

Supplementary Information

PAF-1@Cellulose Nanofibril Composite Aerogel for Highly-efficient Removal of Bisphenol A

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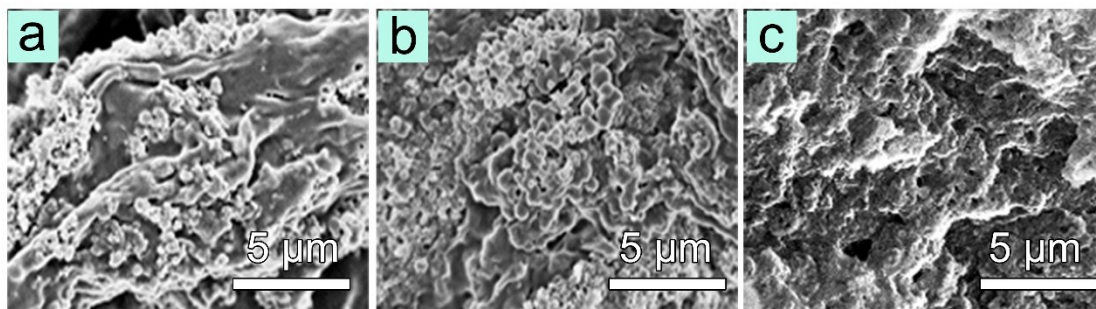


Fig. S1 SEM images of PAF-1@CNF composite aerogels with the PAF-1 content of (a) 20 wt%, (b) 33.3%, and (c) 50 wt%.

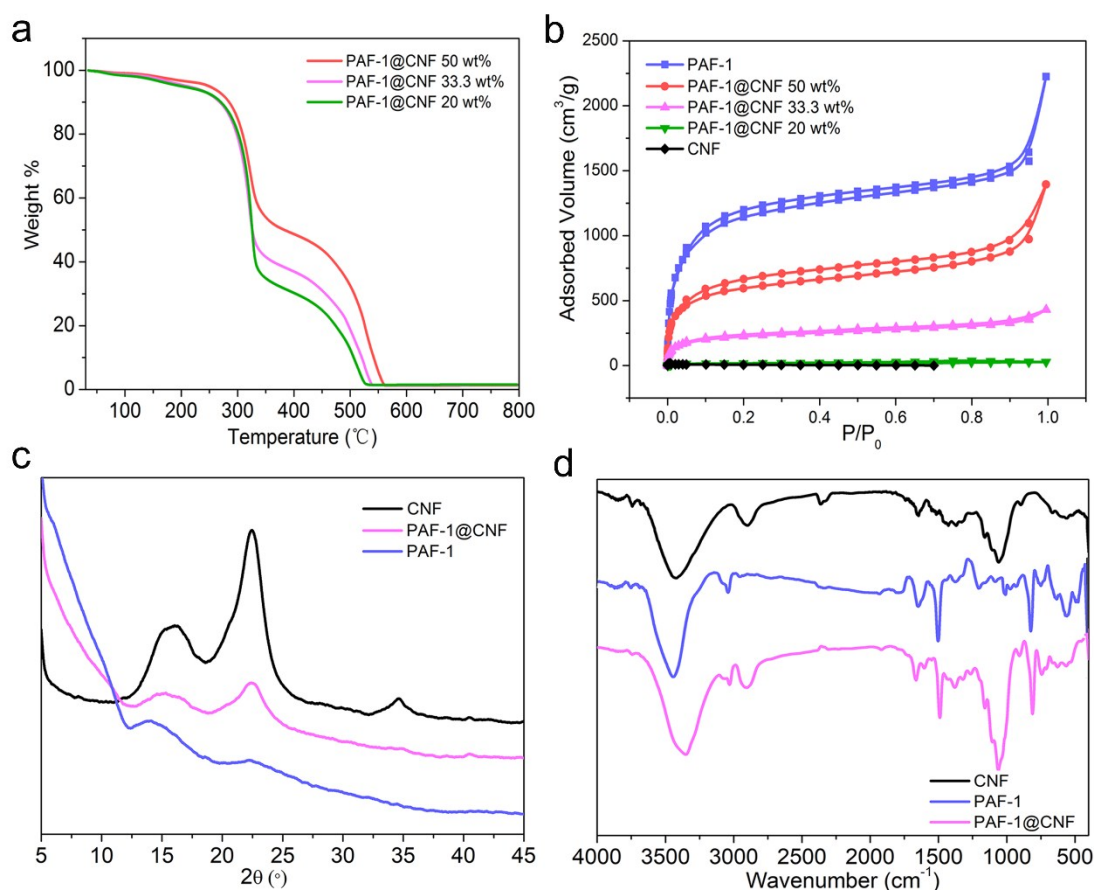


Fig. S2 (a) TGA curves for PAF-1@CNF composite aerogels with PAF-1 content of 20, 33.3, and 50 wt%. (b) Nitrogen adsorption-desorption isotherms of double-cross-linked CNF aerogel, PAF-1, and PAF-1@CNF composites with the PAF-1 content of 20, 33.3, and 50 wt%. (c) XRD patterns of PAF-1, physically cross-linked CNF aerogel, and PAF-1@CNF composite aerogel with the PAF-1 content of 33.3 wt%. (d) FTIR spectra of PAF-1, physically cross-linked CNF aerogel, and PAF-1@CNF composite aerogel with PAF-1 content of 33.3 wt%.

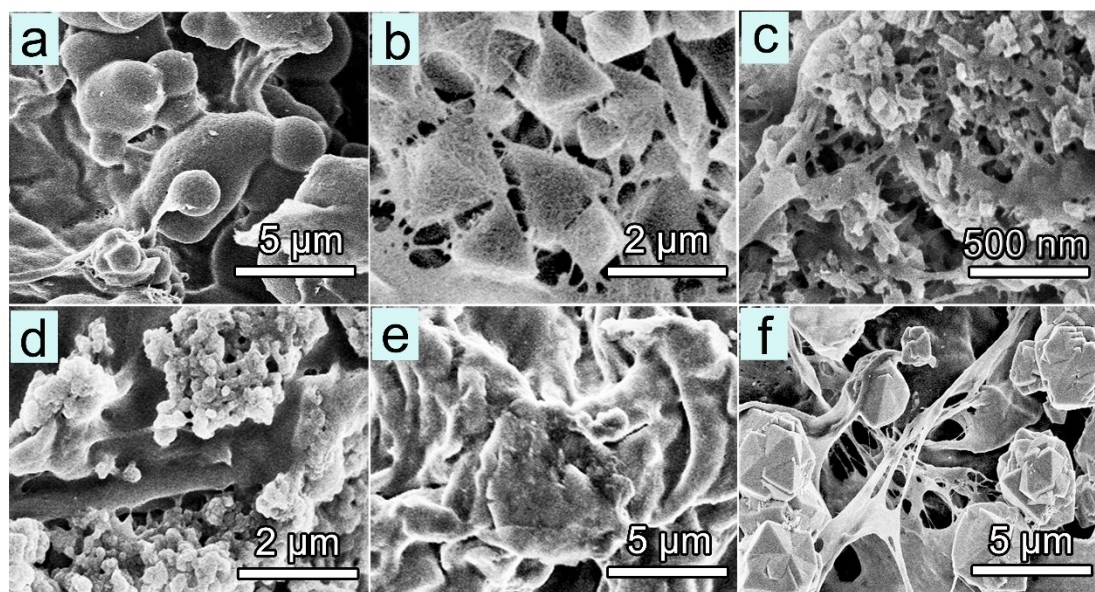


Fig. S3 SEM images of CNFs-based composite aerogels loaded with (a) PAF-45, (b) MIL-101, (c) MIL-53, (d) PIM-1, (e) activated carbon, and (f) zeolite-X.

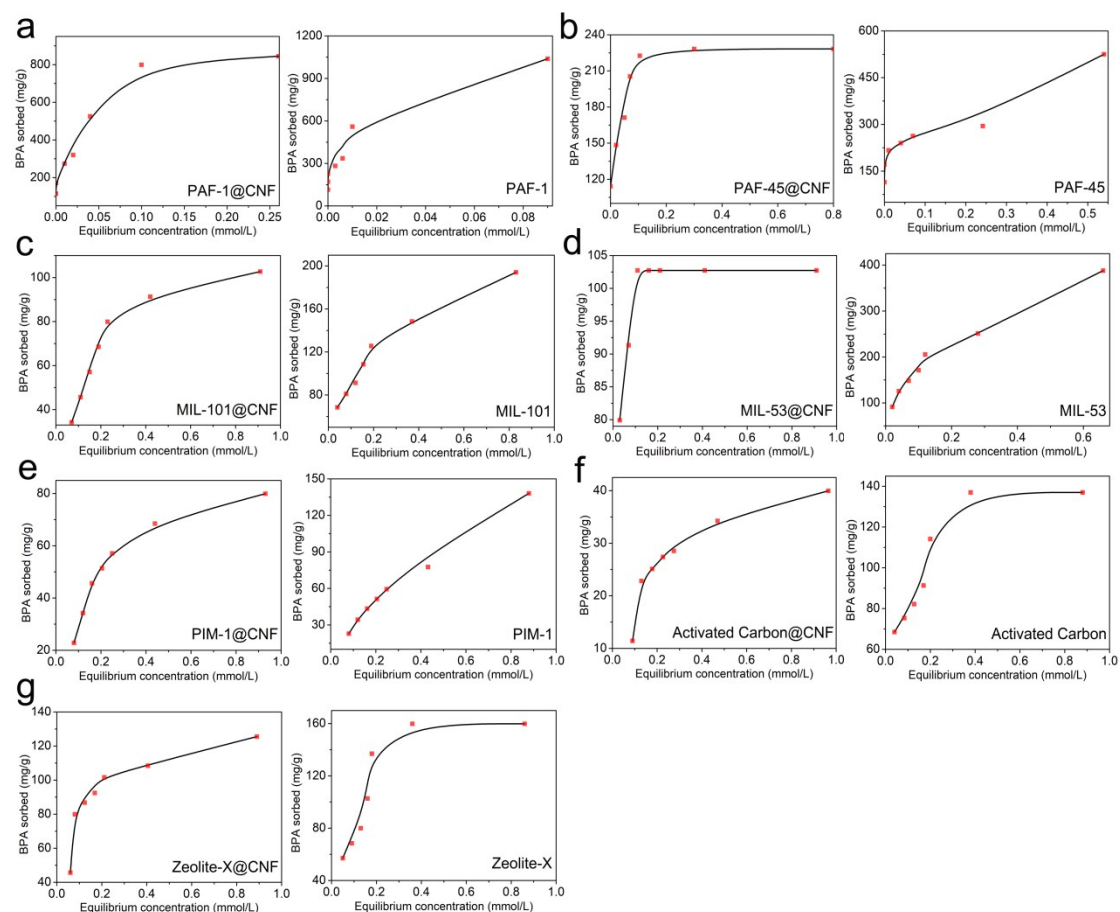


Fig. S4 Plot of the adsorption equilibrium uptake of CNF composite aerogels loaded with (a) PAF-1, (b) PAF-45, (c) MIL-101, (d) MIL-53, (e) PIM-1, (f) activated carbon, and (g) zeolite-X composite aerogels (left panels) and their corresponding powders (right panels) versus the initial concentration of BPA in water.

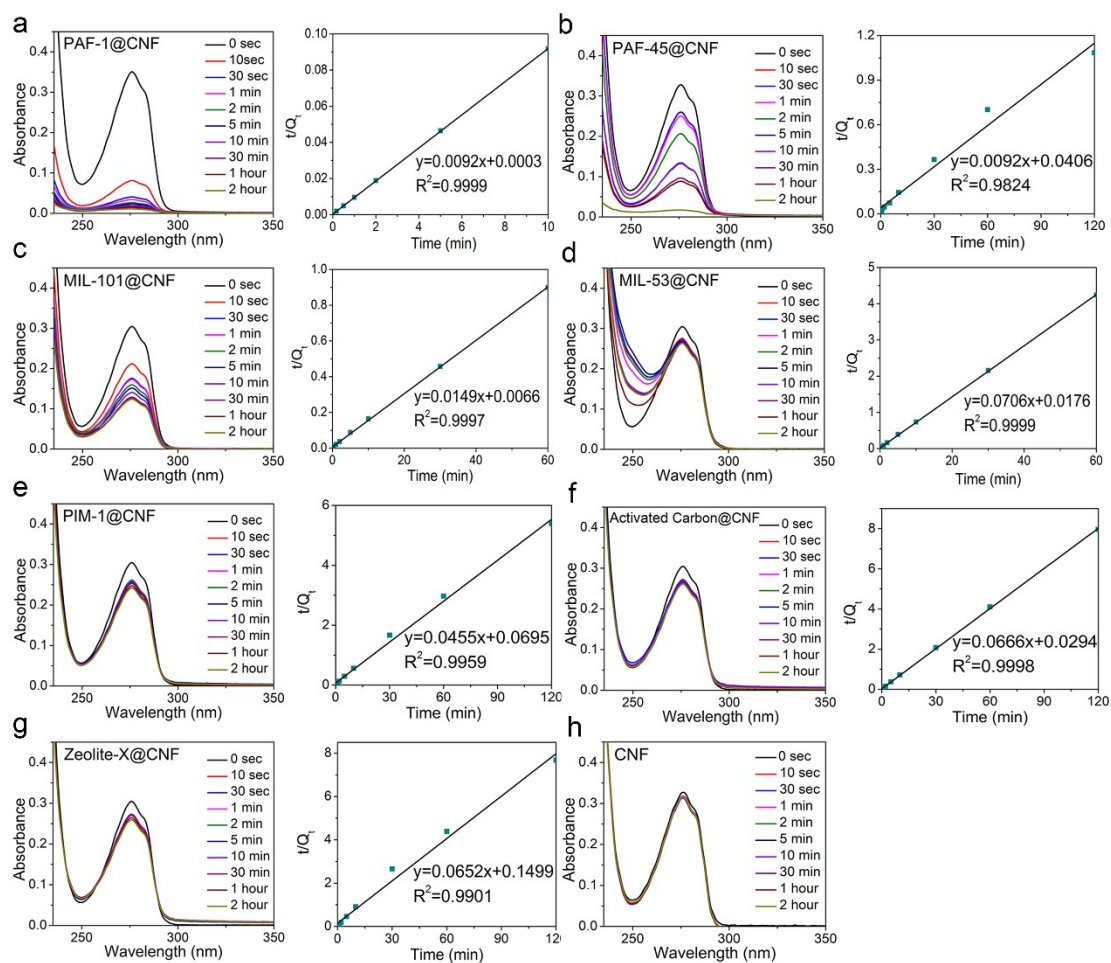


Fig. S5 Temporal evolution of the UV-vis spectra of the aqueous solution of BPA during adsorption of CNF composite aerogels loaded with (a) PAF-1, (b) PAF-45, (c) MIL-101, (d) MIL-53, (e) PIM-1, (f) activated carbon, and (g) zeolite-X, and (h) pristine CNF aerogel (left panels) and the pseudo-second-order plots (right panels) of the corresponding CNF composite.

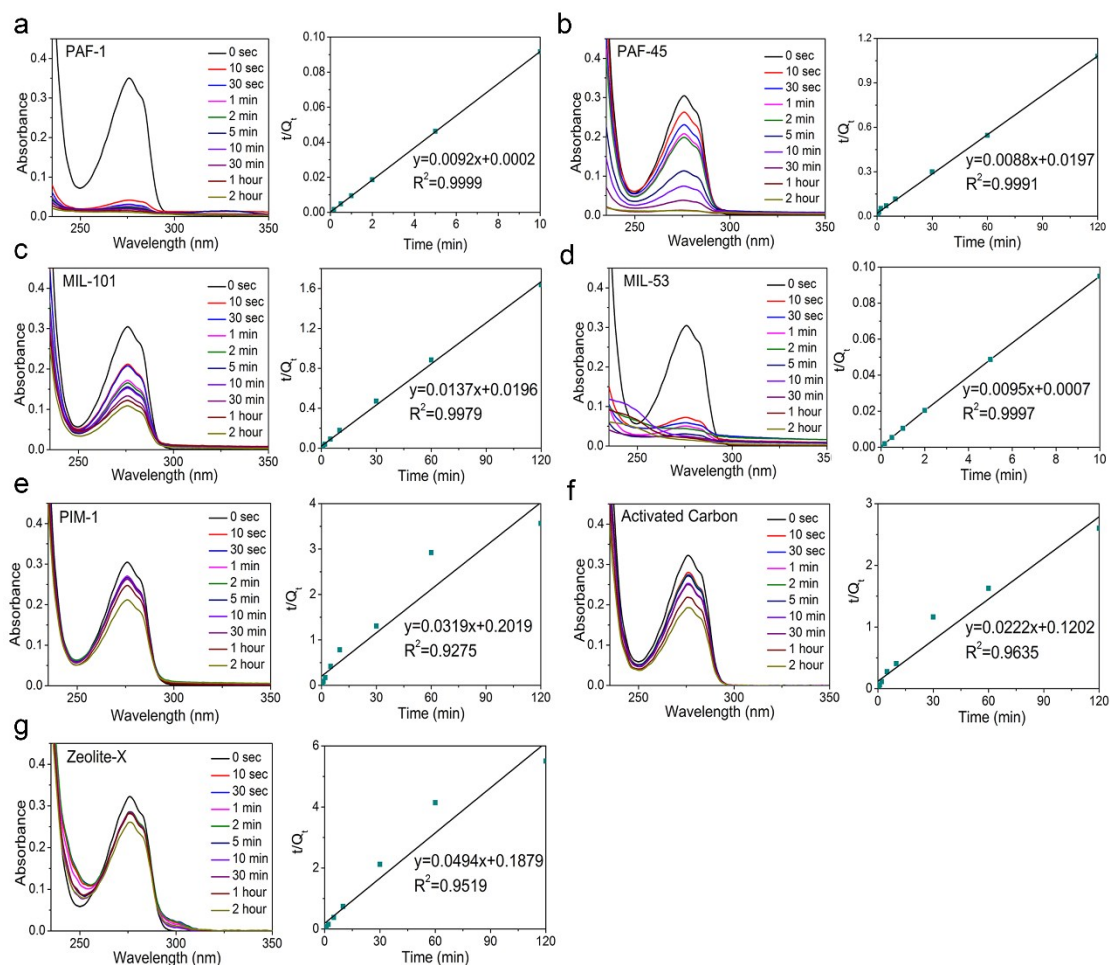
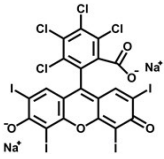
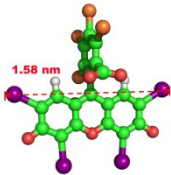
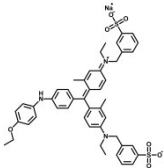
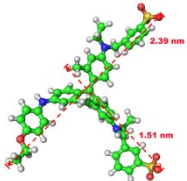
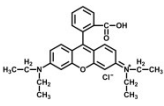
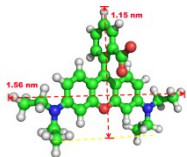
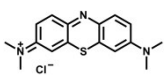
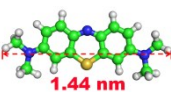
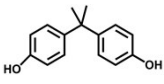
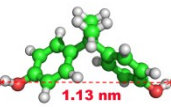


Fig. S6 Temporal evolution of the UV-vis spectra of the aqueous solution of BPA during adsorption of the powders of (a) PAF-1, (b) PAF-45, (c) MIL-101, (d) MIL-53, (e) PIM-1, (f) activated carbon, and (g) zeolite-X (left panels) and the pseudo-second-order plots (right panels) of the corresponding porous material powder.

Table S1 BET surface areas, maximum sorption capacities and BPA removal performance of CNF composite aerogels loaded with PAF-1, PAF-45, MIL-101, MIL-53, PIM-1, activated carbon, and zeolite-X and their corresponding porous materials.

Adsorbent	BET surface area (m ² g ⁻¹)	Maximum sorption capacity (mg g ⁻¹)	%Removal efficiency		Time to reach equilibrium (min)	K_{obs} (g/mg min)	Correlation coefficient R ²
			in 10 sec	in 1 h			
PAF-1@CNF	2203	1000	77.7	97.0	10	0.282	0.9999
PAF-1	4308	1428.6	89.2	97.0	10	0.423	0.9999
PAF-45@CNF	523	238.1	20.0	75.0	120	0.002	0.9824
PAF-45	1045	322.6	12.2	96.3	120	0.004	0.9991
MIL-101@CNF	2508	140.8	29.6	58.5	60	0.034	0.9997
MIL-101	2883	161.3	29.3	59.5	>120	0.010	0.9979
MIL-53@CNF	47	109.9	8.5	12.4	60	0.283	0.9999
MIL-53	1035	277.8	76.4	93.7	10	0.129	0.9997
PIM-1@CNF	17	135.1	13.1	17.7	120	0.030	0.9959
PIM-1	732	232.6	10.2	18.0	>120	0.005	0.9275
Activated carbon@CNF	30	65.4	9.5	12.8	120	0.151	0.9998
Activated carbon	704	125	12.9	32.3	>120	0.004	0.9635
Zeolite-X@CNF	35	144.9	9.2	12.0	>120	0.028	0.9901
Zeolite-X	672	185.2	11.1	12.7	>120	0.013	0.9519

Table S2 Structure, model, molecular weight, and charge of selected dyes and BPA.

No.	Dye	Structure	Model	M _w (g mol ⁻¹)	Charge
Dye-1	Acid red 94 (AR-94)			1017.63	-2
Dye-2	Brilliant blue G (BB-G)			854.02	-1
Dye-3	Rhodamine B (R-B)			479.1	+1
Dye-4	Methylene blue (MB)			319.85	+1
5	Bisphenol A (BPA)			228.29	

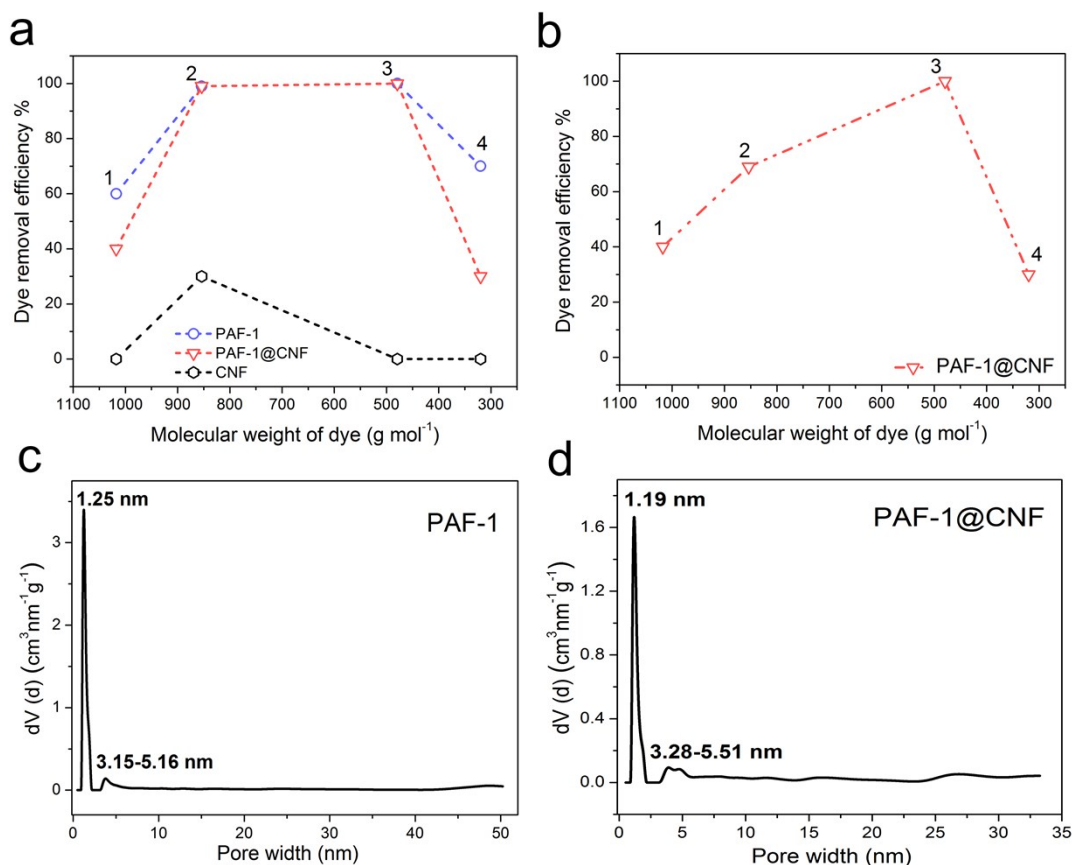


Fig. S7 (a) Dye removal efficiencies of different materials (PAF-1, PAF-1@CNF aerogel, and CNF aerogel) versus the molecular weight of selected dyes in water. (b) Dye removal efficiencies of PAF-1@CNF aerogel deducting CNF versus the molecular weight of dyes in water. Pore size distribution of (c) PAF-1 and (d) PAF-1@CNF aerogel.