Achieving Long-Term Anticorrosion via the Inhibition of

Graphene's Electrical Activity

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Table S1. Chemical composition (wt %) of the Q235 used.										
Element	В	C O Si		Cr	Cr Mn					
Weight%	0.56	5.60	0.45	0.26	0.02	0.43	Bal.			



Figure S1. High-resolution TEM images of GO sheets (a-b); SEM images of GO sheets (d); The size (e) and thickness (f) distributions of the GO sheets.

The single-layered feature of the GO sheets was confirmed by high-resolution transmission electron microscopy (HR-TEM) and scanning electron microscopy (SEM) images as shown in Fig.S1. Under the HR-TEM inspection (Fig. S1a-c), the GO sheets exhibit typical single-layer structures and the selected area electron diffraction patterns (corresponding insets) further reveal their single-layer nature. In SEM images (Fig. S1d), the GO sheets exhibit typical wrinkles, demonstrating their fine flexibility. Based on the statistics from the SEM and TEM images, where a statistical analysis of over 100 flakes shows that > 99 % of the sheets are single layer (<1 nm in thickness) with a lateral size of $8\pm0.5\mu$ m (Fig. S1e, f). These results reveal that large-size and few-layered GO was successfully prepared in our work.



Figure S2. The size distributions of the G (a), BG (b) and NG (c) nanosheets, respectively.

Table S2 . Relative percentages of the elemental composition of the doped graphene materials.									
Sample	С	0	Н	Ν	В	S			
G	97.32	1.94	0.74	0	0	0			
BG	94.71	2.14	1.28	0	1.87	0			
NG	93.19	2.22	2.13	2.46	0	0			

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Figure S3. FTIR spectroscopy of BG (a), NG (b), and G sheets treated at 60 °C.



Figure S4. Zeta potential measurement for low and high temperature sonicated graphene-based materials.



Figure S5. The gas barrier property results for the neat PU and their nanocomposite coating

systems.



Figure S6. Open corrosion potential of neat PU, G/PU, BG/PU, and NG/PU nanocomposite coatings during 240 h immersion in 3.5 wt.% NaCl_(aq).

Table S3. The impedance modulus at the lowest frequency $(Z_{f=0.01 \text{ Hz}})$ for different coatings with different immersion time.

Sample	$Z_{ ext{f=0.01 Hz}} \left(\Omega \text{ cm}^2 \right)$										
	2h	24h	48h	72h	96h	120h	144h	168h	192h	216h	240h
neat PU	6.05×10 ⁹	4.46×10 ⁹	3.93×10 ⁹	2.16×10 ⁹	1.24×10 ⁹	7.33×10 ⁸	5.01×10 ⁸	4.08×10 ⁸	6.38×10 ⁷	3.89×10 ⁷	9.25×10 ⁶
G/PU	8.92×10 ⁹	8.15×10 ⁹	4.76×10 ⁹	4.21×10 ⁹	4.25×10 ⁹	2.91×10 ⁹	1.15×10 ⁹	5.88×10 ⁶	4.05×10 ⁶	5.25×10 ⁶	5.51×10 ⁶
BG/PU	1.29×10 ¹⁰	9.96×10 ⁹	9.58×10 ⁹	9.27×10 ⁹	8.98×10 ⁹	8.80×10 ⁹	8.87×10 ⁹	7.20×10 ⁹	6.69×10 ⁹	6.92×10 ⁹	6.67×10 ⁹
NG/PU	7.2×10 ⁹	418×10 ⁹	3.5×10 ⁹	1.85×10 ⁹	8.97×10 ⁸	7.74×10 ⁸	6.97×10 ⁸	3.85×10 ⁶	3.64×10 ⁶	2.53×10 ⁶	2.15×10 ⁶

Sample	$R_c (\Omega cm^2)$										
	2h	24h	48h	72h	96h	120h	144h	168h	192h	216h	240h
neat PU	2.29×10 ⁸	1.01×10 ⁸	7.19×10 ⁷	7.07×10 ⁷	5.89×10 ⁷	3.25×10 ⁷	1.01×10 ⁷	6.88×10 ⁶	3.32×10 ⁶	9.33×10 ⁵	5.12×10 ⁵
G/PU	3.35×10 ⁹	1.73×10 ⁹	6.54×10 ⁸	5.27×10 ⁸	3.17×10 ⁸	4.22×10 ⁷	4.76×10 ⁵	3.13×10 ⁵	3.17×10 ⁵	2.15×10 ⁵	2.55×10 ⁵
BG/PU	6.82×10 ⁹	4.55×10 ⁹	2.67×10 ⁹	1.86×10 ⁹	2.96×10 ⁹	1.03×10 ⁹	9.21×10 ⁸	7.34×10 ⁸	5.65×10 ⁸	4.76×10 ⁸	2.54×10 ⁸
NG/PU	3.62×10 ⁹	1.02×10 ⁹	5.43×10 ⁸	4.23×10 ⁸	2.83×10 ⁸	1.54×10 ⁷	4.43×10 ⁵	2.76×10 ⁵	2.96×10 ⁵	2.07×10 ⁵	2.13×10 ⁵

Table S4. The fitting R_c results of the collected EIS results of different coatings as a function of immersion time.



Figure S7. Electrical conductivity of the PU nanocomposits coatings.



Figure S8. The XPS spectra and Polarization curves of neat PU and their nanocomposite coatings with different B doping concentration graphene.



Figure S9. XRD patterns spectra for rust regions on the steel beneath the neat PU and their nanocomposite coatings.