

Electronic Supplementary Information (ESI)

Pyrazine-incorporated graphdiyne nanofilm as a metal-free electrocatalyst for hydrogen evolution reaction

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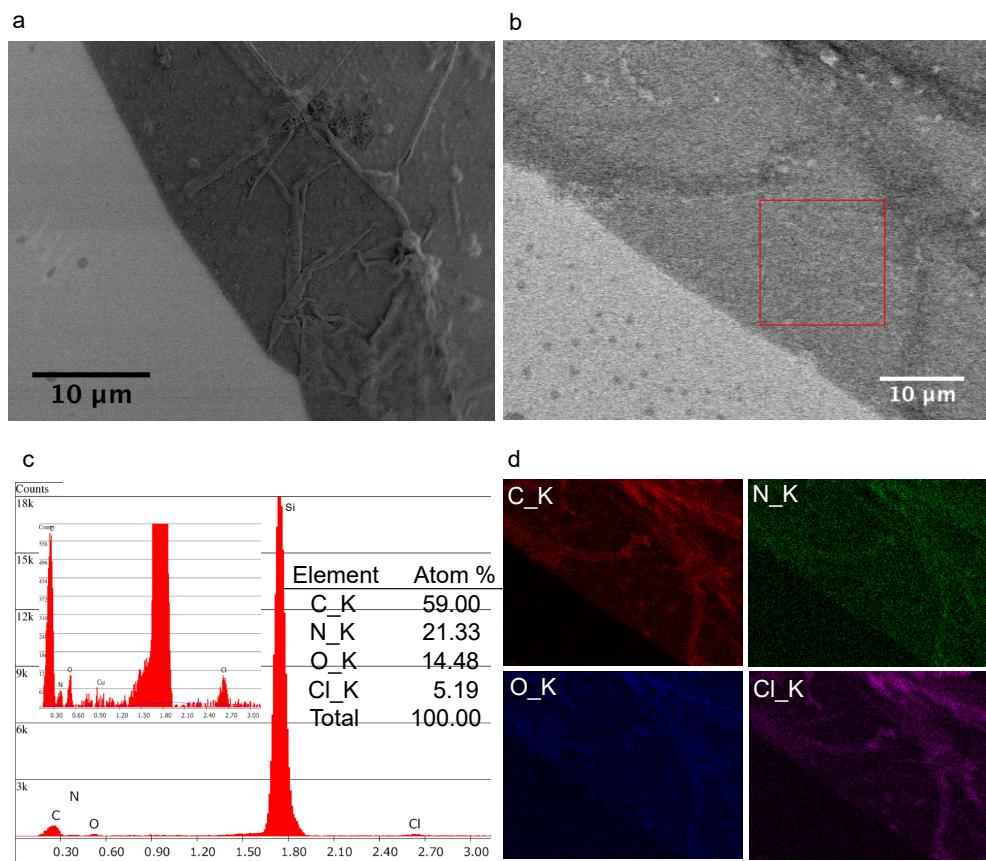


Fig. S1 (a,b) SEM image of a **PR-GDY** nanofilm on a Si/SiO₂ substrate. (c,d) EDX spectrum [red square region in (b)] and mapping.

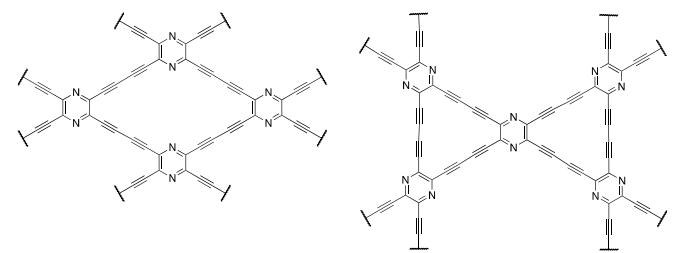


Fig. S2 Two possible 2D lattices for **PR-GDY**.

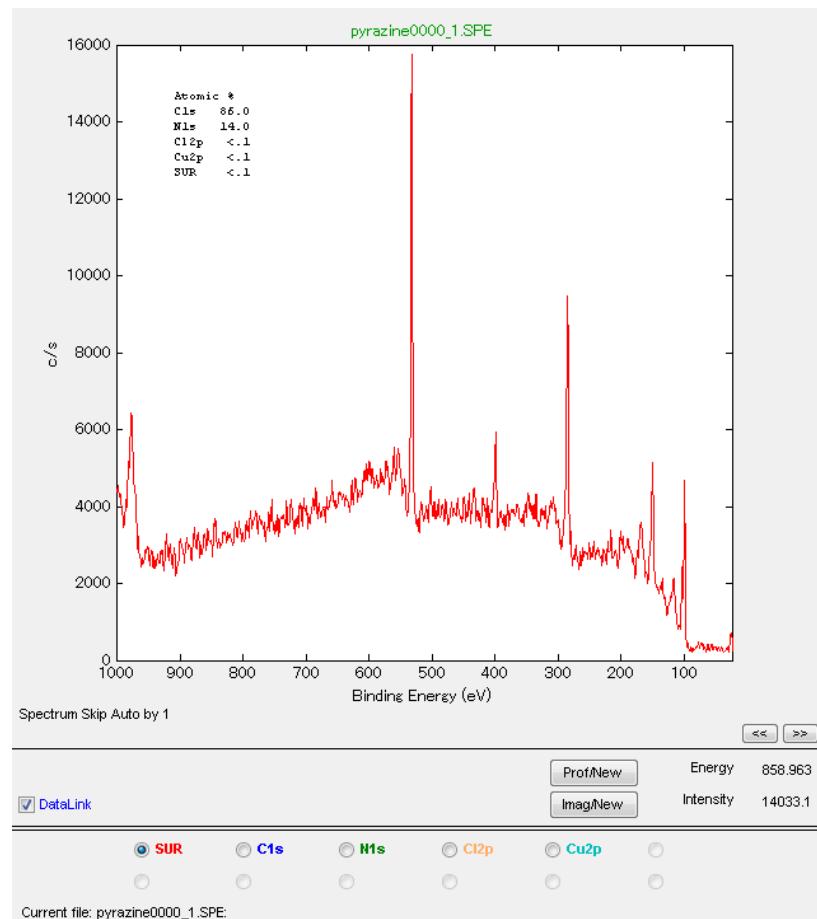


Fig. S3 Quantification of the C/N ratio of PR-GDY in XPS.

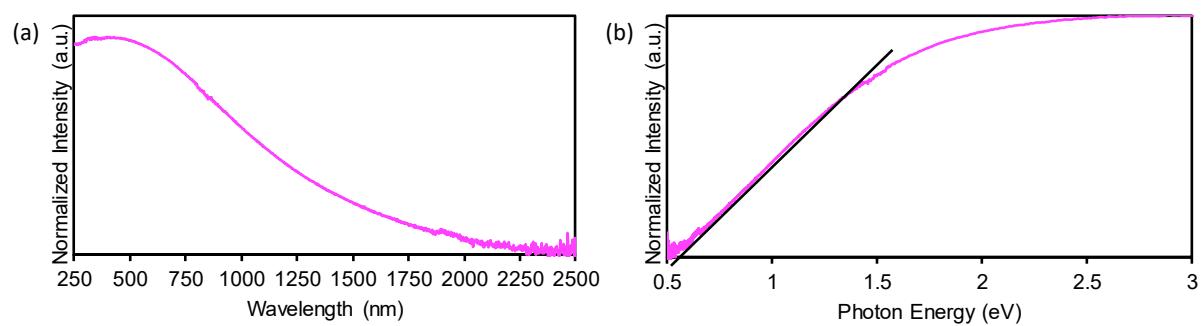


Fig. S4 (a,b) Absorption spectra of **PR-GDY** (as a pellet; magenta solid lines) with horizontal axes in nm and eV, respectively.

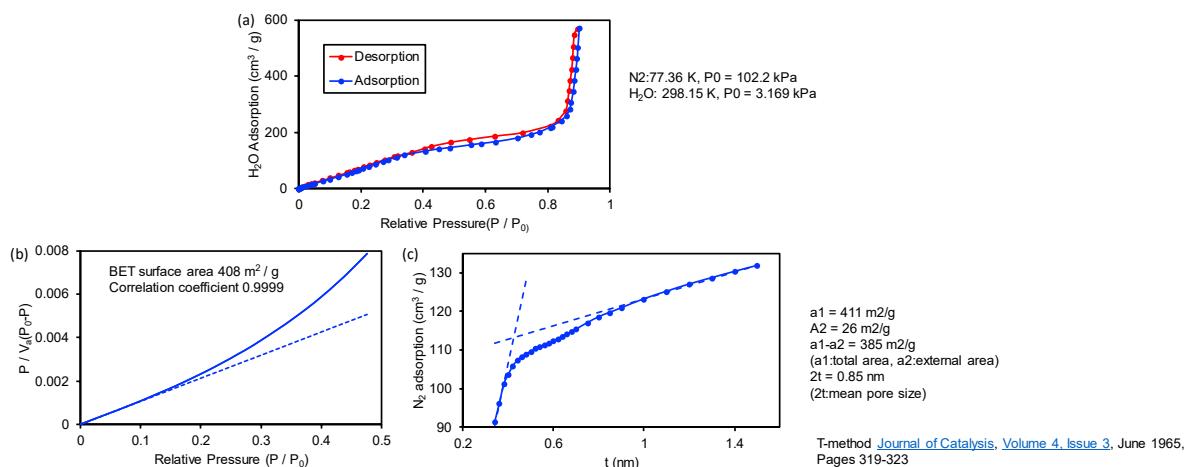


Fig. S5 (a) H_2O adsorption (blue symbols) and desorption (red symbols) isotherms at 298 K for **TP-GDY**. (b) BET analysis for the N_2 isotherm. (c) t-plot for the N_2 isotherm.

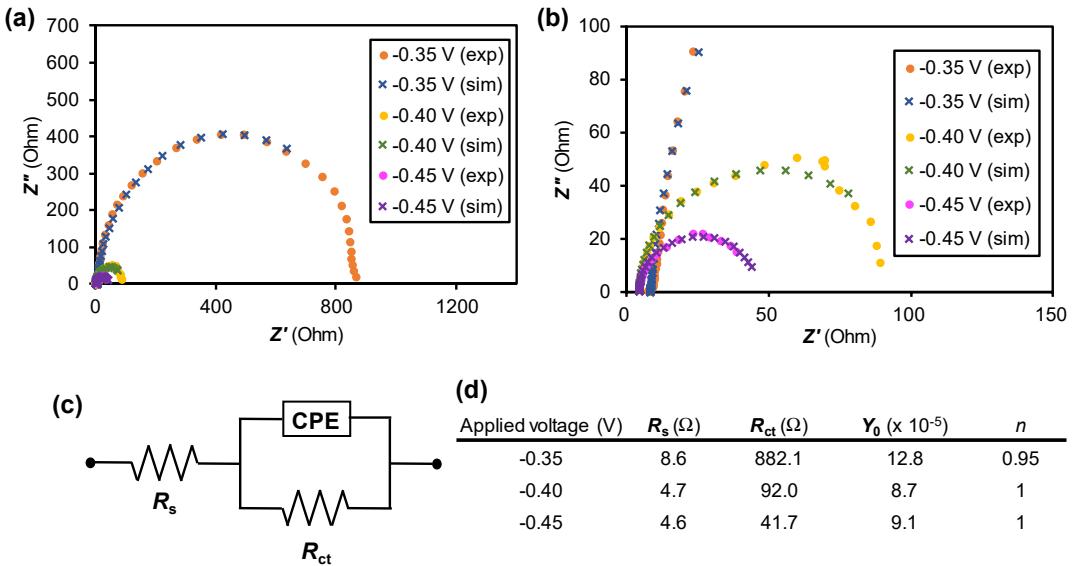


Fig. S6 (a) Nyquist plots for **TP-GDY** on GC in 0.5M H₂SO₄ at various electrode potentials. (b) Close-up of the Nyquist plots at around the origin. (c) Equivalent circuit for the simulation of the Nyquist plots. R_{ct} and R_s denote charge transfer and internal resistances, respectively, while CPE indicates a constant phase element, the impedance (Z_{CPE}) of which is represented using equation $Z_{CPE} = 1 / (j\omega)^n Y_0$, where j , ω , n , Y_0 are the imaginary unit, frequency of the alternating current, CPE exponent and constant, respectively. (d) Parameters extracted from the Nyquist plot simulation.

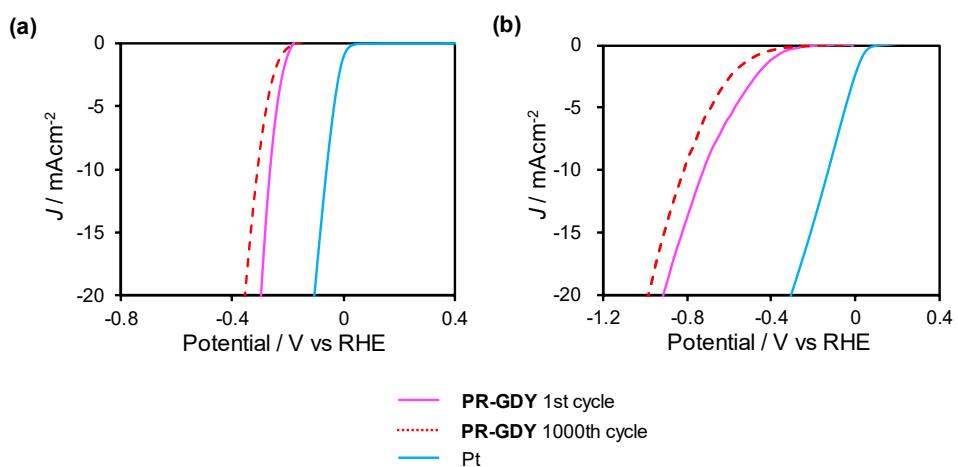


Fig. S7 (a,b) Linear sweep voltammograms (1st and 1000th sweeps) for GC modified with **PR-GDY** in (a) 0.5 M H₂SO₄ and (b) 0.1M NaHCO₃ + 0.1M Na₂CO₃.