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# Supplementary material

## Synthesis of N-doping carbon nanosheets with controllable porosity

### derived from bio-oil for high-performance supercapacitors

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# **Content:**

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Fig. S1 Cyclic voltammetry curves for ABF-9%M with the scan rate of 50 mV s<sup>-1</sup>

(a) and 10 mV s<sup>-1</sup> using 2M KCl and 6M KOH as the electrolyte.

Fig. S2 Plot of Bode phase angle with frequency.

 Table S1 Comparison of the specific capacitance of ABF-9%M with reported

 biomass-derived carbon materials previously.

 Table S2. Equivalent circuit parameters of the ABFs.

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**Fig. S1** Cyclic voltammetry curves for ABF-9%M with the scan rate of 50 mV s<sup>-1</sup> (a) and 10 mV s<sup>-1</sup> using 2M KCl and 6M KOH as the electrolyte.



Fig. S2 Plot of Bode phase angle with frequency.

| Precursors | $S_{ m BET}$   | C/S                     | С                    | Electrolyte                               | Ref. |
|------------|----------------|-------------------------|----------------------|---|------|
|            | $(m^2 g^{-1})$ |                         | (F g <sup>-1</sup> ) |   |      |
| D 1        | 1500           | 0 <b>7</b> 4 1h         |                      |   |      |
| Bamboo     | 1732           | $0.5 \text{ A g}^{-10}$ | 222                  | 6 M KOH                                   | 1    |
| Bark       | 1721           | 0.5 A g <sup>-1b</sup>  | 206                  | $1 \text{ M H}_2\text{SO}_4$              | 2    |
| Rapeseed   | 1417           | 5 mV s <sup>-1b</sup>   | 171                  | $1 \mathrm{M} \mathrm{H}_2 \mathrm{SO}_4$ | 3    |
| Carrot     | 1899           | 0.2 A g <sup>-1b</sup>  | 276                  | 6 M KOH                                   | 4    |
| Chitosan   | 1582           | 0.5 A g <sup>-1b</sup>  | 252                  | 6 M KOH                                   | 5    |
| tofu       | 1208           | 0.5 A g <sup>-1b</sup>  | 262                  |   |      |
|            |                | 50 A g <sup>-1b</sup>   | 145                  | 6 M KOH                                   | 6    |
| Raspberry  | 1234           | 0.1 A g <sup>-1b</sup>  | 213                  | 6 M KOH                                   | 7    |
| Bagasse    | 1360           | 5 mV s <sup>-1a</sup>   | 173                  | 6 M KOH                                   | 8    |
| Agar       | 1672           | 1 A g <sup>-1b</sup>    | 226                  | 6 M KOH                                   | 9    |
| Soybean    | 1749           | 0.5 A g <sup>-1b</sup>  | 243                  | 6 M KOH                                   | 10   |
| catkin     | 1462           | 0.5 A g <sup>-1b</sup>  | 251                  | 6 M KOH                                   | 11   |
| Bio-oil    | 2566           | 0.5 A g <sup>-1b</sup>  | 289                  | 6 M KOH                                   | This |
|            |                | 20 A g <sup>-1b</sup>   | 227                  |   | work |
|            |                | 0.5 A g <sup>-1a</sup>  | 256                  |   |      |

**Table S1** Comparison of the specific capacitance of ABF-9%M with reported

 biomass-derived carbon materials previously.

C/S represents current density or sweep rate; <sup>a</sup> two electrode; <sup>b</sup> three electrode

| Sample  | $R_{\rm s}(\Omega)$ | $R_{\rm ct}(\Omega)$ | $f_{\rm o}({\rm Hz})$ | $\tau_0(s)$ |
|---------|---------------------|----------------------|-----------------------|-------------|
| ABF     | 0.94                | 0.71                 | 0.16                  | 6.10        |
| ABF-9%M | 0.64                | 0.87                 | 0.20                  | 5.08        |

#### Reference

- 1 Y. Gong, D. Li, C. Luo, Q. Fu and C. Pan, Green Chem., 2017, 19, 4132-4140.
- 2 Z. Sun, J. Liao, B. Sun, M. He, X. Pan, J. Zhu, C. Shi and Y. Jiang, *Int. J. Electrochem. Sci.*, 2017, **12**, 12084-12097.
- 3 X. Kang, H. Zhu, C. Wang, K. Sun and J. Yin, *J. Colloid Interface Sci.*, 2018, **509**, 369-383.
- 4 Y. Cheng, B. Li, Y. Huang, Y. Wang, J. Chen, D. Wei, Y. Feng, D. Jia and Yu. Zhou, *Appl. Surf. Sci.*, 2018, **439**, 712-723.
- 5 X. Deng, B. Zhao, L. Zhu and Z. Shao, Carbon, 2015, 93, 48-58.
- 6 T. Ouyang, K. Cheng, F. Yang, L. Zhou, K. Zhu, K. Ye, G. Wang and D. Cao, J. Mater. Chem. A, 2017, 5, 14551–14561.
- 7 Y. Yao, Q. Zhang, P. Liu, L. Yu, L. Huang, S. Z. Zeng, L. Liu, X. Zeng and J. Zou, *RSC Advances*, 2018, 8, 1857-1865.
- 8 S. S. Gunasekaran, S. K. Elumalali, T. K. Kumaresan, R. Meganathan, A. Ashok, V. Pawar, K. Vediappan, G. Ramasamy, S. Z. Karazhanov, K. Raman and R. S. Bose, *Materials Letters*, 2018, 218, 165-168.
- 9 L. Zhang, H. Gu, H. Sun, F. Cao, Y. Chen and G. Z. Chen, *Carbon*, 2018, 132, 573-579.
- 10 G. Lin, R. Ma, Y. Zhou, Q. Liu, X. Dong and J. Wang, *Electrochim. Acta*, 2018, 261, 49-57.
- 11 S. Gao, X. Li, L. Li and X. Wei, Nano Energy, 2017, 33, 334-342.