

Engineering of TiO_2 Anode toward a Record High Initial Coulombic Efficiency Enabling High-Performance Low-Temperature Na-ion Hybrid Capacitors

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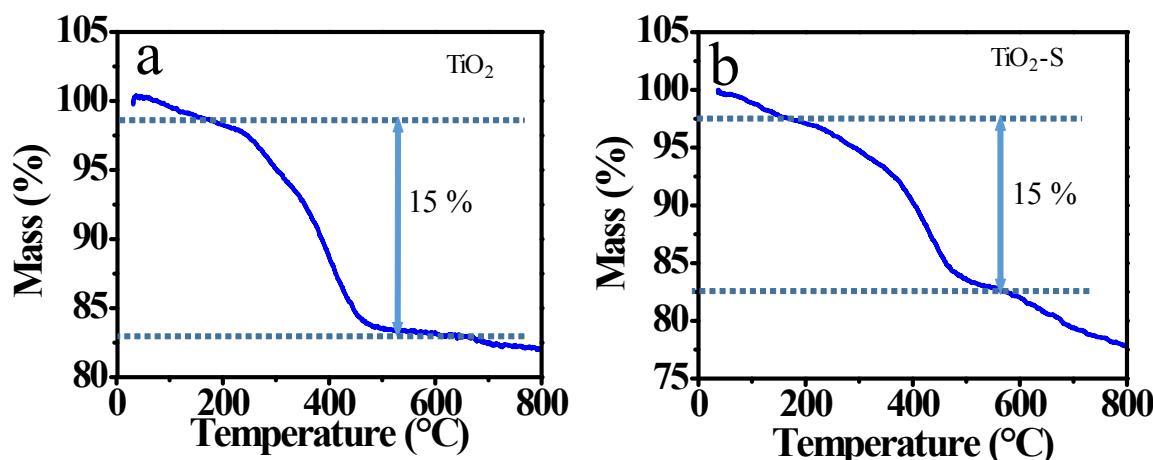


Fig. S1. TG curves of TiO_2 (a) and $\text{TiO}_2\text{-S}$ (b).

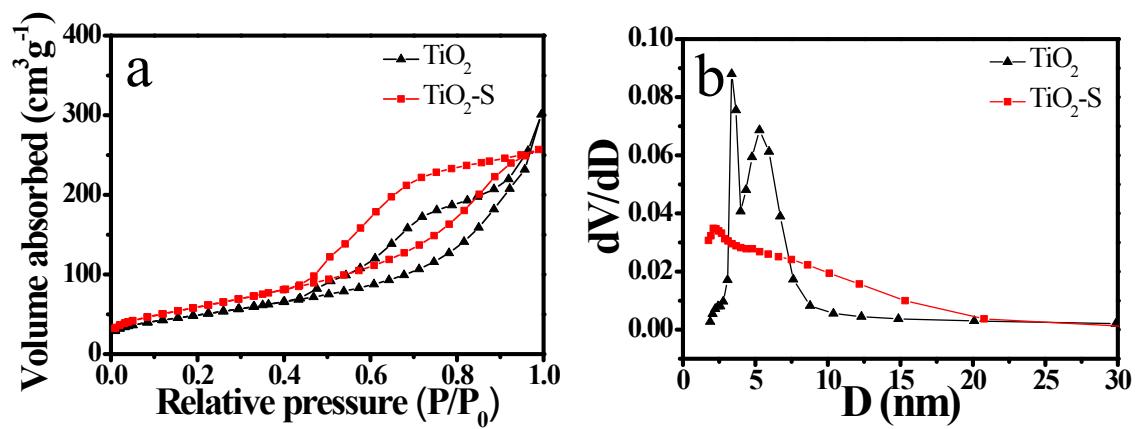


Fig. S2. (a) Nitrogen adsorption-desorption curves of TiO_2 and $\text{TiO}_2\text{-S}$, (b) the corresponding BJH pore size distribution.

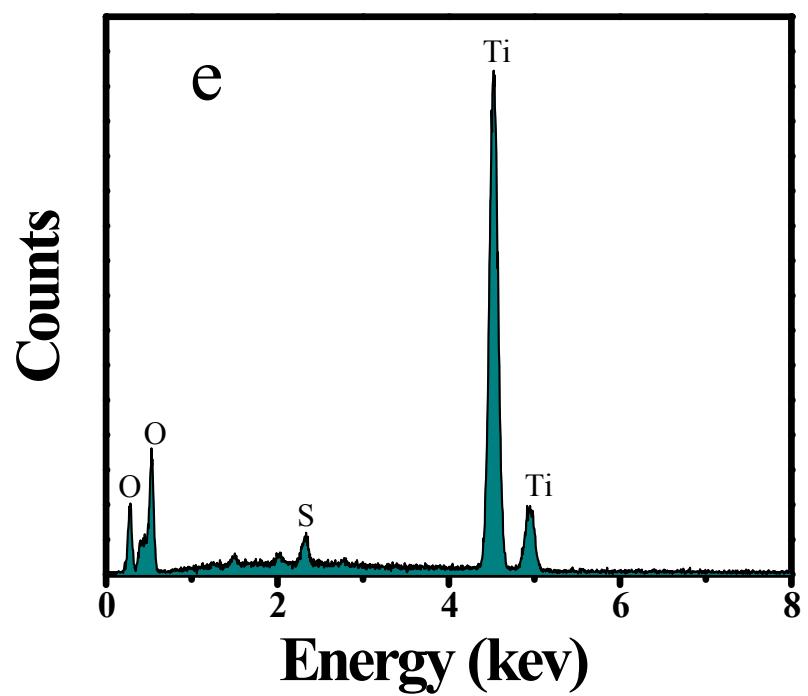
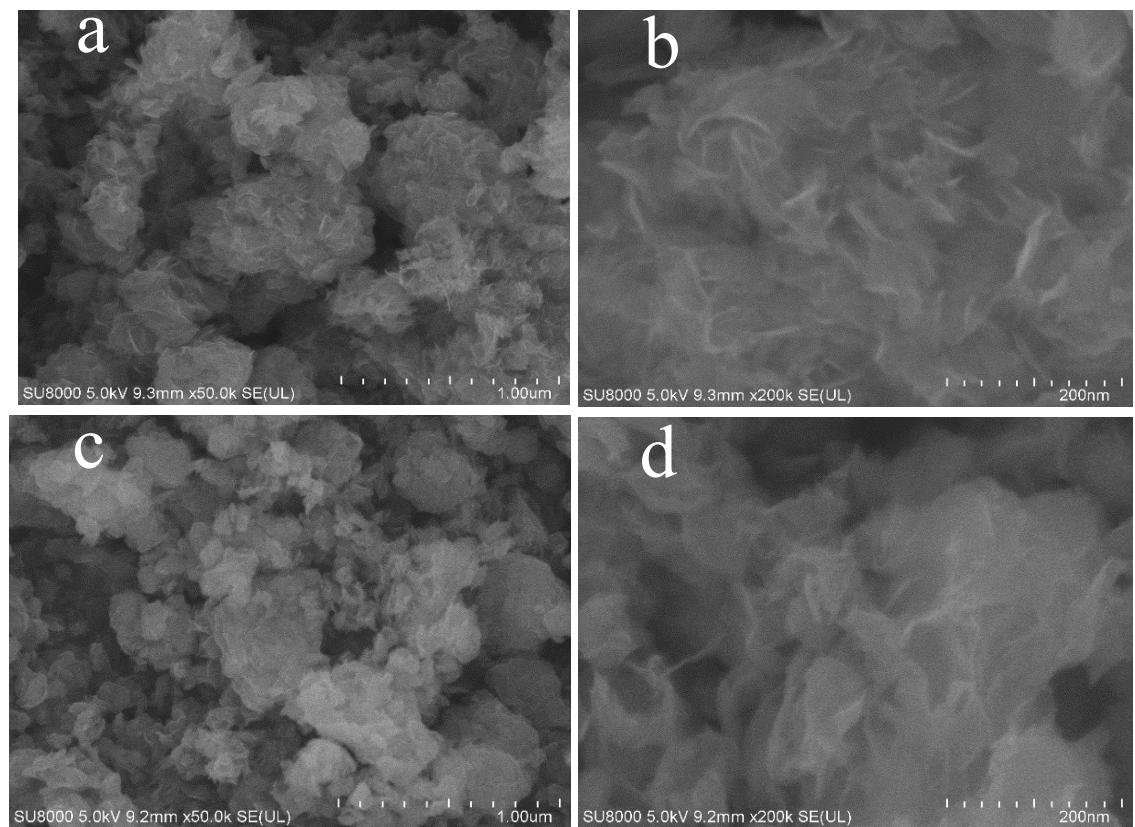


Fig. S3. SEM images of (a, b) the as-made product, (c, d) TiO_2 (e) EDS pattern of $\text{TiO}_2\text{-S}$.

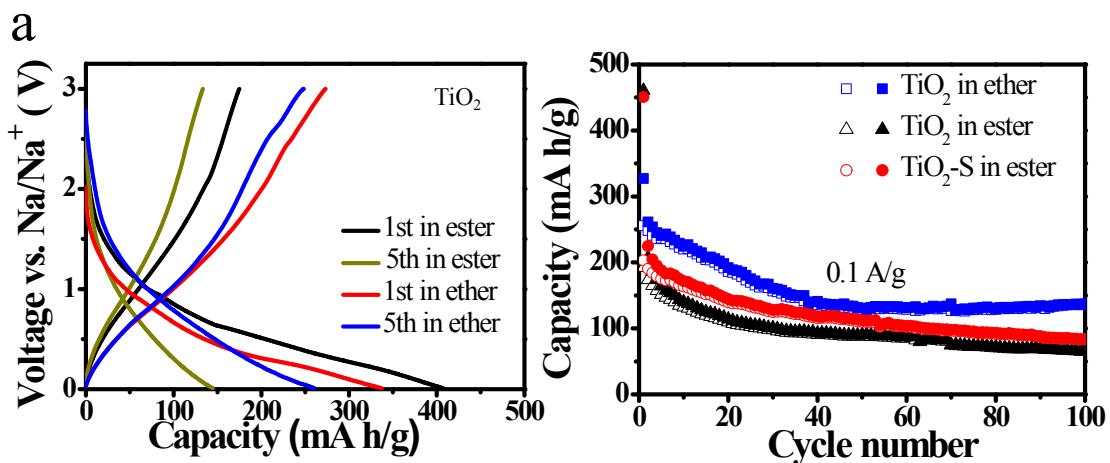


Fig. S4. a) Charge-discharge profiles for TiO_2 in ether and ester under 0.05 A/g . b) Cycling stability of the different electrodes in different electrolytes.

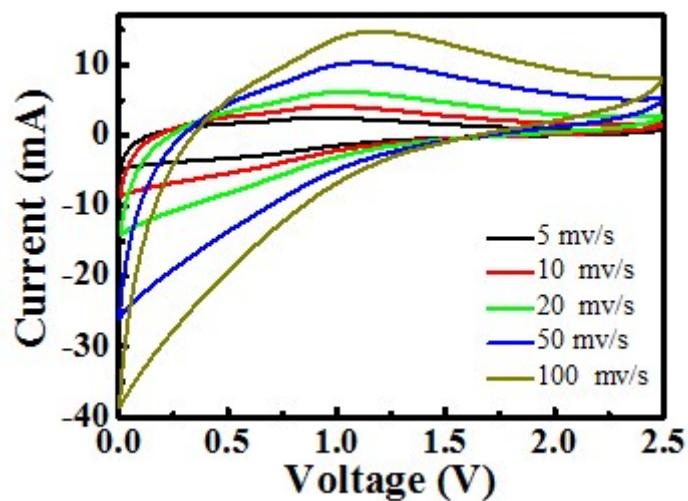
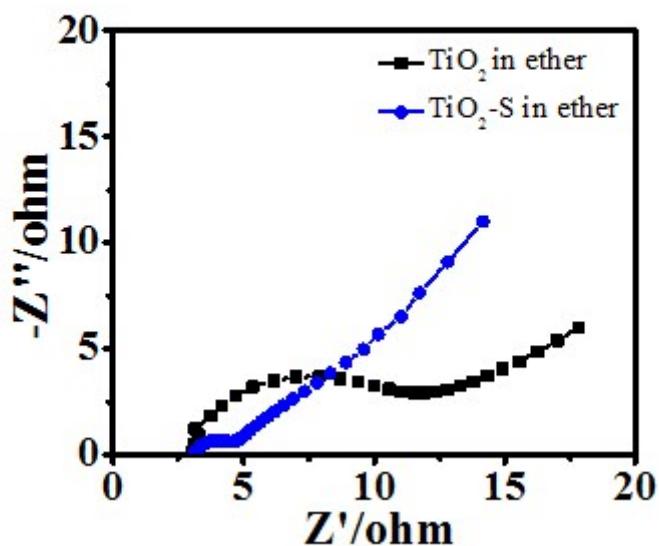


Fig. S5. CV curves of $\text{TiO}_2\text{-S}$ electrode in ether electrolyte under high sweep rates.

a



b

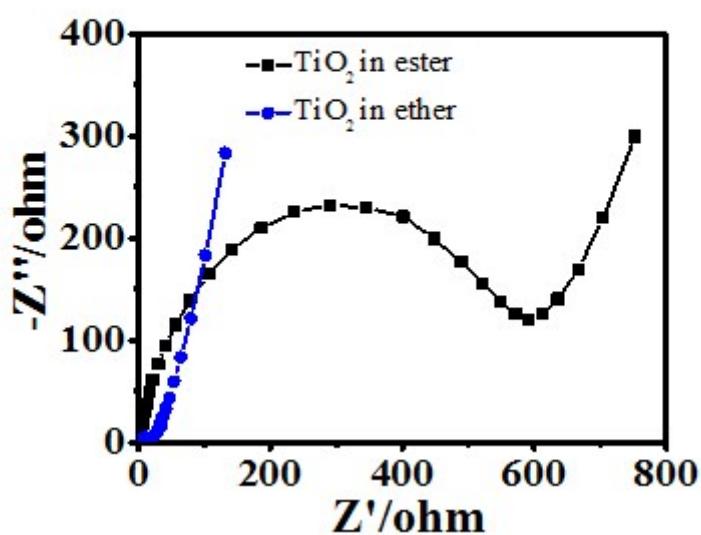


Fig. S6. EIS plots for a) different electrodes in ether and b) the TiO_2 electrode in different electrolytes.

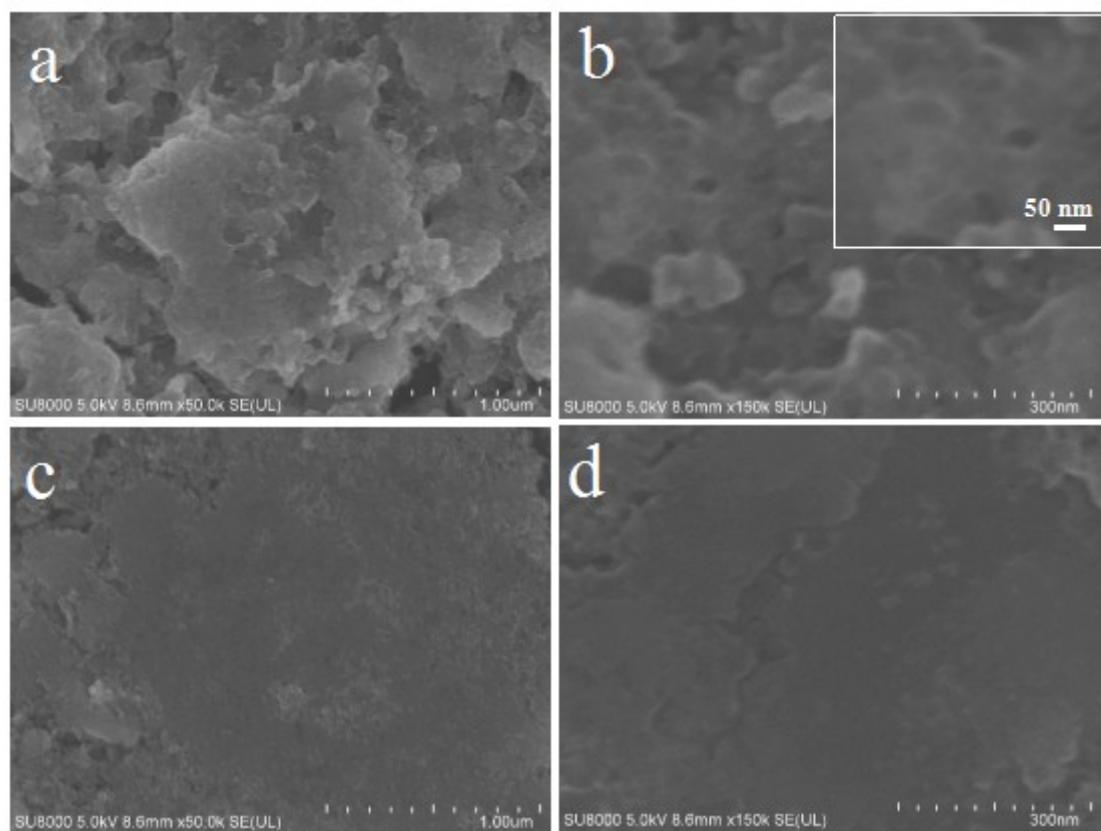


Fig. S7. SEM images for TiO_2 -S electrode in (a, b) ether electrolyte and (c, d) in ester after the first discharge. The inset in b is the enlarged SEM image.

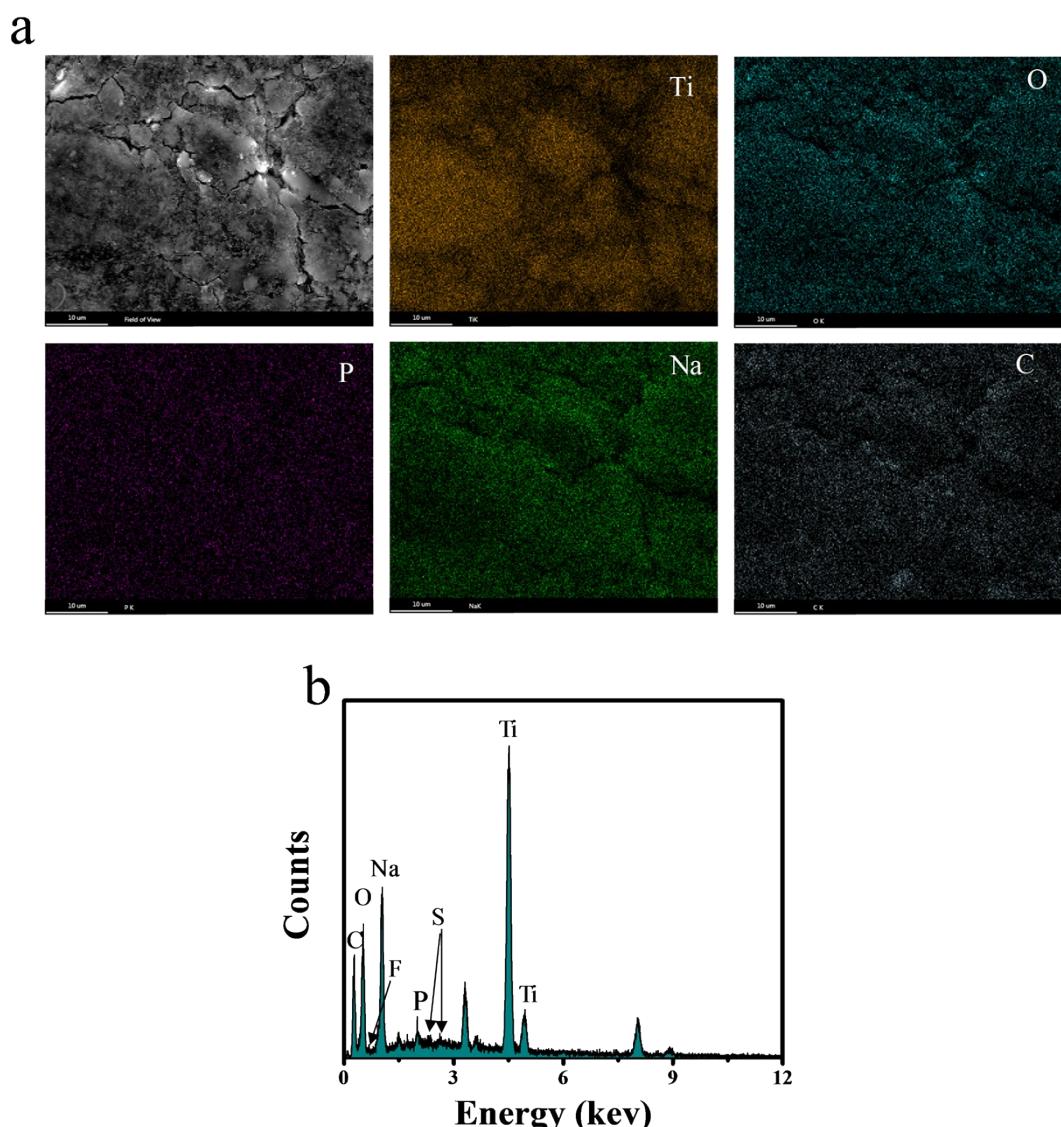
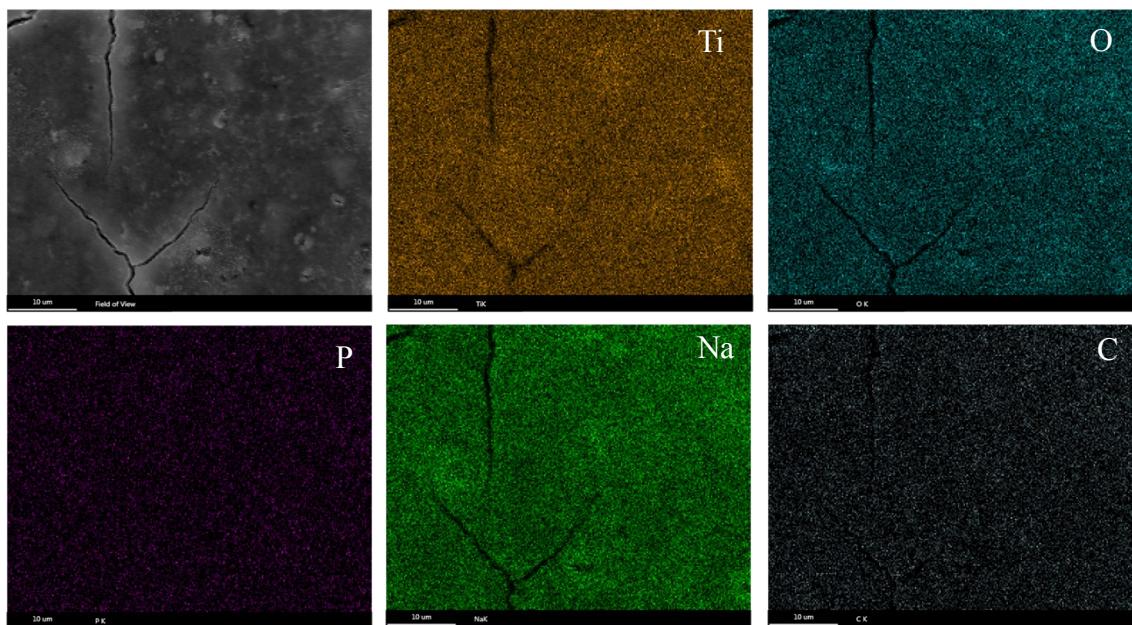


Fig. S8. The elements mapping (a) as well as correspnding full area EDS (b) of $\text{TiO}_2\text{-S}$ electrodes after first discharge in ester.

a



b

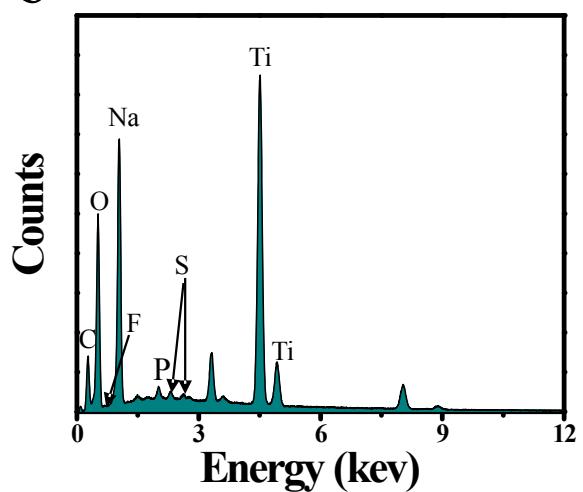


Fig. S9. The elements mapping (a) as well as correspindng full area EDS (b) of $\text{TiO}_2\text{-S}$ electrodes after first discharge in ether.

Table S1 The various element mass percent in titanium dioxide nanosheets with sulfur doing in ester electrodes after first discharge.

Element	Ti	O	C	Na	F	P	S
Ether (%)	10.1	47.03	20.37	20.9	1.16	0.29	0.16
Ester (%)	10.22	37.99	35.12	14.07	2.01	0.41	0.18

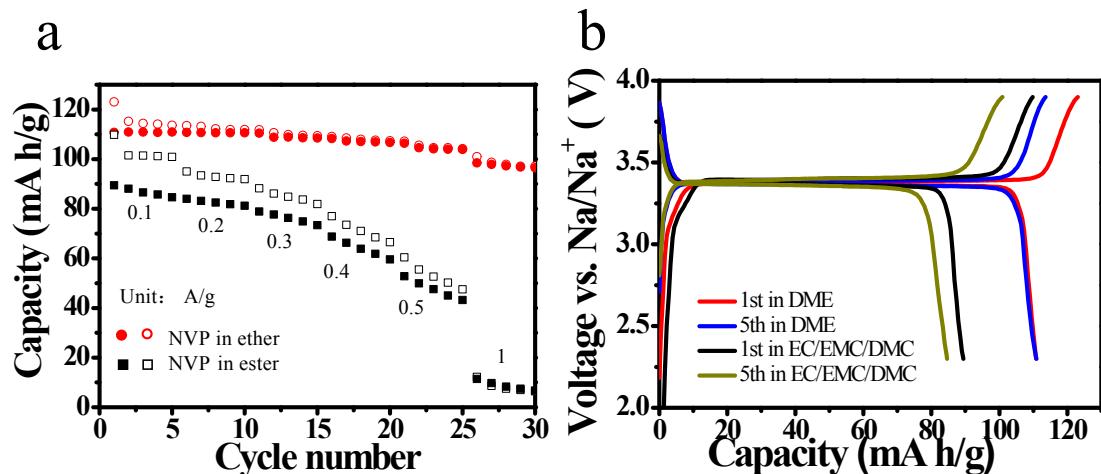


Fig. S10. The electrochemical performance of NVP cathode evaluated by half-cells within different electrolytes: a) rate capability combining with b) charge/discharge profiles at the selective cycles at 0.1 A/g.

Table S2 Comparison of representative electrochemical energy storage devices.

Anode//Cathode	Potential range (V)	Energy density (Wh/kg)	Power density (W/kg)	Cycling life	Reference
V ₂ O ₅ -CNT//AC	0-2.8	38	140	≈80% at 60C after 900 cycles	
MWTOG//AC	0.01-3	25.8	1357	≈90% at 10C after 10000 cycles	
Na ₃ V ₂ (PO ₄) ₃ @C//AC	0-3	118	96	≈95% at 1.1 mA cm ⁻² after 10 000 cycles	1
Na ₂ Ti ₉ O ₁₉ //PC	0-2.5	54	687	≈75% at 2 A g ⁻¹ after 2000 cycles	2
Nb ₂ O ₅ @C/rGO//AC	1-4.3	76	80	≈100% at 1 A g ⁻¹ after 3000 cycles	3
SCN-A//SCN-A	0-4	112	67	≈85% at 5 A g ⁻¹ after 3000 cycles	4
Na-TNTs//AC	0-3	34	889	≈90% at 0.25 A g ⁻¹ after 1000 cycles	5
NaTi ₂ O ₄ (OH) ₂ //PC	0-3	65	500	≈93% at 1 A g ⁻¹ after 3000 cycles	6
Nb ₂ O ₅ //PSC	1-3	43.2	5760	≈80% at 1.2 A g ⁻¹ after 3000 cycles	7
N-TiO ₂ //AC	1-4	80.3	500	≈80% at 1 A g ⁻¹ after 6500 cycles	
TiO₂-S//Na₂V₃(PO₄)₃	0.01-3.5	158	1075	≈83% at 1 A g ⁻¹ after 1000 cycles	this work

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