

Using inorganic dynamic porogens for preparing high surface area capacitive carbons with tailored micropores

Xiao-Ling Dong, Wen-Cui Li, Biao Jiang, Yu-Qi Zhou, An-Hui Lu*

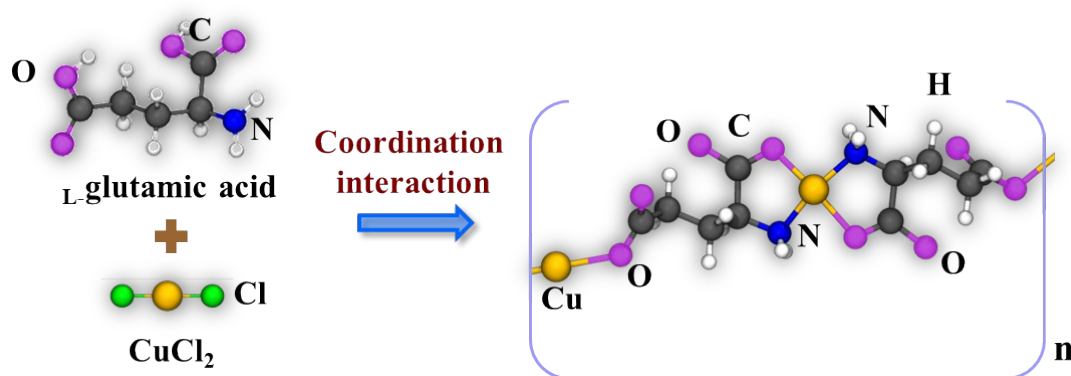


Fig.S1. Schematic illustration for preparing the complex.

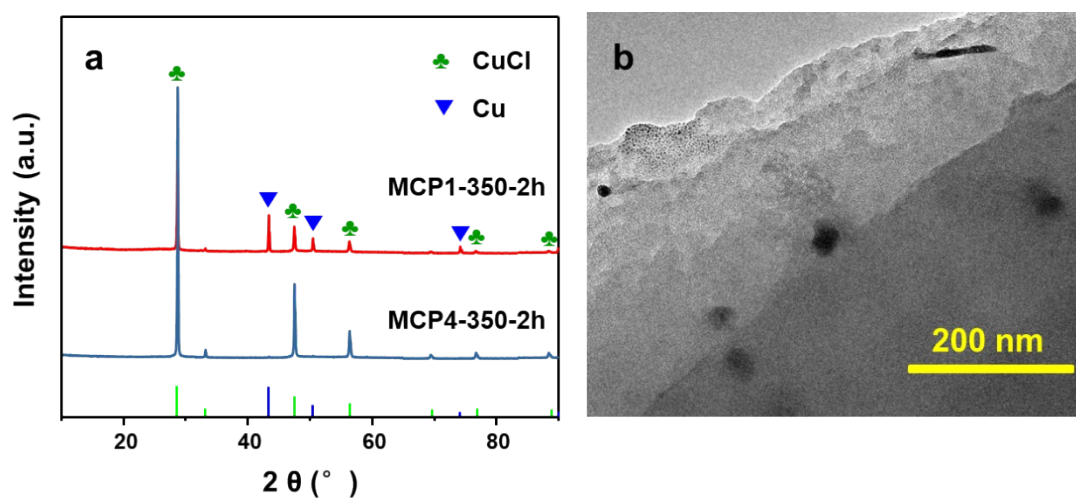


Fig. S2. (a) XRD patterns of MCP1-350-2h and MCP4-350-2h. (b) TEM image of MCP1-350-2h.

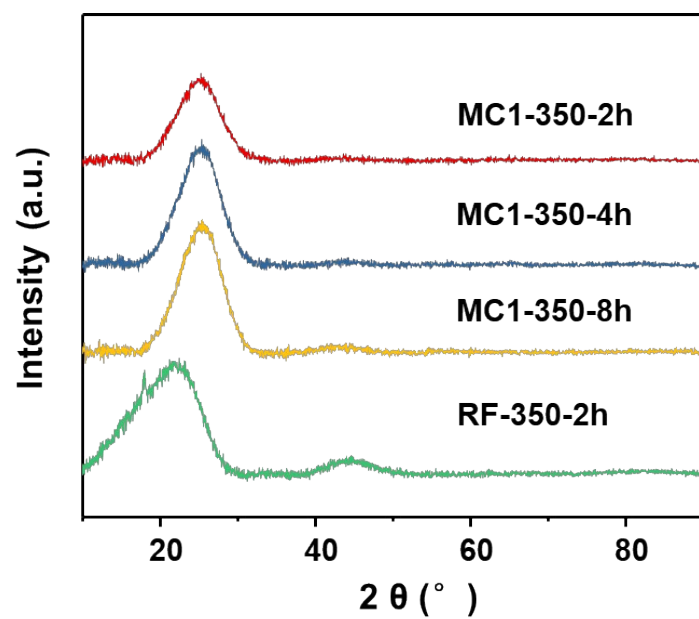


Fig.S3 XRD patterns of MC1-350-2h, MC1-350-4h, MC1-350-8h and RF-350-2h.

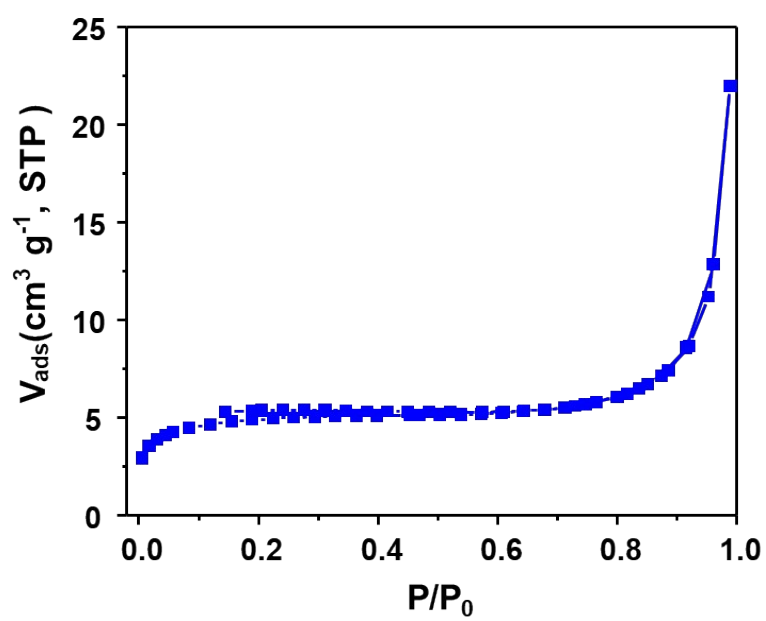


Fig. S4. Nitrogen sorption isotherm of RF-350-2h.

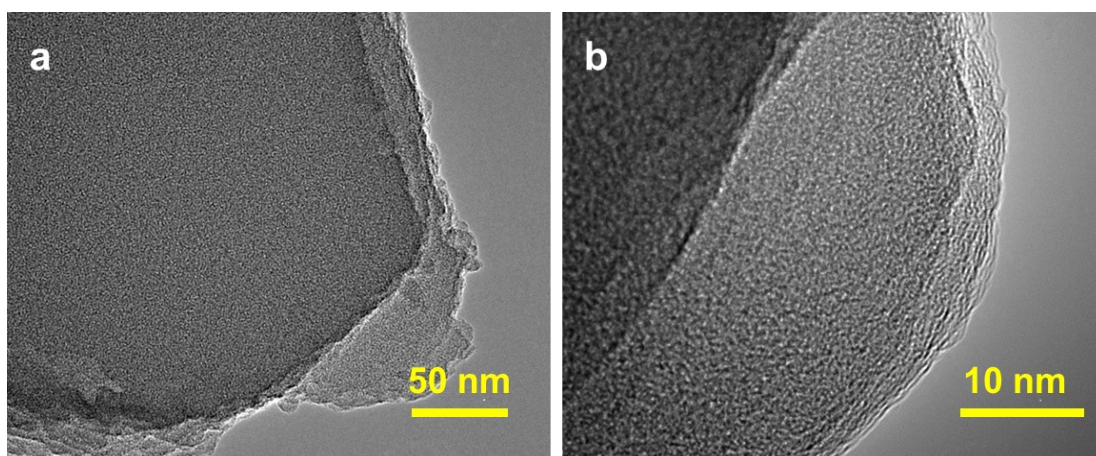


Fig. S5. TEM images of MC1-350-4h.

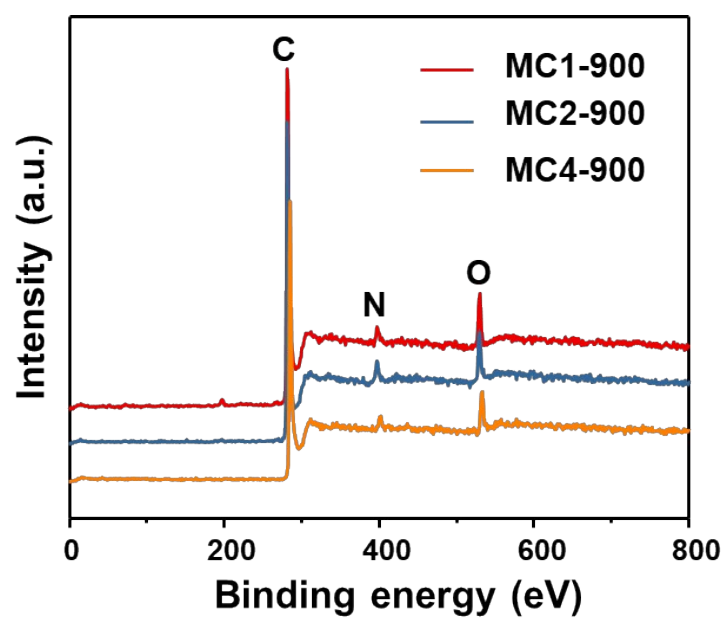


Fig. S6 XPS survey spectra of MC1-900, MC2-900 and MC4-900.

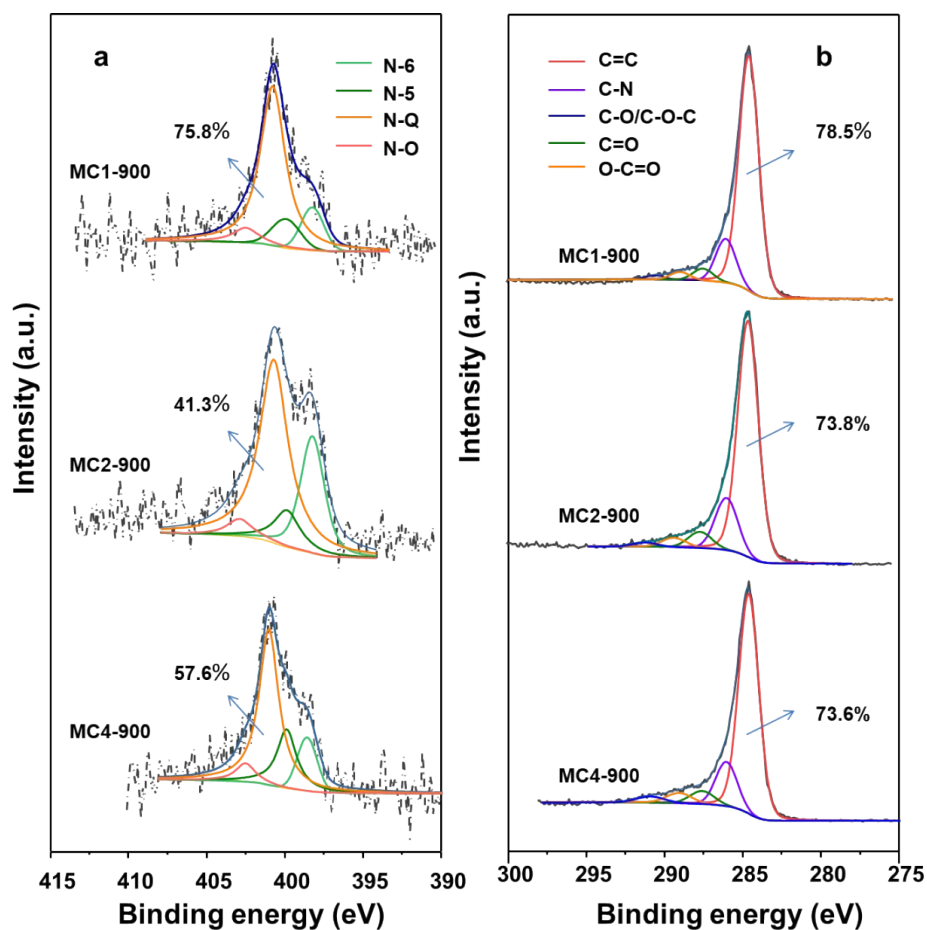


Fig. S7. High-resolution XPS spectra of the deconvoluted N1s peak (a) and C1s peak (b) for MC1-900, MC2-900 and MC4-900.

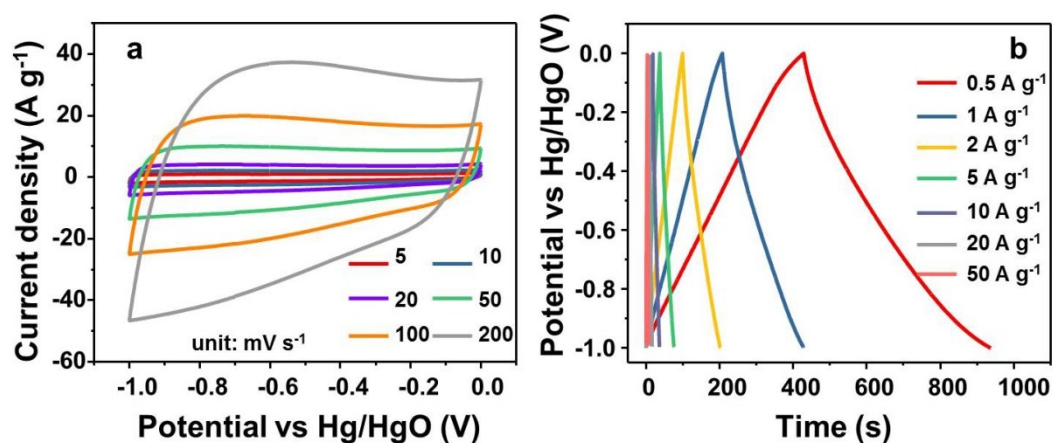


Fig. S8. (a) CV curves and (b) GC curves of MC4-900 electrode in 6 M aqueous KOH electrolyte.

Table S1. Tap densities and chemical compositions of the samples quantified by XPS.

Sample	n(CuCl ₂)/n (L-glutamic acid)	Tap density	XPS atom%		
		g cm ⁻³	C	O	N
MC1-900	1:1	0.58	79.87	16.34	2.77
MC2-900	2:1	0.45	78.67	15.88	3.64
MC4-900	4:1	0.35	73.31	21.56	3.22

Table S2. XPS results about surface atomic percentages of carbon and nitrogen

species of the samples.

Samples	% of total O1s					% of total N1s			
	C=C	C-N	C-O/ C-O-C	C=O	O-C=O	N-6	N-5	N-Q	N-X
MC1-900	78.5	13.9	3.8	2.6	1.2	12.9	6.7	75.8	4.7
MC2-900	73.8	16.4	5.3	3.1	1.5	14.0	37.6	41.3	7.1
MC4-900	73.6	14.9	5.0	3.9	2.6	13.5	22.0	57.6	6.9

Table S3. Specific capacitances of recently reported carbons in aqueous electrolytes.

Materials	Carbon precursors	Capacitance (F g ⁻¹)	Ref.
N, S, P co-doped graphene	graphene oxide	196.4 (1 A g ⁻¹)	1
PU sponge templated activation of GO	graphene oxide	260 (1 A g ⁻¹)	2
porous carbon	ZIF-67	272 (5 mV s ⁻¹) 108 (200 mV s ⁻¹)	3
nitrogen-doped carbon	azine-linked covalent organic framework	234 (1 A g ⁻¹) 167 (50 A g ⁻¹)	4
A850-6-3	liquefied wood	225 (0.5 A g ⁻¹)	5
HTC-JG-900	Jujun grass	220 (1.0 A g ⁻¹)	6
HPC25 (hierarchical porous carbon)	PVDC-b-PS copolymers	241.9 (0.5 A g ⁻¹) 171 (20 A g ⁻¹)	7
zeolite-templated nanoporous carbon	ethylene	250 (1 A g ⁻¹)	8
hierarchical porous carbon	starch	229 (1 A g ⁻¹)	9
highly cross-linked porous carbon microspheres	polyacrylonitrile microspheres	290 (0.5 A g ⁻¹)	10
porous nitrogen self-doped carbon aerogels	benzoxazine	199 (0.5 A g ⁻¹)	11
nitrogen-doped porous carbon nanosheets	bio-oil	289 (0.5 A g ⁻¹)	12
carbon nanofiber/graphene nanosheet composites	Bacterial cellulose/graphene hydrogels	215 (1 A g ⁻¹)	13
zeolite templated carbon	pyrrole	253 (1.25 A g ⁻¹)	14
boron doped carbon spheres	glucose	277 (1 A g ⁻¹)	15

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