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Supporting Information

Simultaneously achieving superior foldability, mechanical strength and toughness for transparent healable polysiloxane films through building hierarchical crosslinked networks and dual dynamic bonds

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Scheme S1 Synthetic routes of SS-NCO, LPx, HPSi and HE-NCO.



Fig. S1 FTIR spectrum of SS-NCO.



Fig. S2 TGA curves of LPx-SS-HP and LPx-SS under N₂ atmosphere.



Fig. S3 Digital images during the scratch healing process on the same location of LP2-SS-HP film left in air for 24 h before starting self-healing (a: the 1st healing process; b: the 2nd healing; c: the 3rd healing process).



Fig. S4 Tensile stress-strain curves of LP2-IP-HP, LP2-HE-HP and LP2-SS-HP films.



Fig. S5 Digital images during the healing process of LP2-IP-HP film.



Fig. S6 Digital images during the healing process of LP2-HE-HP film.

| Sample name | Self-healing component | <i>T_g</i> (°C) | <i>Т_{di}</i> (°С) | Self-healing condition | Self-healing Efficiency (%) | Photo ^b | σ _b ^c (MPa) | ε _b (%) | Ref |
|----------------------------|--|---------------------------|-------------------------------|---------------------------------|--------------------------------|--------------------|--------------------------------------|-----------------------|------------|
| P1 | Hydrogen bonds | a | | 1 h | | Yes | | | C1 |
| P2 | | | | 2 h | | Yes | | | 51 |
| SESi-3 | Hydrogen bonds | -112.7 | | 25 °C/24 h | 90 | Yes | 0.45 | 400 | S2 |
| SESi ₁ | Hydrogen bonds | -118.3 | | 100 °C/16 h | 100 | No | 2.75 | 225 | S3 |
| Co-TIA-PDMS | Coordination bonds | <-100 | <350 | 140 °C/24 h | 52.2 | Yes | 1.12 | 560 | S4 |
| Fe-TIA-PDMS | Coordination bonds | <-90 | | 60 °C/20 h | 94.3 | Yes | 0.35 | 2500 | S5 |
| PDMS-Boroxine | Boroxine | 65 | | 70 °C/12 h | 100 | Yes | 9.46 | 10 | S6 |
| Fe-Hpdca-PDMS | Coordination bonds | <-90 | | r.t./48 h | 90 | Yes | 0.225 | 1880 | S7 |
| PDMS-PtL | Pt \cdots Pt and π - π interaction | <-50 | | r.t./12 h | 100 | Yes | 0.3 | 1390 | S 8 |
| Zn(OTf) ₂ -PDMS | Coordination bonds | <-50 | | r.t./48 h | 76 | Yes | 0.6 | 310 | S9 |
| HSE-0.65 | Hydrogen bonds | -120 | | 80 °C/24 h | >90 | Yes | 0.15 | 550 | S10 |
| PDMS-DA-PU | - Diels-Alder reaction | | | 140 °C/0.5 h | 99 | Yes | 1.04 | 108 | S11 |
| PDMS/PCL-DA-PU-10% | | | | 80 °C/24 h | 90 | Yes | 3.25 | 244 | |
| PM2FS | - Diels–Alder reaction | | | 140 °C/0.5 h | 85 | Yes | 0.13 | 35.5 | S12 |
| PM3FS | | | | 80 °C/24 h | 95 | Yes | 0.61 | 50.9 | |
| Py-PDMS-Co-0.5 | Coordination bonds | -114 | 233 | r.t./24 h | 91 | Yes | 0.13 | 560 | S13 |
| Eu(OTf) ₃ -PDMS | - Coordination bonds | -19.1 | | r.t./24 h | ~100 | No | 0.2 | 300 | S14 |
| Tb(OTf) ₃ -PDMS | | -18.1 | | | ~100 | No | 0.25 | 290 | |
| PDMS-network | Vinylogous urethane | -118 | | 75 °C/3 h | | Yes | 0.08 | 67 | S15 |
| SR-SH | Disulfide | | | light radiation/48 h | 84 | Yes | 1.03 | 178 | S16 |
| PDMS-TFB | Imine | -117 | | 25 °C/1 h | 98.3 | Yes | 0.035 | 135 | S17 |
| C1 | Boronic ester bonds | | | r.t./24 h | 85 | Yes | 1.28 | 1000 | S18 |
| DAE@AS | Imine | | | UV-light/2 h | 91 | Yes | 0.35 | 55 | S19 |
| PAPMS-25-OA-2.5 | Ionic bonds | | 200 | methanol and chloroform mixture | 81.5 | Yes | 4.43 | 1150 | S20 |
| A_4B_2 | Acylhydrazone; Hydrogen bonds | -120 | | 120 °C/1 h | 99 | Yes | 1.75 | 150 | S21 |
| PY-PSBTh | Lewis acid- base adduct | | | 70 °C/1 h | | Yes | | | S22 |
| PDMS-MeNNN-Zn | Coordination bonds | -69.6 | | r.t./1 h | 99.3 | Yes | 0.066 | 456 | - S23 |
| PDMS-NNN-Zn | | -68.7 | | r.t./24 h | 100.3 | Yes | 0.091 | 230 | |
| PDMS-PUa | Hydrogen bonds | | | r.t./48 h | 90 | Yes | 0.81 | 551 | S24 |
| Si-A_IN30 | Ionic bonds | | | 120 °C/12 h | 77 | Yes | 3.08 | 387 | S25 |
| PDMS-1 | Disulfide | | | r.t./4 h | 95 | Yes | 0.15 | 700 | S26 |
| LP2-SS-HP | Disulfide | 96 | 282 | 100 °C/0.5 h | 96.5 | Yes | 8.6 | 224.2 | This work |

 Table S1 Typical self-healing properties of polydimethylsiloxane polymers

a: data not given in the reference. b: "Yes" means that optical microscopic photos representing the self-healing of the sample were provided in the reference. "No" means that optical microscopic photos representing the self-healing of the sample were not provided in the reference. c: the value of the original sample without scratch.

Film σ_b (MPa) ε_b (%)E (MPa)Toughness (MJ m⁻³)LP2-IP-HP 8.8 ± 0.5 191.4±4.1196.6±12.714.4±0.2LP2-HE-HP 8.5 ± 0.4 213.8±3.1181.2±13.515.4±0.3

Table S2 Tensile properties for LP2-IP-HP and LP2-HE-HP films

References

- [S1] N. Roy, E. Buhler, J.-M. Lehn, Chem. Eur. J. 2013, 19, 8814.
- [S2] A. Zhang, L. Yang, Y. Lin, L. Yan, H. Lu, L. Wang, J. Appl. Polym. Sci. 2013, 129, 2435.
- [S3] L. Yang, Y. Lin, L. Wang, A. Zhang, Polym. Chem. 2014, 5, 153.
- [S4] X.-Y. Jia, J.-F. Mei, J.-C. Lai, C.-H. Li, X.-Z. You, Chem. Commun. 2015, 51, 8928.
- [S5] X.-Y. Jia, J.-F. Mei, J.-C. Lai, C.-H. Li, X.-Z. You, *Macromol. Rapid Commun.* 2016, 37, 952.
- [S6] J. C. Lai, J. F. Mei, X. Y. Jia, C. H. Li, X. Z. You, Z. Bao, Adv. Mater. 2016, 28, 8277.
- [S7] C.-H. Li, C. Wang, C. Keplinger, J.-L. Zuo, L. Jin, Y. Sun, P. Zheng, Y. Cao, F. Lissel, C. Linder, X.-Z. You, Z. Bao, *Nat Chem* 2016, 8, 618.
- [S8] J. F. Mei, X. Y. Jia, J. C. Lai, Y. Sun, C. H. Li, J. H. Wu, Y. Cao, X. Z. You, Z. Bao, *Macromol. Rapid Commun.* 2016, 37, 1667.
- [S9] Y.-L. Rao, A. Chortos, R. Pfattner, F. Lissel, Y.-C. Chiu, V. Feig, J. Xu, T. Kurosawa, X. Gu, C. Wang, M. He, J. W. Chung, Z. Bao, *J. Am. Chem. Soc.* 2016, 138, 6020.
- [S10] Y. You, W. Huang, A. Zhang, Y. Lin, J. Polym. Sci., Part A: Polym. Chem. 2016, 54, 3760.
- [S11] J. Zhao, R. Xu, G. X. Luo, J. Wu, H. S. Xia, Polym. Chem. 2016, 7, 7278.
- [S12] J. Zhao, R. Xu, G. Luo, J. Wu, H. Xia, J. Mater. Chem. B 2016, 4, 982.
- [S13] L. Liu, S. Liang, Y. Huang, C. Hu, J. Yang, Chem. Commun. 2017, 53, 12088.
- [S14] Y.-L. Rao, V. Feig, X. Gu, G.-J. Nathan Wang, Z. Bao, J. Polym. Sci., Part A: Polym. Chem. 2017, 55, 3110.
- [S15] T. Stukenbroeker, W. D. Wang, J. M. Winne, F. E. Du Prez, R. Nicolay, L. Leibler, *Polym. Chem.* 2017, 8, 6590.
- [S16] H. P. Xiang, M. Z. Rong, M. Q. Zhang, Polymer 2017, 108, 339.
- [S17] B. Zhang, P. Zhang, H. Zhang, C. Yan, Z. Zheng, B. Wu, Y. Yu, *Macromol. Rapid Commun.* 2017, 38, 1700110.
- [S18] Y. Zuo, Z. Gou, C. Zhang, S. Feng, Macromol. Rapid Commun. 2016, 37, 1052.
- [S19] M. Kathan, P. Kovaricek, C. Jurissek, A. Senf, A. Dallmann, A. F. Thunemann, S. Hecht, Angew. Chem. Int. Ed. Engl. 2016, 55, 13882.
- [S20] H. Lu, S. Feng, J. Polym. Sci., Part A: Polym. Chem. 2017, 55, 903.

- [S21] D.-D. Zhang, Y.-B. Ruan, B.-Q. Zhang, X. Qiao, G. Deng, Y. Chen, C.-Y. Liu, *Polymer* 2017, 120, 189.
- [S22] F. Vidal, H. Lin, C. Morales, F. Jakle, *Molecules* 2018, 23, 405.
- [S23] D. P. Wang, J. C. Lai, H. Y. Lai, S. R. Mo, K. Y. Zeng, C. H. Li, J. L. Zuo, *Inorg. Chem.* 2018, 57, 3232.
- [S24] C. Liu, C. Ma, Q. Xie, G. Zhang, J. Mater. Chem. A 2017, 5, 15855.
- [S25] F. B. Madsen, L. Yu, A. L. Skov, ACS Macro Lett. 2016, 5, 1196.
- [S26] C. Lv, K. Zhao, J. Zheng, Macromol. Rapid Commun. 2018, 39, 1700686.