

Supporting Information

Efficient Polysulfides Barrier of Graphene Aerogel-Carbon Nanofibers-Ni Network for High-Energy-Density Lithium-Sulfur Batteries with Ultrahigh Sulfur Content

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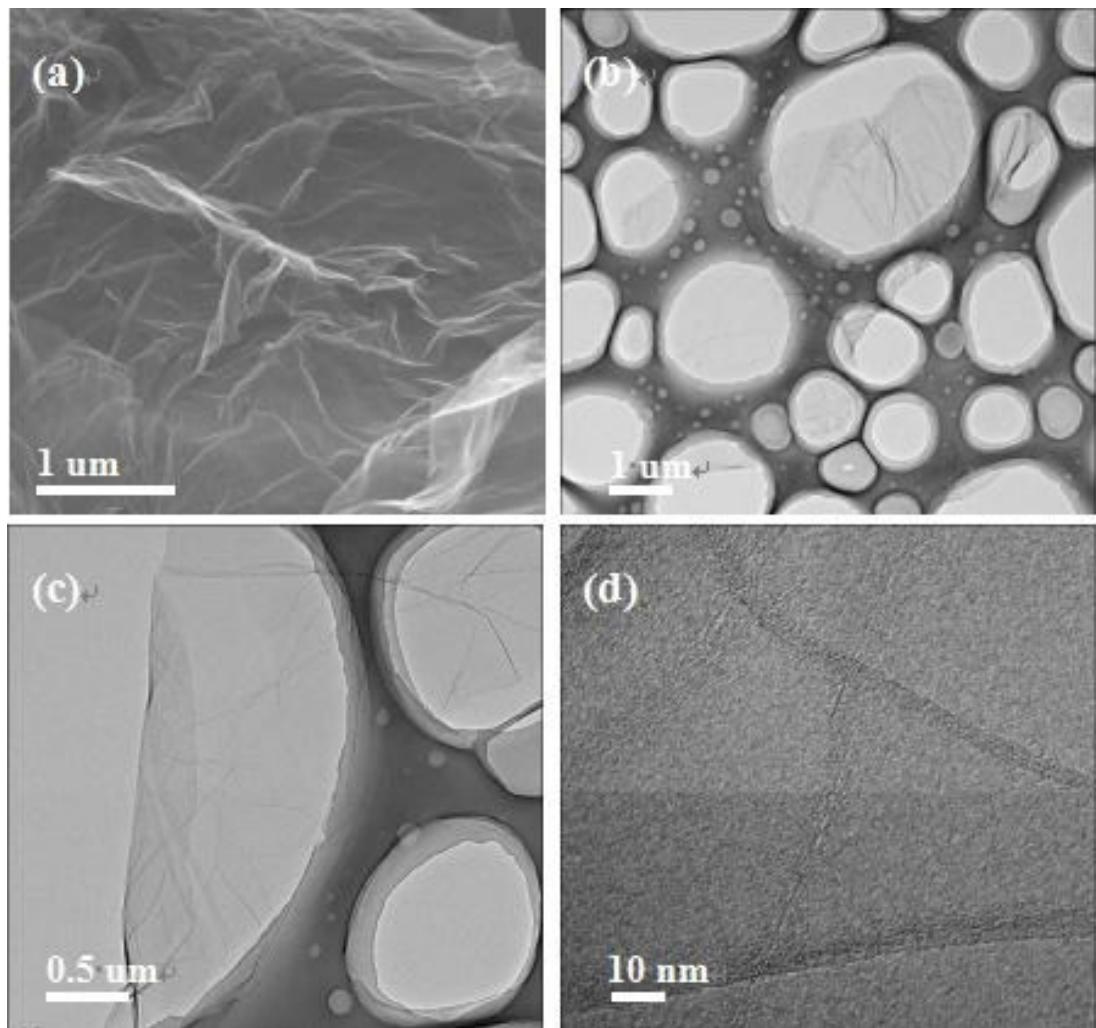


Figure S1. (a) SEM, (b, c) TEM and d) HRTEM images of exfoliated graphene oxide

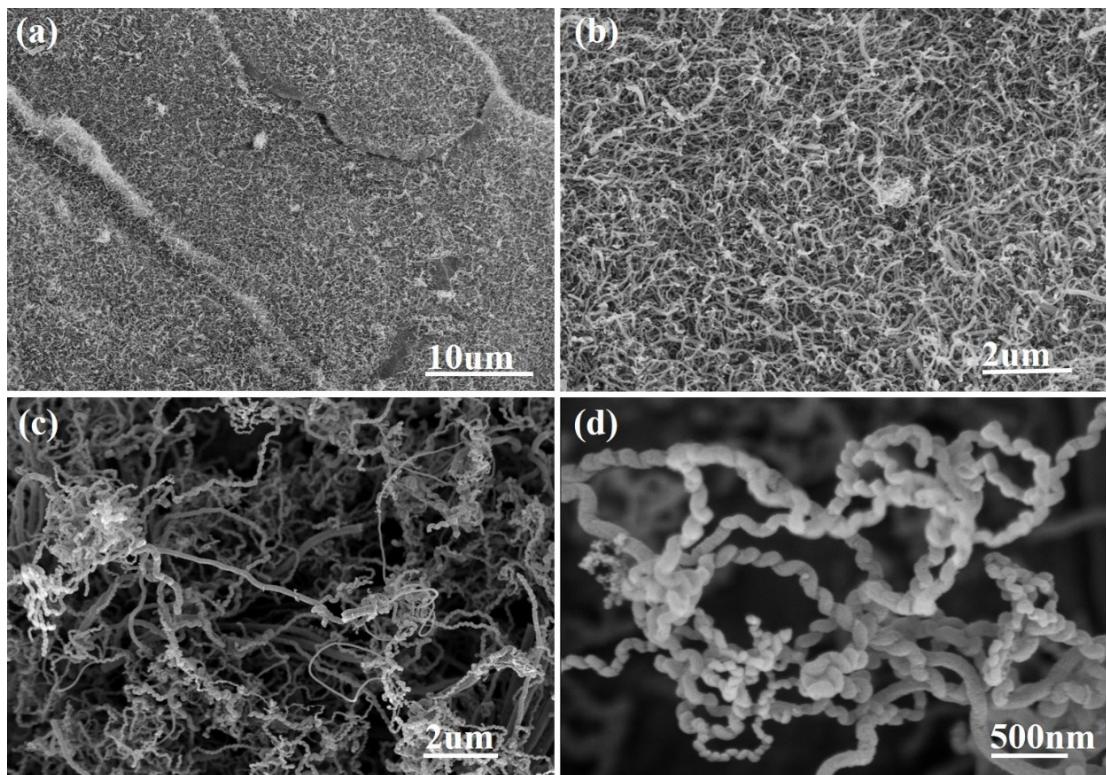


Figure S2. SEM images of GA-CNF-Ni hybrids synthesized with growth time of (a, b) 30 min and (c, d) 2 h.

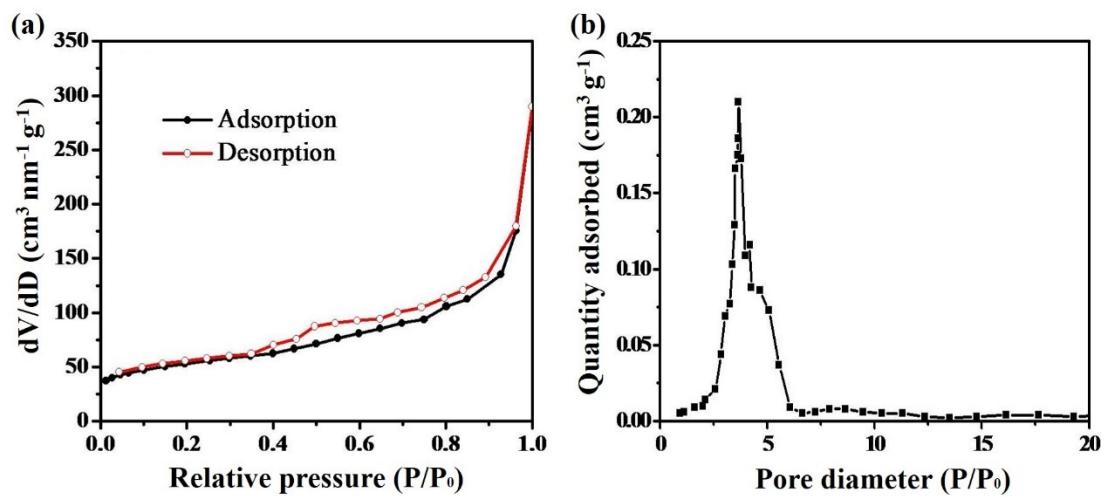


Figure S3. (a) N₂ adsorption-desorption isotherms of GA and (b) corresponding pore size distribution.

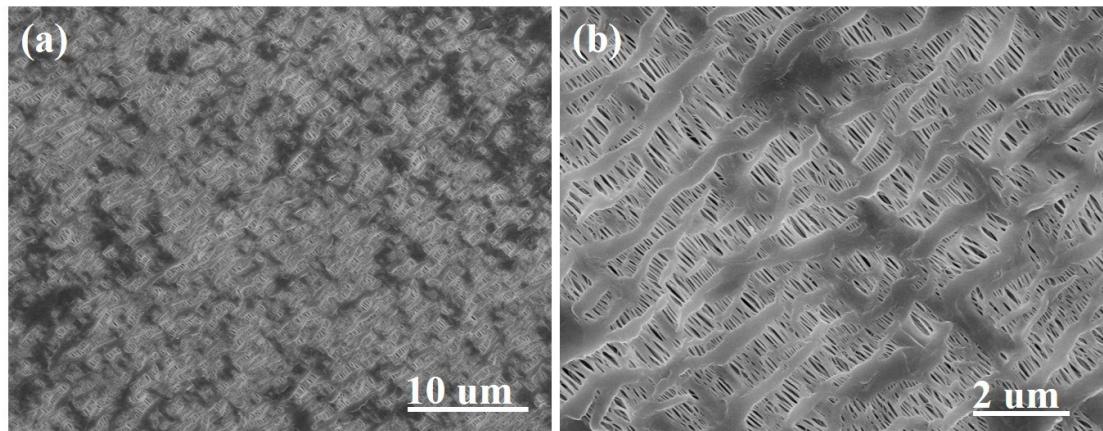


Figure S4. SEM images of pristine separator.

Table S1. Comparisons of electrical properties between GA-CNFs-Ni and GA/CNT coating layers

Coating layer	Resistivity ($\Omega \cdot cm$)	Conductivity ($S \text{ cm}^{-1}$)
GA-CNFs-Ni	6.8×10^{-3}	147
GA/CNT	1.2×10^{-2}	83

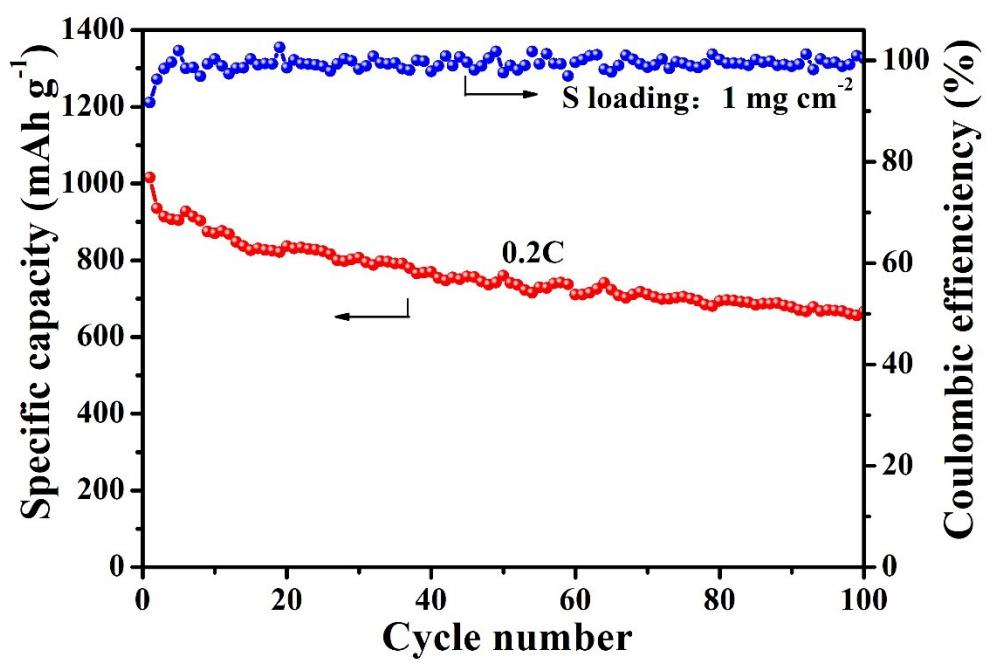


Figure S5. The cycle performance of GA-CNFs-Ni/S cathode with pristine separator at 0.2 C

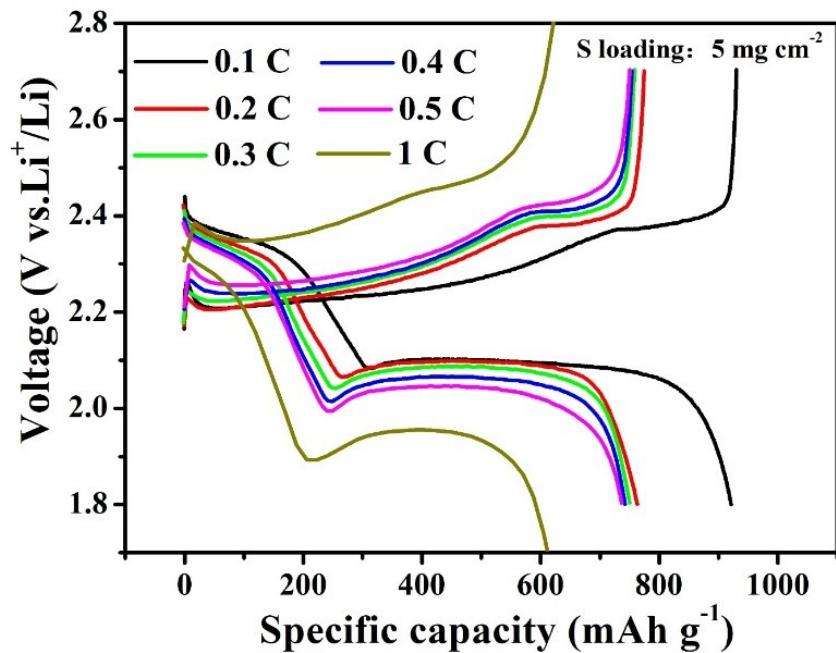


Figure S6. Charge-discharge voltage profiles of the cells fabricated with GA-CNFs modified separator at 0.2 C



Figure S7. Optical image of LED lamps with the pattern of “ECUST” lighted by the cell based on GA-CNFs-Ni coated separator and ultra-high loading sulfur cathode.

Table S2. Comparison of electrochemical properties functional separators in Li–S cells

Coating or interlayer	Sulfur loading	Sulfur content	Cathode	Discharge Capacity	Areal capacity (mAh cm ⁻²)	E/S ratio	Ref
MoO ₃ @CNT	1.0	60%	Super P/S	~755 mAh g ⁻¹ , 200 th , 0.3 C	<2.0	N/A	[S1]
SPEEK	6.0	45%	CNT/S	650 mAh g ⁻¹ , 500 th , 0.2 C	~3.9	N/A	[S2]
SSNS/CNT	1.0	65%	KB/S	680 mAh g ⁻¹ , 100 th , 0.2 C	<2.0	N/A	[S3]
Ultrathin hollow carbon sphere	3.0	57%	CB/S	730 mAh g ⁻¹ , 100 th , 0.2 C	~2.2	~13	[S4]
EUC-CNF	1.5	70%	Super P/S	~1038 mAh g ⁻¹ , 50 th , 0.2 C	<2.0	N/A	[S5]
GO@MoS ₂	3.64	70%	CB/S	~600 mAh g ⁻¹ , 95 th , 0.2 C	~2.2	~14	[S6]
Fe-N-carbon nanofiber	1.0	77%	S/KB	~601.9 mAh g ⁻¹ , 500 th , 0.5 C	<2.0	N/A	[S7]
MoO ₃	0.9-1.0	49%	CNT/S	~782 mAh g ⁻¹ , 100 th , 0.5 C	<2.0	40	[S8]
Co ₉ S ₈	2.0-5.6	70%	Super P/S	~985 mAh g ⁻¹ , 200 th , 0.1 C	5.5	10	[S9]
Nano TiO ₂	2.0	63%	AB/S	~883 mAh g ⁻¹ , 180 th , 0.1 C	<2.0	N/A	[S10]
CNF@VS ₂ /CNT@GN	5.6	60%	Super P/S	~750 mAh g ⁻¹ , 43 th , 0.3 C	4.1	N/A	[S11]
OCNT	2.5	65%	CB/S	~820 mAh g ⁻¹ , 60 th , 0.2 C	~2.0	N/A	[S12]
TiO ₂ /C	1.1	54%	AB/S	~1000 mAh g ⁻¹ , 60 th , 0.1 C	<2.0	36	[S13]
Gr-CNT-Ni	7.68	64%	polysulfide catholyte	~696 mAh g ⁻¹ , 100 th , 0.2 C	~5.3	10	[S14]
CNT@ZIF	1.2	56%	AB/S	~1100 mAh g ⁻¹ , 60 th , 0.2 C	<2.0	36	[S15]
CNF@ZrO ₂	2.7	70%	CB/S	~800 mAh g ⁻¹ , 60 th , 0.2 C	~2.1	40	[S16]
g-C ₃ N ₄	4.0	45%	GO/S	~600 mAh g ⁻¹ , 60 th , 0.2 C	~2.4	20	[S17]
aCNT	6.0	67%	aCNT/S	617 mA h g ⁻¹ , 60, 0.1 C	3.7	N/A	[S18]
CoP	3.24	56%	rGO/S	~800 mAh g ⁻¹ , 100 th , 0.2 C	~2.7	N/A	[S19]
MnO ₂	1.5-2.5	66%	KB/S	~603 mAh g ⁻¹ , 500 th , 0.5 C	<2.0	7.0-8.8	[S20]

rGO@SL	3.8	75%	CB/S	700 mAh g ⁻¹ 50 th , 0.05 C	2.7	8	[21]
MoP/rGO	3.6-3.9	85.6%	CNTs-AB/S	760 mAh g ⁻¹ 300 th , 0.5 C	3.2	15	[22]
TiN	1.3	70%	Super P/S	744 mAh g ⁻¹ 200 th , 0.5 C	<2.0	46	[23]
Sb ₂ S ₃	1.0	65%	KB/S	680 mAh g ⁻¹ 100 th , 0.2 C	<2.0	50	[24]
GA-CNF-Ni	5.0 10.0	90%	CB/S	890 mAh g ⁻¹ , 90 th , 0.2 C 550 mAh g ⁻¹ , 60 th , 0.1 C	4.5 5.5	12 6	This work

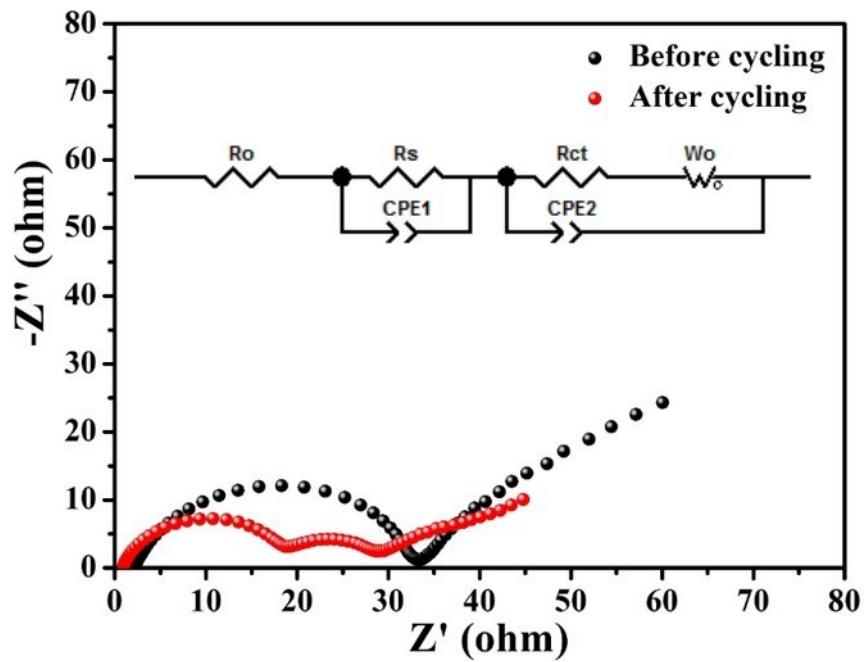


Figure S8. Nyquist plots and corresponding equivalent circuit (insert) of the cells based on “double high” sulfur cathode and GA-CNFs-Ni modified separator before and after cycling.

Table. S3. The parameters of Nyquist plot of the cells based on “double high” sulfur cathode and GA-CNFs-Ni modified separator before and after cycling

Samples	R_o (Ω)	R_s (Ω)	R_{ct} (Ω)
Before cycling	2.2	30.4	-
After cycling	0.9	18.4	8.2

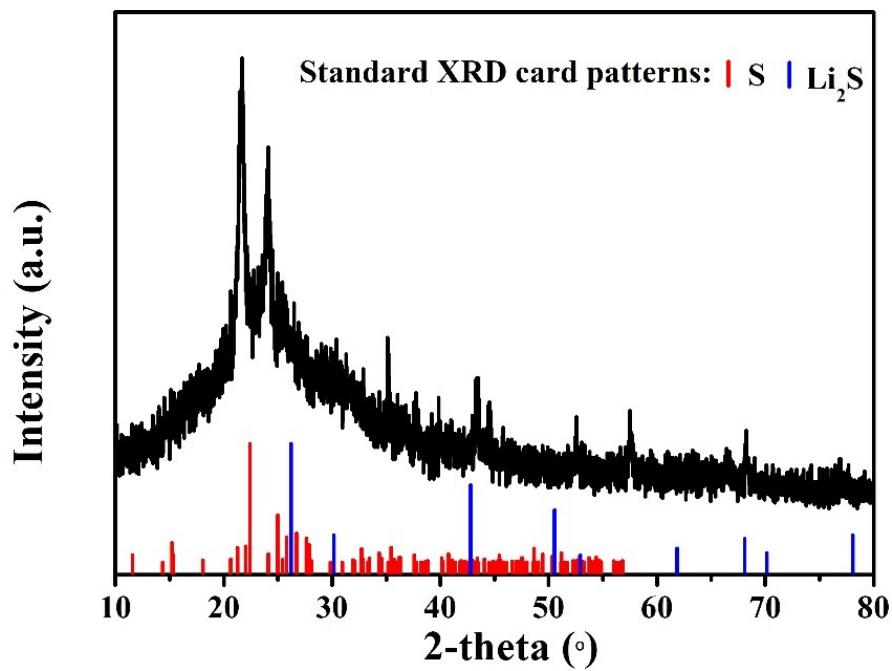


Figure S9. XRD patterns of GA-CNFs-Ni coated separator after cycling and standard XRD patterns sulfur and Li_2S powder.

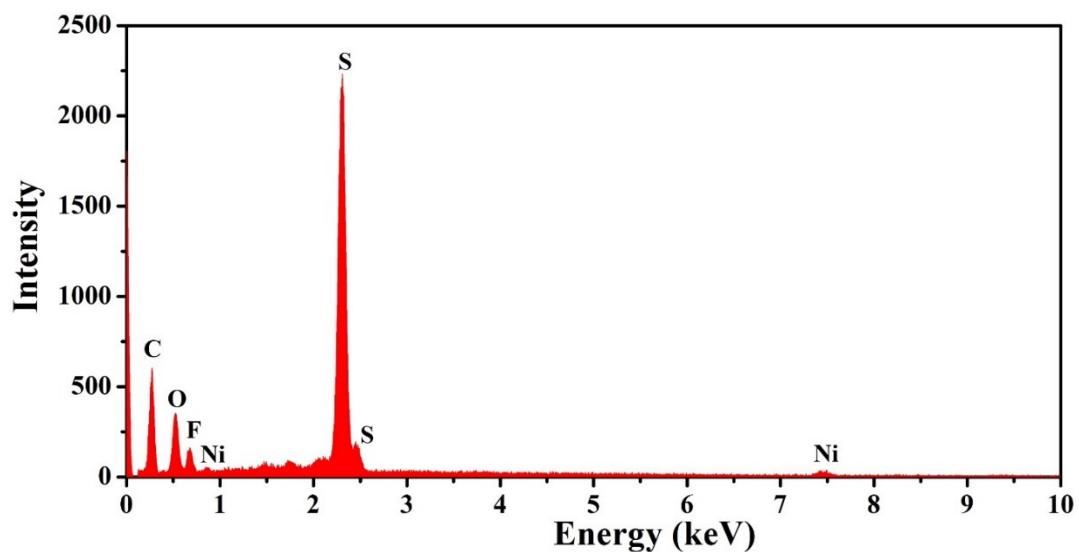


Figure S10. EDX spectra of GA-CNFs-Ni coated separator after 90 cycles at 0.2 C

References

- [S1] L. Y. Luo, X. Y. Qin, J. X. Wu, G. M. Liang, Q. Li, M. Liu, F. Y. Kang, G. H. Chen, B. H. Li, *J. Mater. Chem. A*, 2018, **6**, 8612-8619.
- [S2] D. B. Babu, K. Giribabu, and K. Ramesha, *ACS Appl. Mater. Interfaces*, 2018, **10**, 19721-19729.
- [S3] S. S. Yao, J. C. J. Q. Huang, Z. H. Lu, Y. Deng, W. G. Chong, J. X. Wu, M. Haq, F. Ciucci and J. K. Kim, *Adv. Energy Mater.*, 2018, 1800710.
- [S4] J. J. Song, C. Y. Zhang, X. Gao, J. Q. Zhang, L. Q. Luo, H. Liu, F. Y. Wang, G. X. Wang, *J. Mater. Chem. A*, 2018, **6**, 16610-16616.
- [S5] K. S. Wu, Y. Hu, Z. Shen, R. Z. Chen, X. He, Z. L. Cheng and P. Pan, *J. Mater. Chem. A*, 2018, **6**, 2693-2699.
- [S6] L. Tan, X. H. Li, Z. X. Wang, H. J. Guo, J. X. Wang, *ACS Appl. Mater. Interfaces*, 2018, **10**, 3707-3713.
- [S7] X. Song, S. Q. Wang, G. P. Chen, T. Gao, Y. Bao, L. X. Ding, H. H. Wang, *Chem. Eng. J.*, 2018, 333, 564-571.
- [S8] S. Imtiaz, Z. A. Zafar, R. Razaq, D. Sun, Y. Xin, Q. Li, Z. L. Zhang, L. Zheng, Y. H. Huang and J. A. Anderson, *Adv. Mater. Interfaces*, 2018, 1800243.
- [S9] J. R. He, Y. F. Chen, A. Manthiram, *Energy Environ. Sci.*, DOI:10.1039/c8ee00893k.
- [S10] H. Y. Shao, W. K. Shao, H. Zhang, A. B. Wang X. N. Chen, Y. Q. Huang, *J. Power Sources*, 2018, **378**, 537-545.
- [S11] L. H. Wang, Y. B. He, L. Shen, D. N. Lei, J. M. Ma, H. Ye, K. Shi, B. H. Li, F. Y. Kang, *Nano energy*, 2018, **50**, 367-375.
- [S12] P. Kim, K. Kim, V. Pol, *Carbon*, 2018, **131**, 175-183.
- [S13] S. Y. Liu, C. Y. Fan, Y. H. Shi, H. C. Wang, X. L. Wu, J. P. Zhang, *ACS Appl. Mater. Interfaces*, 2018, **10**, 509-516.

- [S14] G. Gnana Kumar, S. H. Chung, T. Raj Kumar, A. Manthiram, *ACS Appl. Mater. Interfaces*, 2018, **10**, 20627-20634.
- [S15] F. Wu, S. Zhao, L. Chen, Y. Lu, Y. Su, Y. Jia, L. Bao, J. Wang, S. Chen, R. Chen, *Energy Storage Mater.*, 2018, **14**, 383-391.
- [S16] Y. Li, J. Zhu, R. Shi, M. Dirican, P. Zhu, C. Yan, H. Jia, J. Zang, J. He, X. Zhang, *Chem. Eng. J.*, 2018, **349**, 376-387.
- [S17] Y. Huangfu, T. Zheng, K. Zhang, X. She, H. Xu, Z. Fang, K. Xie, *Electrochim. Acta*, 2018, **272**, 60-67.
- [S18] J. Q. Huang, W. G. Chong, Q. B. Zheng, Z. L. Xu, J. Cui, S. S. Yao, C. W. Wang, J. Kim, *Electrochim Acta*, 2018, 268, 1-9.
- [S19] X. X. Chen, X. Y. Ding, C. S. Wang, Z. Y. Feng, L. Q. Xu, X. Gao, Y. J. Zhai and D. B. Wang, *Nanoscale*, DOI:10.1039/c8nr03854f.
- [S20] X. Song, G. P. Chen, S. Q. Wang, Y. P. Huang, Z. Y. Jiang, L. X. Ding, H. H. Wang, *ACS Appl. Mater. Interfaces*, 2018, **10**, 26274-26282.
- [21] T. Y. Lei, W. Chen, W. Q. Lv, J. W. Huang, J. Zhu, J. W. Chu, C. Y. Yan, C. Y. Wu, Y. C. Yan, W. D. He, J. Xiong, Y. R. Li, C. L. Yan, J. B. Goodenough, X. F. Duan, Joule, 2018, doi.org/10.1016/j.joule.2018.07.022.
- [22] M. Li, C. Wang, L. X. Miao, J. W. Xiang, T. Y. Wang, K. Yuan, J. T. Chen, Y. H. Huang, *J. Mater. Chem. A*, 2018, **6**, 5862-5869.
- [23] B. Qi, X. S. Zhao, S. G. Wang, K. Chen, Y. J. Wei, G. Chen, Y. Gao, D. Zhang, Z. H. Sun, F. Li, *J. Mater. Chem. A*, 2018, **6**, 14359-14366.
- [24] S. S. Yao, J. Cui, J. Q. Huang, Z. H. Lu, Y. Deng, W. G. Chong, J. X. Wu, M. Haq, F. Ciucci, J. Kim, *Adv. Energy Mater.*, 2018, 1800710.