

Electrochemical Reduction of CO₂ by Single Atom Catalyst TM-TCNQ Monolayers

Supplemental Information

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Table. S1 Lattice constant a and b of TM–TCNQ in Å, where TM are the metal atoms of the first transition metal series.

TM–TCNQ	Lattice (a)	Lattice (b)
Sc–TCNQ	7.2383	11.6652
Ti–TCNQ	7.1576	11.5513
V–TCNQ	7.1767	11.4626
Cr–TCNQ	7.1495	11.4075
Mn–TCNQ	6.9973	11.3408
Fe–TCNQ	7.1285	11.2672
Co–TCNQ	6.9116	11.1595
Ni–TCNQ	7.1073	11.2714
Cu–TCNQ	6.8897	11.1371
Zn–TCNQ	7.0104	11.2765

Table. S2 E_c is the cohesive energy of the bulk TM, E_b is the binding energy between the TM and the TM–TCNQ, and E_f is the formation energy of TM–TCNQ, where TM are the metal atoms of the first transition metal series.

TM–TCNQ	E_b / eV	E_c / eV	E_f / eV
Sc	−9.194	−4.518	−3.148
Ti	−8.596	−5.282	−2.523
V	−8.317	−5.632	−2.117
Cr	−6.907	−4.547	−0.848
Mn	−6.741	−4.247	−0.630
Fe	−6.815	−5.283	−1.013
Co	−7.463	−5.633	−1.320
Ni	−6.312	−4.645	0.023
Cu	−5.088	−3.857	0.859
Zn	−3.665	−1.402	2.461

¹ E_b is binding energy of metal atoms on the TCNQ, which is calculated by: $E_b = E_{\text{TM-TCNQ}} - E_{\text{TM}} - E_{\text{TCNQ}}$, where the $E_{\text{TM-TCNQ}}$, E_{TM} and E_{TCNQ} are the energies of TM–TCNQ, TCNQ and metal atoms.

² E_c is the cohesive energy of metals, which is calculated by $E_c = (E_{\text{M(bulk)}} - nE_{\text{M}}) / n$, where the $E_{\text{M(bulk)}}$ is the energy of metal crystal, E_{M} is the energy of single metal atoms and n is the number of metal atoms in the crystal.

³ E_f is the formation energy of TM–TCNQ, which is calculated by $E_f = E_{\text{TM-TCNQ}} - n_{\text{TM}}\mu_{\text{TM}} - n_{\text{C}}\mu_{\text{C}} - n_{\text{N}}\mu_{\text{N}} - n_{\text{H}}\mu_{\text{H}}$, where the $E_{\text{TM-TCNQ}}$ are the energies of TM–TCNQ, n_{TM} , n_{C} , n_{N} and n_{H} are the number of TM, C, N and H, respectively. μ_{TM} , μ_{C} , μ_{N} and μ_{H} are the chemical potential of TM, C, N and H, respectively.

Table S3 Adsorption Energy (E_{ads}/eV) of different CO₂ reduction products and the bond length between metal atom and the carbon or oxygen atom ($R_{\text{M-C/M-O}}/\text{\AA}$).

TM-TCNQ		CO	HCOOH	HCHO	CH ₃ OH	CH ₄
Sc-TCNQ	$R_{\text{Sc-C/Sc-O}}/\text{\AA}$	2.296	2.121	2.011	2.161	2.561
	E_{ads}/eV	-0.731	-1.002	-1.171	-1.141	-0.306
Ti-TCNQ	$R_{\text{Ti-C/Ti-O}}/\text{\AA}$	2.066	2.017	1.916	2.089	2.571
	E_{ads}/eV	-1.397	-1.181	-1.466	-1.291	-0.408
V-TCNQ	$R_{\text{V-C/V-O}}/\text{\AA}$	1.923	2.026	1.816	2.066	3.126
	E_{ads}/eV	-1.165	-0.481	-0.615	-0.582	-0.161
Cr-TCNQ	$R_{\text{Cr-C/Cr-O}}/\text{\AA}$	1.817	2.325	1.906	2.044	3.185
	E_{ads}/eV	-0.824	-0.336	0.029	0.147	-0.175
Mn-TCNQ	$R_{\text{Mn-C/Mn-O}}/\text{\AA}$	1.747	1.982	1.894	2.029	3.111
	E_{ads}/eV	-1.204	0.172	0.155	0.060	-0.200
Fe-TCNQ	$R_{\text{Fe-C/Fe-O}}/\text{\AA}$	1.736	2.083	1.969	2.249	3.086
	E_{ads}/eV	-1.259	-0.053	-0.172	-0.552	-0.197
Co-TCNQ	$R_{\text{Co-C/Co-O}}/\text{\AA}$	1.796	2.182	2.154	2.180	3.345
	E_{ads}/eV	-0.716	-0.205	-0.254	-0.346	-0.153
Ni-TCNQ	$R_{\text{Ni-C/Ni-O}}/\text{\AA}$	1.844	2.345	2.296	2.219	3.199
	E_{ads}/eV	-0.701	-0.034	-0.044	-0.390	-0.265
Cu-TCNQ	$R_{\text{Cu-C/Cu-O}}/\text{\AA}$	1.845	3.012	2.413	2.315	3.218
	E_{ads}/eV	-0.674	-0.418	-0.263	-0.492	-0.148
Zn-TCNQ	$R_{\text{Zn-C/Zn-O}}/\text{\AA}$	2.229	2.128	2.188	2.126	3.251
	E_{ads}/eV	-0.398	-0.589	-0.723	-0.876	-0.466

Table S4 Gibbs free energy change ($\Delta G/\text{eV}$) of the first protonation step in the CO_2 reduction reaction (CRR) and H_2 evolution reaction (HER) on the TM–TCNQ

M–TCNQ	ΔG_{H}	$\Delta G_{\text{C}^*\text{OOH}}$	$\Delta G_{\text{O}^*\text{CHO}}$
Sc–TCNQ	0.074	–0.662	–1.745
Ti–TCNQ	–0.383	–0.711	–1.848
V–TCNQ	0.319	–0.043	–0.657
Cr–TCNQ	0.982	0.694	–0.138
Mn–TCNQ	0.594	0.260	0.373
Fe–TCNQ	0.196	–0.133	0.284
Co–TCNQ	0.465	0.203	0.630
Ni–TCNQ	1.233	0.673	0.452
Cu–TCNQ	1.239	0.812	0.520
Zn–TCNQ	0.640	0.501	–0.226

Table S5. The calculated free energy corrections (G_c) for adsorbates. The temperature of the reaction is 298.15 K.

Sc-TCNQ	E_{ZPE} (kcal/mol)	C_v (cal/mol·k)	S (cal/mol·k)	H (kcal/mol)	G_c (kcal/mol)
C*OOH	95.012	67.159	82.663	106.358	81.712
O*CHO	93.307	63.730	77.835	104.067	80.861
C*O	82.834	60.512	67.753	92.840	72.639
O*CHOH	99.777	67.981	88.448	111.406	85.035
O*CH	88.157	60.400	64.537	98.013	78.771
C*OH	89.022	60.396	71.725	98.896	77.511
C*HO	91.066	61.861	72.603	101.303	79.656
O*CH₂	96.242	61.033	64.620	106.139	86.873
O*CH₃	104.853	60.964	68.092	114.889	94.587
O*HCH₂	105.645	66.893	73.019	116.653	94.883
CH₃O*H	112.077	66.799	78.273	123.065	99.728
O*	81.309	57.304	66.144	90.805	71.084
O*H	86.008	57.735	74.513	95.516	73.300

Ti-TCNQ	E_{ZPE} (kcal/mol)	C_v (cal/mol·k)	S (cal/mol·k)	H (kcal/mol)	G_c (kcal/mol)
C*OOH	90.367	61.177	62.787	100.130	81.410
O*CHO	93.533	65.108	73.922	104.348	82.308
C*O	83.448	61.596	72.328	93.509	71.945
O*CHOH	100.125	75.017	65.713	110.918	88.552
O*CH	88.786	65.723	77.425	99.777	76.693
C*OH	90.367	61.177	62.787	100.130	81.410
C*HO	88.856	61.489	75.666	99.094	76.534
O*CH₂	96.383	64.916	71.147	107.022	85.809
O*CH₃	105.123	64.391	69.231	115.616	94.975
O*HCH₂	105.758	64.170	60.985	115.655	97.472
CH₃O*H	111.800	66.904	77.823	122.778	99.575
O*	81.231	57.303	68.903	90.630	70.086
O*H	86.755	60.720	62.527	96.499	77.857

V-TCNQ	E_{ZPE} (kcal/mol)	C_v (cal/mol·k)	S (cal/mol·k)	H (kcal/mol)	G_c (kcal/mol)
C*OOH	93.637	65.571	69.870	104.084	83.252
O*CHO	94.191	64.228	63.902	104.316	85.264
C*O	84.521	60.218	58.360	93.852	76.452
O*CHOH	101.098	66.501	69.221	111.712	91.074
O*CH	90.172	62.287	61.091	99.943	81.729
C*OH	90.912	60.364	57.834	100.232	82.989
C*HO	90.191	61.992	62.508	99.961	81.324
O*CH₂	97.013	61.860	60.557	106.644	88.588
O*CH₃	106.563	64.869	63.476	116.772	97.847
O*HCH₂	104.461	67.899	73.868	115.275	93.251
CH₃O*H	112.985	67.668	71.232	123.859	102.621
O*	81.937	56.058	52.112	90.495	74.957
O*H	87.668	59.232	59.898	96.801	78.942

Cr-TCNQ	E_{ZPE} (kcal/mol)	C_v (cal/mol·k)	S (cal/mol·k)	H (kcal/mol)	G_c (kcal/mol)
C*OOH	94.103	62.405	56.745	103.390	86.472
O*CHO	93.785	64.667	67.153	104.142	84.121
C*O	85.115	60.845	58.981	94.461	76.875
O*CHOH	100.687	67.161	84.439	111.976	86.800
O*CH	89.497	62.017	59.128	99.049	81.420
C*OH	90.697	61.947	63.239	100.380	81.525
C*HO	90.651	60.909	55.599	99.869	83.292
O*CH₂	97.997	64.306	62.956	107.955	89.185
O*CH₃	104.988	63.340	58.901	114.644	97.082
O*HCH₂	104.170	67.553	68.644	114.767	94.301
CH₃O*H	112.759	67.010	66.686	123.231	103.349
O*	81.540	57.858	55.921	90.476	73.803
O*H	87.237	57.734	54.465	96.077	79.838

Mn-TCNQ	E_{ZPE} (kcal/mol)	C_v (cal/mol·k)	S (cal/mol·k)	H (kcal/mol)	G_c (kcal/mol)
C*OOH	94.553	65.746	64.537	104.628	85.386
O*CHO	94.304	65.417	66.121	104.572	84.858
C*O	85.553	60.237	55.114	94.610	78.178
O*CHOH	102.230	66.944	64.281	112.553	93.387
O*CH	90.381	62.969	58.699	99.968	82.467
C*OH	91.763	62.702	59.628	101.404	83.626
C*HO	91.097	62.089	58.038	100.517	83.213
O*CH₂	98.677	63.629	58.542	108.306	90.852
O*CH₃	105.209	64.950	61.626	115.186	96.812
O*HCH₂	103.776	65.792	64.477	113.970	94.746
CH₃O*H	113.673	66.140	61.609	123.767	105.398
O*	81.823	57.622	51.987	90.515	75.016
O*H	87.436	58.906	53.730	96.328	80.308

Fe-TCNQ	E_{ZPE} (kcal/mol)	C_v (cal/mol·k)	S (cal/mol·k)	H (kcal/mol)	G_c (kcal/mol)
C*OOH	94.452	64.177	59.742	104.080	86.268
O*CHO	91.742	64.217	56.472	101.306	84.469
C*O	84.933	59.339	52.859	93.805	78.045
O*CHOH	101.154	63.939	62.874	110.913	92.167
O*CH	90.314	63.401	59.598	100.015	82.246
C*OH	90.335	59.908	54.599	99.295	83.016
C*HO	91.019	60.542	53.345	99.994	84.089
O*CH₂	97.835	62.937	57.124	107.370	90.339
O*CH₃	105.610	64.744	63.114	115.577	96.759
O*HCH₂	103.047	67.178	69.916	113.828	92.983
CH₃O*H	112.481	65.557	65.944	122.705	103.044
O*	81.247	54.307	47.751	89.335	75.098
O*H	87.520	56.932	48.843	95.865	81.303

Co-TCNQ	E_{ZPE} (kcal/mol)	C_v (cal/mol·k)	S (cal/mol·k)	H (kcal/mol)	G_c (kcal/mol)
C*OOH	94.517	65.011	64.671	104.678	85.397
O*CHO	93.977	64.920	63.530	104.187	85.246
C*O	85.140	62.080	61.440	94.955	76.637
O*CHOH	101.016	67.373	70.805	111.931	90.820
O*CH	89.363	63.238	62.435	99.307	80.692
C*OH	90.911	64.063	65.844	101.133	81.502
C*HO	91.096	61.701	59.308	100.670	82.987
O*CH₂	97.648	75.542	66.234	108.491	85.968
O*CH₃	105.149	65.819	68.838	115.688	95.164
O*HCH₂	104.054	63.721	73.897	114.311	92.279
CH₃O*H	113.195	67.985	71.276	124.197	102.946
O*	81.473	59.231	66.730	91.230	71.334
O*H	88.360	61.020	67.973	97.985	77.719

Ni-TCNQ	E_{ZPE} (kcal/mol)	C_v (cal/mol·k)	S (cal/mol·k)	H (kcal/mol)	G_c (kcal/mol)
C*OOH	93.835	65.973	74.408	104.609	82.424
O*CHO	94.057	65.557	74.086	104.856	82.767
C*O	84.753	63.115	70.325	95.213	74.245
O*CHOH	100.592	61.990	64.156	110.587	91.459
O*CH	88.676	64.205	73.445	99.370	77.473
C*OH	89.626	63.620	78.525	100.233	76.821
C*HO	90.656	60.443	64.869	100.365	81.024
O*CH₂	95.725	58.228	66.901	105.128	85.181
O*CH₃	104.635	63.063	66.914	114.823	94.873
O*HCH₂	104.625	66.325	78.214	115.779	92.460
CH₃O*H	112.331	67.061	84.552	123.616	98.407
O*	80.893	60.064	71.612	91.043	69.692
O*H	87.288	56.744	55.525	96.181	79.627

Cu-TCNQ	E_{ZPE} (kcal/mol)	C_v (cal/mol·k)	S (cal/mol·k)	H (kcal/mol)	G_c (kcal/mol)
C*OOH	93.120	65.348	79.912	104.247	80.421
O*CHO	93.183	66.367	81.747	104.653	80.280
C*O	83.337	60.643	76.609	93.663	70.822
O*CHOH	100.381	65.851	74.829	111.477	89.167
O*CH	87.286	54.159	56.428	96.093	79.269
C*OH	88.759	62.705	79.735	99.459	75.686
C*HO	89.153	59.787	72.039	99.141	77.663
O*CH₂	97.034	59.563	68.791	106.982	86.472
O*CH₃	104.225	67.329	78.029	115.740	92.475
O*HCH₂	103.925	64.649	72.819	114.816	93.105
CH₃O*H	111.502	63.682	71.610	122.177	100.827
O*	79.797	54.980	68.151	89.093	68.774
O*H	86.514	59.617	68.068	96.503	76.208

Zn-TCNQ	E_{ZPE} (kcal/mol)	C_v (cal/mol·k)	S (cal/mol·k)	H (kcal/mol)	G_c (kcal/mol)
C*OOH	93.607	62.877	71.504	103.916	82.597
O*CHO	94.190	63.783	79.125	104.956	81.365
C*O	84.691	57.984	62.890	94.231	75.481
O*CHOH	100.865	60.002	71.461	110.636	89.330
O*CH	88.681	64.520	75.193	99.692	77.273
C*OH	90.408	61.709	72.345	100.715	79.146
C*HO	89.751	61.300	72.053	100.006	78.524
O*CH₂	98.015	64.230	77.397	108.882	85.806
O*CH₃	105.041	64.743	73.954	115.960	93.911
O*HCH₂	103.396	65.954	72.684	114.420	92.749
CH₃O*H	113.534	68.190	87.387	125.153	99.099
O*	80.666	58.606	73.992	90.680	68.619
O*H	87.409	57.059	66.524	96.742	76.907