

Supplementary information

Direct Storage of Holes in Ultrathin Ni(OH)₂ on Fe₂O₃ Photoelectrodes for Integrated Solar Charging Battery-type Supercapacitors

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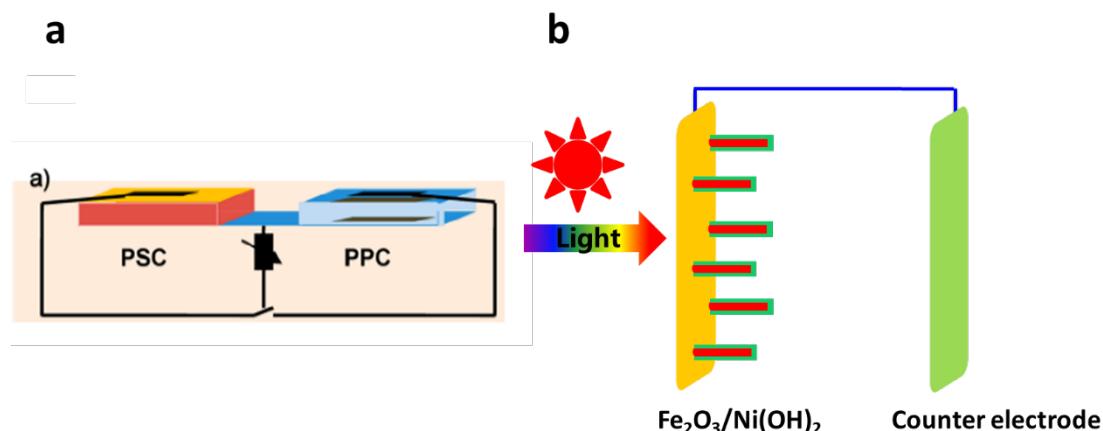


Fig. S1 Diagrammatic sketches of (a) a solar cell + supercapacitor¹ and (b) a photoelectrochemical supercapacitor

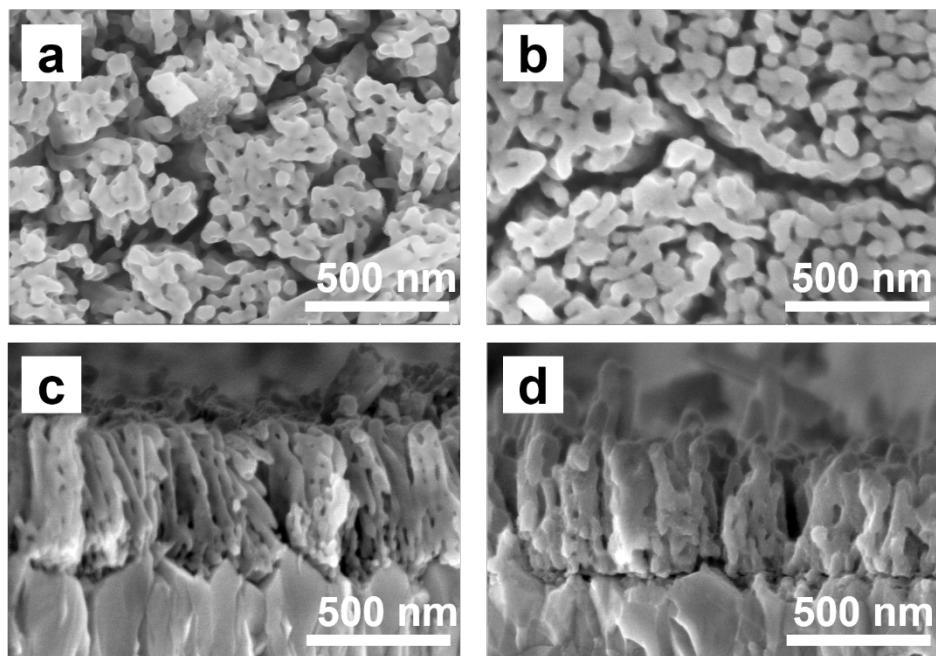


Fig. S2 Surface and cross-section SEM images of bare Fe_2O_3 (a, c) and $\text{Fe}_2\text{O}_3 @ \text{Ni(OH)}_2$ with deposition time of Ni(OH)_2 for 10 min (b, d)

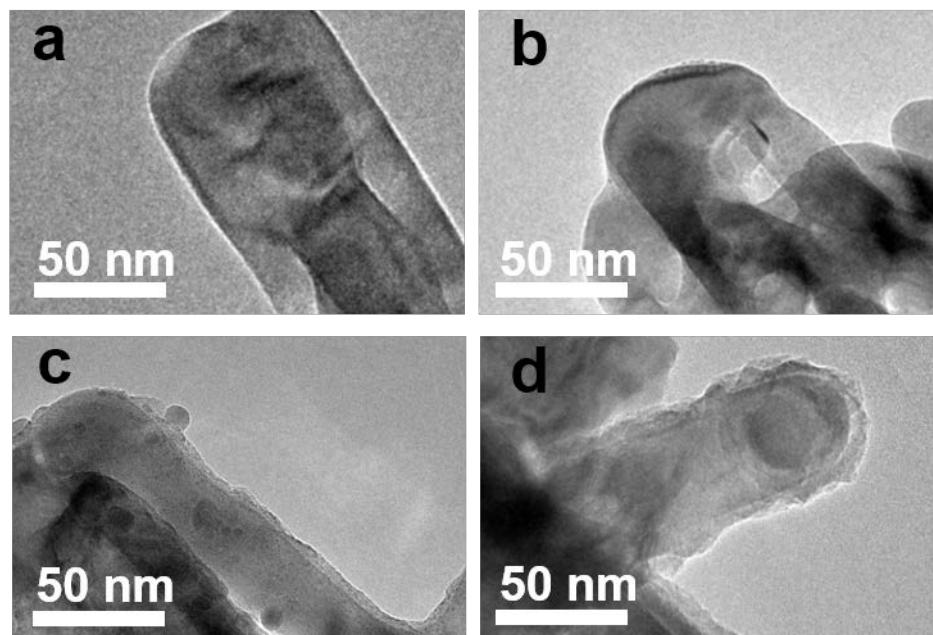


Fig. S3 TEM images of bare Fe_2O_3 (a) and $\text{Fe}_2\text{O}_3 @ \text{Ni(OH)}_2$ with different deposition time of Ni(OH)_2 for 3 min (b), 5 min (c) and 10 min (d).

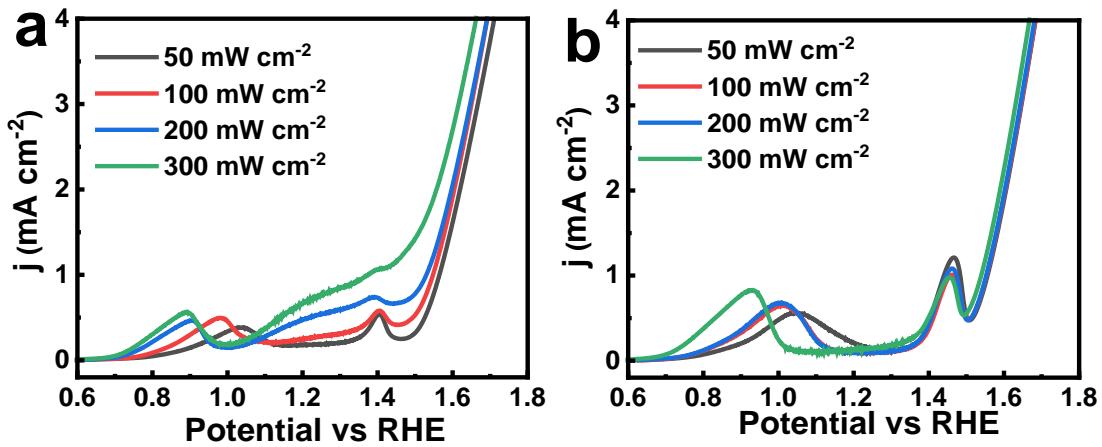


Fig. S4 Photocurrent-potential curves of $\text{Fe}_2\text{O}_3@\text{Ni(OH)}_2$ with deposition time of Ni(OH)_2 for 5 min (a) and 10 min (b) under illumination with different light intensities.

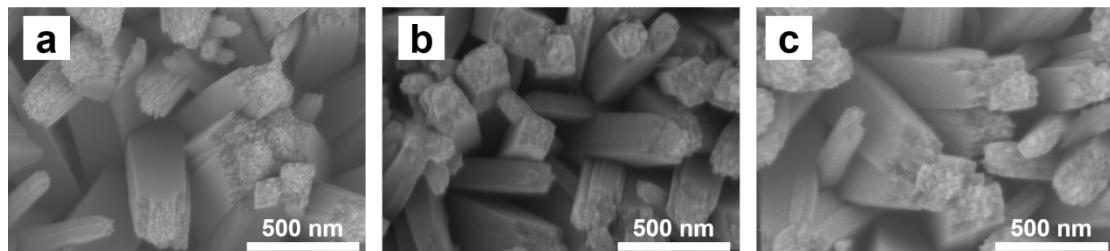


Fig. S5 Surface SEM images of bare TiO_2 (a) and $\text{TiO}_2@(\text{Ni(OH)})_2$ with deposition time of Ni(OH)_2 for 5 min (b) and 10 min (c).

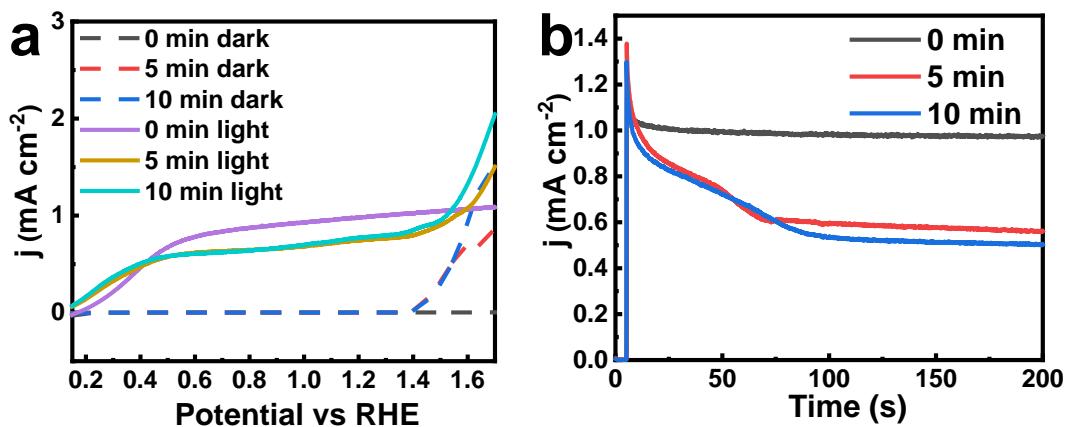


Fig. S6 (a) Dark current-potential and photocurrent-potential curves of bare TiO_2 and $\text{TiO}_2@(\text{Ni(OH)})_2$ with deposition time of Ni(OH)_2 for 5 min and 10 min; (b) I-t curves at 1.1 V_{RHE} of $\text{TiO}_2@(\text{Ni(OH)})_2$ with deposition time of Ni(OH)_2 for 10 min under illumination with different light intensities. Electrolyte: 1M KOH aqueous solution.

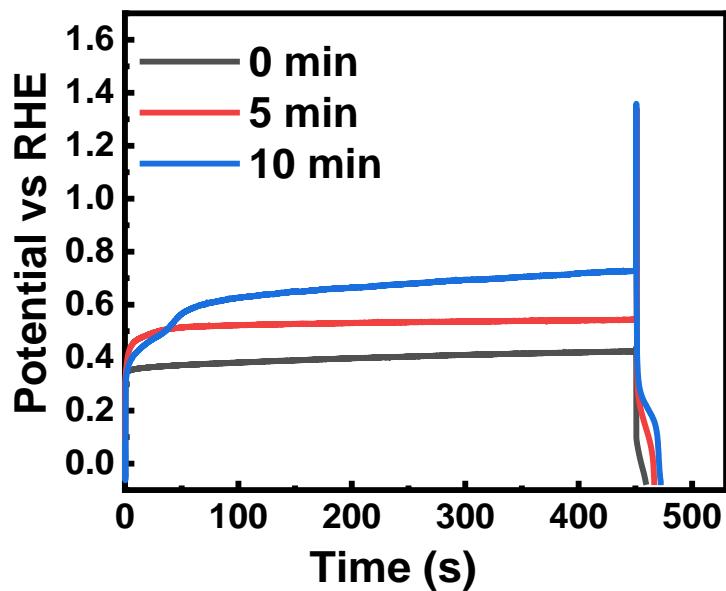


Fig. S7 Galvanostatic photo-charge at 0.6 mA cm^{-2} and dark-discharge curves of bare TiO_2 and $\text{TiO}_2 @ \text{Ni(OH)}_2$ with depositon time of Ni(OH)_2 for 5 min and 10 min

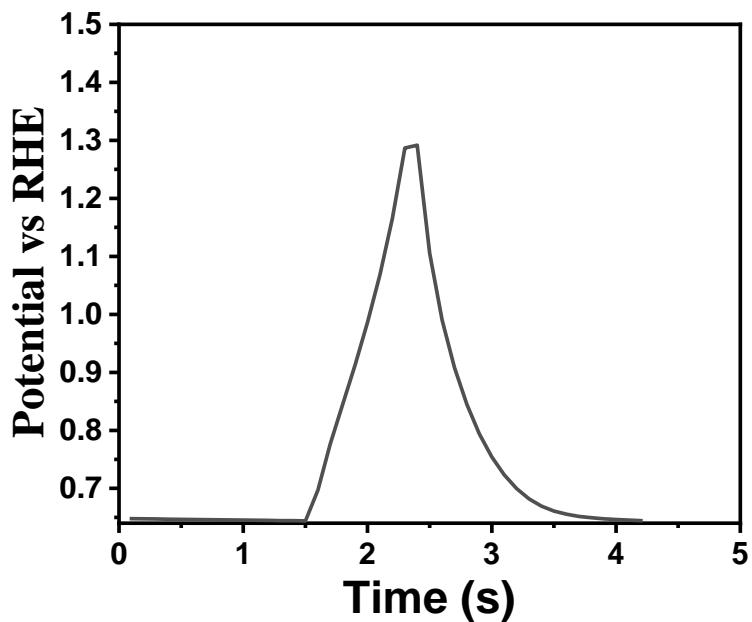


Fig. S8 Galvanostatic dark charge and discharge curves of $\text{Fe}_2\text{O}_3 @ \text{Ni(OH)}_2$ with depositon time of Ni(OH)_2 for 10 min.

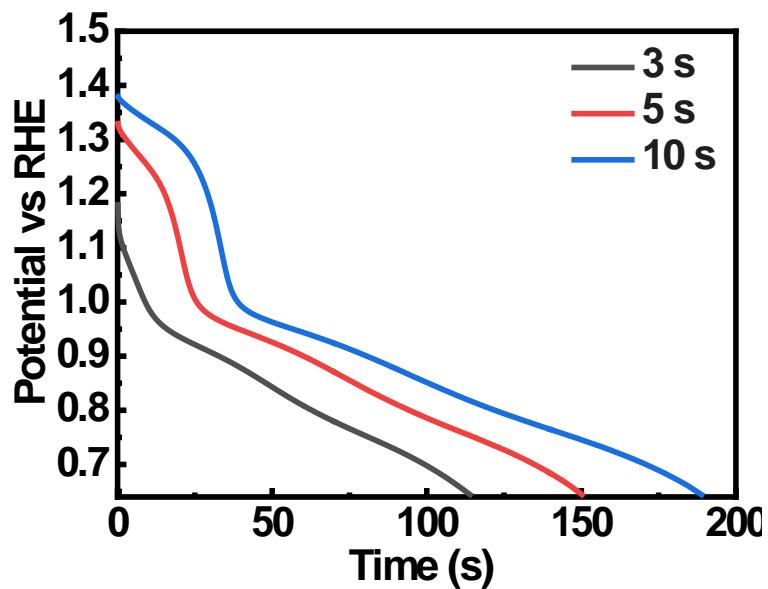


Fig. S9 Discharge curves at current density of 0.1 mA cm^{-2} of $\text{Fe}_2\text{O}_3@\text{Ni(OH)}_2$ with deposition time of Ni(OH)_2 for 10 min after photo-charge at $1.35 \text{ V}_{\text{RHE}}$ for different times.

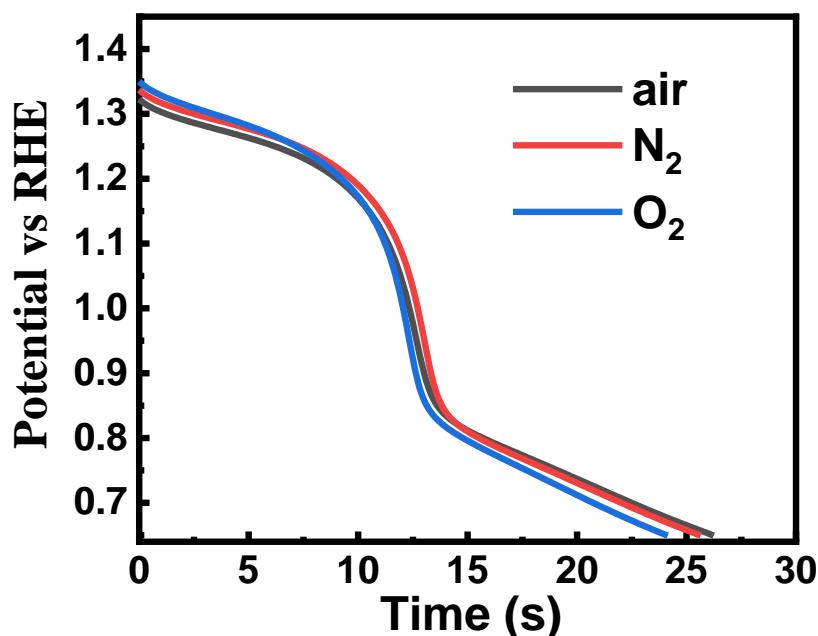


Fig. S10 Discharge curves (0.6 mA cm^{-2}) of $\text{Fe}_2\text{O}_3@\text{Ni(OH)}_2$ with depositon time of Ni(OH)_2 for 10 min at different atmosphere.

Table S1 The performances of different photo-capacitors in this study and literatures

Photoelectrochemical	Specific capacitance	Reference
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supercapacitors		
Fe ₂ O ₃ @ Ni(OH) ₂	20.6 mF cm ⁻²	This work
BiVO ₄ /PbO _x	4.5 mF cm ⁻²	2
PV+ Supercapacitors		
DSSC/TiO ₂ /SC	1.1 mF cm ⁻²	3
PSC/ PEDOT-carbon/SC	11.5 mF cm ⁻²	4
DSSC-SC	3.3 mF cm ⁻²	5
DSSC-fiber SC	41 mF cm ⁻²	6
PSC-SC	28.7 mF cm ⁻²	7

Note: PV: Photovoltaics

PSC: Perovskite Solar Cell

SC: Supercapacitor

DSSC: Dye Sensitized Solar Cells

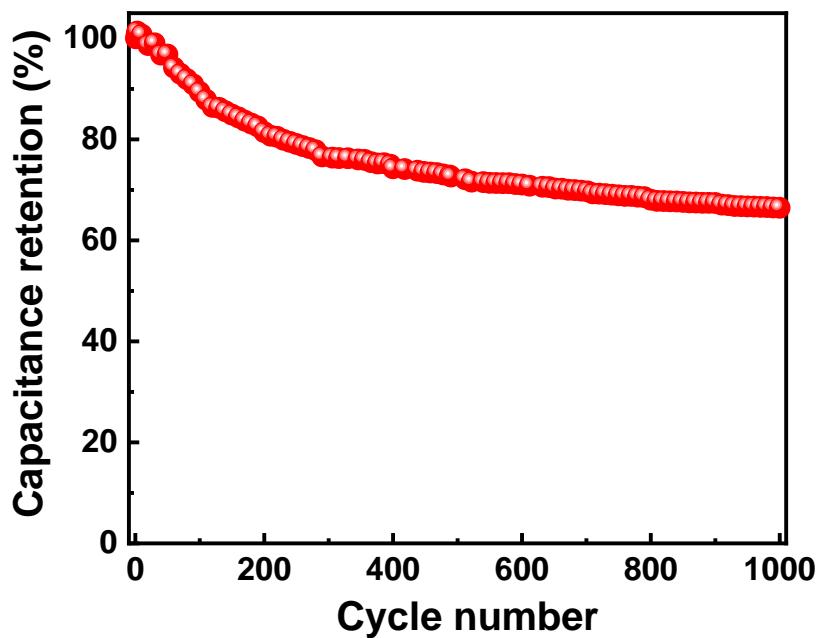


Fig. S11 Cycling capability of Fe₂O₃@Ni(OH)₂ with depositon time of Ni(OH)₂ for 10 min at 50 mV s⁻¹.

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