

Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A.
This journal is © The Royal Society of Chemistry 2018

Supplementary Information

Adjusting the Yolk-shell Structure of Carbon Spheres to Boost the Capacitive K⁺ Storage Ability

Hehe Zhang,^a Hanna He,^a Jingyi Luan,^a Xiaobing Huang,^b Yougen Tang^a and Haiyan Wang^{*a}

^a*Hunan Provincial Key Laboratory of Chemical Power Sources, College of Chemistry and Chemical Engineering,
Central South University, Changsha, 410083, P.R. China*

^b*College of Chemistry and Chemical Engineering, Hunan University of Arts and Science, Changde, 415000,
P.R. China*

*Corresponding author: Haiyan Wang

Tel: +86 0731 8830886; Fax: +86 0731 8879616.

E-mail: wanghy419@csu.edu.cn (H. Wang)

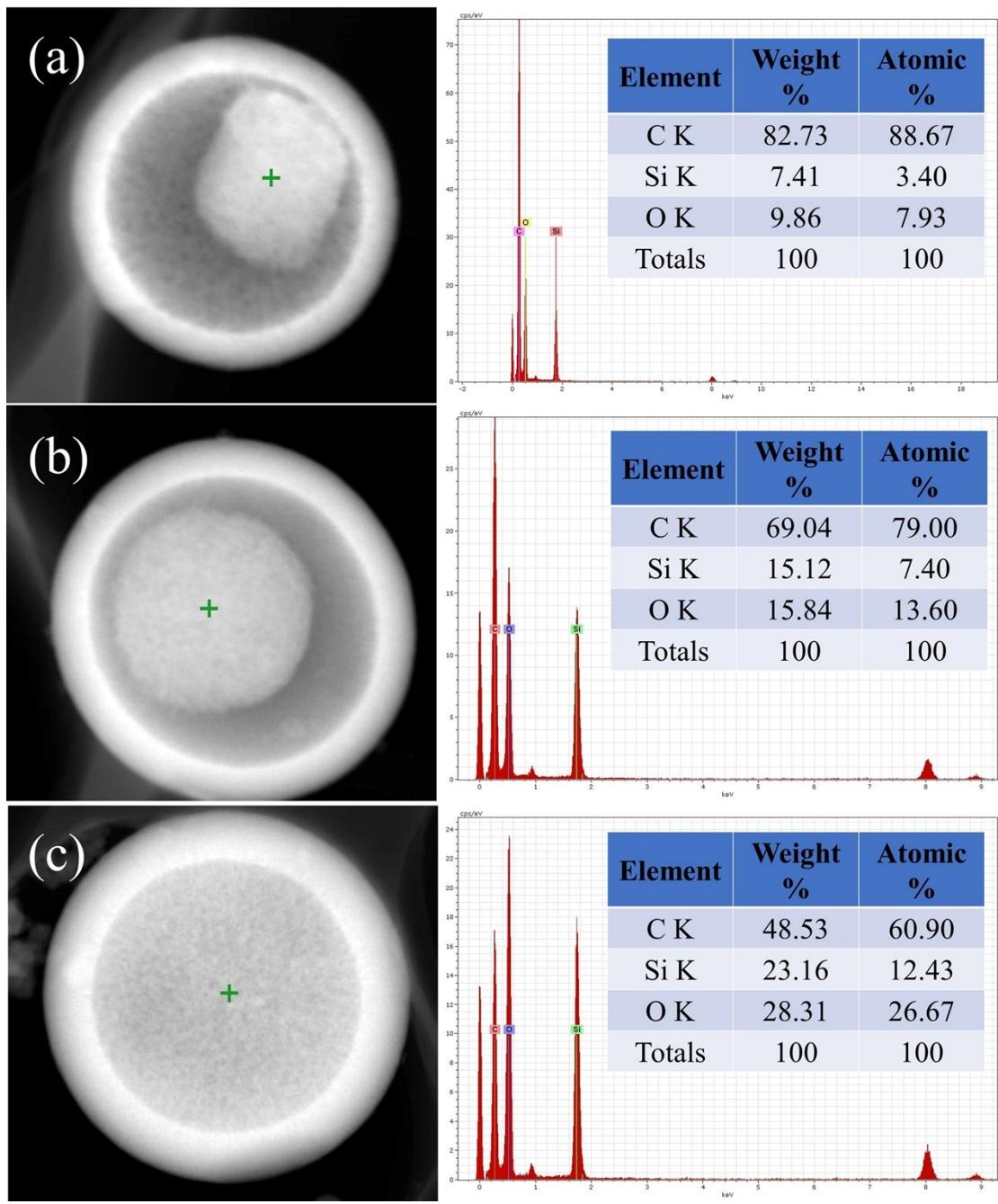


Figure S1 EDX spectra and the element composition of the yolk of SiO_2/C nanospheres: (a) $\text{SiO}_2/\text{C}-1$: 0.08 mmol L⁻¹ TEOS; (b): $\text{SiO}_2/\text{C}-2$: 0.16 mmol L⁻¹ TEOS and (c): $\text{SiO}_2/\text{C}-3$: 0.24 mmol L⁻¹ TEOS.

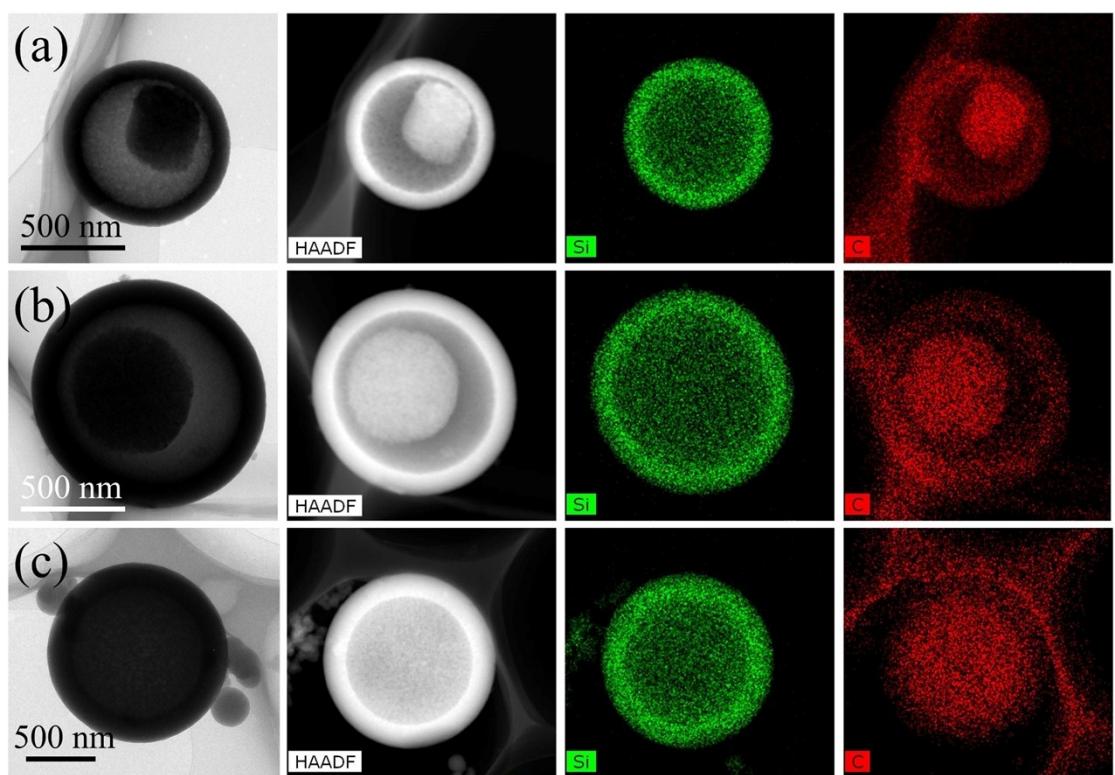


Figure S2 TEM images and elemental mapping images of (a) $\text{SiO}_2/\text{C}-1$, (b) $\text{SiO}_2/\text{C} -2$ and (c) $\text{SiO}_2/\text{C}-3$.

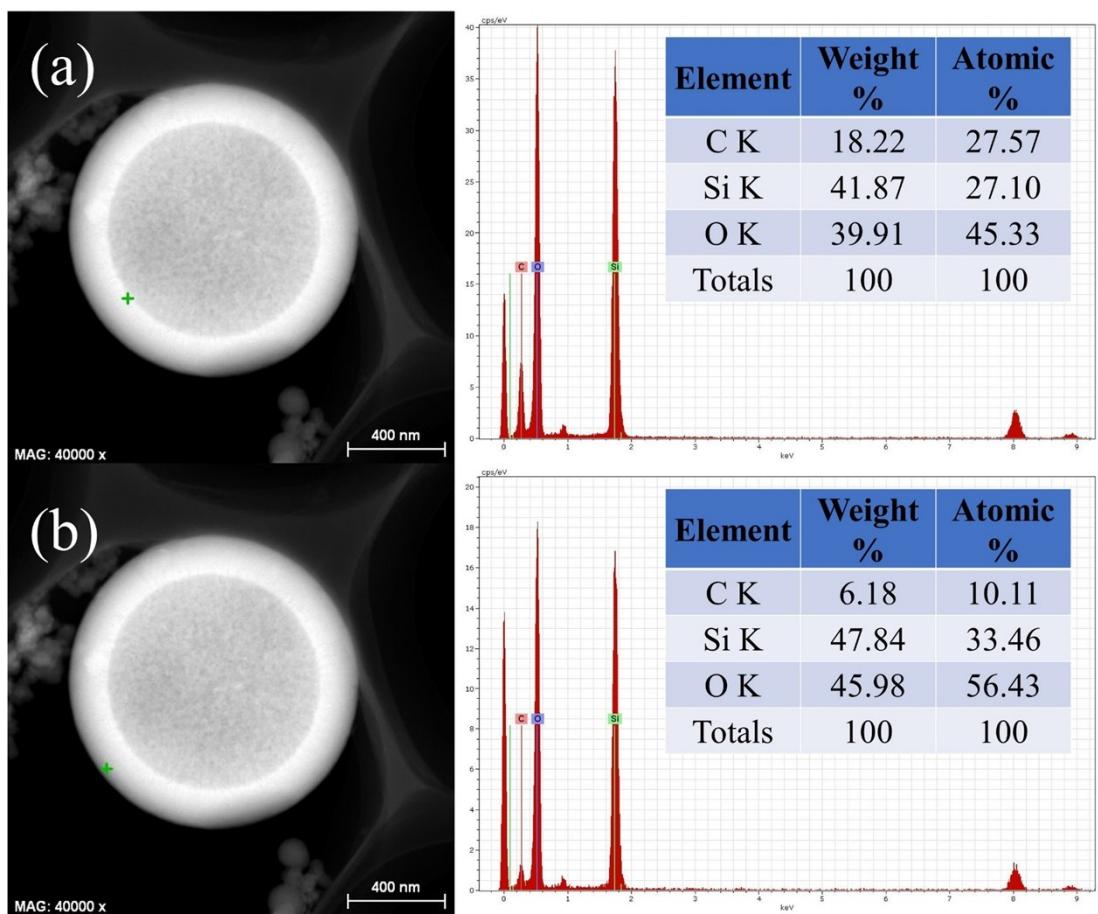


Figure S3 EDX spectra and the element composition of the inner and outer shell of SiO₂/C-3.

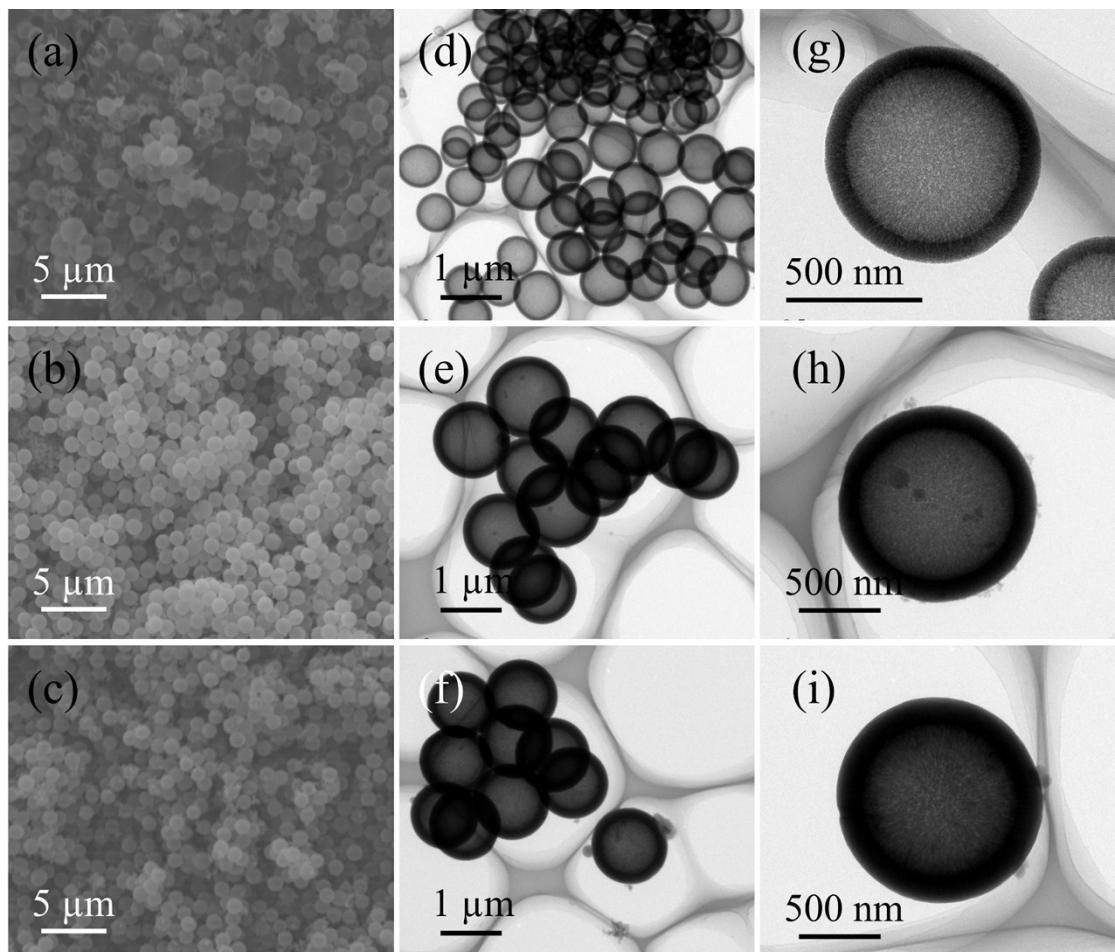


Figure S4 (a-c) SEM and (d-i) TEM images of the SiO₂ spheres: (a, d and g) SiO₂-1: 0.08 mmol L⁻¹ TEOS; (b, e and h): SiO₂-2: 0.16 mmol L⁻¹ TEOS and (c, f and i): SiO₂-3: 0.24 mmol L⁻¹ TEOS.

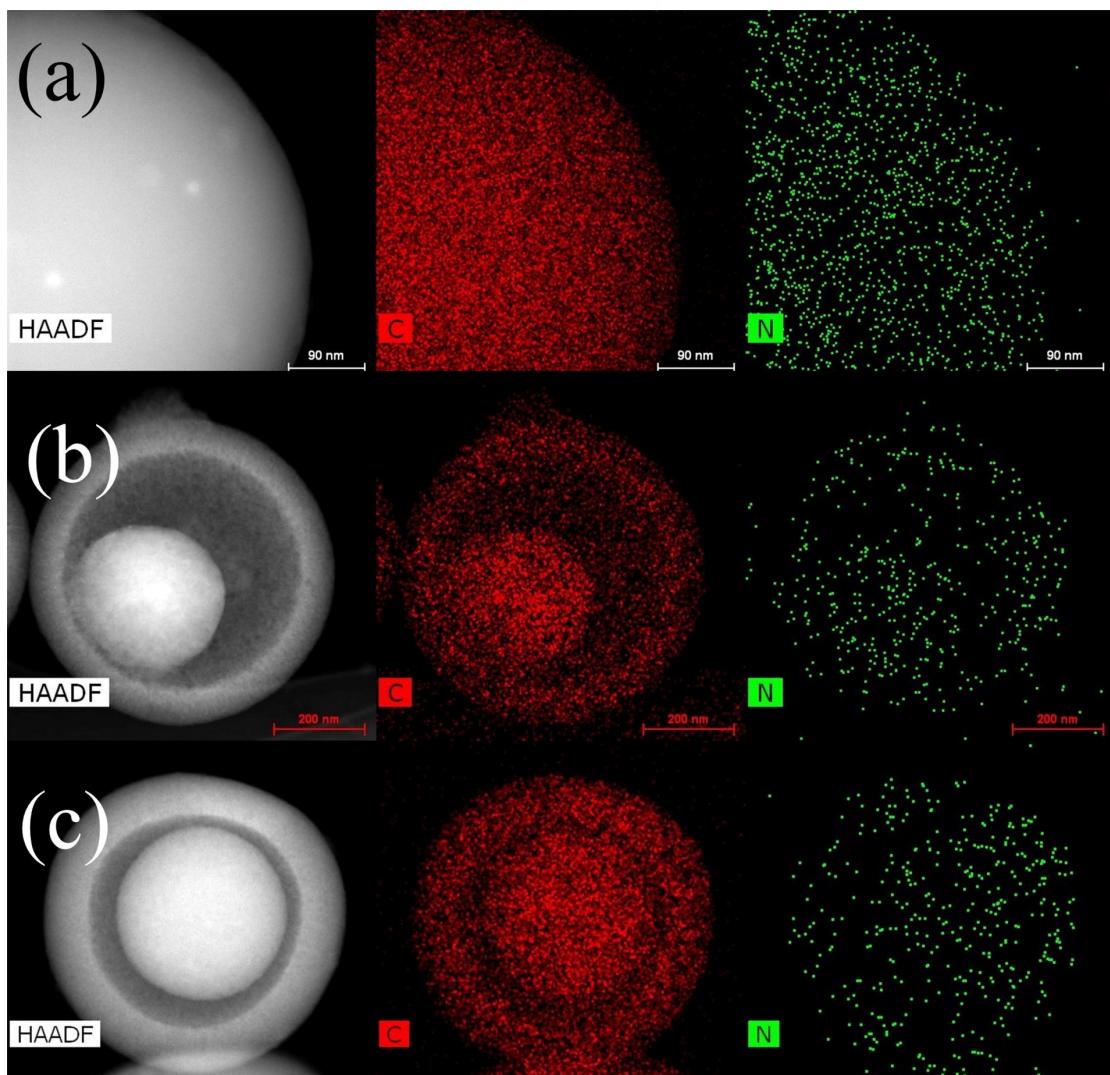


Figure S5 Element maps of (a) CS, (b) HYCS-1 and (c) HYCS-2.

Table S1. Atomic percentage of different samples calculated from the results of XPS

Materials	Atomic % (XPS)		
	C	N	O
CS	93.56	1.58	4.86
HYCS-1	94.29	1.65	4.05
HYCS-2	93.46	1.57	4.98
HYCS-3	93.73	1.64	4.63

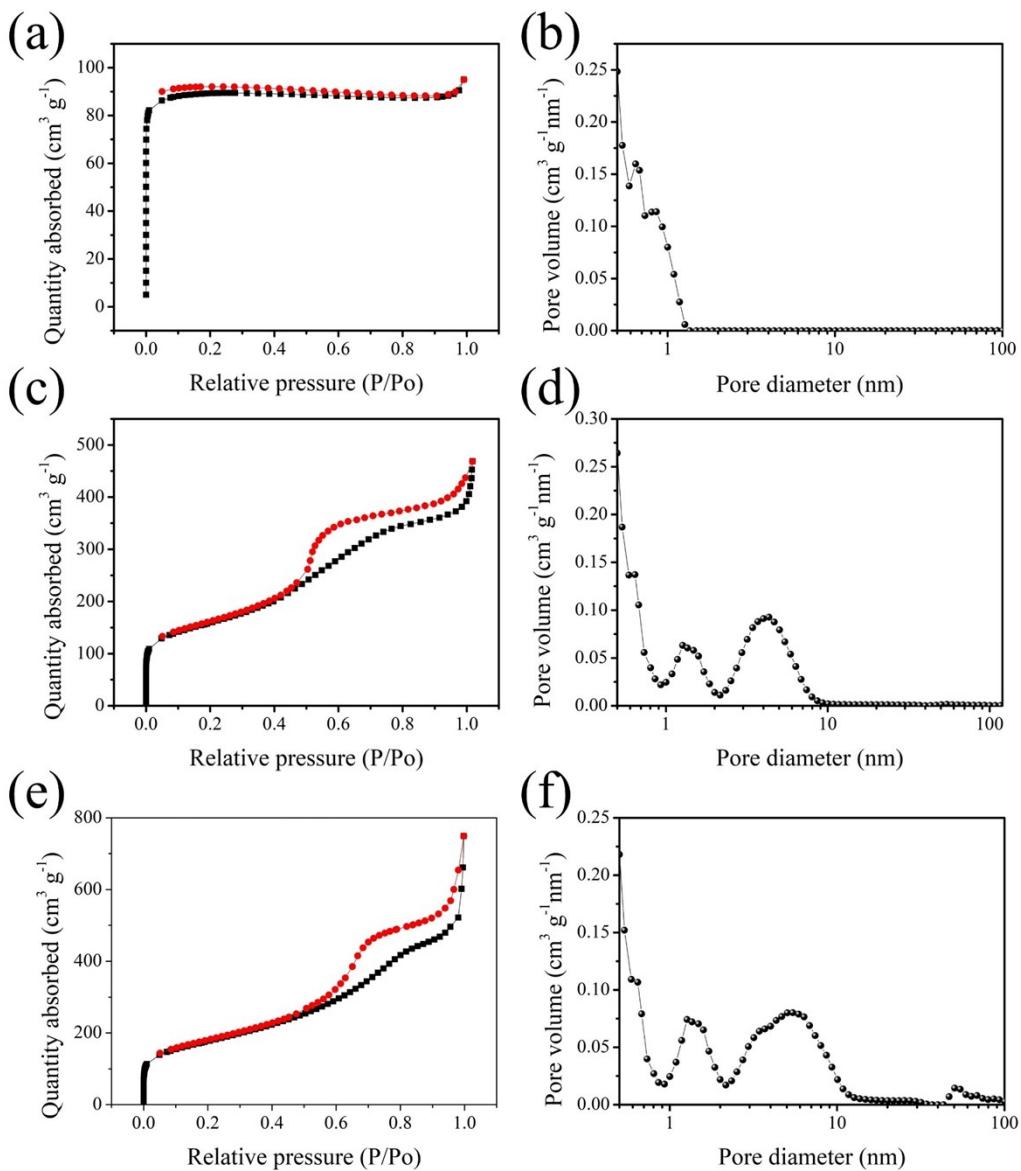


Figure S6 N_2 adsorption/desorption isotherm and pore size distribution of: (a, b) CS; (c, d) HYCS-1 and (e, f) HYCS-2.

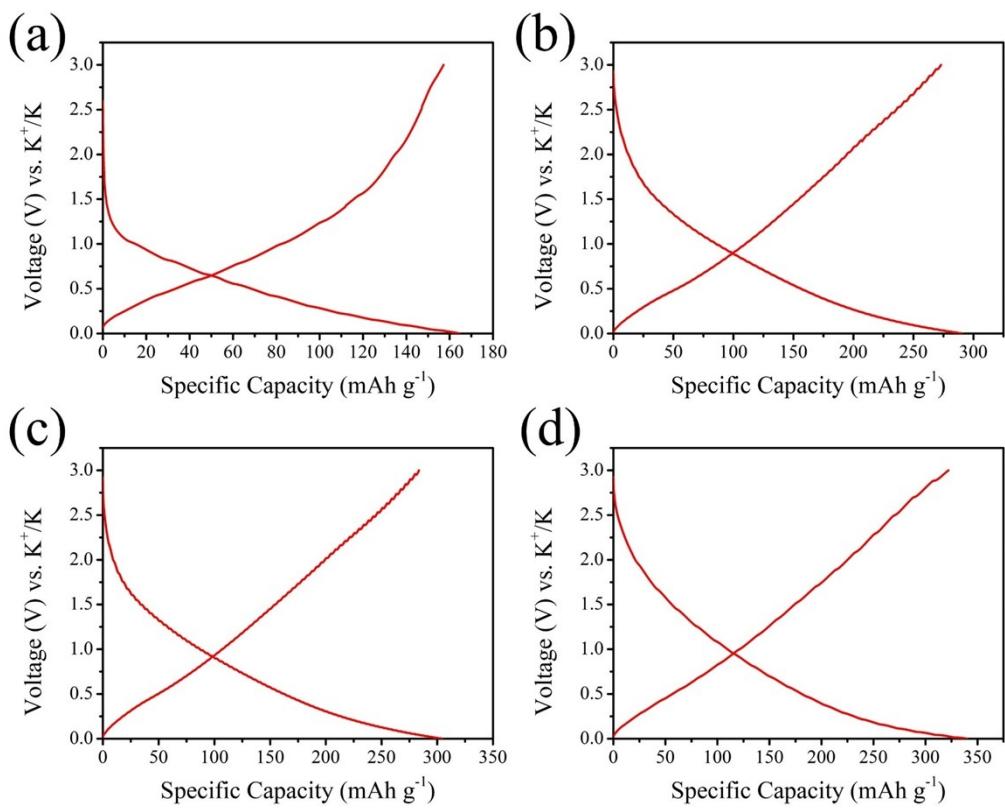


Figure S7 Charge-discharge curves at 50 mA g⁻¹ of (a) CS, (b) HYCS-1, (c) HYCS-2 and (d) HYCS-3.

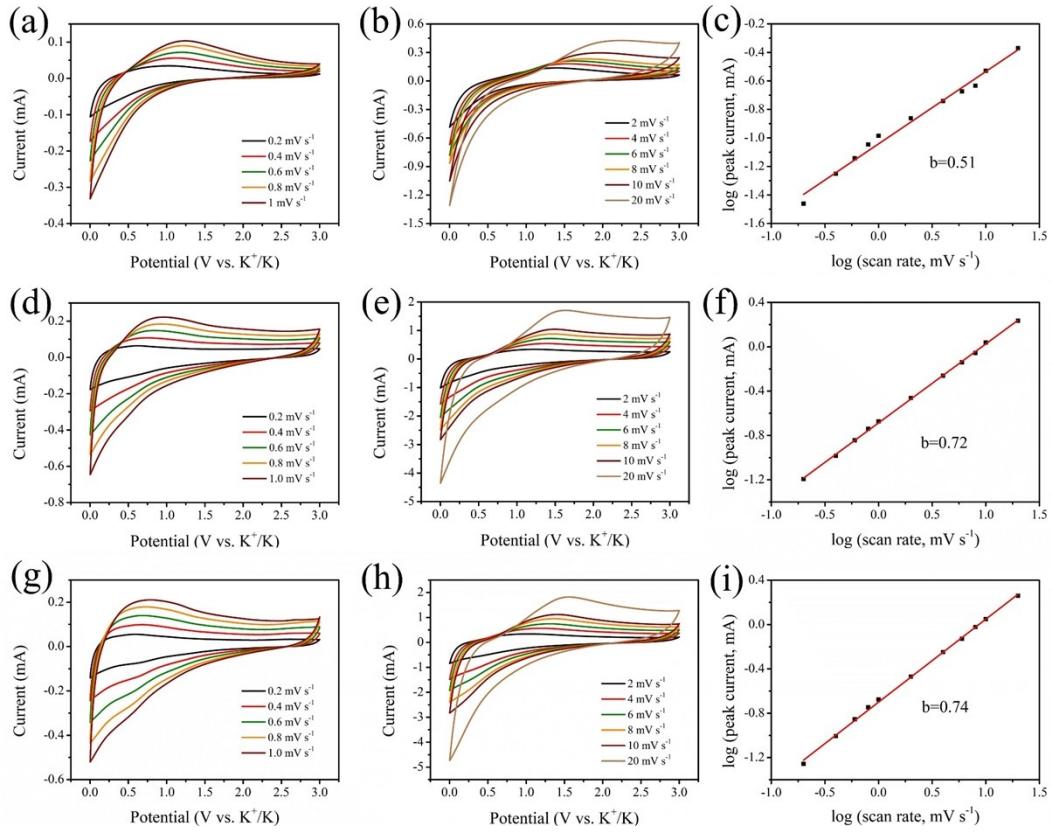


Figure S8 CV curves at different scan rates between 0.001 and 3.0 V for (a, b) CS, (d, e) HYCS-1 and (g, h) HYCS-2. The log(v)–log(i) profiles of (c) CS, (f) HYCS-1 and (i) HYCS-2.

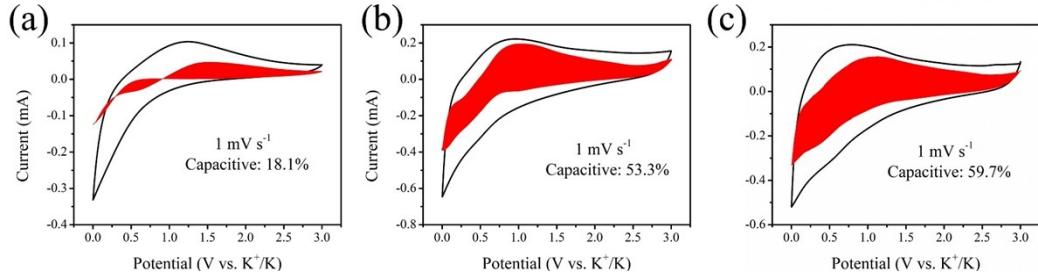


Figure S9 Separation of the capacitive and diffusion-controlled charges at 1 mV s⁻¹ of (a) CS, (b) HYCS-1 and (c) HYCS-2.

Table S2. A comparison of potassium storage performance for several reported carbon materials

Negative electrode materials	Cyclability (mA h g ⁻¹)	Rate performance (mA g ⁻¹)	Reference
Graphite	100 at 140 mA g ⁻¹ after 50 cycles	263 at 28 mA g ⁻¹ ; 172 at 140 mA g ⁻¹ ; 80 at 280 mA g ⁻¹	1
Hard carbon microspheres	216 at 28 mA g ⁻¹ after 100 cycles	262 at 28 mA g ⁻¹ ; 205 at 280 mA g ⁻¹ ; 136 at 1400 mA g ⁻¹	2
Hard-soft composite carbon	200 at 280 mA g ⁻¹ after 200 cycles	230 at 140 mA g ⁻¹ ; 190 at 190 mA g ⁻¹ ; 81 at 2800 mA g ⁻¹	3
Porous carbon nanofiber	211 at 20 mA g ⁻¹ after 1200 cycles	270 at 20 mA g ⁻¹ ; 190 at 2000 mA g ⁻¹ ; 140 at 5000 mA g ⁻¹	4
N-doped carbon microsphere	180 at 500 mA g ⁻¹ after 4000 cycles	200 at 34 mA g ⁻¹ ; 190 at 202 mA g ⁻¹ ; 156 at 5040 mA g ⁻¹	5
N/O dual-doped carbon	130 at 1050 mA g ⁻¹ after 1100 cycles	315 at 50 mA g ⁻¹ ; 230 at 200 mA g ⁻¹ ; 118 at 3000 mA g ⁻¹	6
Hollow carbon nanospheres	212 at 560 mA g ⁻¹ after 100 cycles	298 at 28 mA g ⁻¹ ; 210 at 280 mA g ⁻¹ ; 155 at 1400 mA g ⁻¹	7
Activated carbon	100 at 200 mA g ⁻¹ after 100 cycles	209 at 100 mA g ⁻¹ ; 159 at 200 mA g ⁻¹ ; 30 at 1000 mA g ⁻¹	8
N-doped carbon nanotube	236 at 20 mA g ⁻¹ after 100 cycles	338 at 10 mA g ⁻¹ ; 98 at 600 mA g ⁻¹ ; 75 at 1000 mA g ⁻¹	9
S/O codoped porous hard carbon microspheres	226.6 at 50 mA g ⁻¹ after 100 cycles	230 at 50 mA g ⁻¹ ; 213 at 200 mA g ⁻¹ ; 158 at 1000 mA g ⁻¹	10
highly N-doped carbon nanofibers	248 at 25 mA g ⁻¹ after 100 cycles	238 at 100 mA g ⁻¹ ; 217 at 200 mA g ⁻¹ ; 101 at 20 A g ⁻¹	11
N-doped hierarchical porous yolk-shell spheres	218 at 200 mA g ⁻¹ after 500 cycles	314 at 50 mA g ⁻¹ ; 227 at 200 mA g ⁻¹ ; 121 at 5000 mA g ⁻¹	this work

Supplementary References

1. Z. Jian, W. Luo and X. Ji, *J. Am. Chem. Soc.*, 2017, **137**, 11566.
2. Z. Jian, Z. Xing, C. Bommier, Z. Li and X. Ji, *Adv. Energy Mater.*, 2016, **6**, 1501874.
3. Z. Jian, S. Hwang, Z. Li, A. S. Hernandez, X. Wang, Z. Xing, D. Su and X. Ji, *Adv. Funct. Mater.*, 2017, **27**, 1700324.
4. X. Zhao, P. Xiong, J. Meng, Y. Liang, J. Wang and Y. Xu, *J. Mater. Chem. A*, 2017, **5**.
5. C. Chen, Z. Wang, B. Zhang, L. Miao, J. Cai, L. Peng, Y. Huang, J. Jiang, Y. Huang and L. Zhang, *Energy Storage Mater.*, 2017, **8**, 161-168.
6. J. Yang, Z. Ju, Y. Jiang, Z. Xing, B. Xi, J. Feng and S. Xiong, *Adv. Mater.*, 2018, **30**, 1700104.
7. D. S. Bin, Z. X. Chi, Y. Li, K. Zhang, X. Yang, Y. G. Sun, J. Y. Piao, A. M. Cao and L. J. Wan, *J. Am. Chem. Soc.*, 2017, **139**, 13492.
8. Z. Tai, Q. Zhang, Y. Liu, H. Liu and S. Dou, *Carbon*, 2017.
9. X. Zhao, Y. Tang, C. Ni, J. Wang, A. Star and Y. Xu, *ACS Appl. Energy Mater.*, 2018, **1**, 1703-1707.
10. M. Chen, W. Wang, X. Liang, S. Gong, J. Liu, Q. Wang, S. Guo and H. Yang, *Adv. Energy Mater.*, 2018, 1800171.
11. Y. Xu, C. Zhang, M. Zhou, Q. Fu, C. Zhao, M. Wu and Y. Lei, *Nat. Commun.*, 2018, **9**.