.



Figure S8. TGA curves of Hf-VPI-100 (Cu) and Hf-VPI-100 (Ni).



Figure S9. N_2 adsorption isotherms of (A) VPI-100 (Cu) and (B) VPI-100 (Ni) at 77 K, 1 atm after activation at 100 °C for 24 h.

MOF	Activation Condition	BET surface areas (N ₂ , 77 K)	Pore size	CO ₂ sorption at 273 K unde 1 atm	
				STP (cm ³ /g)	wt. %
VPI-100 (Cu)	373 K	399 m ² /g	6.37 Å	33.85	6.65
VPI-100 (Ni)	373 K	344 m ² /g	5.77 Å	28.2	5.52

Hf- VPI-100 (Cu)	373 K	396 m ² /g	7.7 Å	41.60	8.25
Hf- VPI-100 (Ni)	373 K	399 m²/g	7.2 Å	46.05	9.13

Section 3. Catalytic studies of cycloaddition of CO₂ to epoxides using Hf-VPI-100

Determination of the catalytic conversion.



Scheme S1. Synthesis of cyclic carbonates from CO₂ and epoxides catalyzed by Hf-VPI-100.

For each reaction, conversion was determined by comparison of the ¹H NMR integrals of the corresponding proton in the starting material (¹Hc) and in the product (¹Hc') according to equation 1 and Table S2.

$$Conversion = \frac{I_{Hc}'}{(I_{Hc} + I_{Hc}')}$$
(1)

Table S3. Chemical shifts (δ , ppm) for the integrated proton in the epoxides and in the corresponding carbonates (in CDCl₃).

Epoxides	δH (CDCl ₃)(epoxide, ¹ Hc)	δH (CDCl ₃)(carbonate, ¹ Hc')
а	2.65	4.35
b	2.45	4.05



Figure S10. ¹H NMR (400 MHz, CDCl₃) spectrum of reaction mixture in Table 1 entry 1 only using TBAB as catalysts. Conversion = 1/6.18 = 16.2%



Figure S11. ¹H NMR (400 MHz, CDCl₃) spectrum of reaction mixture in Table 1 entry 2 using $[CuL(ClO_4)_2]$ (1) and TBAB as catalysts. Conversion = 1/2.12, = 90.2%.



Figure S12. ¹H NMR (400 MHz, CDCl₃) spectrum of reaction mixture in Table 1 entry 3, run 1 using Hf-VPI-100 (Cu) and TBAB as catalysts. Conversion = 1.05/1.08, = 97.2%.



Figure S13. ¹H NMR (400 MHz, CDCl₃) spectrum of reaction mixture in Table 1 entry 4, run 2 using Hf-VPI-100 (Cu) and TBAB as catalysts. Conversion = 1.11/1.16, = 95.6%.



Figure S14. ¹H NMR (400 MHz, CDCl₃) spectrum of reaction mixture in Table 1 entry 5, run 1 using Hf-VPI-100 (Ni) and TBAB as catalysts. Conversion = 1.11/1.23, = 89.5%.



Figure S15. ¹H NMR (400 MHz, CDCl₃) spectrum of reaction mixture in Table 1 entry 6, run 2 using Hf-VPI-100 (Ni) and TBAB as catalysts. Conversion = 1.12/1.3, = 86.2%.



Figure S16. PXRD of Hf-VPI-100 (Cu) after catalysis.



Figure S17. ¹H NMR (400 MHz, CDCl₃) spectrum of reaction mixture in run 1 using Hf-VPI-100 (Cu) and TBAB as catalysts. Conversion = 1.09/2.46, = 43.8%.



Figure S18. ¹H NMR (400 MHz, CDCl₃) spectrum of reaction mixture in run 1 using Hf-VPI-100 (Ni) and TBAB as catalysts. Conversion = 1.09/2.77, = 39.4%.



Figure S19. ¹H NMR (400 MHz, CDCl₃) spectrum of reaction mixture in run 1 using Zr-VPI-100 (Cu) and TBAB as catalysts. Conversion = 1.1/1.99, = 55.3%.



Figure S20. ¹H NMR (400 MHz, CDCl₃) spectrum of reaction mixture in run 1 using Zr-VPI-100 (Ni) and TBAB as catalysts. Conversion = 1.09/2.70, = 40.3%.



Figure S21. ¹H NMR (400 MHz, CDCl₃) spectrum of reaction mixture using only TBAB as catalysts. Conversion = 1.09/8.04, = 13.5%.

Section 4. The structure of metallo-cyclam ligands



Figure S22. Anisotropic displacement ellipsoid drawings (30%) of (A) $[CuL(ClO_4)_2]$ and (B) $[NiL](ClO_4)_2$. Hydrogen atoms were omitted forclarity. Color scheme: C, gray; O, red; N, blue; Ni, light green; Cu, yellow; Cl, green.

Section 5. Xyz coordinates of optimized structures

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NU-1000(Hf)--EPO TS

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0	1.299658	1.312189	-3.174497
0	-0.360420	3.899175	-1.063331
С	4.109178	3.122760	0.292396
С	4.977946	2.905607	1.372433
С	6.109893	3.697781	1.537560
C	6 407406	4 738663	0 636972
C	5 543606	4 930881	-0 459256
C	1 115000	1 133085	-0 630633
с u	9.415000	2 929027	4 669509
	0.391040	2.929027	4.000500
н	0.481443	-2.390662	4.844422
H	-5.82/561	-2.985033	-0.243057
H	-2.581196	-3.114165	-4.391998
Н	-2.703739	2.727806	-4.610877
H	-5.882795	2.854503	-0.436079
Н	4.756040	2.100437	2.073825
Н	3.746284	4.292880	-1.479409
Н	5.758340	5.719606	-1.184919
Н	6.776589	3.509870	2.383171
Н	4.841611	-1.746412	2.203773
Н	3.875050	-4.241846	-1.158966
Н	5.872097	-5.648308	-0.701204
н	6 849558	-3 129919	2 668825
н	7 290989	5,364832	0.772755
н	7 384314	-5 117676	1 231503
н Ц	-0 867828	-0 002120	-3 386/15
11 11	0.00/020	0.092130 2 0222305	1 160603
n U	0.433773	Z.YZZJYJ 1 feoded	T.TQAQA2
н	-2.9439/9	-1.339233	4.040235
H	-3.46/059	-0.2/3081	3.2/0991
H	2.318132	-1.378795	-3.091314
Н	-0.769972	-4.051042	-0.118665

Н	-2.528632	-4.418229	1.038642
Н	-0.624533	-4.375353	-1.661190
Н	-4.320540	-0.056704	1.116151
Н	0.592953	-2.783123	1.287802
Н	-3.206246	1.738012	3.921513
Н	1.377442	0.285832	-3.208848
Н	0.999394	1.589811	-4.048745
Н	-2.803817	4.320570	0.695195
Н	-0.707784	4.364994	-1.832829
Н	-1.333157	4.053852	0.159504
0	3.913146	-0.958877	-3.347511
С	5.450469	-0.840945	-2.175861
С	4.243159	-0.044801	-2.377221
Н	6.219129	-0.784660	-2.944094
Н	5.352559	-1.788004	-1.647696
Н	4.425245	0.999517	-2.715766
Н	3.562380	0.006941	-1.495986
Br	6.918993	0.147670	-0.447571

Cy(Ni)-EPO

С	2.725269	1.369079	0.191798
Ν	1.299366	1.577029	0.478319
Ni	0.074650	0.007916	-0.193114
Ν	-1.207106	-1.550097	-0.762411
С	-2.631983	-1.353273	-0.493246
С	-3.166294	-0.059694	-1.086235
Н	-4.263347	-0.093757	-1.028769
С	3.263065	0.071820	0.772158
С	2.803290	-1.209257	0.096319
Ν	1.396309	-1.523673	0.378545
С	0.883365	-2.710068	-0.321056
С	-0.624785	-2.765830	-0.188769
Н	4.359406	0.108507	0.699840
С	0.724538	2.771203	-0.156408
С	-0.785037	2.739469	-0.036786
N	-1.296542	1.518714	-0.665204
С	-2.711877	1.232468	-0.427839
0	-0.818951	-0.204807	1.877322
Cl	0.877923	0.112042	-2.463025
Н	1.303661	-1.683264	1.384051
Н	1.182937	1.663302	1.490897
Н	1.344885	-3.641614	0.046779
Н	1.161002	-2.605173	-1.381835
Н	2.897569	-1.113266	-0.996129
Н	3.451709	-2.048140	0.405400
Н	3.053768	0.024555	1.856681
Н	3.318817	2.220978	0.568505
Н	2.831476	1.360717	-0.903849
Н	1.013493	2.741126	-1.218869
Н	1.130878	3.705123	0.266149
Н	-1.128531	1.576651	-1.674062
Н	-1.050447	-1.556262	-1.774382

Η	-1.226938	3.651278	-0.473199
Н	-1.086636	2.728720	1.023883
Н	-2.869522	1.180621	0.662208
Н	-3.340207	2.068162	-0.785064
Н	-2.937657	-0.024300	-2.166064
Н	-3.219625	-2.204725	-0.881343
Н	-2.767991	-1.357847	0.600506
Н	-0.916792	-2.807500	0.873843
Н	-1.021052	-3.682085	-0.658496
С	-1.567463	0.329346	2.964548
С	-0.406134	-0.510566	3.206022
Н	0.538080	-0.051063	3.519553
Н	-0.539828	-1.557345	3.499034
Н	-1.487982	1.413085	3.102754
Н	-2.574326	-0.086378	3.074255

Cy(Ni)-EPO TS

С	-0.532650	2.667924	1.408575
Ν	-0.320750	1.226200	1.589285
Ni	-1.097586	0.061299	0.009722
Ν	-1.899225	-1.115581	-1.542101
С	-1.713168	-2.556917	-1.389531
С	-2.264165	-3.082870	-0.073488
Н	-2.279010	-4.180615	-0.133304
С	0.028919	3.195607	0.098178
С	-0.731919	2.820979	-1.163953
Ν	-0.580825	1.408666	-1.518647
С	-1.373675	0.973190	-2.672252
С	-1.345237	-0.539462	-2.767397
Н	0.038678	4.293243	0.163613
С	-0.961115	0.671931	2.789440
С	-0.956923	-0.842743	2.730965
Ν	-1.650349	-1.287256	1.521950
С	-1.505117	-2.702278	1.188029
0	0.705414	-0.936345	-0.405637
Cl	-3.355616	1.038734	0.255903
Н	0.404197	1.211682	-1.708462
Н	0.685200	1.053473	1.649178
Н	-1.026559	1.429828	-3.615511
Н	-2.406835	1.317618	-2.503965
Н	-1.808328	3.000823	-1.016301
Н	-0.406429	3.468537	-1.998806
Н	1.093112	2.913745	-0.000286
Н	-0.092925	3.228714	2.253161
Н	-1.620317	2.834816	1.440270
Н	-2.000316	1.037043	2.794096
Н	-0.484878	1.026957	3.719363
Н	-2.643378	-1.051874	1.605445
Н	-2.894261	-0.883871	-1.482446
Н	-1.401812	-1.262316	3.650375
Н	0.076810	-1.223322	2.690197
Н	-0.430031	-2.910399	1.059307

Η	-1.846051	-3.338189	2.026197
Н	-3.323875	-2.788027	0.031699
Н	-2.186057	-3.102424	-2.227490
Н	-0.631324	-2.753719	-1.450539
Н	-0.305244	-0.893398	-2.858165
Н	-1.879440	-0.879397	-3.672249
С	1.815318	-1.283911	0.363150
С	2.408929	-0.363620	-0.582685
Н	2.347602	0.706281	-0.398644
Н	2.672373	-0.707649	-1.578903
Н	1.809414	-0.961460	1.420578
Н	2.119137	-2.342258	0.299703
Br	4.985656	-0.157118	-0.041805

Cy(Cu)-EPO

С	-2.673528	-1.400906	0.477758
N	-1.219319	-1.573665	0.552706
Cu	-0.142509	0.005180	-0.229843
Ν	1.056614	1.579869	-0.837018
С	2.504995	1.397703	-0.751051
С	2.981040	0.155785	-1.480478
Н	4.076168	0.205121	-1.552132
С	-3.153005	-0.143496	1.175466
С	-2.790228	1.157437	0.486389
N	-1.360101	1.471160	0.593068
С	-0.964420	2.699033	-0.111065
С	0.542178	2.777553	-0.166710
Н	-4.248480	-0.194603	1.242405
С	-0.725272	-2.754371	-0.167385
С	0.781353	-2.703822	-0.239780
N	1.185714	-1.452346	-0.887343
С	2.619141	-1.168687	-0.836696
0	1.063721	-0.089304	2.035012
Cl	-1.216574	-0.014144	-2.512950
Н	-1.113016	1.558238	1.582247
Н	-0.925458	-1.623169	1.532369
Н	-1.394703	3.598419	0.358712
Н	-1.375058	2.629768	-1.130957
Н	-3.021094	1.094529	-0.588783
Н	-3.387637	1.986672	0.903347
Н	-2.802518	-0.131818	2.223214
H	-3.178967	-2.284961	0.904213
H	-2.932273	-1.360840	-0.592285
Н	-1.152422	-2.712990	-1.182009
H	-1.070065	-3.692371	0.297666
H	0.862361	-1.456495	-1.861997
H	0.759119	1.588672	-1.819353
H	1.178276	-3.586984	-0.767576
H	1.217565	-2.718333	0.772323
H	2.924719	-1.174791	0.221681
H	3.182419	-1.983438	-1.325158
Н	2.623146	0.175738	-2.524514

Н	3.021370	2.285415	-1.157303
Н	2.770790	1.343765	0.317709
Н	0.967730	2.804642	0.849752
Н	0.869441	3.702671	-0.669526
С	2.102162	-0.466871	2.930448
С	1.108397	0.516018	3.321720
Н	0.296828	0.232180	4.000704
Н	1.373349	1.579340	3.329054
Н	2.038360	-1.490865	3.313815
Н	3.107576	-0.136239	2.646293

Cy(Cu)-EPO TS

С	-1.787430	-2.597444	-1.331181
Ν	-1.256180	-1.261295	-1.585810
Cu	-1.249515	0.022776	0.032616
Ν	-1.148050	1.324095	1.642086
С	-0.639032	2.668258	1.384364
С	-1.362622	3.362392	0.245463
Н	-1.062479	4.419434	0.251342
С	-1.073368	-3.303275	-0.193633
С	-1.335706	-2.740948	1.190981
Ν	-0.703375	-1.439064	1.393926
С	-0.940664	-0.840302	2.708415
С	-0.449869	0.588810	2.697811
Н	-1.392425	-4.354930	-0.196877
С	-1.930200	-0.529354	-2.658135
С	-1.443234	0.900947	-2.665370
Ν	-1.698852	1.497902	-1.353220
С	-1.094310	2.810197	-1.142397
0	1.010422	0.012373	-0.467450
Cl	-3.787860	-0.161649	0.501652
Н	0.302919	-1.495872	1.212480
Н	-0.245213	-1.288656	-1.757001
Н	-0.464329	-1.416855	3.519595
Н	-2.028762	-0.862640	2.883466
Н	-2.417920	-2.597687	1.343194
Н	-0.988626	-3.456778	1.957969
Н	0.013961	-3.330650	-0.389788
Н	-1.727562	-3.212266	-2.247735
Н	-2.856334	-2.477308	-1.089760
Н	-3.011175	-0.558374	-2.443688
Н	-1.777817	-1.002867	-3.643191
Н	-2.709798	1.535428	-1.180548
Н	-2.153523	1.352927	1.846735
Н	-1.917395	1.474543	-3.480098
Н	-0.355241	0.932942	-2.840144
Н	-0.009278	2.707892	-1.309826
Н	-1.459206	3.524856	-1.902226
Н	-2.449294	3.368569	0.441253
Η	-0.709707	3.285010	2.298544
Н	0.435150	2.578986	1.154394
Н	0.628866	0.621803	2.470020

Н	-0.582664	1.058387	3.687450
С	2.153697	0.779601	-0.617546
С	2.787452	-0.447708	-0.172045
Н	2.920948	-1.261708	-0.879709
Н	2.781628	-0.685419	0.889260
Н	2.390919	1.104963	-1.648847
Н	2.257212	1.646005	0.065577
Br	5.356293	-0.125134	-0.060155