

Electronic Supplementary Information

Nitrogen- and iodine-doped microporous carbon derived from a hydrogen-bonded organic framework: an efficient metal-free electrocatalyst for oxygen reduction reaction

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Table S1. The yield of pyrolyzed carbon synthesized at different temperatures.

Carbon Materials	M(HOF-8)/mg	M(Carbon)/mg	Yield/%
C _{KI}	300	30	10.0
C ₇₀₀	300	125	41.7
C ₈₀₀	300	121	40.3
C ₉₀₀	300	116	38.7
C ₁₀₀₀	300	100	33.3

Table S2. Content of elements in carbon materials calculated from XPS.

Carbon materials	C content/at%	O content/at%	N content/at%	Pyridinic N/at%	I content/at%
C_{KI}	82.97	8.89	7.42	1.97	0.71
C₇₀₀	84.99	10.23	4.78	1.22	0
C₈₀₀	83.39	12.20	4.42	1.16	0
C₉₀₀	80.95	15.27	3.78	1.09	0
C₁₀₀₀	85.42	12.17	2.41	0.29	0

Table S3. BET surface areas and pore volumes of carbon materials.

Carbon materials	BET Surface Areas/m ² g ⁻¹	Total Pore Volumes/cm ³ g ⁻¹
C_{KI}	1140	0.562
C₇₀₀	22	0.026
C₈₀₀	327	0.147
C₉₀₀	282	0.138
C₁₀₀₀	142	0.078

Table S4. I_D/I_G values calculated from Raman spectra.

Carbon materials	I _D /I _G
C_{KI}	1.10
C₇₀₀	1.10
C₈₀₀	1.08
C₉₀₀	1.06
C₁₀₀₀	1.05

Table S5. Various parameters of carbon materials for ORR catalysis obtained from RDE experiments.

Carbon materials	Onset potential/V vs. RHE	Halfwave potential/V vs. RHE	Limited current density (@0.213 V vs. RHE)/mA cm ⁻²	Kinetic current density (@0.830 V vs. RHE) /mA cm ⁻²
C_{KI}	0.944	0.828	4.96	4.470
C₇₀₀	0.744	0.539	2.02	0.004
C₈₀₀	0.862	0.705	3.64	0.112
C₉₀₀	0.850	0.704	2.76	0.077
C₁₀₀₀	0.836	0.703	2.57	0.050
Pt/C	0.936	0.816	5.45	2.742

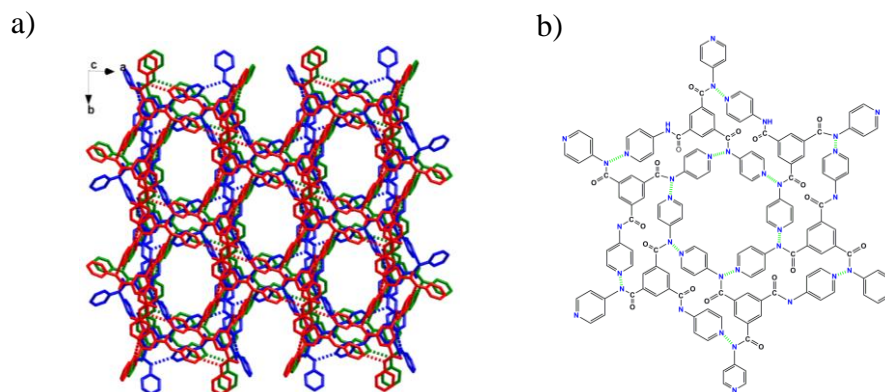


Fig. S1 a) The 3D supramolecular structure of HOF-8, and b) the chemical structure of HOF-8.

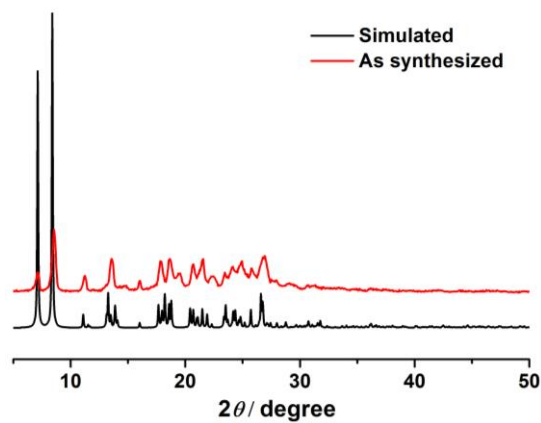


Fig. S2 Powder XRD of HOF-8 at room temperature.

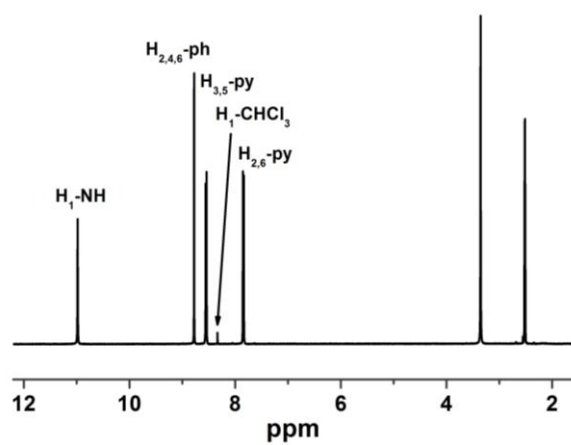


Fig. S3 ^1H NMR spectrum of HOF-8 (DMSO- d_6 , 400 MHz, 298 K).

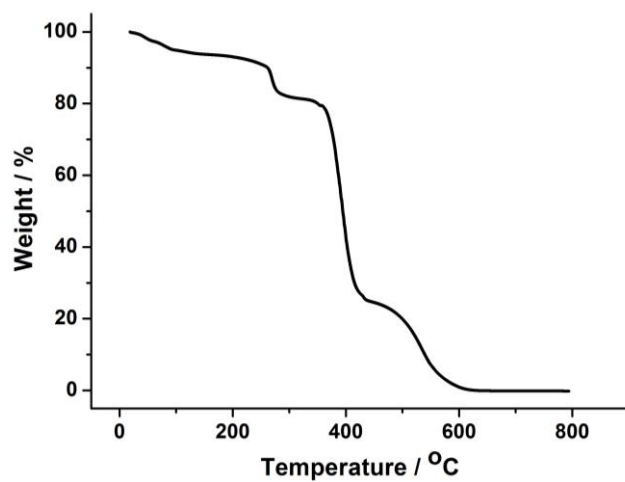


Fig. S4 TGA curve of HOF-8 at the N₂ atmosphere.

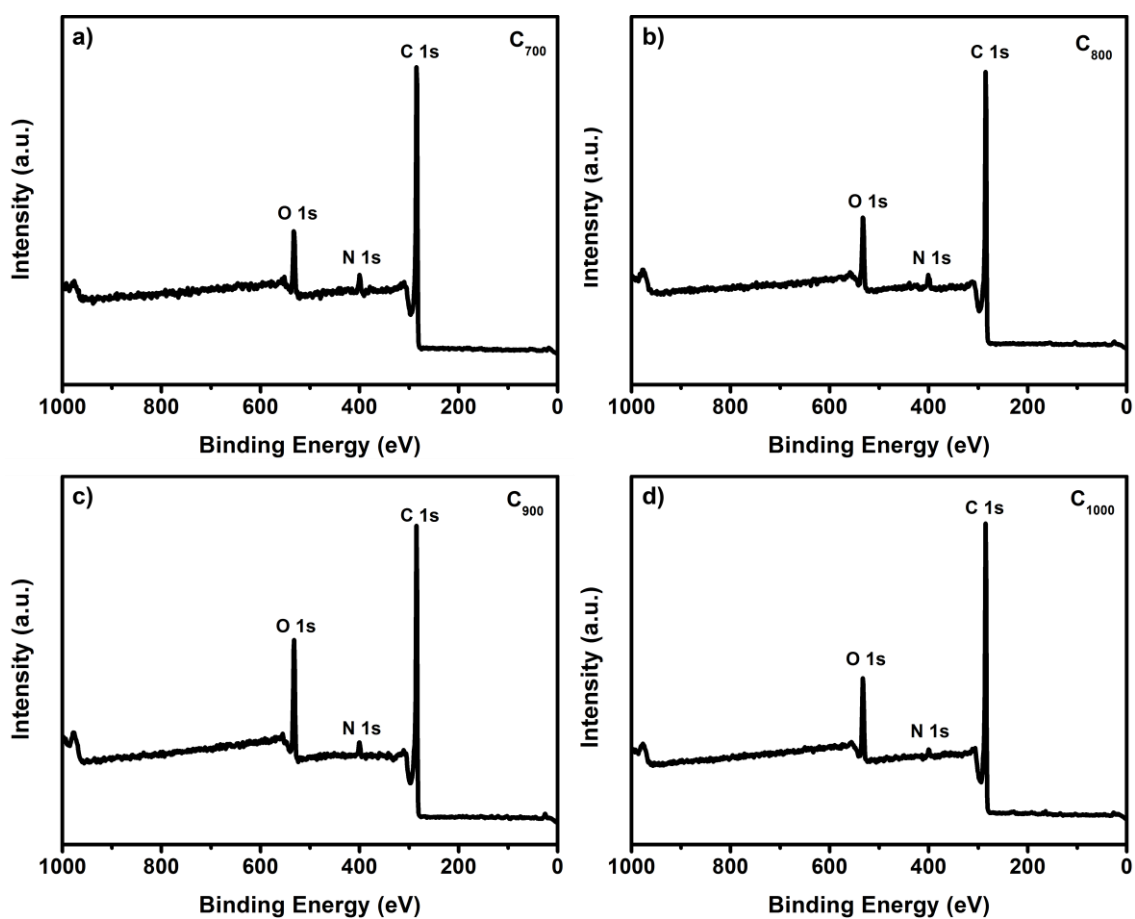


Fig. S5 XPS spectra of a) C₇₀₀, b) C₈₀₀, c) C₉₀₀ and d) C₁₀₀₀.

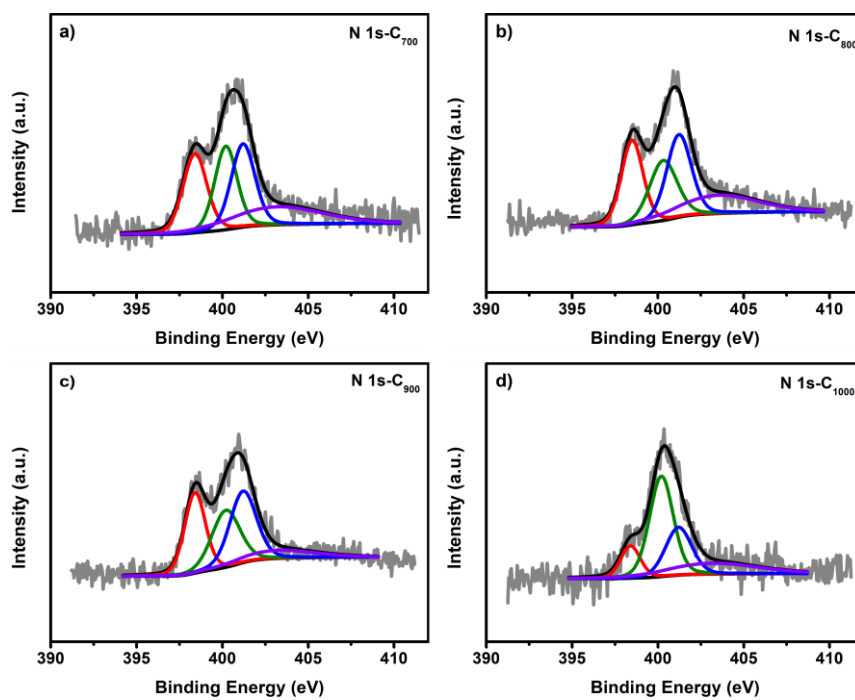


Fig. S6 High resolution N 1s spectra of a) C₇₀₀, b) C₈₀₀, c) C₉₀₀ and d) C₁₀₀₀.

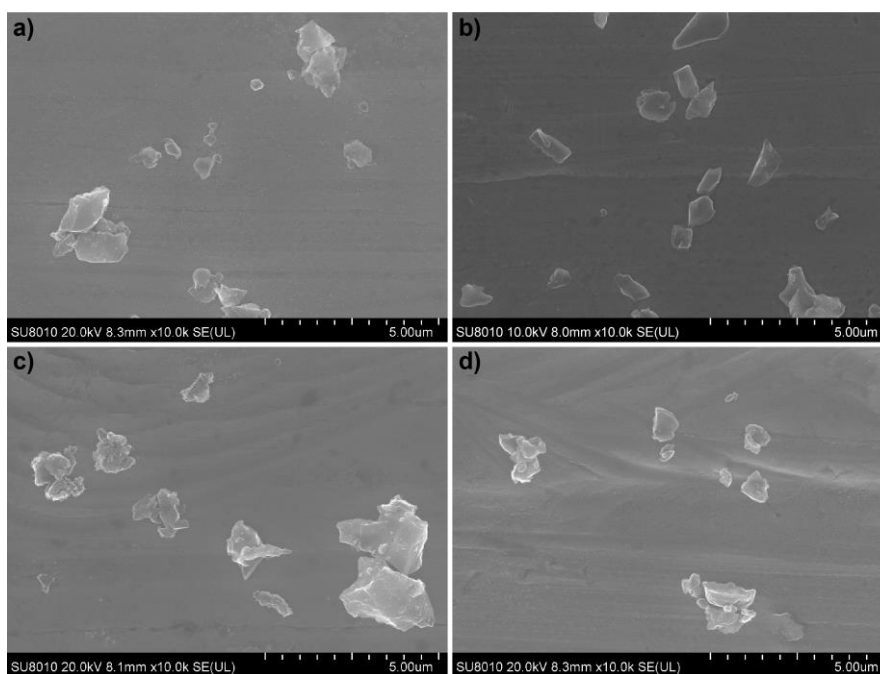


Fig. S7 SEM images of a) C₇₀₀; b) C₈₀₀; c) C₉₀₀; d) C₁₀₀₀.

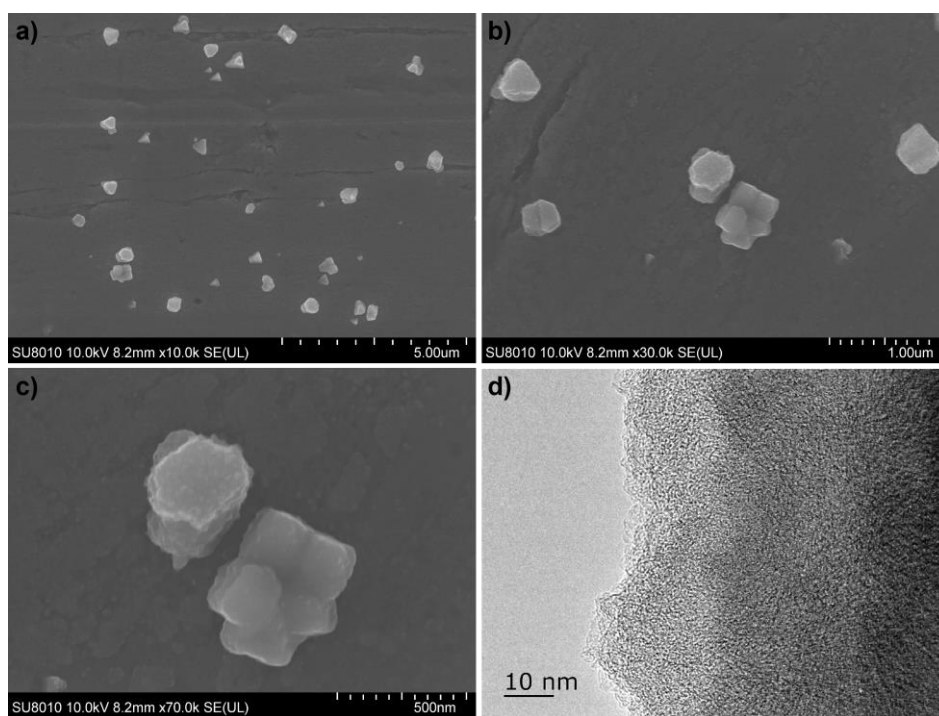


Fig. S8 a), b), c) SEM images showing the regular polyhedral shape and hierarchical structure of C_{KI}; d) TEM image of C_{KI}.

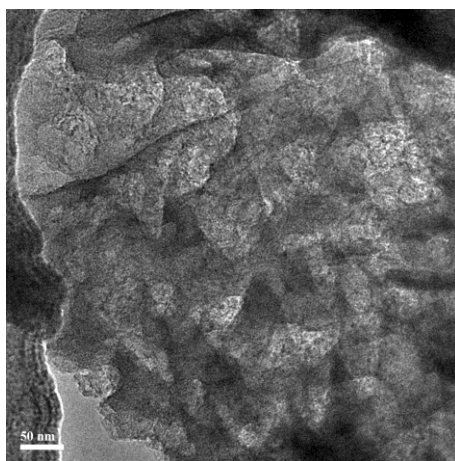


Fig. S9 TEM image of C_{KI}.

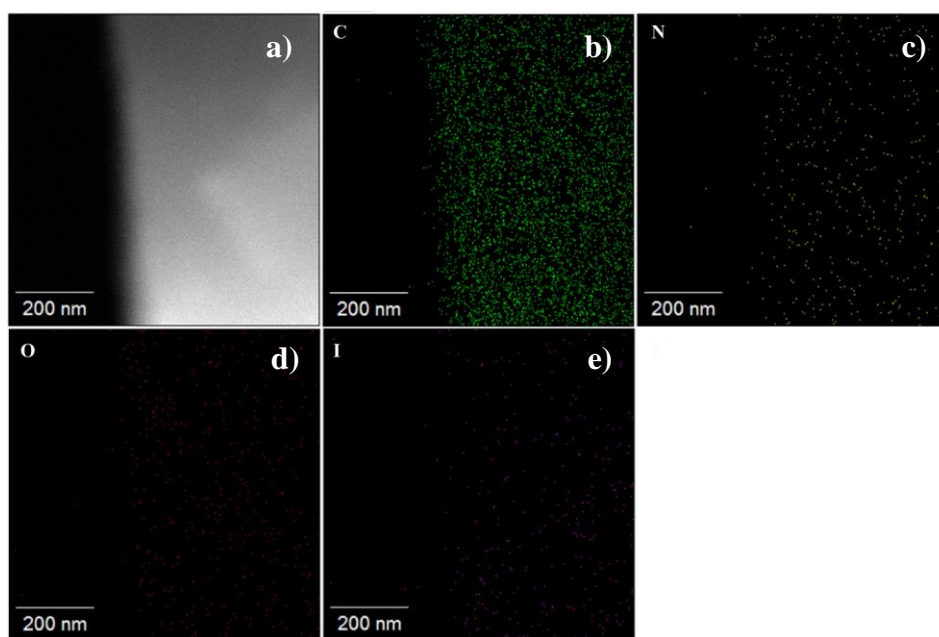


Fig. S10 a) TEM image of C_{KI} and the corresponding elemental mapping of b) C, c) N, d) O and e) I.

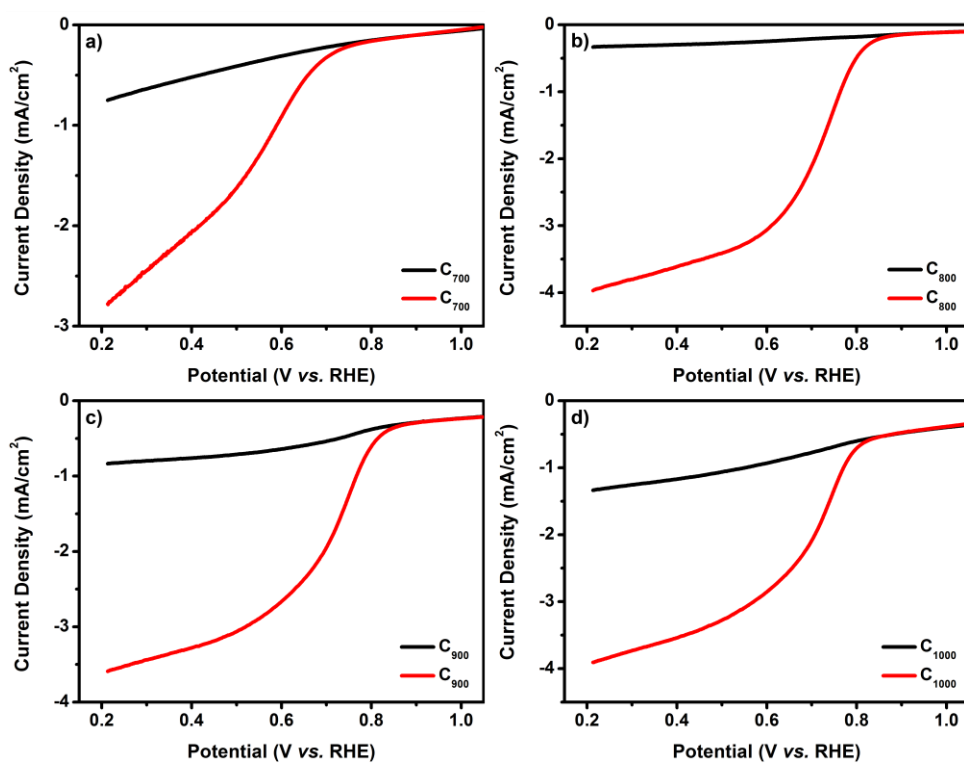


Fig. S11 RDE curves of a) C_{700} , b) C_{800} , c) C_{900} and d) C_{1000} in 0.1 M KOH solution saturated with N_2 (black) and O_2 (red).

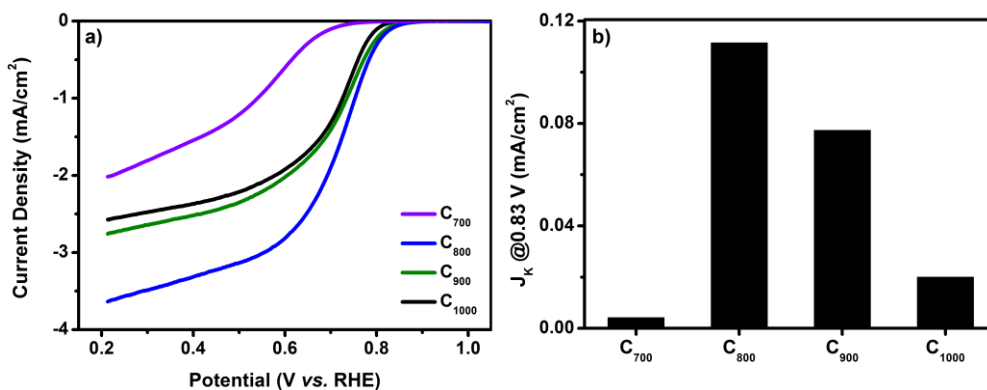


Fig. S12 a) LSV polarization curves of C_T at scanning speed of 10 mV s^{-1} and rotating speed of 1600 rpm in 0.1 M KOH solution with background subtraction. b) Comparison of kinetic current density of C_T at 0.83 V .

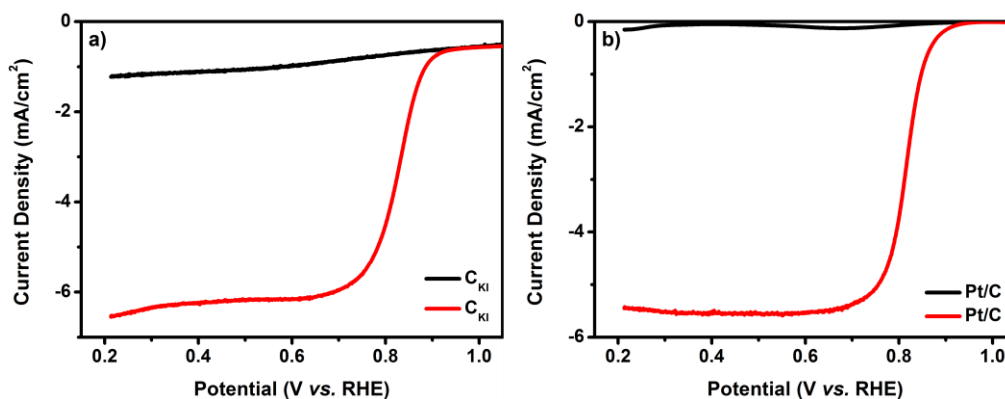


Fig. S13 RDE curves of a) C_{KI} and b) Pt/C in 0.1 M KOH solution saturated with N_2 (black) and O_2 (red).

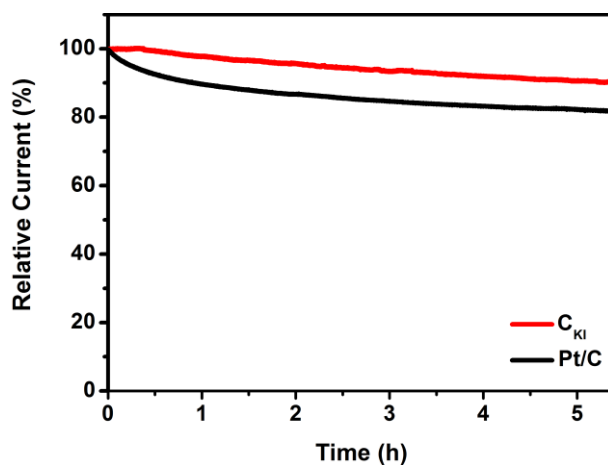


Fig. S14. Chronoamperometric responses of C_{KI} and 20 wt\% Pt/C in O_2 -saturated 0.1 M KOH at 0.764 V with 400 rpm rotating speed.

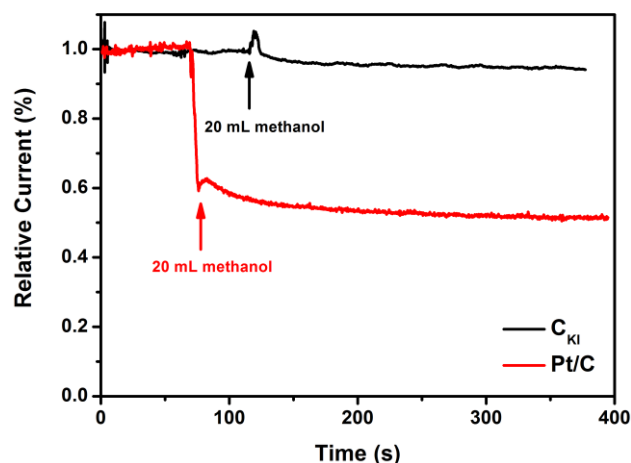


Fig. S15 Chronoamperometric responses of C_{KI} and 20 wt% Pt/C in O₂-saturated 0.1 M KOH under methanol addition at 0.764 V with 400 rpm rotating speed.

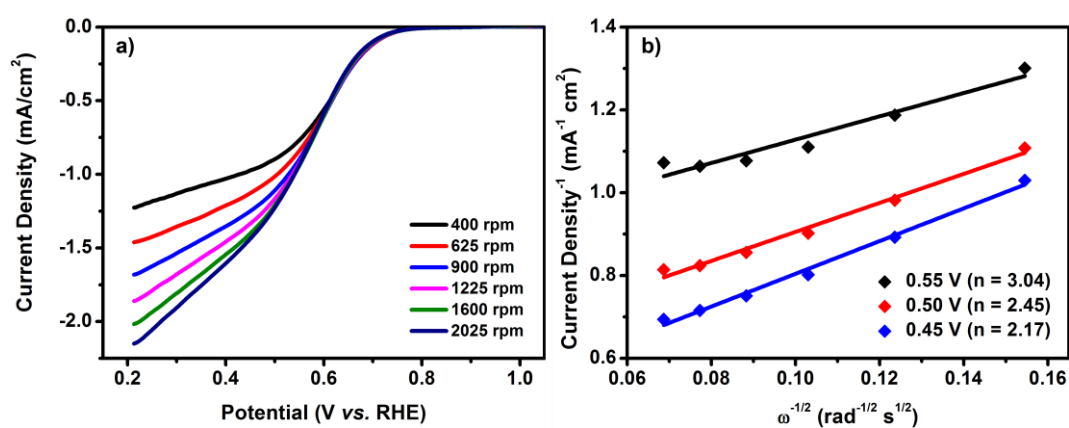


Fig. S16 a) LSV polarization curves of C₇₀₀ in 0.1 M KOH at various rotating speeds. b) K–L plots of C₇₀₀ at various potentials.

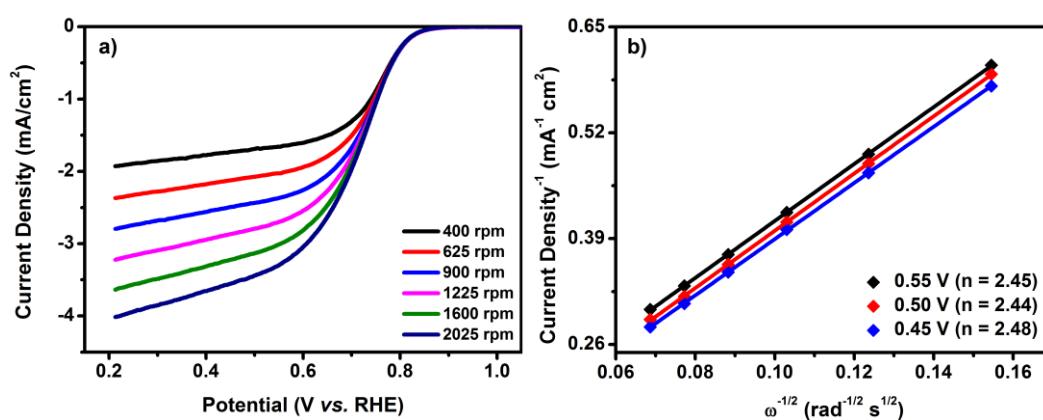


Fig. S17 a) LSV polarization curves of C₈₀₀ in 0.1 M KOH at various rotating speeds. b) K–L plots of C₈₀₀ at various potentials.

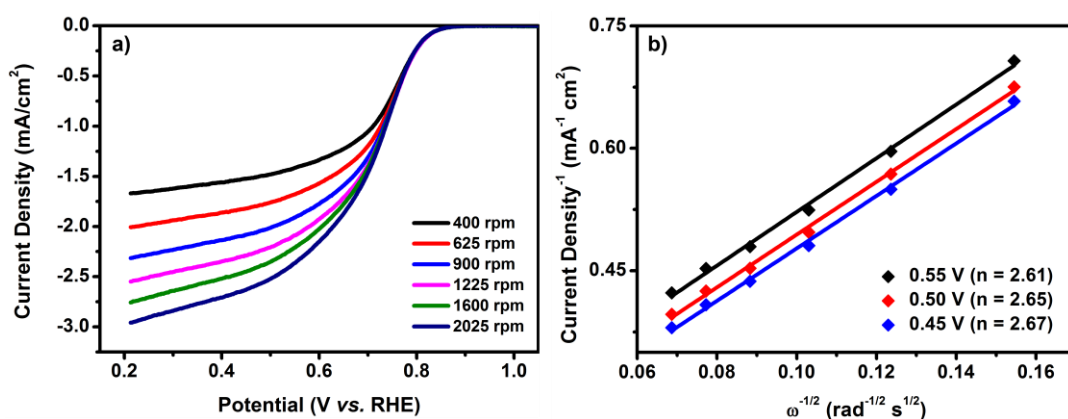


Fig. S18 a) LSV polarization curves of C₉₀₀ in 0.1 M KOH at various rotating speeds. b) K–L plots of C₉₀₀ at various potentials.

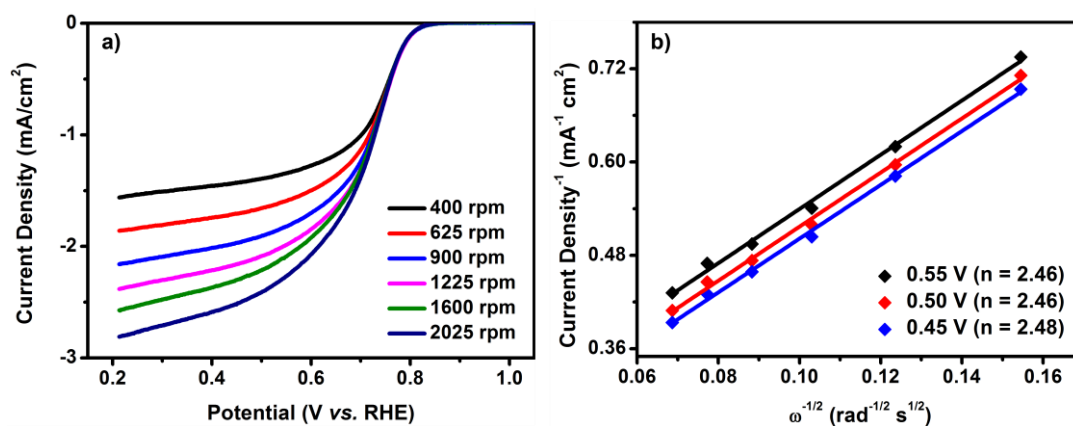


Fig. S19 a) LSV polarization curves of C₁₀₀₀ in 0.1 M KOH at various rotating speeds. b) K–L plots of C₁₀₀₀ at various potentials.