

Electronic Supplementary Information (ESI)

General and Facile Synthesis of Hollow Metal Oxide Nanoparticles Coupling with Graphene Nanomesh Architectures for Highly Efficient Lithium Storage

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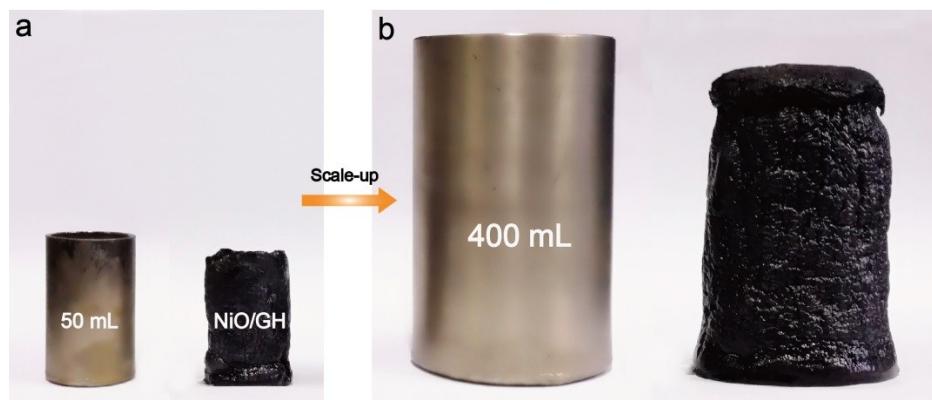


Figure S1. The photograph of a monolithic NiO/GH with (a) 50 mL Ni vessel and (b) 400 mL Ni vessel.

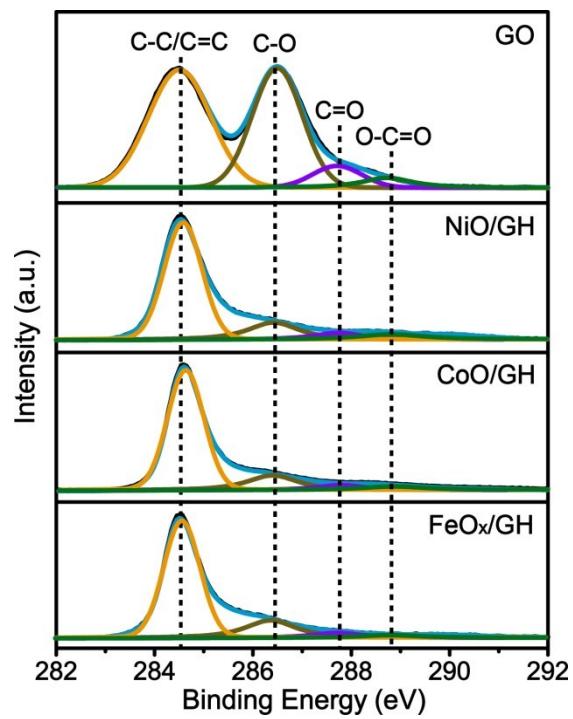


Figure S2. XPS C 1s spectra of GO, NiO/GH, CoO/GH and FeO_x/GH.

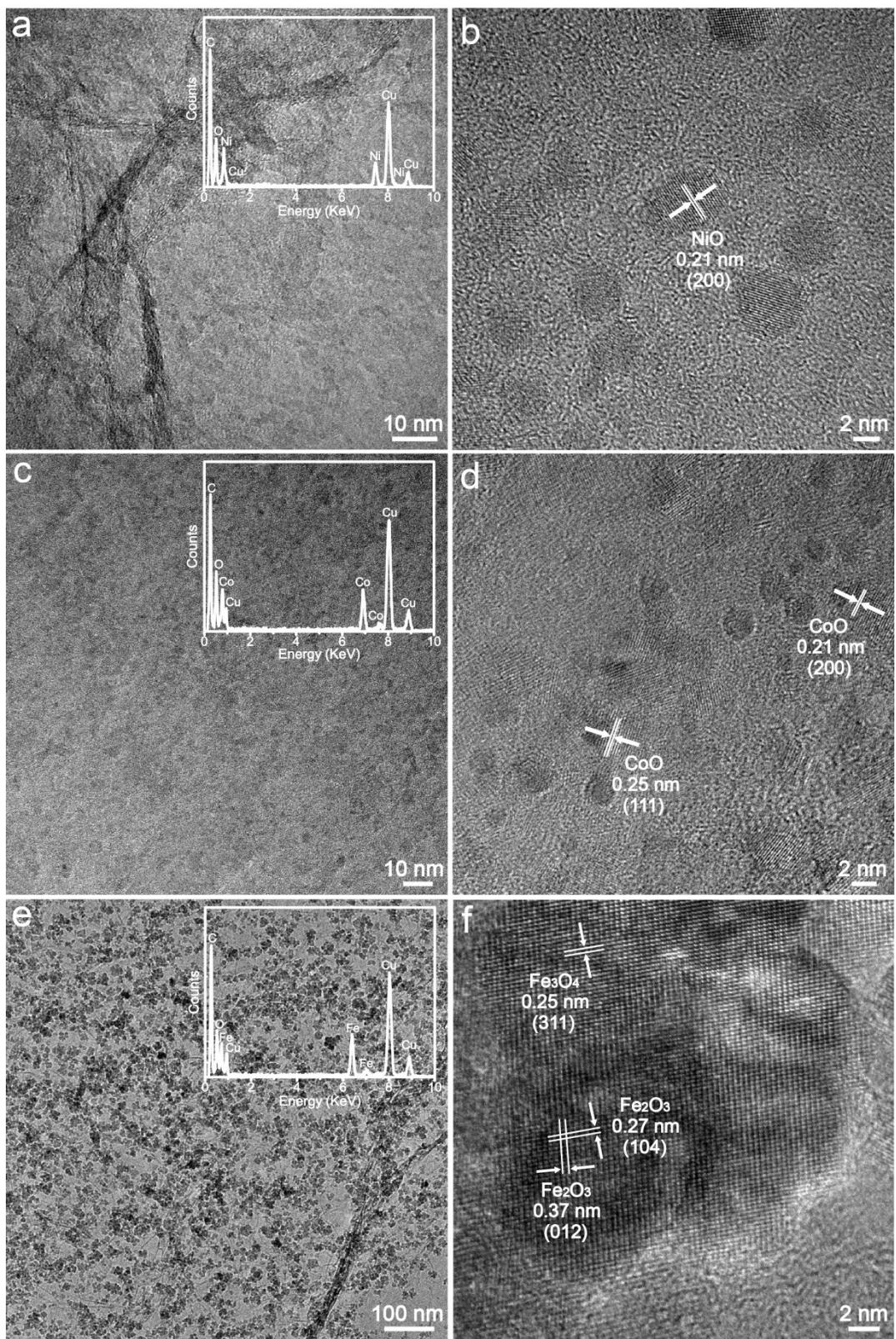


Figure S3. TEM images of (a,b) NiO/GH, (c,d) CoO/GH and (e,f) FeO_x/GH. Inset in (a), (c) and (e) are the corresponding EDX spectra.

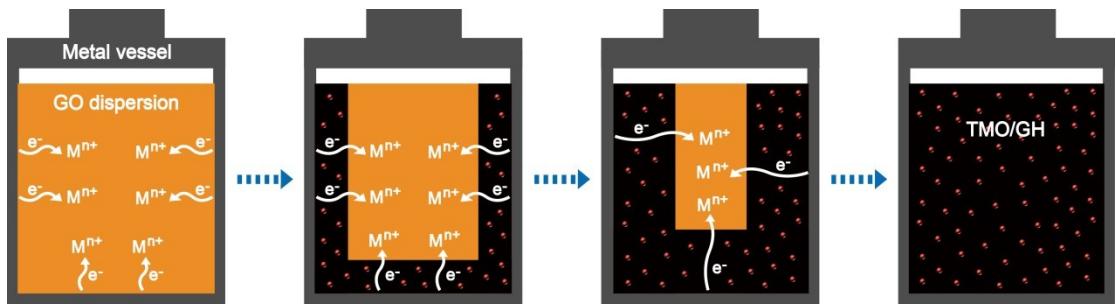


Figure S4. Schematic illustration of the formation process of TMO/GH.

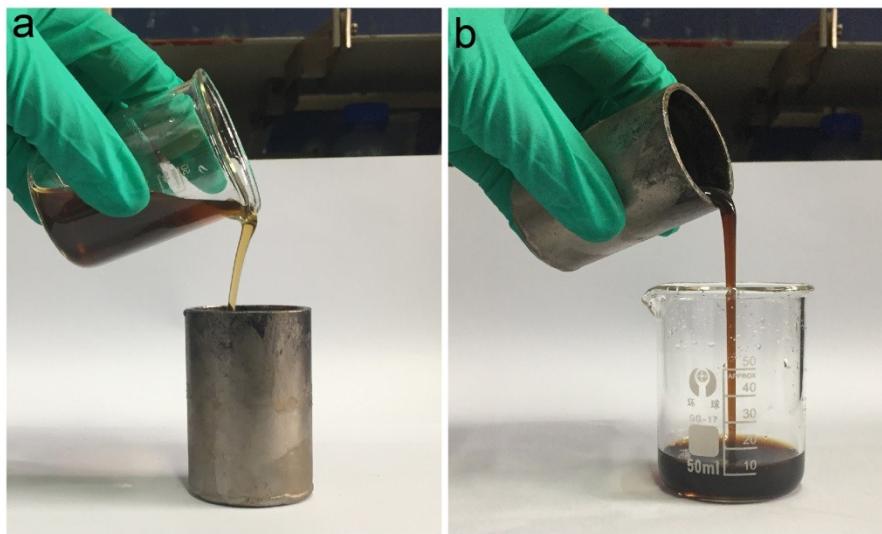


Figure S5. Room temperature reaction. (a) before and (b) after 40 h reaction. Only a dark-brown despersioin can be observed after reaction.

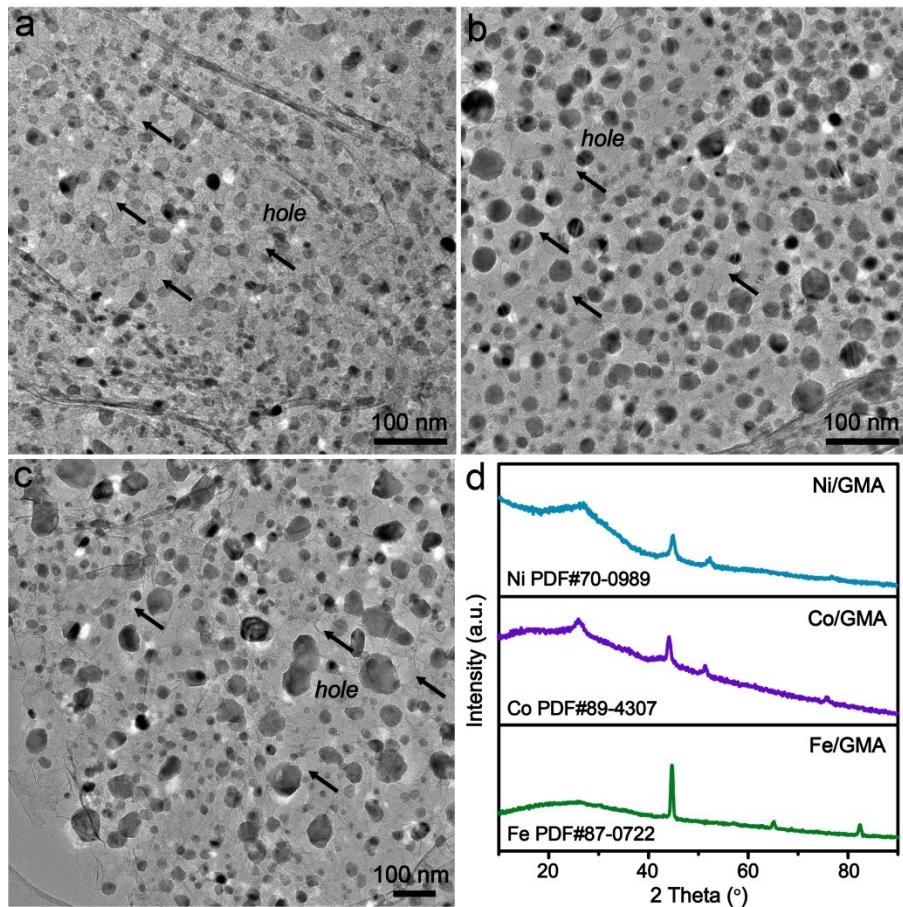


Figure S6. TEM images of (a) Ni/GMA, (b) Co/GMA and (c) Fe/GMA. (d) XRD patterns of various TM/GMAs.

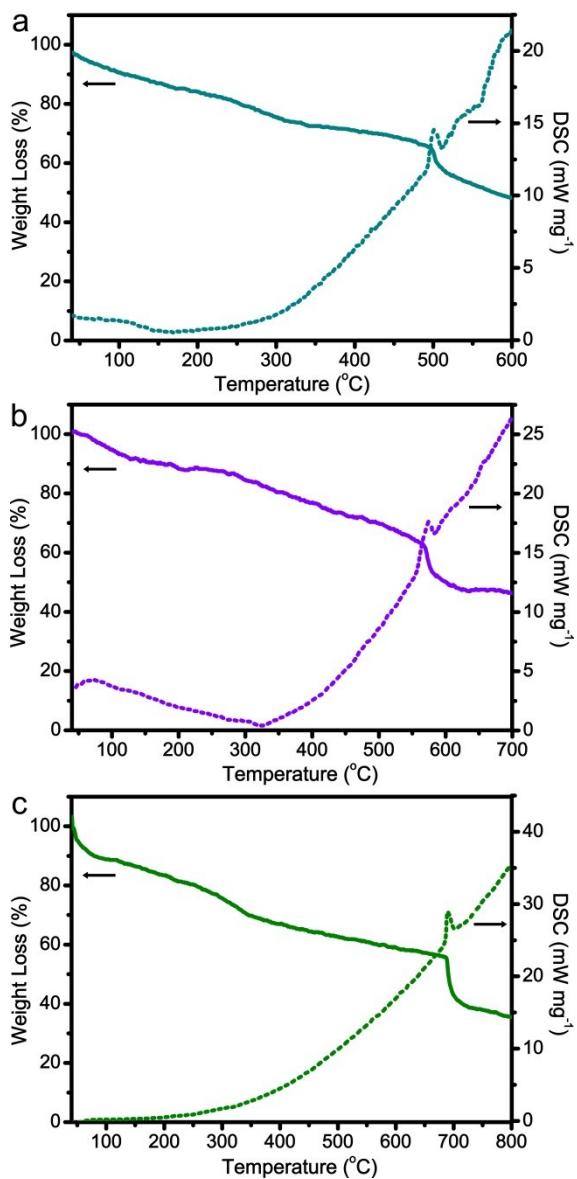


Figure S7. TGA/DSC curves of (a) NiO/GH, (b) CoO/GH and (c) FeO_x/GH.

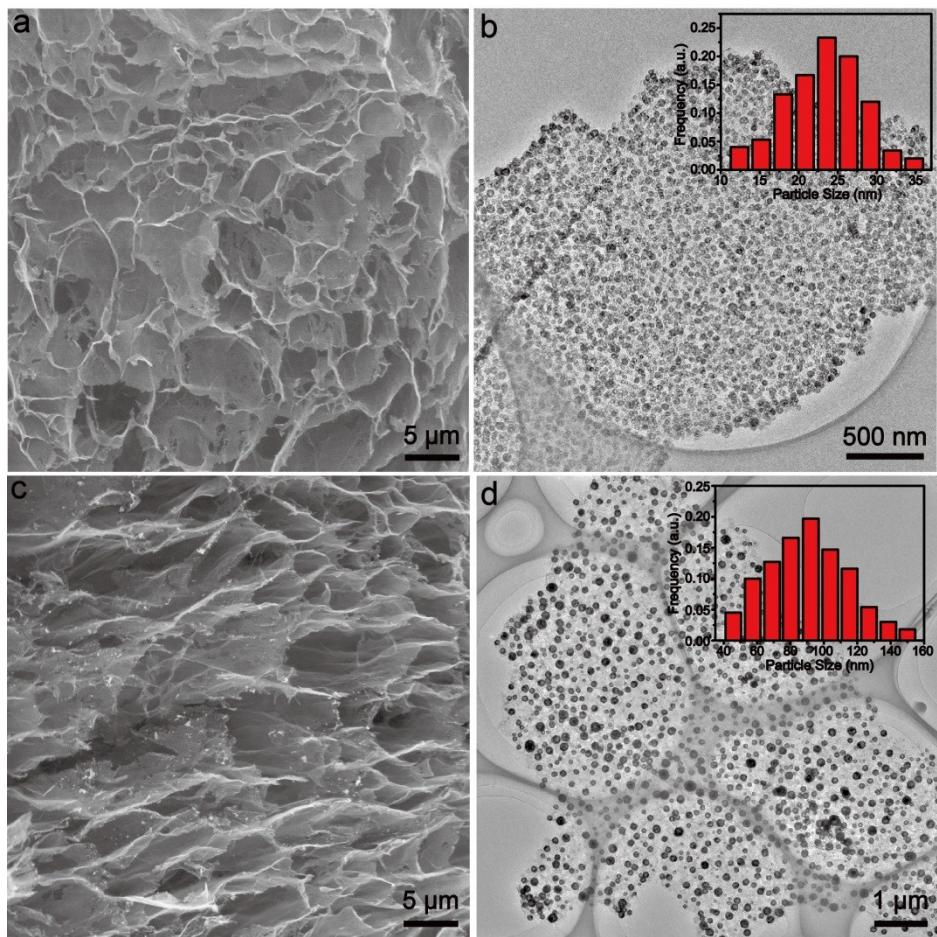


Figure S8. (a) SEM and (b) TEM images of H-Co₃O₄/GMA. (c) SEM and (d) TEM images of H-FeO_x/GMA. Inset in (b) and (d) are the corresponding particle size distribution histograms.

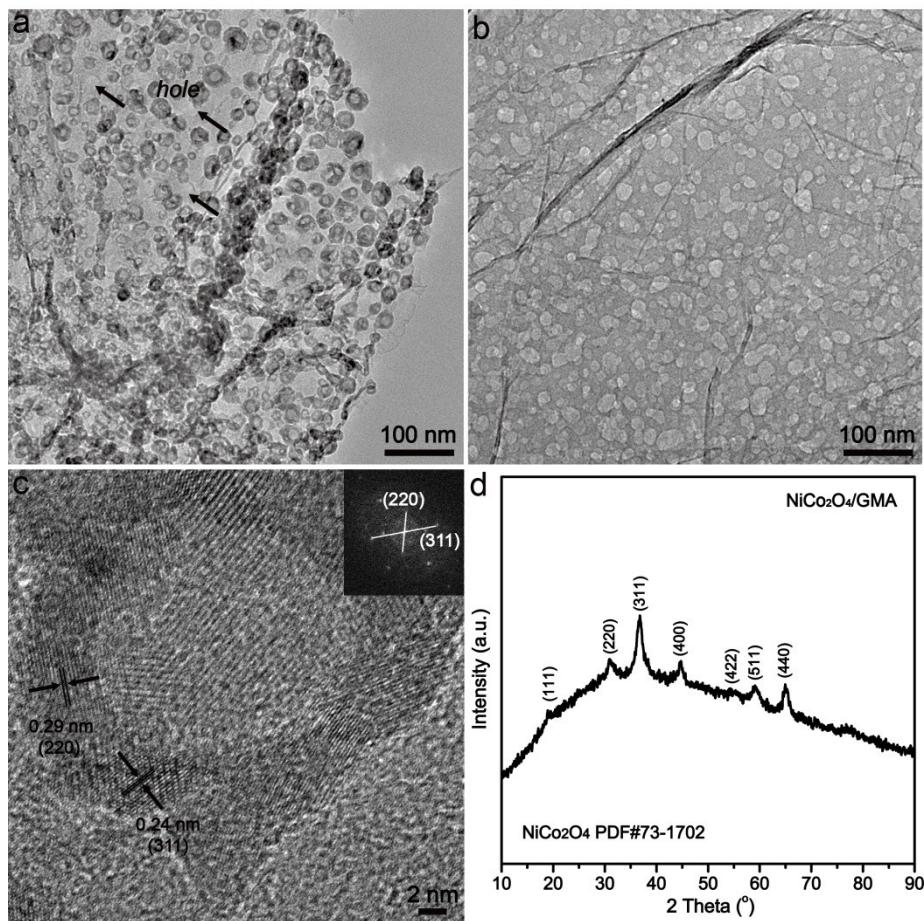


Figure S9. (a-c)TEM images of H-NiCo₂O₄/GMA. (d) XRD pattern of H-NiCo₂O₄/GMA. Inset in (c) is the corresponding FFT pattern.

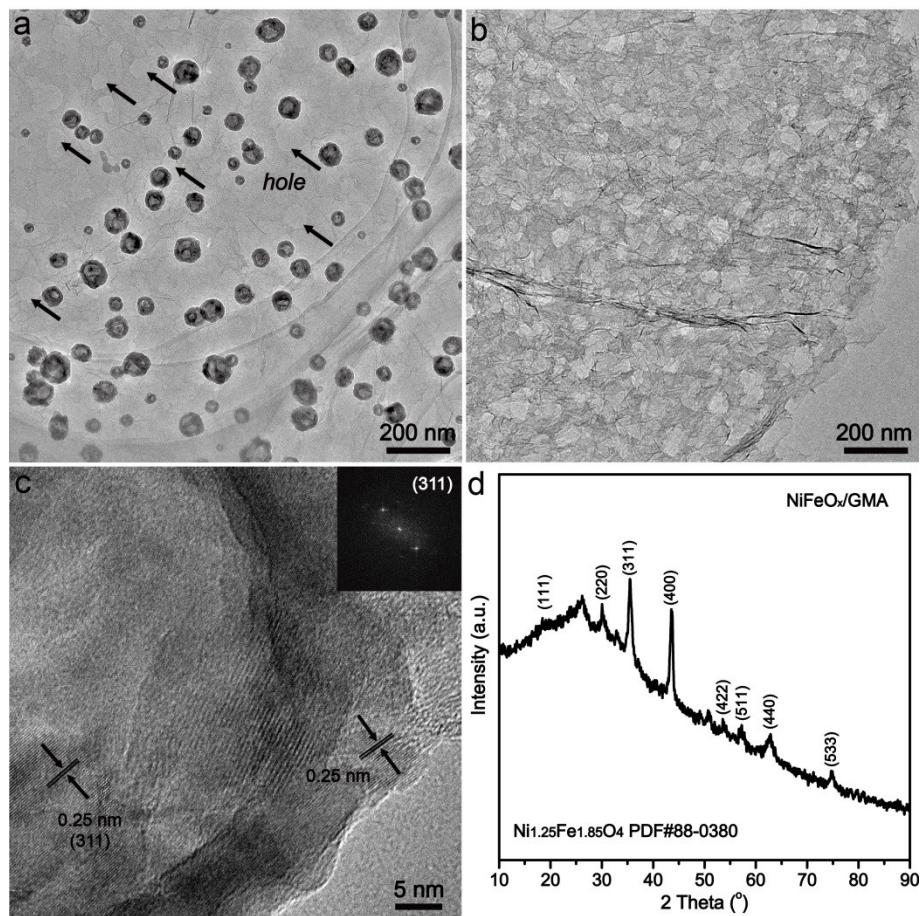


Figure S10. (a-c)TEM images of H-NiFeO_x/GMA. (d) XRD pattern of H-NiFeO_x/GMA. Inset in (c) is the corresponding FFT pattern.

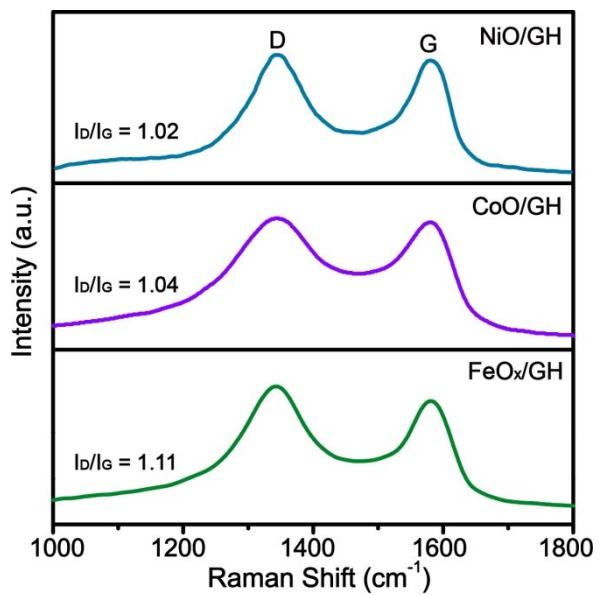


Figure S11. Raman spectra of NiO/GH, CoO/GH and FeO_x/GH.

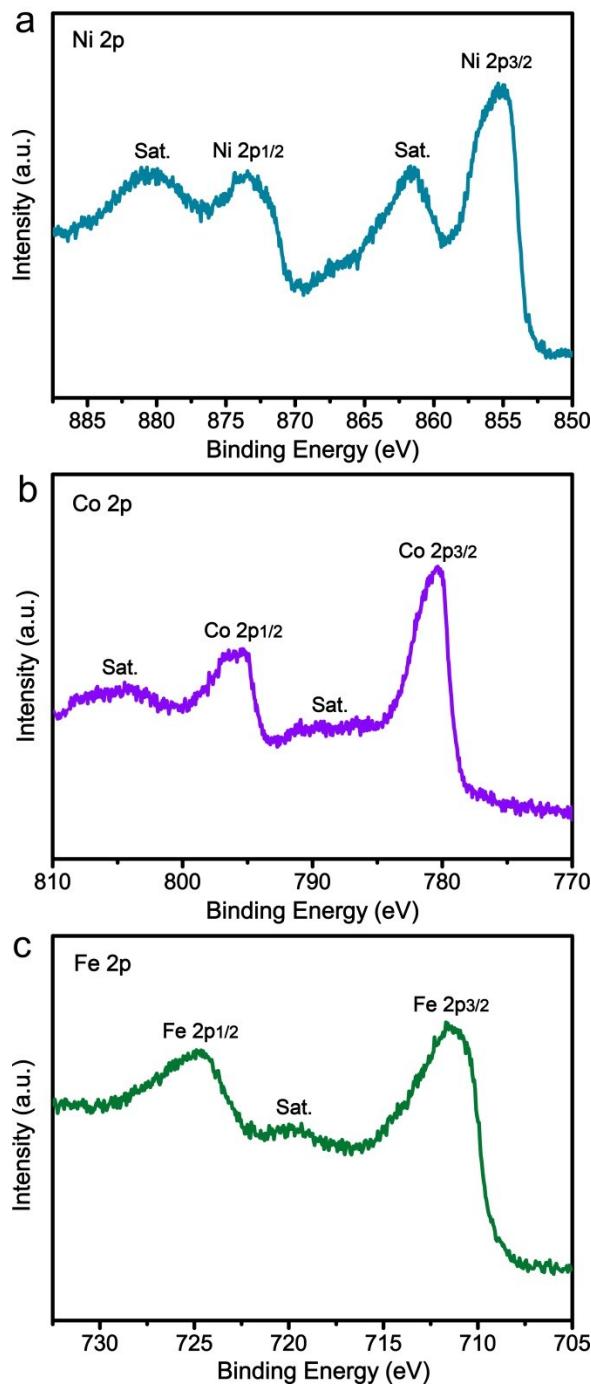


Figure S12. (a) Ni 2p XPS spectra of H-NiO/GMA. (b) Co 2p XPS spectra of H-Co₃O₄/GMA. (c) Fe 2p XPS spectra of H-FeO_x/GMA.

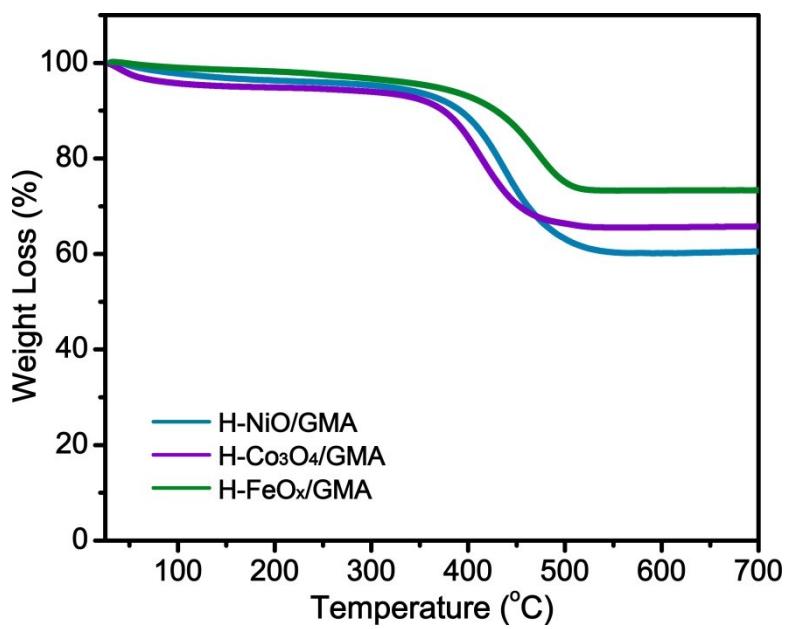


Figure S13. TGA curves of H-NiO/GMA, H-Co₃O₄/GMA and H-FeO_x/GMA in air atmosphere.

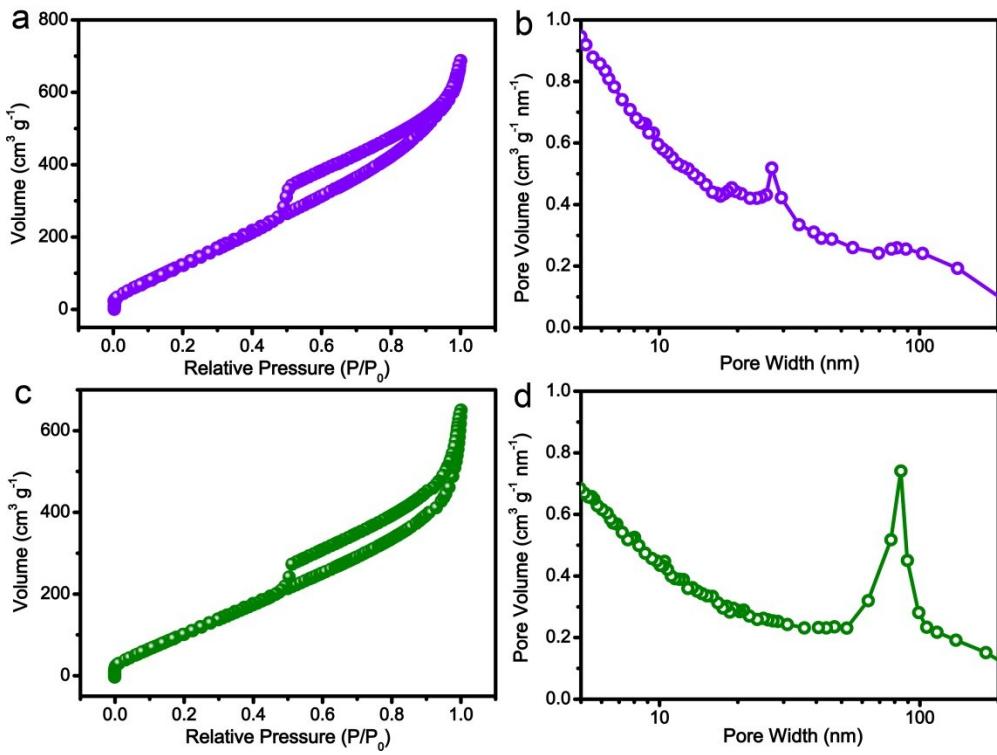


Figure S14. N_2 adsorption-desorption isotherms of (a) H-Co₃O₄/GMA and (c) H-FeO_x/GMA. Pore size distribution of (b) H-Co₃O₄/GMA and (d) H-FeO_x/GMA.

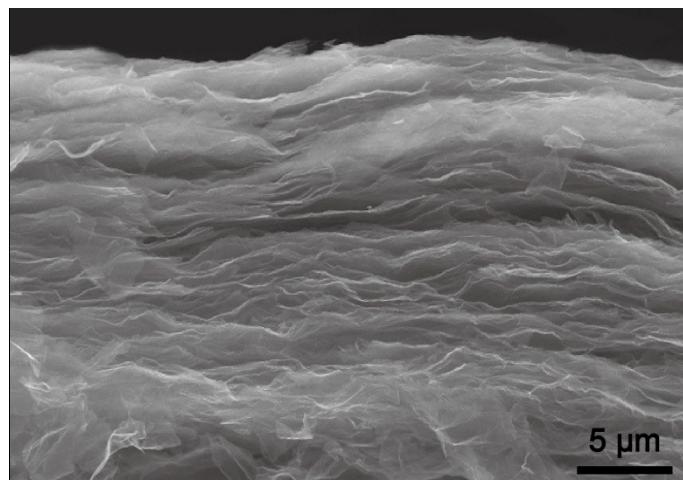


Figure S15. Cross-sectional SEM image of the compressed H-NiO/GMA.

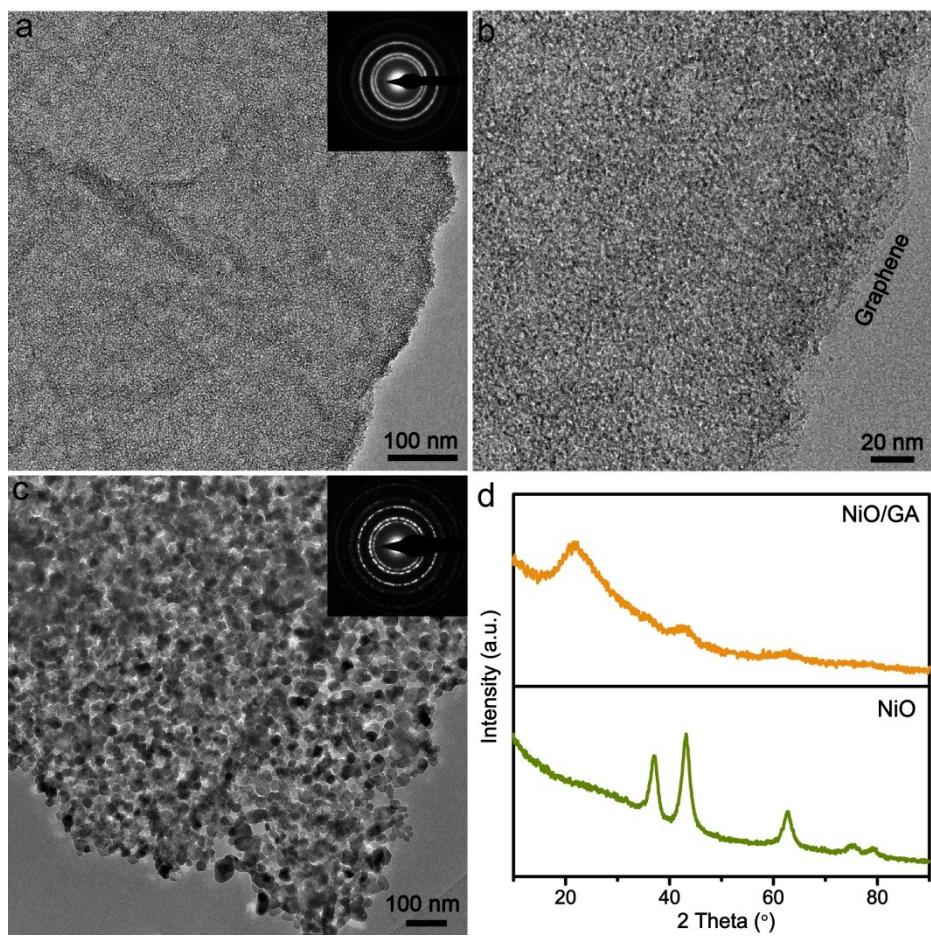


Figure S16. TEM images of (a,b) NiO/GA and (c) pure NiO. (d) XRD patterns of NiO/GA and pure NiO. Inset in (a) and (c) are the corresponding SAED patterns.

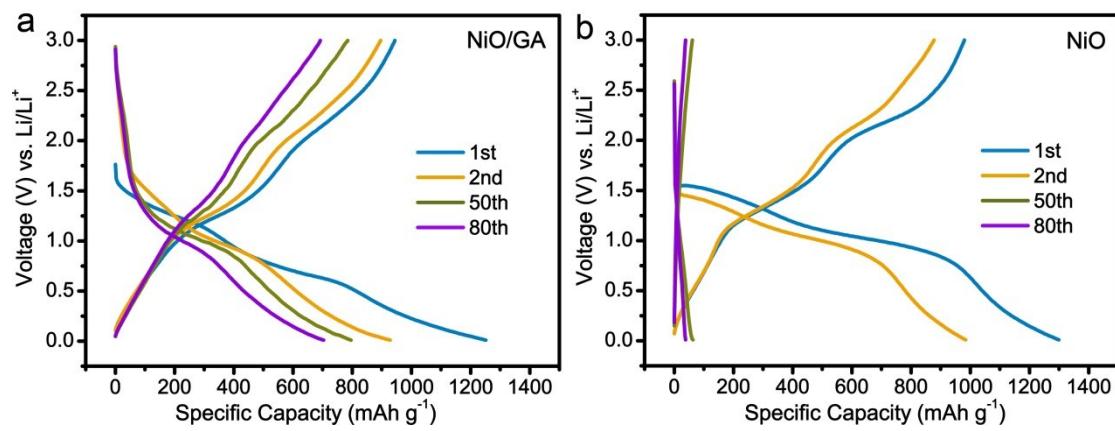


Figure S17. Discharge/charge profiles of (a) NiO/GA and (b) pure NiO electrodes at a current density of 0.2 A g⁻¹.

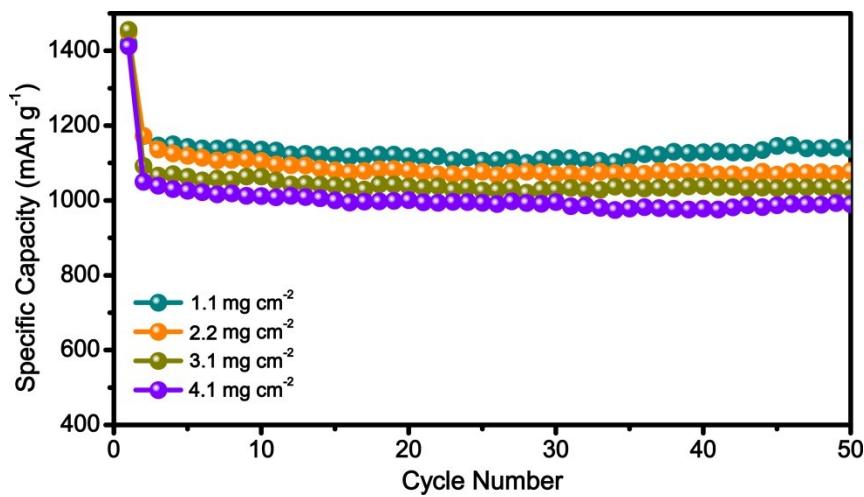


Figure S18. Cycling performance of H-NiO/GMA electrode with different loading amounts at a current density of 0.2 A g⁻¹.

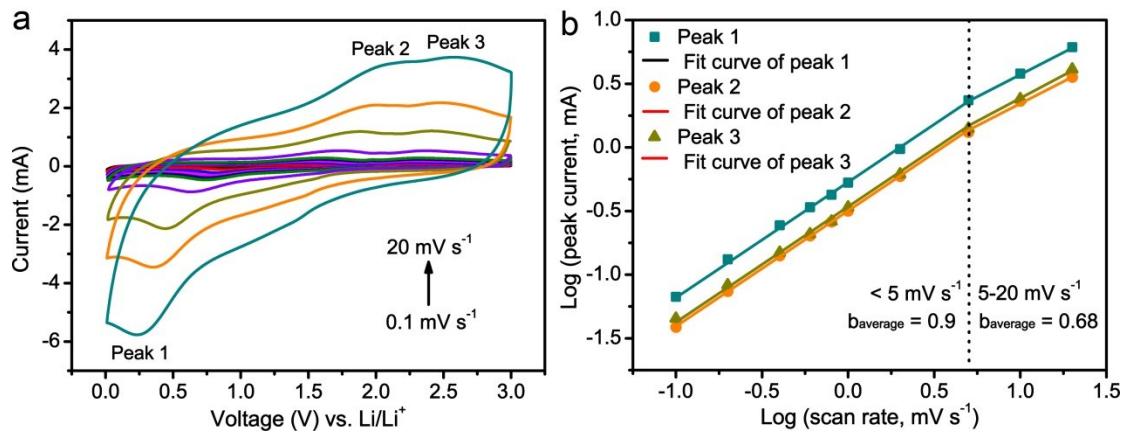


Figure S19. Kinetics analysis of the electrochemical behavior of the NiO/GA electrode. (a) CV curves at various scan rates from 0.1 to 20 mV s⁻¹. (b) b-value analysis using the relationship between the peak currents and the scan rates.

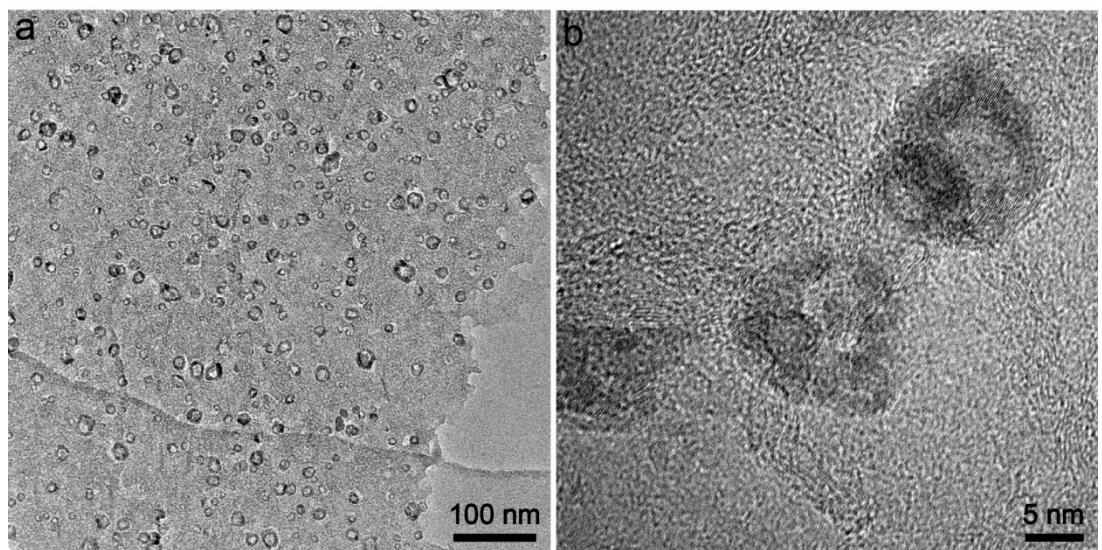


Figure S20. (a) TEM and (b) HRTEM images of H-NiO/GMA after cycling test.

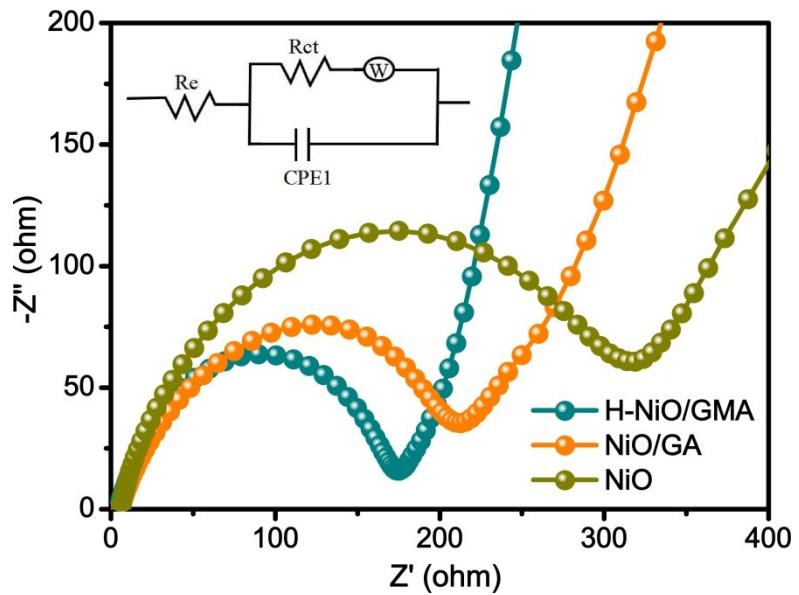


Figure S21. EIS spectra of H-NiO/GMA, NiO/GA and pure NiO electrodes. Inset is the corresponding equivalent circuit model. R_e shows the resistance of electrolyte, R_{ct} is the charge transfer resistance, CPE shows the double layer capacitance and W is the Warburg impedance.

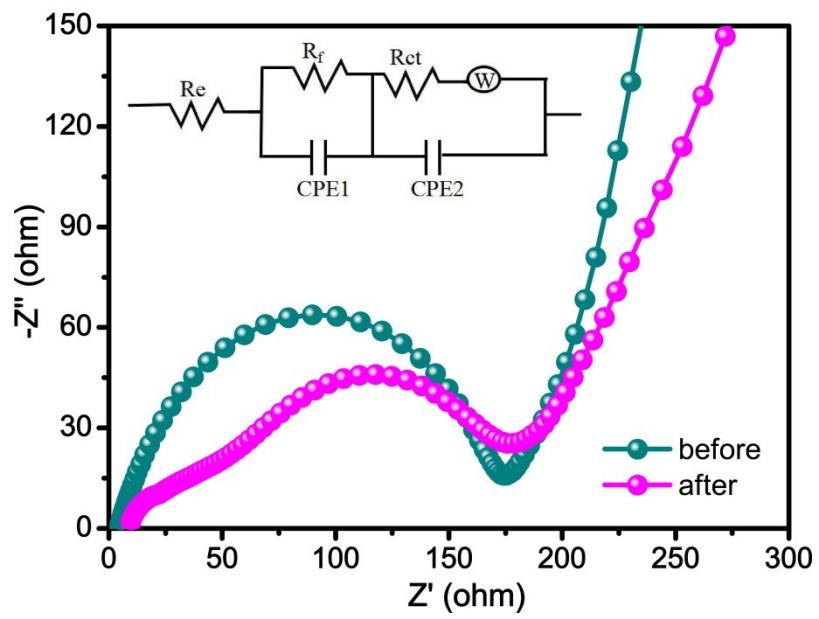


Figure S22. EIS spectra of H-NiO/GMA electrode before and after cycling. Inset is the equivalent circuit model after cycling. R_f is the resistance of SEI film.

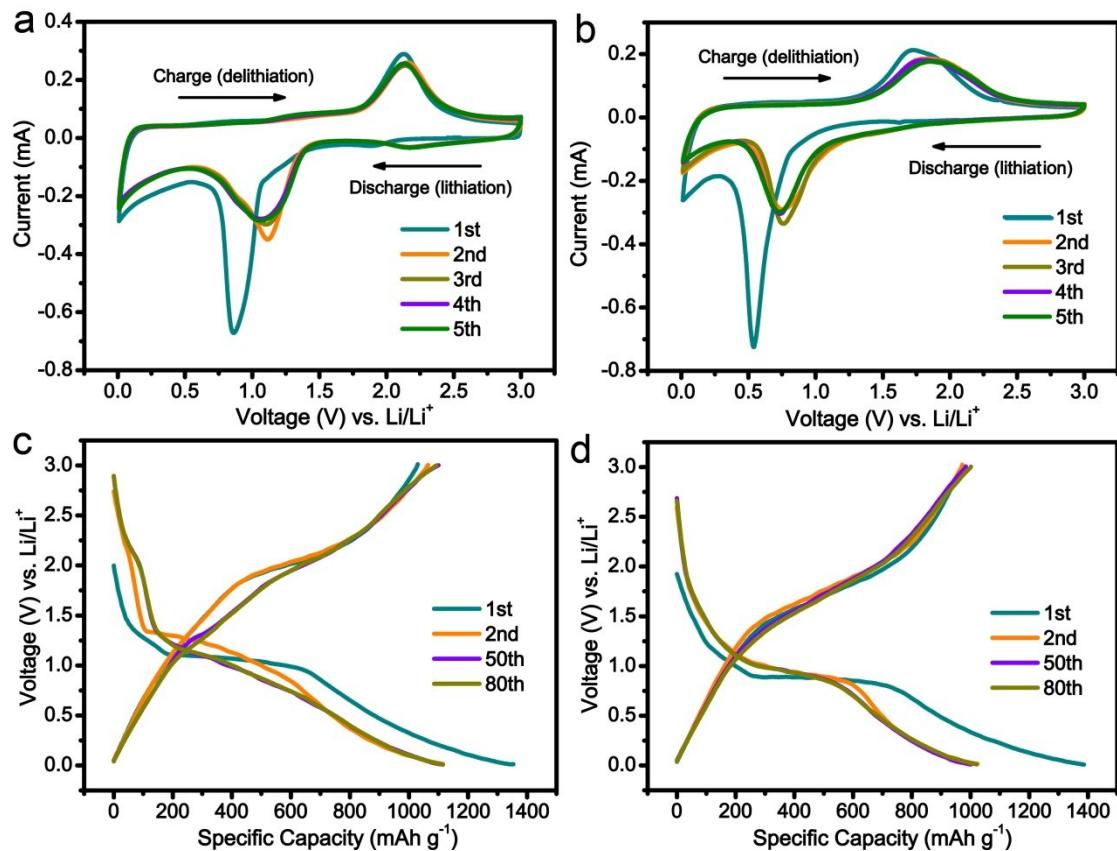


Table S1. Comparison of electrochemical performance of H-NiO/GMA and recently reported typical NiO-based anode materials.

Electrodes	Mass Loading (mg cm ⁻²)	Storage Capacity (mAh g ⁻¹)	Rate Capability (mAh g ⁻¹)	Capacity Retention	References
NiO/graphene composites	1-2	883 (0.05 A g ⁻¹)	550 (2.5 A g ⁻¹)	90 % (50 cycles at 0.05 A g ⁻¹)	1
Triple-shelled NiO microfibers	0.7	920 (0.1 A g ⁻¹)	540 (5 A g ⁻¹)	87 % (200 cycles at 1 A g ⁻¹)	2
Mesoporous NiO microspheres	N.A.	800 (0.5 A g ⁻¹)	620 (1 A g ⁻¹)	114 % (100 cycles at 0.5 A g ⁻¹)	3
Yolk-shell NiO/GQDs microspheres	1.5	1182 (0.1 A g ⁻¹)	200 (5 A g ⁻¹)	121 % (250 cycles at 0.1 A g ⁻¹)	4
RGO/Ni foam	1.5	1200 (0.1 A g ⁻¹)	335 (3 A g ⁻¹)	74.3 % (300 cycles at 0.2 A g ⁻¹)	5
Diaper derived NiO/Ni composites	0.8	1075 (0.1 A g ⁻¹)	440 (4 A g ⁻¹)	97.6 % (400 cycles at 1 A g ⁻¹)	6
NiO nanowire foam	1.3	577 (0.14 A g ⁻¹)	75 (35.9 A g ⁻¹)	68 % (1000 cycles at 0.359 A g ⁻¹)	7
NiO nanosheets	1.5-1.8	892 (0.1 A g ⁻¹)	298 (5 A g ⁻¹)	78 % (150 cycles at 0.7 A g ⁻¹)	8
Porous NiO nanorods	N.A.	700 (0.1 A g ⁻¹)	150 (2 A g ⁻¹)	134 % (60 cycles at 0.1 A g ⁻¹)	9
Hollow NiO nanospheres/ N-doped graphene	1	1104 (0.08 A g ⁻¹)	422 (4 A g ⁻¹)	135 % (150 cycles at 0.08 A g ⁻¹)	10
NiO/graphene/carbon Fiber	0.7	834 (0.1 A g ⁻¹)	380 (2 A g ⁻¹)	94 % (350 cycles at 0.5 A g ⁻¹)	11
NiO hollow microspheres	0.8	766 (0.5 A g ⁻¹)	457 (10 A g ⁻¹)	76.4 % (100 cycles at 1 A g ⁻¹)	12
Porous NiO-wrapped graphene	N.A.	705 (0.2 A g ⁻¹)	403 (1.6 A g ⁻¹)	47 % (50 cycles at 0.2 A g ⁻¹)	13
Hollow NiO/C hybrid nanoparticle	N.A.	622 (0.2 A g ⁻¹)	500 (1 A g ⁻¹)	88.7 % (100 cycles at 1 A g ⁻¹)	14
Ni/MoO ₂ microflowers	1	994 (0.1 A g ⁻¹)	406 (5 A g ⁻¹)	90.7 % (100 cycles at 1 A g ⁻¹)	15
NiO-Co ₃ O ₄ @C nanocomposites	1	870 (0.1 A g ⁻¹)	315 (5 A g ⁻¹)	109 % (100 cycles at 0.1 A g ⁻¹)	16
SnO ₂ /NiO@Ag nanotubes	3	1150 (0.1 A g ⁻¹)	300 (10 A g ⁻¹)	99 % (500 cycles at 1 A g ⁻¹)	17
CuO@NiO hollow spheres	2	1061 (0.1 A g ⁻¹)	N.A.	124 % (200 cycles at 0.1 A g ⁻¹)	18
TiC/NiO core/shell	N.A.	636 (0.05 A g ⁻¹)	369 (3 A g ⁻¹)	90 % (60 cycles at 0.2 A g ⁻¹)	19
H-NiO/GMA	1.1	1202 (0.1 A g⁻¹)	574 (10 A g⁻¹)	98 % (1000 cycles at 10 A g⁻¹)	This work

Table S2. The parameters obtained from the EIS on various electrodes.

Electrodes	R_e (Ω)	R_{ct} (Ω)	R_f (Ω)
H-NiO/GMA	4.8	168.6	-
(after 80 cycles)	8.5	156.4	31.5
NiO/GA	5.5	201.4	-
NiO	6.1	308.2	-
H-Co ₃ O ₄ /GMA	5.3	203.4	-
H-FeO _x /GMA	5.8	238.6	-

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